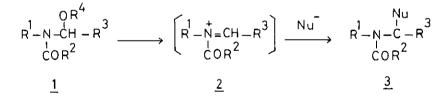
REDUCTION OF N-ALKOXYCARBONYLLACTAMS WITH NaBH<sub>4</sub>/EtOH-H<sup>+</sup>: A FACILE SYNTHESIS OF  $\alpha$ -ETHOXYURETHANES

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<u>Abstract</u> — Reduction of N-alkoxycarbonyllactams with  $NaBH_4/H^+$  in ethanol (Speckamp's condition) afforded  $\alpha$ -ethoxyurethanes (Shono's compounds) in good yields.

New methods for the carbon-carbon bond formation at the  $\alpha$ -position of amines <u>via</u> acyliminium ion (<u>2</u>) have been developed in recent years.<sup>1</sup> Among these,  $\alpha$ -alkoxyamides<sup>2</sup> (<u>1</u>; R<sup>2</sup> = alkyl) and  $\alpha$ -alkoxyurethanes<sup>3</sup> (<u>1</u>; R<sup>2</sup> = alkoxy), precursors for <u>2</u>, are conveniently available from the reduction of imides with NaBH<sub>4</sub>/EtOH-H<sup>+</sup> or from the anodic oxidation of urethanes, respectively. In this paper, a convenient alternative method for the synthesis of cyclic N-alkoxycarbonyl-2-ethoxyamines (7 - 9) is described.



Although the reduction of N-acyllactams (e. g. <u>4a</u>, <u>4b</u>, <u>5a</u>) with NaBH<sub>4</sub> did not afford  $\alpha$ -ethoxyamides (e. g. <u>7a</u>, <u>7b</u>, <u>8a</u>), the reductinon of N-alkoxycarbonyllactams (<u>4c-q</u> - <u>6c-q</u>) in a similar condition gave the desired  $\alpha$ -ethoxyurethanes (<u>7</u> - <u>9</u>) in good yields. The results are shown in Table 1. The reaction was carried out by the modification of Speckamp's procedure<sup>2</sup>: A small amount of bromocresol green (pH 3.8 yellow - 5.4 blue) was used as an internal indicator and the optimum reaction temperature was between -6° and 0°.

The investigation of the carbon-sulfur, carbon-nitrogen and carbon-carbon bonds formation at the  $\alpha$ -position of amines using these  $\alpha$ -ethoxyurethane are now in progress.

Table 1. Reduction of N-Alkoxycarbonyllactams<sup>a</sup> to  $\alpha$ -Ethoxyurethanes<sup>b</sup>

(CH <sub>2</sub> )n N 0 C=0 R	NaBH₄ ──EtOH/H <sup>+</sup>	CH2)n NOEt C=O R
4 n = 1		<u>7</u> n = 1
<u>5</u> n = 2		<u>8</u> n = 2
<u>6</u> n≃3		<u>9</u> n = 3

Yield  $(%)^{C}$  and bp (°C/mmHg) of Products (7, 8, 9)

		R		7	<u>8</u>		9	<u>.</u>
a	:	Me		0 <sup>d</sup>	0 <sup>ć</sup>	1		
b	:	Ph		$0^{\mathbf{d}}$	-		-	
с	:	OMe	80	(120°/3)	<u>-</u> 79	(85°/4)		
d	:	OEt	83	(83°/2)	-		64	(83°/2)
е	:	OCH <sub>2</sub> Ph	83	(110°/2)	80	(_ <sup>e</sup> )	88	( - <sup>e</sup> )
			88	(75°/2)	70	(70°/2)	65	(78°/2)
g	:	$OCH_2CH=CH_2$	88	(105°/2)	-		_	

a) N-Alkoxycarbonyllactams ( $\underline{4} - \underline{6}$ ) were prepared by the alkoxycarbonylation of lactams with chloroformates (or di-<u>tert</u>-butyl dicarbonate for <u>tert</u>-butoxy-carbonylation ) and sodium hydride in 60-80% yields. b) All the products gave satisfactory ir, <sup>1</sup>H-nmr and ms spectra. c) Isolated yield. d) In the case of N-acyllactams, the cleavage of N-acyl groups and the reduction of the ring-opened compounds were observed. e) Purified by chromatography.

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