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An Improved Synthesis of the (E,Z)-Dienoate Precursor of (+)-Damavaricin D Via a Vinylogous Horner-Wadsworth-Emmons Reaction

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Abstract: An improved synthesis of the C(1)-C(5) (E,Z)-dienoate segment of (+)-damavaricin D precursor **3** was accomplished using the unsaturated phosphonate reagent **9** in a (Z)-selective vinylogous Horner-Wadsworth-Emmons reaction. © 1999 Elsevier Science Ltd. All rights reserved.

In our recently completed total synthesis of (+)-damavaricin D (DmD), the C(1)-C(5) (*E*, *Z*)-dienoate unit of macrocyclization precursor **3** was installed in 24% overall yield via an eight step sequence starting from **1**.¹ The (*Z*)-enoate unit of **2** was generated with 8 : 1 selectivity using Still's (*Z*)-selective olefination procedure,² and the (*E*)-trisubstituted enoate was introduced subsequently by a conventional Horner-Wadsworth-Emmons reaction.³ The most difficult step of this sequence was the selective DIBAL reduction of **2** that had to be performed under experimentally stringent conditions (THF, -100 °C to -78 °C) in order to minimize competitive reduction of other carbonyl functions in the molecule; the desired allylic alcohol was obtained in 62% yield along with 16% of the corresponding enal after one recycle of recovered starting material.¹ We describe herein the development of a more concise five-step route (30% overall yield) to the key dienoate intermediate **3** involving a (*Z*)-selective vinylogous Horner-Wadsworth-Emmons reaction using the unsaturated phosphonate reagent **9**.



In the early stages of this program we explored the Horner-Wadsworth-Emmons reaction of aldehyde 4 with the unsaturated phosphonate 5 as a route to the targeted (E,Z)-dienoate.⁴ Our best result at that time was obtained by using LHMDS in Et₂O. However, because these conditions provided the (E,Z)-dienoate 6 with only 1.5 : 1 selectivity, we were discouraged from attempting to apply this method to the DmD synthesis.



0040-4039/99/\$ - see front matter © 1999 Elsevier Science Ltd. All rights reserved. *PII:* S0040-4039(98)02638-0 Evans and co-workers subsequently reported a successful application of this methodology in their total synthesis of (+)-macbecin.⁵ Although they also obtained low selectivity (1.5:1) in the Horner-Wadsworth-Emmons reaction of *i*-PrCHO and 5, somewhat better selectivity (2.7:1) favoring (E,Z)-8 was achieved when macbecin precursor 7 was treated with a large excess (8 equiv) of the anion generated from phosphonate 5.

This result encouraged us to reexamine the viability of this method as a route to the damavaricin precursor **3**. To do so, however, required that we first develop a synthesis of the unsaturated phosphonate **9** reagent containing a β -trimethylsilylethyl ester protecting group. We had already spent considerable effort orchestrating the protecting group combinations in **3** such that the aniline and carboxylic acid units could be liberated in a single step immediately prior to the macrolactamization reaction, and we did not wish to revisit this aspect of the synthesis.¹ The synthesis of phosphonate **9** thus began with γ -bromotiglic acid **10**, obtained by bromination of tiglic acid as previously described.⁶ It was necessary to protect the acid as allyl ester **11** prior to the Arbuzov reaction as trifluoroethyl esters were obtained when acid **10** was used, and the 2-trimethylsilyl ethyl ester of **10** did not survive the Arbuzov conditions (complex mixtures were obtained). Esterification of **10** with allyl alcohol under Mitsunobu conditions⁷ afforded **11** (76%), which was then submitted to the Arbuzov reaction with 2,2,2-trifluoroethyl phosphite (10 equiv) in the presence of catalytic Bu₄NI (0.1 equiv), thereby providing phosphonate **12** in 51% yield.^{8,9} The allyl group was removed¹⁰ (90%) and the resulting acid **13** was re-esterified with β -trimethylsilylethanol, DCC, and DMAP to provide the targeted phosphonate reagent **9** (84%).



Reactions of phosphonate 9 and aldehyde 14 were explored to develop conditions that maximized selectivity for the (E,Z)-diene. Of the various base and solvent combinations examined, best results were obtained by use of potassium hexamethyldisilazide (KHMDS) in Et₂O: this combination provided (E,Z)-15 with 4.7 : 1 selectivity (entry 1). Interestingly, addition of 18-crown-6 to this reaction, which in other systems serves to enhance (Z-)selectivity,² gave an 11 : 1 mixture favoring(E,E)-15 (entry 2). The (E,E)-diene also predominated by a significant margin in reactions using MeMgBr as the base (entry 8). Reactions performed using *n*-BuLi to deprotonate 9 were only marginally selective: when performed in Et₂O or toluene (E,Z)-15 was obtined with 1.3-1.6 : 1 selectivity, while the reaction with *n*-BuLi in THF gave a 2 : 1 mixture favoring (E,E)-15. While the stereoselectivity of the olefination reactions of 14 was insensitive to the number of equivalents of the phosphonate reagent employed (entries 3-5), the stereoselectivity of these reactions is temperature sensitive and best results were obtained when the aldehyde was added slowly to the phosphonate anion at -78 °C over a 0.5 to 1.0 h period.

TBDPS	Me ₃ S		i(CH₂)₂0	2 ^{CF} 3 ² TBDPSO	TBDPSO	
14	м́е		-78 to -45 °C	Me ₃ SiCH ₂ CH ₂	Me3SICH2CH2O	
-	Entry	Equiv 9	Base, Solvent	Selectivity E,Z : E,E	Yield	-
	1	1.4	KHMDS, Et ₂ O	4.7 : 1	84%	
	2	2.9	KHMDS, Et ₂ O, 18-crown	-6 1:11		
	3	1.4	n-BuLi, Et₂Õ	1.6 : 1	73%	
	4	5	n-BuLi, Et ₂ O	1.3 : 1		
	5	8	n-BuLi, Et ₂ O	1.3 : 1	80%	
	6	5	<i>n</i> -BuLi, TĤF	1:2		
	7	5	n-BuLi, toluene	1.3 : 1		
	8	2.0	MeMgBr, Et ₂ O	1:9		

Encouraged by the results obtained in the reaction of 9 and 14, we explored the olefination reactions of 9 with progressively more complex substrates. However, we were quickly disappointed to discover that the reaction of 9 and 16 was non-selective using KHMDS under the conditions developed for the reaction with 14. However, a 2 : 1 mixture favoring (E,Z)-17 was obtained using 2 equiv of 9 with *n*-BuLi as the base. When the olefination of 18 was performed using 2-4 equiv of 9, (E,Z)-19 was also obtained with 2 : 1 selectivity. Fortunately, selectivity in this case increased to 4 : 1 when 18 was treated with 8 equiv of the anion generated from 9. Under these conditions, (E,Z)-19 was obtained in 45% yield along with 32% of aldehyde 18 which could be recycled. Interestingly, the olefination of 18 with phosphonate reagent 5 under comparable conditions provided a 2.7 : 1 mixture of (E,Z) : (E,E) olefin isomers, suggesting that further modification of the ester unit may constitute a strategy to increase the stereoselectivity of these Horner-Wadsworth-Emmons reactions.



Application of this methodology to an improved synthesis of **3** proceeded as follows. The carbamate functionality of **20** was introduced in 71% yield via a Curtius rearrangement¹¹ of the carboxylic acid generated by tris(dimethylamino)sulfonium difluorotrimethylsilicate (TAS-F, 2.2 equiv) mediated selective deprotection¹² of **1**. Treatment of **20** with HF•Et₃N in refluxing CH₃CN afforded primary alcohol **21** (91%), which was subsequently oxidized using Swern conditions¹³ to the corresponding aldehyde. Horner-Wadsworth-Emmons olefination of the crude aldehyde with 10 equiv of the lithium anion of **9** in Et₂O afforded, after one recycle of the recovered aldehyde intermediate, 60% of a 4 : 1 mixture of (E,Z) : (E,E) dienoates **3** along with 13% of recovered

aldehyde. The desired DmD intermediate, (E,Z)-3 was isolated in 47% yield by HPLC, and proved to be identical in all respects to the intermediate utilized in our DmD total synthesis.¹ This five step sequence thus provides DmD intermediate 3 in 30% overall yield from 1.



In summary, we have developed an improved synthesis of DmD intermediate 3 from precursor 1 using the vinylogous Horner-Wadsworth-Emmons reagent 9 in a (Z)-selective olefination sequence. Although it is not clear at present to what extent the reaction stereoselectivity of these vinylogous Horner-Wadsworth-Emmons reactions is governed by kinetic or thermodynamic considerations, 5,14 it is clear that synthetically useful results may be obtained by judicious selection and optimization of reaction conditions.

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