Synthesis and Biological Activity of 1,5-Diphenyl-4-arylazopyrazoles and 5,5-Dimethylcyclohexane-1,2,3-trione Bishydrazones

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Abstract A series of 1,5-diphenyl-4-arylazopyrazoles was syn-
thesized and evaluated for antimalarial activity. The compound
possessing the best antimalarial activity of the entire study was 1,5-
diphenyl-4-methoxyphenylazopyrazole. Another series of 5,5-
dimethylcyclohexane-1,2,3-trione bishydrazones was prepared and
tested as anthelmintics. No significant activity was observed.

Keyphrases 1,5-Diphenyl-4-arylazopyrazoles—synthesized and screened as potential antimalarial agents 5,5-Dimethylcyclohexane-1,2,3-trione bishydrazones—synthesized and screened as potential anthelmintics Anthelmintic agents, potential—synthesis and screening of 5,5-dimethylcyclohexane-1,2,3-trione bishydrazones Antimalarial agents, potential—synthesis and screening of 1,5-diphenyl-4-arylazopyrazoles

In a previous paper (1) the synthesis and biological data of N^1 -substituted arylsulfonylpyrazoles (Ic) were reported. The test results on rodent malaria performed with some of these derivatives (Ic, X = 2-Cl, 4-OCH₂, or 4-NO₂) were not encouraging; the mean survival time of drug-treated, young female, noninbred ICR-HA swiss mice infected with a standard intraperitoneal inoculum of *Plasmodium berghei* (KBG-173) increased by 0.1 day compared to controls. In a search

for efficient antimalarials, analogs of 1,5-diphenyl-4arylazopyrazoles (Ia) were prepared and are reported in this article. The compounds were found to possess marked improvement in potencies over previous candidates.

In view of the anthelmintic activity displayed by Ic, a series of 5,5-dimethylcyclohexane-1,2,3-trione bishydrazones (IV) containing a few of these groupings was synthesized.

N-N-N-F

Table I—Chemical and Biological Properties of 1,5-Diphenyl-4-arylazopyrazoles

Num- ber	R	R'	х	Yield,	Melting Point	Color	Formula	——Analy Calc.	sis, %—— Found	Anti D	mal TD	arial T/C	Activity ^o Remark
1	C ₄ H ₄	Н	н	60	180°	DR	CziHieNe	C 77.77 H 4.93	77.72 4.95	320	2	3.9	Toxic
2	C ₆ H ₆	H	4-Cl	5 0	195°	Y	CnH15CIN4	N 17.28 C 70.29 H 4.18	17.24 70.24 4.28	320	2	3.9	Toxic
3	C ₆ H ₆	Н	2-OCH ₄ , 5-OCH ₄	60	188°	В	C11H10N4O1	N 15.62 C 71.87 H 5.20	15.41 71.67 5.12	320	0	1.1	
4	C ₄ H ₅	н	4-CH;	55	190°	Y	C22H18N4	N 14.58 C 78.18 H 5.32	14.34 78.02 5.22	500	0	0.1	_
5	C ₆ H ₆	H	4-OCH ₃	70	206°	Ο	C22H18N4O	N 16.56 C 74.57 H 5.08	16.46 74.56 5.01	320	3	4.4	Toxic
6	C ₈ H ₅	Н	2-OCH _a	65	193°	LY	C12H18N4O	N 15.81 C 74.57 H 5.08	15.61 74.37 4.98	320	0	2.3	
7	C ₄ H ₄	Н	2-NO ₃	50	184°	ο	C ₂₁ H ₁₅ N ₅ O ₂	N 15.81 C 68.29 H 4.06	15.91 68.01 4.03	320	0	0.1	_
8	C ₄ H ₅	H	4-OCH,CH,	50	170°	R	C22H20N4O	N 15.21 C 75.00 H 5.43	15.11 74.92 5.31	320	3	3.9	Toxic
9 10 11	C ₆ H ₆ CHOHCH ₉ C ₆ H ₆ CHOHCH ₂ C ₆ H ₆ CHOHCH ₂	СН, СН, СН,	2-Br 2-NO ₂ 2-CH ₃ , 3-CH ₄		101°d 123°d 78°d	_	=	N 15.21	15.11 — — —	320 640 640	0 0 0	0.3 0.5 0.3	Ξ

^aJ. M. L. Mathur, M.S. thesis, University of Roorkee, Roorkee, India, 1971. Test data supplied by the Medicinal Chemistry Division, Walter Reed Army Institute for Research. ^b B = brown, Bk = brick, Bl = blood, Da = dark, D = dull, L = light, O = orange, R = red, and Y = yellow. ^c D = dose, mg./kg.; TD = toxic deaths attributed to drug toxicity when mice died 2-5 days postinfection of P. berghel (KBG-173); and T/C = increase in survival time of treated mice over controls. ^d Reference 5.

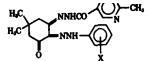


Table II—Chemical and Biological Properties Dimethylcyclohexane-1,2,3-trione Bishydrazones*

Num-		Yield.	Melting			——Analys	—Effic	Dose.				
ber	x	% Tield,	Point	Color	Formula	Calc.	Found	Tg	N	C	0	mg./kg.
1	Н	80	>270°4	DR	C ₂₀ H ₃₂ N ₆ O ₈	C 60.91 H 5.58	60.72 5.34		0	75	0	300*
2	4-CH ₃	78	>270°′	0	CnH24N4O3	N 21.31 C 61.76 H 5.88 N 20.58	21.41 61.56 5.82 20.68	_	0	0	0	300•
3	3-CH ₈	75	137°	DaB	CnH14NeO1	C 61.76 H 5.88	61.66 5.85 20.60	_		_	_	_
4	4-OCH _a	78	264-266°/	LB	CnH14N4O4	N 20.68 C 59.43 H 5.66 N 19.81	59.23 5.43 19.79	_	0	0	0	100
. 5	2-OCH ₃	65	152-154°/	DaB	CnHnNtO4	C 59.43 H 5.66 N 19.81	59.35 5.52 19.69		_	.—	_	_
6	4-Cl	80	248-249°4	BIR	C ₁₀ H ₂₁ ClN ₄ O ₄	C 56.00 H 4.60 N 19.60	56.20 4.59 19.42		_		_	_
7	4-Br	75	255-257°/	DaR	C ₂₆ H ₂₁ BrN ₆ O ₃	C 50.73 H 4.43 N 17.75	50.53 4.22 17.55	_	0	0	0	100
8	3-NO ₂	70	240-241°4	R	C ₂₀ H ₂₁ N ₇ O ₅	C 54.66 H 4.78 N 22.32	54.46 4.52 22.22	0	0	0	0	100
9	2-NO ₂	77	250-251°4	R	C ₁₀ H ₂₁ N ₇ O ₅	N 22.32 C 54.66 H 4.78 N 22.32	54.56 4.42 22.30	0	0	0	50	100/400
10	2-Cl, 3-Cl	68	195–196°⁴	BkR	C ₂₀ H ₂₀ Cl ₂ N ₆ O ₈	C 51.83 H 4.31 N 18.14	51.62 4.28 18.12	_	_	_	_	_
11	2-Cl, 4-Cl	70	180°	BkR	C ₁₀ H ₁₀ Cl ₁ N ₆ O ₂	C 51.83 H 4.31 N 18.14	51.72 4.29 18.04	0	0	0	0	200
12	2-OCH ₂ , 4-OCH ₃	72	215–217°	DaB	C ₂₂ H ₂₆ N ₆ O ₅	C 58.14 H 5.72 N 18.50	58.04 5.70 18.48	_		_		

⁶ A. Singhal, M.S. thesis, University of Roorkee, Roorkee, India, 1971. Test data supplied by Dr. W. C. McGuire, Philips Roxane, Inc. ⁵ See Footnote b of Table I. ⁶ T_g = toxoplasma gondii-RH strain prevention of mortality in mice, N = nematodes, C = cestodes, and O = oxyurids. ⁴ Charring. ⁶ Approximate LD₅₀. ⁷ Decomposed.

Compounds of general structure Ia were synthesized by the reaction of the sodium salt of oxymethyleneacetophenone (2) (II) with aryldiazonium salts to give III, which were then condensed with phenylhydrazine.

Bishydrazones (IV) were obtained by the condensation of 2-methyl-4-hydroxy-5-pyrimidylcarbohydrazide (V) with 5,5-dimethylcyclohexane-1,2,3-trione 2-arylhydrazones (VI). Compounds VI were prepared by the coupling of the diazotized anilines with 5,5-dimethylhexane-1,3-dione (3).

Ia: $R = R'' = C_6H_6$, R' = H

Ib: $R = C_6H_6CHOHCH_2$, $R' = CH_3$, $R'' = C_6H_6$

Ic:
$$R = -SO_2 - O_3$$
, $R' = R'' = CH_8$

PHARMACOLOGY

Pyrazoles of series Ia and Ib appeared weakly active against P. berghei 173 in mice, increasing survival time up to 4.4 hr. over controls.

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Anthelmintic activity testing of bishydrazone derivatives (IV) revealed that all were inactive except 2-phenyl and 2-(2-nitrophenyl) derivatives

EXPERIMENTAL¹

1,5-Diphenyl-4-arylazopyrazoles (Ia)—The appropriate aniline was diazotized as reported previously (4) and treated with sodium salt of oxymethyleneacetophenone (II). The precipitated arylhydrazone (III) (0.01 mole), on treatment with phenylhydrazine (0.01 mole) in ethanol and acetic acid followed by refluxing for several hours and finally cooling to room temperature, afforded pyrazole

derivatives (Ia). All of these compounds were recrystallized from acetone (Table I).

5,5-Dimethylcyclohexane-1,2,3-trione, 1-(2-Methyl-n-hydroxy-5-pyrimidylcarbo)-2-(substituted phenyl) Bishydrazones (IV)—A solution of 2-arylhydrazone of 5,5-dimethylcyclohexane-1,2,3-trione (VI) (3) (0.005 mole) in ethanol (15 ml.) and 2-methyl-4-hydroxy-5-pyrimidylcarbohydrazide (V) (0.005 mole) in glacial acetic acid (1 ml.) was refluxed for 1 hr. On cooling, shining crystals of IV separated out and were recrystallized from dimethylformamide (Table II).

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Synthesis and CNS Depressant Activity of Imidazolinone Glyoxylates

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Abstract
Six new imidazolinone glyoxylates, which exist in tautomeric equilibria with hydroxyimidazole glyoxylates, were synthesized by a new type of Claisen rearrangement and evaluated in a neuropharmacological mouse profile. CNS depression was observed in several members of the class.

Keyphrases ☐ Imidazolinone glyoxylates—synthesized and screened as potential CNS agents ☐ CNS agents, potential—synthesis and screening of imidazolinone glyoxylates ☐ Imidazoles—synthesis and CNS depressant activity of six imidazolinone glyoxylates

A recent publication (1) described the synthesis and characterization of a new type of imidazolinone glyoxylate (II) by the Claisen-like rearrangement of amide oxime—dimethyl acetylenedicarboxylate adducts (I). Pyrolysis of these adducts in refluxing diphenyl ether gave 41–68% yields of the tautomeric methyl 2-aryl-4-oxo-2-imidazoline-5-glyoxylates (II) (Scheme I). The impressive history of imidazoles as therapeutic agents (2, 3) and, more relevantly, the recent reports that several imidazoles function as centrally acting depressants (4–6) prompted this report of the evaluation of a new imidazole class in a primary Irwin neuropharma-cological mouse profile (7).

The amide oximes utilized as starting materials were prepared by addition of hydroxylamine to aromatic

nitriles. These amide oximes were condensed with dimethyl acetylenedicarboxylate to yield 1:1 adducts (I) in which —OH addition to the alkyne had occurred.

$$\begin{array}{c} H-N & C & COOCH_3 \\ Ar-C & NH & COOCH_3 \\ I & & & & & \\ \end{array} \qquad \begin{array}{c} \Delta & \left[\begin{array}{c} O \\ HN \\ Ar-C \\ N \end{array} \right] & COOCH_3 \\ \left[\begin{array}{c} HN \\ Ar-C \\ N \end{array} \right] & COOCH_3 \end{array}$$

II Scheme I

¹ All melting points were determined with a Kofler hot-stage-type apparatus and are uncorrected.