

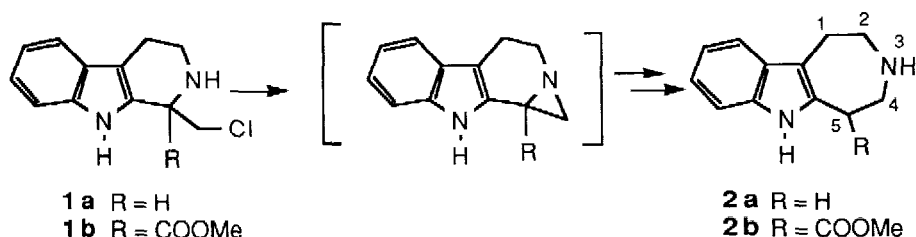
"HOMO-PICTET-SPENGLER" CYCLIZATION OF TRYPTAMINES: AN EASY ACCESS TO THE HEXAHYDROAZEPINO[4,5-b]INDOLE RING SYSTEM.

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Summary: The title compounds were prepared in one step from tryptamines **3** and α -bromoaldehydes **4**.

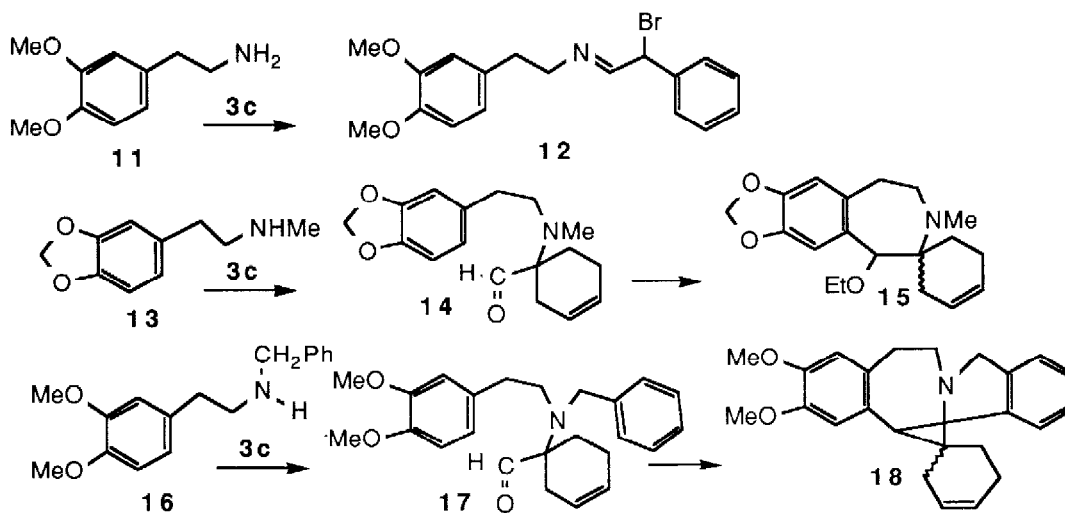
Synthesis of the hexahydroazepino[4,5-b]indole system through ring expansion of a tetrahydro- β -carboline has been little studied until now: the model compound **2a** (Scheme 1) was prepared by Julia¹ from 1-chloromethyl- β -carboline **1a** via the intermediate aziridine. Later, Kuehne obtained compound **1b** through the Pictet-Spengler cyclization of tryptamine with methyl chloropyruvate and further rearranged it to his highly potent azepinoindolic synthon **2b**².



Scheme 1

We have now studied the reaction of tryptamines with some α -bromoaldehydes³ along similar lines, in order to prepare 4,4-disubstituted hexahydroazepino[4,5-b] indoles (Scheme 3, Table). We found that heating tryptamine **3a** with α -bromo aldehydes **4a-d** in acetic acid at 70°C for 15 hrs afforded in one step the 5-acetoxy-4,4-dialkylhexahydroazepinoindoles **5a-d** in yields ranging from 34 to 75%. With 5-methoxytryptamine **3b** and bromoaldehydes **4a** or **4c** the synthesis of **5e** and **5f**, respectively, was completed within only 2 hrs. On the contrary, reaction of *N(b)-methyl* tryptamine **3c** with bromoaldehyde **4c** was very sluggish: compound **5g** was isolated (42%) after a 100 hr heating in AcOH, together with aldehyde **10** (X = H, R₁, R₂ = -CH₂-CH=CH-CH₂-CH₂-; R₃ = Me) (11%) and methyltryptamine. When the reaction was stopped after 6 hrs, aldehyde **10** was isolated (36%) as the sole reaction product. Compound **10** could further be cyclized in AcOH to **5g**.

With N(b)-unsubstituted tryptamines ($R_3 = H$) the initially formed carbinolamine **6** gives iminium **7** which leads to aziridinium **8** along (concerted?) Pictet-Spengler cyclization and bromine elimination. Finally, attack by an acetate gives the 5-acetoxylhexahydroazepinoindoles **5a-f**. Electron enrichment of the indole nucleus by a methoxy group ($X=OMe$) accelerates the Pictet-Spengler cyclization. No diastereoselection was observed in the formation of **5c-d** and **5f-g** with regard to the poor discrimination between R_1 and R_2 . With the N(b)-substituted tryptamine ($R_3 = Me$) carbinolamine **6** rearranges to aldehyde **10**⁶ *via* aziridinium **9**. Further cyclization of **10** to **5g** is preceded⁷.

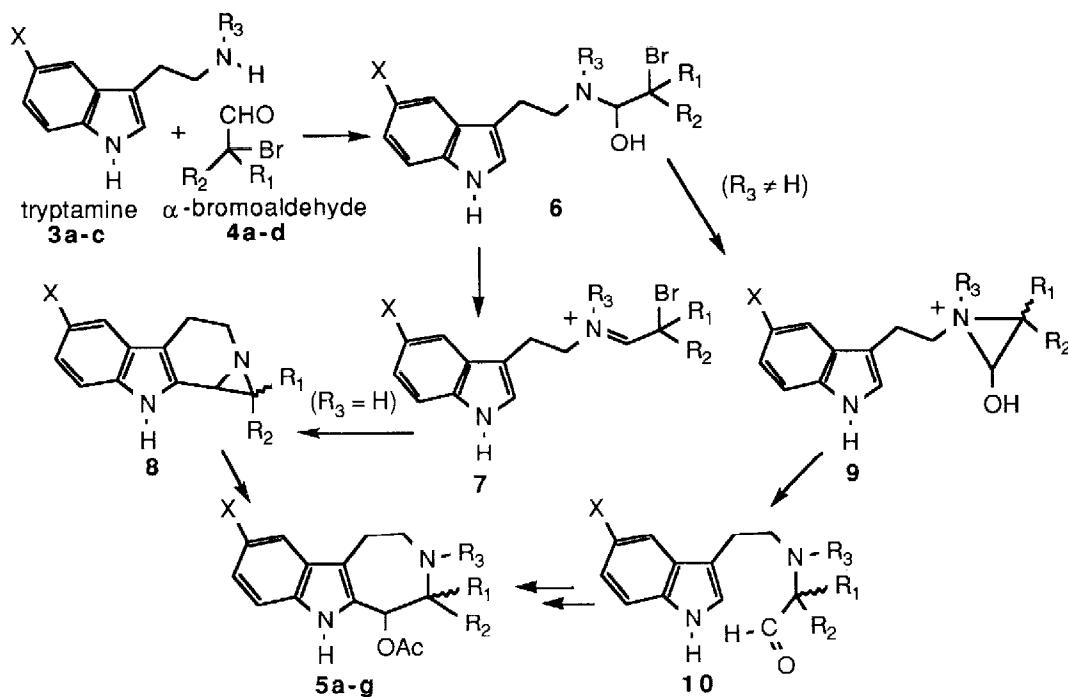


Scheme 2

Intermediacy of **7** or **10** (depending on the substitution of N(b) in tryptamines) was substantiated by the reactions of some α -bromoaldehydes with phenethylamine derivatives (scheme 2). In no case was the one-step cyclization in AcOH observed. Heating homoveratrylamine **11** and bromoaldehyde **3c** in AcOH yielded bromoimine **12** which could not be further cyclized. Treatment of N-methylhomopiperonylamine **13** with **3c** and K_2CO_3 in refluxing DME for 4 days gave aldehyde **14** (48 %), which was cyclized to benzazepine **15** (24%) upon reaction with TFAA and BF_3 etherate (probably through an acylal intermediate⁸). A similar sequence starting from N-benzylhomoveratrylamine **16** and **3c** gave aldehyde **17** (51%) whose cyclization (BF_3 etherate) was accompanied with a supplementary ring closure affecting the benzene ring, which yielded compound **18** (66%) along a not uncommon reaction pathway^{9,10}.

The use of α -bromoaldehydes in such "homo-Pictet-Spengler" cyclizations then appears of interest in the synthesis of fused azepino- ring systems. Further applications in the azepinoindole series are under current study.

Acknowledgements: We thank Drs D Royer, Dr D Cartier and Mr P. Sigaut for spectral measurements, and CNRS and ADIR company for financial support.



Scheme 3

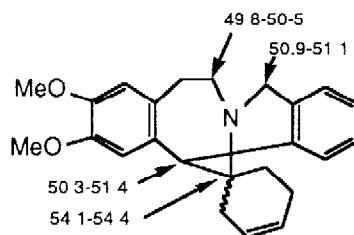
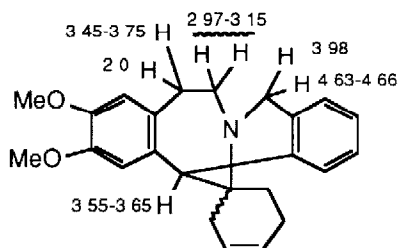
| tryptamine | | | α -bromoaldehyde | | time of reaction | product (yield %) |
|------------|----------------|----|-------------------------|---|------------------|----------------------------|
| X | R ₃ | | R ₁ | R ₂ | | |
| 3a | H | H | 4a | Me | 15 hrs | 5a (34) |
| 3a | H | H | 4b | Et | 15 hrs | 5b (56) |
| 3a | H | H | 4c | -CH ₂ -CH=CH-CH ₂ -CH ₂ - | 15 hrs | 5c^a (70) |
| 3a | H | H | 4d⁴ | -CH ₂ -CH=C(Me)-CH ₂ -CH ₂ - | 15 hrs | 5d^a (75) |
| 3b | MeO | H | 4a | Me | 2 hrs | 5e (66) |
| 3b | MeO | H | 4c | -CH ₂ -CH=CH-CH ₂ -CH ₂ - | 2 hrs | 5f^a (72) |
| 3c | H | Me | 4c | -CH ₂ -CH=CH-CH ₂ -CH ₂ - | 100 hrs | 5g^a (42) |

^a mixture of isomers

Table 5

References and Notes

1. Julia, M.; Bagot, J., Siffert, O. *Bull. Soc. Chim. France* **1973**, 1424.
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3. The starting aldehydes were brominated in nearly quantitative yield by using the CuBr₂ bromination method developed for ketones by Brown: Brown, E. C.; Rogié, M. M.; Rathke, M. M. *J. Amer. Chem. Soc.* **1968**, *90*, 6218.
4. 1-methyl-4-bromo-4-formylcyclohexene **4d** was prepared by Diels-Alder cyclization of isoprene with 2-bromoacrolein.
5. All new products were fully characterized by their UV, IR, mass, ¹H (300 MHz) and ¹³C (75 MHz) NMR spectra. Most salient features for compounds **5a-g** were an easy loss of AcOH from the molecular ion in the mass spectra (M-60) and a low field proton at ~ 5.7 ppm (H-5) in the ¹H NMR spectra. C-4 and C-5 were seen at 42-44 and 72-75 ppm, respectively, in the ¹³C NMR spectra.
6. Reaction of α-halogenocarbonyl compounds with secondary amines is known to yield α-aminocarbonyl derivatives: De Kimpe, N.; Verhe, R. *The Chemistry of α-haloketones, α-haloaldehydes and α-haloimines* J Wiley, Ed., New-York, **1988**.
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9. Obtained as a mixture of isomers. The following ¹H and ¹³C data were supported by COSY measurements.



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(Received in France 27 November 1990)