



STRUCTURE AND SYNTHESIS OF [*n*]-DEHYDROSHOGAOLS FROM *ZINGIBER OFFICINALE*

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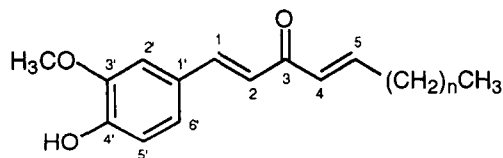
Abstract—Three new dehydroshogaols have been isolated from the rhizomes of *Zingiber officinale*. Their structures were established by spectroscopic analysis and synthesis. © 1998 Elsevier Science Ltd. All rights reserved

INTRODUCTION

Ginger (Chinese name: shengjiang), the rhizomes of *Zingiber officinale*, is a well-known spice and frequently prescribed in traditional Chinese medicine as a stomachic, antiemetic, antidiarrheal, expectorant, antiasthmatic, haemostatic and cardiogenic, for the treatment of gastrointestinal and respiratory diseases [1]. Numerous chemical investigations of the pungent and bioactive principles of ginger have been carried out [2–14]. In the course of our continuing research for novel biologically active compounds from natural sources, bioassay-directed fractionation led to the isolation and characterization of three new dehydroshogaols, [6]-dehydroshogaol (**1**), [8]-dehydroshogaol (**2**) and [10]-dehydroshogaol (**3**), from a diethyl ether extract of the rhizomes of *Z. officinale*. We describe herein the structural elucidation and the synthesis of these compounds.

RESULTS AND DISCUSSION

[6]-Dehydroshogaol (**1**) was isolated as a yellow syrup which showed the molecular formula, $C_{17}H_{22}O_3$,



- 1** : $n = 4$
2 : $n = 6$
3 : $n = 8$

as determined by HR mass spectrometry. The IR spectrum showed hydroxyl absorption at 3400 cm^{-1} and carbonyl absorption at 1654 cm^{-1} . The ^1H NMR spectrum apparently exhibited an ABC-pattern signal at δ 6.93 (*d*, $J = 8.4\text{ Hz}$), 7.04 (*d*, $J = 1.6\text{ Hz}$) and 7.04 (*dd*, $J = 8.4, 1.6\text{ Hz}$), indicating the presence of a 1',3',4'-trisubstituted benzene nucleus (Table 1). Two of the substituents were suggested to be a phenolic group (δ 5.87, D_2O -exchangeable) at C-4' and a methoxyl group (δ 3.95) at C-3', whose regiochemistry was confirmed by a clear NOE between OMe (δ 3.95) and H-2' (δ 7.07) in a NOESY experiment. The latter substituent was identified as a *trans*- α,β -unsaturated carbonyl group at δ 6.82 and 7.58 (*d*, $J = 16.2\text{ Hz}$) from the downfield signals and its large coupling constant. The other set of deshielded vinyl proton signals at δ 6.84 (*dt*, $J = 15.6, 1.6\text{ Hz}$) and 7.0 (*dt*, $J = 15.6, 7.6\text{ Hz}$), as well as unresolvable multiplets between δ 0.8–1.8, suggested the existence of an alkenyl group bearing a five-carbon long-chain residue attached to the open end of the carbonyl group. A NOESY experiment, H-2 showing NOE to H-4, supported this connectivity. Based on the above analyses, the structure of [6]-dehydroshogaol was established as **1**.

[8]-Dehydroshogaol (**2**) and [10]-dehydroshogaol (**3**) exhibited similar spectroscopic properties to that of **1** (Table 1). The major difference was in their mass spectra which showed a $[\text{M}]^+$ at m/z 302 and 330, respectively, corresponding to the addition of 28 and 56 amu to that of **1** ($[\text{M}]^+$ m/z 274). Consequently, these data led us to deduce the structure of [8]-dehydroshogaol as **2** and [10]-dehydroshogaol as **3**.

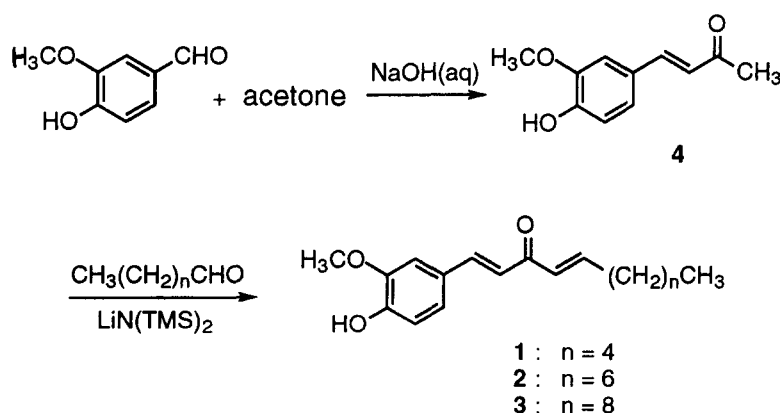
In order to confirm these structures, synthesis of [*n*]-dehydroshogaols was carried out, as summarized in Scheme 1. Aldol condensation between vanillin and acetone in sodium hydroxide gave dehydrogingerone (**4**), and further condensation of **4** with alkanal in the

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Table 1. ^1H NMR spectral data of compounds 1–3 (400 MHz in CDCl_3 , J in Hz)

	1	2	3
H-2'	7.07 (<i>d</i> , 1.6)	7.07 (<i>d</i> , 2.4)	7.08 (<i>d</i> , 2.0)
H-5'	6.93 (<i>d</i> , 8.4)	6.93 (<i>d</i> , 8.0)	6.93 (<i>d</i> , 8.0)
H-6'	7.14 (<i>dd</i> , 8.4, 1.6)	7.13 (<i>dd</i> , 8.0, 2.4)	7.14 (<i>dd</i> , 8.0, 2.0)
3'-OMe	3.95 <i>s</i>	3.94 <i>s</i>	3.94 <i>s</i>
4'-OH†	5.87 <i>br</i>	5.89 <i>br</i>	5.90 <i>br</i>
H-1	7.58 (<i>d</i> , 16.2)	7.57 (<i>d</i> , 16.0)	7.58 (<i>d</i> , 15.6)
H-2	6.82 (<i>d</i> , 16.2)	6.84 (<i>d</i> , 16.0)	6.44 (<i>d</i> , 15.6)
H-4	6.48 (<i>dt</i> , 15.6, 1.6)	6.43 (<i>dt</i> , 16.0, 2.4)	6.82 (<i>d</i> , 16.0)
H-5	7.00 (<i>dt</i> , 15.6, 7.6)	6.99 (<i>dt</i> , 16.0, 7.2)	7.00 (<i>dt</i> , 16.0, 6.8)
H-6	2.27 (<i>qd</i> , 7.6, 1.6)	2.26 (<i>qd</i> , 7.2, 2.8)	2.27 (<i>q</i> , 6.8)
H-7	1.51 <i>m</i>	1.50 <i>m</i>	1.50 <i>m</i>
H-8-H-n	1.33 <i>m</i> (4H)	1.32 <i>m</i> (8H)	1.28 <i>m</i> (12H)
Me (terminal)	0.90 (<i>t</i> , 6.8)	0.90 (<i>t</i> , 7.2)	0.88 (<i>t</i> , 6.8)

† ??

Scheme 1. Synthesis of $[n]$ -dehydroshogaols.

presence of lithium *bis*(trimethylsilyl)amide ($\text{LiN}(\text{TMS})_2$) afforded dehydroshogaols 1–3 in moderate yields [15]. The spectral data (UV, IR, EI, ^1H and ^{13}C NMR) and TLC of the synthetic compounds 1–3 were consistent with the naturally occurring dehydroshogaols.

EXPERIMENTAL

Mps: uncorr. UV: MeOH IR: KBr. MS: direct inlet system. NMR: TMS as int. standard.

Plant material

Fresh ginger, rhizomes of *Z. officinale* Roscoe, were purchases from a market in Tainan, Taiwan.

Extraction and separation

The ginger (64.4 kg) was chopped and then filtered. The filtrate was partitioned between Et_2O and H_2O . The acetone extracts were combined, concentrated, and partitioned between Et_2O and H_2O . The ethereal

solution was subjected to silica gel CC using a gradient of C_6H_6 and Me_2CO as eluent to yield twenty-four frs. The combination of frs 5–8 was repeated rechromatographed to afford 1 (2 mg), 2 (3 mg) and 3 (2 mg), successively.

[6]-Dehydroshogaol (1)

Yellow syrup. HRMS: calcd for $\text{C}_{17}\text{H}_{22}\text{O}_3$, m/z 274.1568 $[\text{M}]^+$, found 274.1571. UV λ_{max} nm ($\log \epsilon$): 258 (3.93), 355 (4.02). IR ν_{max} cm^{-1} : 3354, 2956, 1654, 1625. EIMS m/z (rel. int.): 274 ($[\text{M}]^+$, 56), 217 (50), 177 (100), 145 (29), 137 (91), 117 (13), 89 (16), 77 (15), 55 (33). ^{13}C NMR (CDCl_3): δ 14.0, 22.4, 27.9, 31.4, 32.7, 56.0, 109.7, 114.8, 122.8, 123.3, 127.4, 129.0, 143.3, 147.2, 148.0, 148.1, 189.3.

[8]-Dehydroshogaol (2)

Yellow syrup. HRMS: calcd for $\text{C}_{19}\text{H}_{26}\text{O}_3$, m/z 302.1882 $[\text{M}]^+$, found 302.1881. UV λ_{max} nm ($\log \epsilon$): 258 (3.96), 357 (4.10). IR ν_{max} cm^{-1} : 3395, 2925, 1660, 1614. EIMS m/z (rel. int.): 302 ($[\text{M}]^+$, 26), 217 (67),

204 (25), 177 (100), 150 (25), 145 (32), 137 (83). ^{13}C NMR (CDCl_3): δ 14.0, 22.6, 28.1, 29.0, 29.1, 31.7, 32.7, 55.9, 109.7, 114.8, 122.7, 123.2, 127.3, 129.0, 143.3, 146.8, 148.0, 148.2, 189.3.

[10]-Dehydroshogaol (**3**)

Yellow syrup. HRMS: calcd for $\text{C}_{21}\text{H}_{30}\text{O}_3$, m/z 330.2195 $[\text{M}]^+$, found 330.2193. UV λ_{max} nm ($\log \epsilon$): 257 (4.04), 355 (4.18). IR ν_{max} cm^{-1} : 3533, 2925, 1660, 1614. EIMS m/z (rel. int.): 330 ($[\text{M}]^+$, 26), 217 (82), 204 (29), 177 (100), 150 (32), 145 (34), 137 (98), 117 (21), 69 (20), 57 (30), 55 (46). ^{13}C NMR (CDCl_3): δ 14.1, 22.7, 28.2, 29.2, 29.3, 29.4, 29.5, 31.8, 32.7, 55.9, 109.7, 114.8, 122.7, 123.3, 127.3, 129.0, 143.4, 146.8, 148.0, 148.2, 189.3.

General procedure for synthesis of [n]-dehydroshogaols

Dehydrogingerone (**4**) was obtained by the aldol condensation of vanillin (2.5 g, 16.4 mmol) with Me_2CO (100 ml) in 10% aq. NaOH [15]. Then, **4** (2 g, 10.4 mmol) in 10 ml of THF was added dropwise, over 10 min, to a THF soln (75 ml) of $\text{LiN}(\text{TMS})_2$ (20.8 mmol) at 0° under Ar. After a further 1 h, alkanal (10.4 mmol) was added and the mixt. stirred at 0° for 3 h. EtOAc was then added and the resulting mixt. washed with 5% aq. HCl and satd aq. NaCl. The organic layer was dried (Na_2SO_4) and concd *in vacuo*. The residue was purified by CC to afford dehydroshogaols **1** (1 g, 35% yield), **2** (1.2 g, 38% yield) and **3** (1.1 g, 32% yield) and identified by comparison with the corresponding naturally occurring compounds.

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