is eliminated by the preparations studied in doses of more than 50 mg per kg body weight of the animal.

In a dose consisting of 10% of  $LD_{50}$ , the preparations increase the hypothermia induced by a reserpine-like compound PO-4-1284, but do not change, or inappreciably potentiate, phenamine stereotypia in mice. In experiments on narcotized cats, the preparations (5 mg/ kg, intravenously) slightly increase (by 10-20%) the effect of noradrenaline on the arterial pressure.

The compounds II-V studied have an inherent central neurotropic action. In contrast to other known tricyclic systems (for example,  $\beta$ -carbonyls) exhibiting antidepressant properties, the carbazole derivatives studied have inappreciable adrenopotentiating properties and a more pronounced depressant action on the central nervous system.

#### LITERATURE CITED

- 1. R. M. Ferris, M. Harfenist, G. M. McKenzie, et al., J. Pharm. Pharmacol., <u>34</u>, 388-390 (1982).
- I. P. Zherebtsov, V. P. Lopatinskii, N. M. Rovkina, et al., Izv. Tomsk. Politekh. Inst., 272, 189-194 (1974).
- 3. D. R. Dauer, Applications of Absorption Spectroscopy of Organic Compounds [in Russian], Moscow (1970), p. 104.
- 4. N. K. Barkov and V. V. Zakusov, Farmakol. Toksikol., No. 6, 730-739 (1973).

# SYNTHESIS AND ANTISPASMODIC ACTIVITY OF 2-ALKYL SUBSTITUTED

THIENO[2,3-d]PYRIMIDIN-4-ONES

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A. P. Mkrtchyan, S. G. Kazaryan,A. S. Noravyan, S. A. Vartanyan,I. A. Dzhagatspanyan, N. E. Akopyan,and I. M. Nazaryan

During recent years a systematic search has been carried out for new antispasmodic compounds in the condensed thieno[2,3-d]pyrimidine derivatives. The present work is a continuation of the previous studies in [1, 2] and is devoted to the synthesis of new 2-alkyl substituted pyrano(thiopyrano)[4', 3':4,5]thieno[2,3-d]pyrimidin-4-ones.

2-Amino-3-carbethoxythiophenes served as starting materials for synthesis and were condensed with six-membered heterocycles containing sulfur or oxygen (I, II) [3]. By reacting the latter with acid chlorides a series of acylated derivatives (III-XV) was obtained. On heating in methanolic ammonia solution (V, VI, VIII-X) were converted into the corresponding 2-alkyl substituted thieno[2,3-d]pyrimidin-4-ones (XVI-XXI).



$$\begin{split} & 1:X = 0; \ II:X = S; \ III:X = O, \ R = C_2H_5; \ IV:X = O, \ R = C_3H_7; \ V, \ XVI:X = O, \\ & R = C_4H_9; \ VI, \ XVII:X = O, \ R = C_5H_{11}; \ VII:X = S, \ R = C_2H_5; \ VIII, \ XVIII:X = S, \\ & R = C_3H_7; \ IX, \ XIX:X = S; \ R = C_4H_9; \ X, \ XX:X = S, \ R = C_5H_{11}; \ XI, \ XXI:X = O, \\ & R = coumaryl_XII:X = S, \ R = iso-C_3H_7; \ XIII:X = O, \ R = p-iso-C_4H_9O) \ C_6H_4; \\ & XIV:X = O, \ R = iso-C_4H_9; \ XV:X = O, \ R = C_9H_{19}. \end{split}$$

# EXPERIMENTAL PHARMACOLOGY

Experiments were carried out on mice of weight 18-22 g using procedures for the assessment of substances possessing antispasmodic properties, viz., the maximum electroshock test

A. L. Mndzhoyan Institute of Fine Organic Chemistry, Academy of Sciences of the Armenian SSR, Erevan. Translated from Khimiko-farmatsevtichevskii Zhurnal, Vol. 18, No. 4, pp. 451-454, April, 1984. Original article submitted July 29, 1983.



Fig. 1. Comparative anticorazole activity of 2-substituted thieno[2,3-d]pyrimidin-4-ones. The  $ED_{50}$  (mg/kg) is given on the ordinate in a logarithmic scale and compounds are shown on the abscissa.

[4] and the subcutaneous anticorazole test [5]. The central m-cholinolytic effect was studied in the prevention of arecoline tremor [6] and the n-cholinolytic action in protection from nicotine tremor [7]. In addition the appearance of "neurological deficit", an undesirable action of preparations, was determined. For this purpose procedures were used for "mice laying down on a grid" for recording the orientation reaction [8] and the "rotating rod" test of [9] for clarifying disturbed motor coordination. The daily toxicity in mice was also determined.

Substances were administered intraperitoneally in the form of a Tween-80 suspension. The emulsifying agent was administered to control animals. Statistical treatment of data was carried out by the probit analysis method of [10]. The known anticonvulsants luminal and zarontin were used as reference substances.

Study of the antispasmodic activity of nine of the indicated substances showed that at a dose of 200 mg/kg no substance gave a protective effect in relation to nicotine, arecoline, and electroshock. Their action in relation to corazole was displayed somewhat differently. It is evident from Fig. 1 that compounds (V, VI, XI) possessed insignificant antispasmodic activity. The presence of a pyrimidine ring in (XVI, XVII, XXI) led to a sharp growth in the effect for (XVI) and (XVII). The ED<sub>50</sub> for these compounds was 32 (13.3-76.8) and 32 (12.3-83.2) mg/kg, respectively. However the addition of a coumaryl radical was accompanied once again by a weakening anticorazole action. In the series of thiopyranothieno[2,3-d]pyrimidines (XVIII-XX) the most marked antagonism of corazole was possessed by by compound (XVIII) the ED<sub>50</sub> of which was 56 (30.7-84) mg/kg. An increase in the radical to C<sub>4</sub>H<sub>9</sub> or C<sub>5</sub>H<sub>11</sub> led to a weakening of the antispasmodic action. The most effective among all the studied compounds (XVI-XVIII) differed statistically insignificantly from one another. In the anticorazole test they superseded zarontin [ED<sub>50</sub> = 155 (117.5-204.5) mg/kg] but were superseded by luminal [ED<sub>50</sub> = 12.5 (8.3-18.7) mg/kg] (see Fig. 1).

At doses of 700-2000 mg/kg the studied substances caused disturbances of motor coordination and depressed the orienting reaction in 100% animals. Death of all experimental mice was recorded at a dose of 1000 mg/kg for compounds (V, XVIII-XX) and at 2000-3000 mg/kg for compounds (XVI-XVII).

Thus substances have been discovered among the thienopyrans and thienothiopyrans condensed with pyrimidine which possess marked anticorazole activity. It should be recorded that the derivatives of thiopyranothieno[2,3-d]pyrimidin-4-one were somewhat more toxic than the pyranothieno[2,3-d]pyrimidin-4-ones.

### EXPERIMENTAL CHEMISTRY

IR spectra were taken on a UR-20 instrument (East Germany) in Nujol mulls. PMR spectra were measured on a Varian T-60 (USA) instrument in chloroform, internal standard was TMS.

Compounds (III-XV)	Calculated, 🌾	s	7,42 7,42 7,42 7,42 7,42 7,42 7,42 7,42		Calculated, %	s	$\begin{array}{c} 10,96\\ 10,46\\ 21,63\\ 20,79\\ 19,88\\ 9,09\end{array}$
		z	4,4,4,0,4,4,0,3,3,4,9,4,6,4,4,4,4,4,4,4,4,4,4,4,4,4,4,4,4			z	9,57 9,14 9,44 8,68 7,79
		н	$\begin{array}{c} 6,79\\ 6,74\\ 6,74\\ 6,78\\ 6,78\\ 6,78\\ 6,78\\ 6,78\\ 6,78\\ 6,78\\ 7,36\\ 6,78\\ 7,36\\$			Н	6,89 6,80 6,53 6,53 6,53 77
		υ	57,85 57,85 60,15 61,16 61,15 55,27 58,502 58,502 58,27 63,14 63,14 64,01 64,01 64,01 64,01 64,01 64,01			υ	61,61 62,70 56,72 58,40 59,59 64,75
	Empirical formula		$C_{16}^{1} H_{21}^{1} NO_{4}^{1} S$ $C_{16}^{1} H_{23}^{2} NO_{4}^{1} S$ $C_{16}^{1} H_{23}^{2} NO_{4}^{1} S$ $C_{16}^{1} H_{23}^{2} NO_{3}^{2} S$ $C_{16}^{1} H_{23}^{2} NO_{3}^{2} S$ $C_{16}^{1} H_{23}^{2} NO_{5}^{2} S$ $C_{17}^{1} H_{23}^{2} NO_{5}^{2} S$ $C_{17}^{1} H_{23}^{2} NO_{5}^{2} S$ $C_{17}^{2} H_{23}^{2} NO_{5}^{2} S$ $C_{17}^{2} H_{23}^{2} NO_{5}^{2} S$ $C_{17}^{2} H_{23}^{2} NO_{5}^{2} S$		Emninical formula		$C_{15}H_{20}N_2O_2S$ $C_{10}H_{22}N_2O_2S$ $C_{14}H_{16}N_2O_2S$ $C_{14}H_{16}N_2OS_2$ $C_{16}H_{20}N_2OS_2$ $C_{16}H_{22}N_2OS_2$ $C_{19}H_{20}N_2OS_2$
	Found, do	s .	10,106 10,106 10,10 19,25 19,25 19,26 19,26 19,26 19,26 19,26 19,10 8,26 8,26 8,26 8,26 8,26 8,26 8,26 8,26		Empirical formula	s	10,80 10,55 20,89 19,90 9,27
		N	4,121 4,16 4,50 3,87 3,87 3,80 3,80 3,16 3,133 3,16 3,16 3,16 3,16 3,16 3,1			z	9,68 9,24 3,56 7,88 8,56
		Η	6,62 7,723 6,42 6,42 6,86 6,86 6,86 8,55 7,7,7 7,51 8,86 8,86 8,86 8,86 8,86 8,86 8,86 8,8			H	6,76 7,10 6,76 6,71 6,73 4,36
		υ	57,70 55,000 56,00 58,73 58,73 58,73 58,73 58,73 58,73 58,73 58,73 59,48 56,48 56,48 56,48 56,48 56,48 56,48 56,48 56,48 56,53 57,09 56,48 56,53 57,09 56,53 57,70 56,53 57,70 56,53 57,70 56,53 57,70 56,53 57,70 56,53 57,70 56,53 57,70 56,53 57,700 57,700 57,700 57,700 57,700 57,70000000000			U	61,50 62,83 56,65 58,50 59,74 64,60
	mp, °C		$\begin{array}{c} 119-20\\ 138-9\\ 107\\ 90\\ 68-70\\ 68-70\\ 68-70\\ 64-5\\ 64-5\\ 56\\ 141-43\\ 64-5\\ 64-5\\ 64-5\\ 64-5\\ 64-5\end{array}$	(IXX-IV)	mp, °C		239 212 262—3 243—4 232—1 232—3
	Yield, %		99995388886667,07 7,09999538888867,07 7,0621444,000 7,06214444,000	Compounds (X	Yield, %		51,3 51,3 46,6 64,9 64,9 59,0 59,0
TABLE 1.	Compound	ninadiiiaa	HAVE A CONSTRUCT OF CONSTRUCT O	TABLE 2.	patioamo	authonin	IXX IIIXX XIXX XIXX XXX XXX

The value of signals is given on the  $\delta$  scale. TLC was carried out on Silufol-254 (Czechos-lovakia) plates.

<u>2-Substituted 3-Carbethoxy-5,5-dimethyl-4,5-dihydro-7H-thieno[2,3-c]pyranes(thiopyranes)</u> (<u>III-XV</u>). Compound (I) or (II) (0.01 mole) was dissolved in dry dioxane (30 ml). After adding acid chloride (0.01 mole) the mixture was boiled for 3 h, cooled, and poured into cold water (150 ml). The crystals were filtered off, washed with water, cold ether, and dried. Constants are shown in Table 1.

IR spectrum,  $v_{max}$ , cm<sup>-1</sup>: IV 1660 (NCO), 1680 (COO), 3280-3200 (NH), PMR spectrum (CDCl<sub>3</sub>),  $\delta$ , ppm: IV 1.00 t (3 H, CH<sub>3</sub>), 1.30 s (6H-2CH<sub>2</sub>), 1.43 s (3H-OCH<sub>2</sub>CH<sub>3</sub>), 1.85 s (2H-CH<sub>2</sub>), 2.46 s (2H-CH<sub>3</sub>C=0), 2.75 s [2H(CH<sub>3</sub>)<sub>2</sub>C-CH<sub>2</sub>], 4.33 s (2H-OCH<sub>2</sub>CH<sub>3</sub>), 4.66 s (2H-CH<sub>2</sub>O), 11.26 s (1H-NH).

IR spectrum,  $v_{max}$ , cm<sup>-1</sup>: III, V-XV 1688-1665 (NCO), 1715-1690 (COO), 3250-3220 (NH).

PMR spectrum (CDCl<sub>3</sub>),  $\delta$ , ppm: V 0.80-2.00 m [16H-(CH<sub>3</sub>)<sub>4</sub>, (CH<sub>2</sub>)<sub>2</sub>]; VI 0.80-2.00 m [18H-(CH<sub>3</sub>)<sub>4</sub>, (CH<sub>2</sub>)<sub>3</sub>]; VII 0.80-2.00 m [12H-(CH<sub>3</sub>)<sub>4</sub>], 2.40 t (2H-CH<sub>2</sub>); VIII 0.80-2.00 m [12H-(CH<sub>3</sub>)<sub>4</sub>]; IX 0.80-2.00 m [16H-(CH<sub>3</sub>)<sub>4</sub>, (CH<sub>2</sub>)<sub>2</sub>]; X 0.80-2.00 m [18H+(CH<sub>3</sub>)<sub>4</sub>, (CH<sub>2</sub>)<sub>3</sub>].

The chemical shifts of the remaining protons were practically indistinguishable from those of compound (IV).

Mass spectrum m/z (I rel.) XI 399(25), 398(100), 384(5), 370(5), 354(6), 353(6), 352(9), 343(11), 341(17), 342(60), 325(7), 324(33), 295(18), 270(8), 269(35), 254(16), 241(22), 146(9), 145(75).

<u>2-Alkyl-4-oxo-6,6-dimethyl-5,6-dihydro-8H-pyrano(thiopyrano)[4',3':4,5]thieno[2,3-d]-pyrimidines (XVI-XXI).</u> A mixture of (V) (or VI, VIII-XI) (0.01 mole) in 25% methanolic ammonia solution (50 ml) was heated at 100-120°C in a metal bomb for 18 h. The bomb was cooled to -10°C, opened, and the contents poured into an equal volume of ether. The precipitated crystals were filtered off, washed with ether, and with ethanol. The white crystals were dried in a vacuum desiccator. Compounds (XVI-XXI) were obtained. Constants are shown in Table 2.

IR spectrum,  $v_{max}$ , cm<sup>-1</sup>: XVI-XXI 1620-1610 (C=N), 1680-1660 (NC=O) 3110-3180 (NH); XXI 1590 (C=C).

PMR spectrum (CDCl<sub>3</sub>), δ, ppm: XVI 0.80-2.00 m [13H+(CH<sub>3</sub>)<sub>3</sub>, (CH<sub>2</sub>)<sub>2</sub>], 2.80 t (2H-CH<sub>2</sub>), 3.00 t (2H-CH<sub>2</sub>), 4.80 t (2H-CH<sub>2</sub>O); XVII 0.70-2.00 m [15H-(CH<sub>3</sub>)<sub>3</sub>, (CH<sub>2</sub>)<sub>3</sub>]; XVIII 1.00-2.00 m [11H-(CH<sub>3</sub>)<sub>3</sub>, CH<sub>2</sub>]; XIX 1.00-2.00 m [13H-(CH<sub>3</sub>)<sub>3</sub>, (CH<sub>2</sub>)<sub>2</sub>]; XX 1.00-2.02 m [15H-CH<sub>3</sub>)<sub>3</sub>, (CH<sub>2</sub>)<sub>3</sub>].

PMR spectrum (CF<sub>3</sub>COOD), δ, ppm: XXI 1.60 s (6H, 2CH<sub>3</sub>), 2.95 t (2H, CH<sub>2</sub>), 5.03 t (2H-CH<sub>2</sub>0), 7.35-7.90 m (5H-5CH).

#### LITERATURE CITED

- A. S. Noravyan, A. P. Mkrtchyan, I. A. Dzhagatspanyan, et al., Khim.-farm. Zh., No. 9, 38-42 (1977).
- 2. A. S. Noravyan, A. P. Mkrtchyan, I. A. Dzhagatspanyan, et al., ibid., No. 2, 37-40 (1980).
- 3. A. S. Noravyan, A. P. Mkrtchyan, I. A. Dzhagatspanyan, et al., ibid., No. 8, 20-24 (1977).
- 4. J. E. P. Toman, E. A. Swineyard, and L. S. Goodman, J. Neurophysiol., 8, 231-239 (1946).
- 5. E. A. Swineyard, W. Brown, and L. S. Goodman, J. Pharmacol. Exp. Ther., <u>106</u>, 319-330 (1952).
- 6. N. V. Golyakhovskii, Farmakol. Toksikol, No. 1, 8-10 (1948).
- 7. D. Bovet and V. G. Longo, J. Pharmacol. Exp. Ther., 102, 22-25 (1951).
- 8. Yu. I. Vikhlyaev and T. A. Klygul', Farmakol. Toksikol., No. 1, 30-31 (1971).
- 9. N. W. Dunham and T. S. Miya, J. Am. Pharm. Assoc., 46, 208-209 (1957).
- 10. J. Litchfield and F. Wilcoxon, J. Pharmacol. Exp. Ther., 96, 99-114 (1949).