Enantioselective Synthesis of D-threo-Methylphenidate

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Methylphenidate (Ritalin) is the most commonly prescribed psychotropic medication for children in the United States.¹ It is currently marketed as a racemate, although the D-threo isomer is ca. 13 times more active than its mirror image.² Various strategies for the preparation of the more active isomer of methylphenidate have been examined, including resolution³ and synthesis from scalemic precursors.⁴ We report herein that the rhodium-mediated C-H insertion reaction of methyl phenyldiazoacetate 1 with N-Boc-piperidine 2 (R = N-Boc) leads to a remarkably efficient and highly stereoselective process for the formation of the bioactive isomer of methylphenidate 3 (Scheme 1).^{5,6}

The pioneering work of Doyle has established that scalemic ligands on rhodium can lead to high levels of enantioselectivity in the C-H insertion reaction.^{5,7} While the intramolecular C-H insertion pathway in our recently disclosed diastereoselective synthesis of methylphenidate⁸ was not sensitive to metal catalysis,⁹ we are delighted to report that the intermolecular C-H insertion reaction proceeds in the presence of the Doyle catalyst Rh₂(5R-MEPY)₄¹⁰ with exceedingly high levels of stereochemical control.¹¹ Reaction of methyl phenyldiazoacetate 1¹² with N-Bocpiperidine 2^{13} (R = N-Boc) in the presence of 1 mol % of Rh₂(5R-MEPY)₄ leads to the selective formation of D-threomethylphenidate in 94% de and 69% ee.14 Two recrystallizations of the crude product from ethanol/diethyl ether gave D-threomethylphenidate in 95% de and >95% ee.

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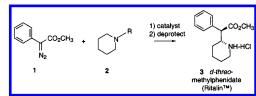
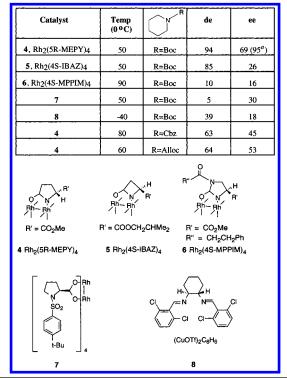


Table 1



^a The initially observed 69% ee was increased to >95% by two recrystallizations from 1:1 v/v diethyl ether/ethanol.

The effect of varying both the catalyst employed and the protecting group on the piperidine nitrogen on the stereochemical outcome of the C-H insertion reaction was next examined, and our results are summarized in Table 1. Both the Rh₂(IBAZ)₄ 5¹⁵ and the Rh₂(MPPIM)₄ 6¹⁶ catalysts led to attenuated diastereoand enantioselectivities. Similarly, the McKervey proline sulfonamide catalyst 717 gave only modest diastereoselectivity and enantioselectivity. Finally, the Jacobsen catalyst 8, which has recently been reported to effect highly enantioselective Si-H insertion,¹⁸ led to only modest stereoselection in this reaction.

The stereoselective formation of the D-threo stereoisomer of methylphenidate 3 with $Rh_2(5R-MEPY)_4$ 4 can be rationalized as shown in Figure 1. The carbene can coordinate to the rhodium

⁽¹⁴⁾ Professor Huw Davies (State University of New York, Buffalo) has observed that the use of the rhodium prolinate catalyst described in ref 11 in this reaction leads to the formation of methylphenidate with attenuated diastereoselectivity. We thank him for sharing his results with us prior to publication.

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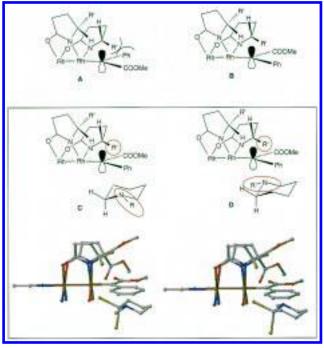


Figure 1.

as indicated in either A or B, and the orientation shown in B minimizes the interaction between the phenyl ring and the indicated ester moiety (R') on the pyrrolidine ring.¹⁹ Approach of the equatorial C–H bond of the N-Boc piperidine²⁰ to the carbene can occur as shown in either C or D, in which the protected piperidine nitrogen is oriented away from or toward

the carbomethoxy group on the indicated pyrrolidinone ring as shown in C and D, respectively. Support for this model for the reaction of **1** and **2** is obtained by decreasing the size of R, the nitrogen protecting group (see Table 1, R = Cbz or Alloc), which leads to an attenuation of the stereoselectivity of the insertion reaction. It is interesting to note that reaction of N-Bocdiethylamine with methyl phenyldiazoacetate **1** in the presence of Rh₂(5R-MEPY)₄ **4** occurs in a stereorandom manner, indicating that the more highly ordered six-membered ring substrate is critical to the stereoselectivity of the reaction.

These results establish a highly stereoselective and efficient pathway for the synthesis of methylphenidate in scalemic form. Further studies on the generality of both the enantioselectivities and diastereoselectivities that have distinguished this work are currently in progress, and our results will be reported in due course.

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Supporting Information Available: Synthetic procedures and spectroscopic data for the preparation of *D-threo*-methylphenidate (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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