1,3,4,5-Tetrahydroazepin-2-ones by Dearomatising Anionic Cyclisation of *N*-Allyl-1-Naphthamides

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Abstract: On treatment with *t*-BuLi and DMPU, 1-naphthamides bearing *N*-allyl or *N*-prenyl substituents cyclise to give a mixture of products from which seven-membered lactams can be isolated in up to 73% yield – the first example of an organolithium–C=C cyclisation leading to a seven-membered ring.

Key words: amide, cyclization, lithium, azepin, dearomatisation

We recently described a cyclisation reaction of *N*-benzyl naphthamides¹ **1** and *N*-benzyl benzamides² which is triggered by deprotonation with strong base (*t*-BuLi) in the presence of an additive such as HMPA or DMPU, and which leads to loss of aromaticity in one ring (Scheme 1). The products of the cyclisation **2** contain a substituted pyrrolidinone ring, formed with high stereoselectivity, but invariably bearing a phenyl group in the 2-position. In an attempt to broaden the scope of this new reaction, we surveyed some other nitrogen substituents with moderate anion stabilising ability,³ and found that allylic *N*-substituents also allowed the cyclisation to take place.



A dearomatising anionic cyclisation; (i) t-BuLi, -78 °C, THF, 2 h; (ii) DMPU or HMPA (6 equiv.), -78 °C, 16 h; (iii) NH₄Cl (82%) Scheme 1

We made *N*-*t*-butyl-*N*-prenyl-1-naphthamide **5** by *N*-alkylation of *N*-*t*-butyl-1-naphthamide **4** (Scheme 2). The amide was treated with *t*-BuLi at -78 °C for 2 h and the resulting organolithium **8** quenched with MeI. Only the 2methyl product **9** was formed, suggesting that the organolithium is principally an ortholithiated and not an α -lithiated species. Prenyl must be a less powerful anionstabiliser than phenyl, since lithiated *N*-benzyl naphthamides react with MeI to give only the α -methylated product.^{1,4}



Synthesis of the starting materials; (i) t-BuNH₂, NaOH, CH_2Cl_2 (98%); (ii) NaH, DMF; (iii) prenyl bromide (46%); (iv) allyl bromide (94%); (v) cinnamyl bromide (68%)

Scheme 2



Cyclisation of *N*-allyl-1-naphthamides; (i) *t*-BuLi, THF, $-78 \degree C$, 2 h; (ii) MeI (91%, $R^1 = R^2 = Me$); (iii) DMPU (6 equiv.), $-50 \degree C$, 16 h; (iv) HMPA (6 equiv.), $-78 \degree C$, 16 h; (v) NH₄Cl Scheme 3

In the *N*-benzyl series, we had found that DMPU or HMPA promotes the cyclisation by favouring an anion translocation from the *ortho* to the α position.⁴ In the presence of DMPU at -50 °C for 16 h, the ortholithiated *N*-prenyl amide **8** must also undergo anion translocation to **10**, because protonation returned a moderate yield of dearomatised cyclised products **13a** and **14a**, presumably via the enolates **11** and **12**. Remarkably, the major product **13** is a tetrahydroazepinone formed by attack at the γ position of the lithiated prenyl group.^{5,6} Seven-membered rings are typically slow to form by cyclisation reactions, and to our knowledge had never been made previously by an organo-lithium cyclisation onto a C=C double bond.⁷ In this case, the probable *Z*-geometry⁶ of the intermediate **10** will favour 7-ring formation.

In view of the structural similarity between these rare⁸ cyclic *N*-acyl enamines and the pharmaceutically important benzazepin-2-ones,^{9,10} we aimed to optimise the formation of the seven- over the five-membered ring product. Removing the geminal methyl groups of the prenyl substituent improved the yield of the reaction and increased the selectivity for seven-membered ring formation. Allylation of *N*-*t*-butyl-1-naphthamide gave the starting material **6**, which was lithiated with *t*-BuLi and cyclised in the presence of DMPU or HMPA at -50 °C or at -78 °C. Protonation gave the seven-membered ring product **13b** in 73% isolated yield,¹¹ along with only 17% of the fivemembered ring product **14b**. The structure and stereochemistry of the product was confirmed by an X-ray crystal structure, shown in the Figure.



X-ray crystal structure of **13b** Figure

The initial products of the cyclisation, lactam enolates **11** and **12**, could be alkylated to give more highly substituted seven-membered ring products, as shown in Scheme 4 and the Table. Five-membered rings were still among the by-products of the reaction, but single diastereoisomers were formed in each case, even with benzaldehyde, which gave a single aldol product **15d** of unknown stereochemistry at the hydroxyl-bearing centre.



Cyclisation–alkylation of **6**; (i) *t*-BuLi, THF, –78 °C, 2 h; (ii) DMPU (6 equiv.), –50 °C, 16 h; (iii) MeI; (iv) EtI; (v) BnBr; (vi) PhCHO **Scheme 4**

entry	electrophile	15, yield (%)	16, yield (%)
1	Mel	15a , 56	16a , 25
2	EtI	15b , 66	16b , 27
3	BnBr	15c , 68	16c , 23
4	PhCHO	15d , 48	16d , 9

The C=C double bonds of **13b** could be reacted selectively under hydrogenation or hydrolysis conditions. Hydrogen over palladium on carbon reduced the styrene double bond of **13b** to give a quantitative yield of **17**. Hydrolysis of **13b**'s *N*-acyl enamine group with aqueous HCl in THF gave the aldehyde **18** in 76% yield (Scheme 5), in which the *trans* stereochemistry presumably arises by epimerisation of a first-formed *cis* isomer.¹² The cyclisation/hydrolysis sequence represents in this case a formal addition of the homoenolate of propionaldehyde¹³ to the naphthalene ring.

In a final attempt to increase still further the selectivity of the reaction for seven-membered ring formation we tried cyclising the *N*-cinnamyl compound **7**. This time, after treatment with *t*-BuLi and DMPU (-50 °C, 16 h) no cyclisation product was observed: instead, we obtained a quantitative yield of the unstable imine **19**, which hydrolysed during purification to give the aldehyde **20** in 46% yield. The imine is formed by an N \rightarrow C acyl transfer:¹⁴ the organolithium attacks the amide carbonyl group instead of the aromatic ring.



Transformations of the seven-membered lactam;(i) H₂ (1 atm.), Pd/C, 16 h (quant.); (ii) HCl, H₂O, THF (76%) Scheme 5



Attempted cyclisation of N-cinnamyl-1-naphthamide 7; (i) *t*-BuLi, THF, -78 °C, 2 h; (ii) DMPU (6 equiv.), -50 °C, 16 h; (iii) NH₄Cl (quant.); (iv) SiO₂ chromatography (46%)

Scheme 6

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References and Notes

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