FACILE SYNTHESIS OF BIOLOGICALLY ACTIVE LINEAR BISAROYL BENZODIFURANS BY PTC AND SOLVENT FREE MICROWAVE IRRADIATION

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Abstract

2,6-Diaroyl/naphthoyl-3,5-dialkyl/phenyl-benzo[1,2-b;5,4-b']difurans (3a-e, 4a-g and 5a-g) have been synthesized by condensing 4,6-diacyl/diaroyl resorcinols (1 i-iii) with various *p*-substituted α bromoketones (2a-g) by (a) phase transfer catalysis method & (b) microwave irradiation. A comparison has been made between the two methods. Microwave irradiation has been found to be an efficient route for the synthesis of bisaroylbenzodifurans. All the compounds have been screened for anti-bacterial (3a-e, 4a-g and 5b-g) and some of the selected compounds have been tested for antiimplantation activity. The benzodifurans 3c, 3d, 3e, 4a, 4b, 4d, 4f and 5g have shown excellent activity against gram positive bacteria and the compound 4c has shown 67% anti-implantation activity at 10 mg/Kg/rat/day on albino rats.

Introduction

The naturally occurring and synthetic benzofuran derivatives are known to be associated with biological and pharmacological activities. Methyl-2-methyl-5-hydroxy-6-acetyl-benzofuran-3-carboxylate (1), 2-ethyl-3-(3',5'-diiodo-4-hydroxybenzoyl)benzofuran (2) have been reported to exhibit anti-inflammatory, analgesic, anti-spasmodic and coronary vasodilatory properties. 2,6,8-trimethyl benzo[1,2-b; 5,4-b']difuran (3) is found to exhibit potent dermal photosensitizing activity.5-acetyl-6-cyanomethoxy-4-methoxybenzofuran,3-acetyl-2-amido-4-methoxydifurobenzene(4), and 2-nitro benzo[1,2-b; 5,4-b']difuran (5) were found active against gram positive and gram negative bacteria. 3-methyl-5-acetyl-6-hydroxybenzofuran (6), 2-(4-hydroxybenzoyl)benzofuran (7), 2-phenyl-3-[p-(β -diethylaminoethoxyphenyl)]-6-methoxybenzofuran (8) have shown potent anti-implantation activity. We have reported 100% anti-implantation activity of 2,6-dibenzoyl-3,5-dimethyl-benzo[1,2-b; 5,4-b']difuran (9) (3a) at 2 mg/kg/rat/day on albino rats.

It is evident from the literature that, 2,6-diaroyl/naphthoyl-3,5-dialkyl/phenyl-benzo[1,2-b;5,4-b']difurans were not reported for anti-bacterial activity.

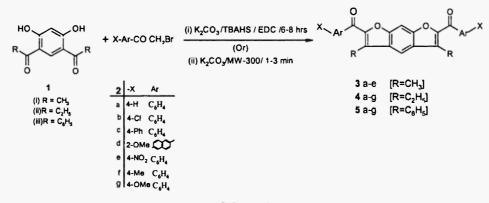
Encouraged by our earlier reported anti-implantation activity results of aroyl benzodifurans^{6,9}, we have designed the synthesis of bisaroylbenzodifurans (3a-e, 4a-g, 5a-g) with ethyl and phenyl substitutions at 3 and 5 positions, to evolve their anti-implantation and anti-bacterial activities. So in the present communication, we report an effective method of synthesis of 2,6-bis aroyl / naphthoyl-3,5-dialkyl/ phenyl benzodifurans.

We have reported the synthesis of 2,6-dibenzoyl-3,5-dimethyl-benzo[1,2-b;5,4-b']difurans under baked K_2CO_3 - anhydrous acetone method, and also under PTC condition (9). We have found that PTC method was an efficient method for the synthesis of bisaroyl benzodifuran.

In the present communication, we report the synthesis of 2,6-diaroyl /naphthoyl-3,5-di diethyl/phenyl-benzo[1,2-b;5,4-b'] difurans (**3a-e**, **4a-g** and **5a-g**) under phase transfer catalytic method, i.e., by condensing 4,6-dipropionyl / benzoyl resorcinol (10) (**1** i-iii) with various *p*-substituted α -bromoketones (11-15) (**2a-g**) in presence of tetrabutyl ammonium hydrogen sulphate as a catalyst in ethylenedichloride for 6-8 hours at 55-60°. This procedure avoids anhydrous conditions and long duration of reaction time.

There has been a growing interest in the use of microwave heating in organic synthesis. This has prompted us to synthesize the linear bisaroylbenzodifurans under microwave irradiation. In the present communication, we have adopted the microwave irradiation technique for the synthesis of bisaroylbenzodifurans under solvent free conditions for the first time. The 4,6-diacyl/aroyl resorcinols and the substituted α -bromoketones (2a-g) were doped with baked K₂CO₃ as a solid support, and the mixture was irradiated at 300W for a period of 1-3 minutes. After irradiation, the inorganic substances were removed and the resulting product was crystallized from methanol. The results were tabulated in Table 2. A comparison was also conducted between the two methods of

synthesis (PTC and MW) from the available data (in **Table 2**). Microwave method is found to be convenient and simple. It does not require toxic solvents and the time of the reaction are also curtailed with reference to PTC method and the yields obtained from MW method are encouraging. The schematic representation of the above methods was depicted in **Scheme 1**.



Scheme 1

The benzodifurans were characterized by UV, IR, NMR and Mass spectral data (**Table 1**). The UV spectra of all the compounds have displayed two absorption bands in the regions λ_{max} 322-327 and 355-359 nm as compared with unsubstituted benzofuran, which showed three absorption bands at λ_{max} 245, 275 and 282 nm. It is evident that all the bands in these benzodifurans (**3a-e**, 4**a-g** and **5a-g**) are observed at longer wavelength region. This bathochromic shift is due to the presence of

In the IR spectra of **3a-e**, **4a-g** and **5a-g** showed three bands were observed at 1600-1680 (C=O str), 1550-1590 (C=C str) and 1230-1260 cm⁻¹.

a second furan ring and two aroyl groups, which may facilitate the extended conjugation.

In the NMR spectra of the compounds, the ring closure and the linear nature of the benzodifurans were confirmed by the following observations.

- (i) The protons at C4 and C8 positions were observed as singlets at δ 7.5-8.3
- (ii) The four protons adjacent to aroyl group i.e., $C_{2',6',2'',6''}$ were observed as doublet at δ 7.8-8.3

	C. H, N, O Analysis	Cal: (%) Found (%)	СН	79 18, 4, 56 79.12, 4.90			67.39, 3.46 67.36; 3.48		83.52, 4.76 83.54; 4.78			77.97; 4.73 77.94; 4.70			31 64.42; 3.34		79.60; 5.25 79.58; 5.24		68.44: 4.10 68.40: 4.08			83,60; 5,26 83,56; 5,25
	C H, N	Cal: (%)	СН	79 18, 4			67.39,3,4		83.52, 4.			77.97;4.			64 46 3.31		79.60; 5.		68.44 4			83,60; 5.2
×	Mass	m/z (% of abundance)		354(6); 365(23): 310(12);	158(22), 130(30);102(14);	105(56); 77(74)	I		1			1			1		422(4),366(25),338(12),	310(20),158(30),130(15), 105(75) 102(8) 77/55)				1
X_A_B_0 2 R_5 2 R_5 3 R_5 3 R_	BDF U.V I.R. ¹ HNMR	δ (ppm)		7.9-8.1(d/H,C::6,2',6) (7.95(s,1H,C1);	7.4-7.6(m 7H,C _{3',4',5',3',4',5'} , c ₈) and	2.7(s, 6H, CH ₃)	8.1(d,4H,C ₂ (t,2,6'),7.95(s,1H,C ₁),7.6(s,1H,C ₁);	7.45(d,4H,C _{3,5,3} ,5) and 2.7(s,6H,CH ₃)	8.15(d,4H,C _{1,6/2} ,6 ⁻),7.95(s,1H,C ₁),7.6(s, ¹ H C ₁);	7.45-7.55(m,10H);7.40(d,4H,C _{3'5;3'5}) and	2.75(s,6HCH ₂)	8 2(d,4H,C _{1,3,1} ,3-),8.1(s,1H,C ₁);	7.7-7.95(m,9H,C4,5,7,8,4, 5,7, 8, & C,);	3.97(s 6H OCH ₃);2.8(s 6H CH ₃)					8.15.8.0(d 4H.C).7.9(s.1H.C.):	7.5-7.15(m,5H,C ₃ , ₅ , ₃ , ₅ , ₆ C ₈);	3 22(q,4H,CH ₂) and 1.55(t,6H, CH ₃)	8.05-8.0(d,4H,C _{2/8.2*6} .); 7.9(s,1H,C.); 7.55(s,1H,C ₈); 7.4-7.6(m,10H);7.36(d,4H,C _{3.5,3*5} *); 3.15(q,4H,CH ₂) and 1.55(t,6H, CH ₃)
1	I.R.	(cm ⁻¹)		1640,	1560, 1250		1680,	1590, 1250	1630,	1580, 1260		1645,	1585, 1260		1620,	1570, 1250	1636,	1599, 1252	1638.	1589, 1281		1642, 1604, 1280
	U.V	Amax.		260,	322,	358	258,	344	262,	346		265,	346		258,	342	258,	324,	256.	340		260, 342
	BDF			3a (X=H)	C25H18'O4		3b (X = CI)	C26H:6Cl204	3c (X=Ph)	C ₃₈ H ₂₆ O ₄		3d	(X=2-OMe)	C ₂ rH ₂ O ₆	3e(X= NO ₁)	C ₂₄ H ₁₆ N ₂ O ₃	4a(X=H)	C ₁₈ H22O4	4h(X=C)	C28H20C1104		4c (X= Ph) C40H30()4

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78 30; 5.15	65,60,3.91	79.95; 5.80	74.64; 5.40	83,36; 4.25	73.56; 3.40	85.94; 4.51	81.35; 4.42	71.02; 3.30	83.45; 4.75	78,85; 4.52
78.33; 5.19 78 30; 5.15	65.62; 3.93 65.60; 3.91	79,98; 5,82 79,95; 5,80	74.67; 5.43 74.64; 5.40	83,38;4,28 83,36;4,25	73.60; 3.43	85,95; 4.51 85,94; 4.51	81,40,4.46	71.05; 3.31 71.02; 3.30	83.50; 4.79	78.88; 4.53 78.85; 4.52
I	1	1	I	518(5),366(23),338(34), 310(12),158(45),130(35),	105(75),102(12),77(55) —	I	1	I	I	1
8.15-8.1(d,4H,Cr,3,r,3,);8.05(s,1H,C ₁);	7.0-7.20H 2H, C4,5,7,8,4-5,7,8,4-6,9,4,0 (5,0H, OCH)), 3.10(q,4H, CH.) and 1.5(t,6H, CH.) 8.3-8 25(d,4H, C _{2,6,2} ,6,) 8 05(s,1H, C4);	8 0-7.95(d AH_C3',5',3",5");7.75(s,1H,C1); 3.2(q,4H,CH ₂) and 1.5(t,6H, CH ₃) 7.95-7.9(d,4H,C _{7622',6} ');7.9(s,1H,C4);	7.4-7.4(m,5H,C ₃ , y, y, &C ₃);3.2(q,4H,CH ₁); 2.75(s,6H,CH ₃), md 1.5(t,6H, CH ₃) 7.95-8.0(d 4H,C ₇₆₂₂ , y,7.8(s,1H,C ₄); 7.45-7.4624 Group PCD-3.2006.6H OCH 33	3.1(q,4H,CH ₂) and 1.4(t,6H, CH ₃) 8.15-8.10(t,4H,C _{2,6,2,5});8.05(s,1H C ₄); 7.95-7.7(m,7H,C ₃ ,s,g,g,r,r,s,&C ₃);7.6-7.5(m,10H)	8.15-8 10(d,4H,C _{r//276} /);8 09(s,1H,C ₂); 7 6 7 66(m 6U C	8.2-8.15(d,4H,C _{2,6,2} ,6,);8.10(s,1H,C4); 7.95-7.75(m,5H,C _{2,5,3} ;3,5,8.10(s,1H,C4);	7.5-7.35(m,10H) 8 05-8.0(d,4H,C _{1/3,1/3} *);7.98(s,1H,C ₁); 7.7-7 55(m,9H,C _{2/3,7,8,4*5*,77,8*&C₃);7.55-7.45(m,10H); 4 1(s,6H,OCH₂)}	8 25-8 20 (1,4H,C _{2,6/2} 6 ⁻);8.15(s,1H,C ₄);7.95(s,1H,C ₈); 7.7-7.5(d 4H,C ₈ ,*,1.7.65-7.6(m 10H)	8 2-8.15(d,4H,C _{2,6/2} ,6/);8.1(s,1H,C ₁); 7.8-7.65(m,5H,C _{2,5/3} ,**&C.);	7.55-7.4(m,10H);2.65(s 6H,CH ₂) 8 1-8.05(d,4H,C _{2,4/2} ,4r); 8.0(s,1H,C.); 7.7-7.f(m,5H,C _{7,2,3,7} ,&C ₈); 7.47.3(m,10H);4.2(<u>s</u> ,6H,OCH ₃)
1622,	1319, 1274 3 1647, 8	1607, 1250 1633,	1604, 1252 1628, 1506	1309, 1257 1652 1598, 1236	1646, 1507 1735	1647, 1602, 1230	1621, 1554, 1272	1635, 1607, 1260	1640, 1606, 1235	1645, 1601, 1260
-	255,	345 256,	347 250, 340	256, 325,	345 258		255 3-14	252, 341	253, 340	256, 342
4d	$C_{18}H_{30}O_{6}$ $C_{18}H_{30}O_{6}$ $4t_{1}(X = NO_{2})$	C ₁₈ H ₂₀ N ₂ O ₃ 41 (X= Me)	C30F/26O4 4g(X=OMe) CHO.	5a(X=H) 5a(X=H) C36H22O1	5b(X = Cl)	5c(X=Ph) C4 ₁ H ₃₀ O ₄	5d (X=2 OMe) C.H., O	5((X=NO ₂)) C ₁ :H-N ₂ O	5 (X=Me) C38H2604	5g(X=OMe) C _{3t} H ₂₈ C ₆

	Anti-implantation	at 10 mg/kg	Efficacy (%)											Inactive		ł		67		Insictive		I		Ina ctive		Inactive		Inactive		1	
			Bac8	1.96	1.96	0.12	0.28	1.96	3.84	0.70	2 82	3.84	6.35	1.96	3.84	1.96	3.84	5.02	635	1.96	3.84	1.96	2.82	1.96	3.84	3.84	5.02	i	ł	5.02	635
			Bac7	0.31	1.2	AN	0.12	1.96	3.84	1.96	5.02	635	7.85	5.02	6.35	6.35	7.85	1.96	3.84	1.96	5.02	1.25	2.82	5.02	635	1.25	1.96	I	i	2.82	5.02
	vity		Bac6	NA	0.12	NA	0.7	NA	0.7	3.84	5.02	5.02	635	5.02	6.35	635	9.49	2.82	5.02	3.84	7.85	1.96	2.82	5.02	635	1.25	1.96	;	ł	1.96	3.84
	crial acti		Bac5	0.94	1.13	1.13	1.32	5.02	635	635	7.85	5.02	6.35	1.96	6.35	3.84	5.02	2.82	3.84	2.82	5,02	0.78	1.13	1.96	6.35	NA	0.7	1	I	1.96	3.84
	Anti - Bacterial activity		Bac4	1.13	1.53	NA	0.7	0.7	3.84	6.35	7.85	5.02	6.35	1.96	5.02	0.7	1.25	1.25	1.96	1.96	2,82	AN	AN	1.96	5.02	0.7	3.84	I	i	5.02	6.85
			Bac3	1.96	7.85	0.7	1.96	1.96	3.84	0.7	3.84	1.25	5.02	0.7	1.25	1.96	5.02	2.82	5.02	NA	1.25	NA	NA	0.7	1.25	0.7	1.96	I	1	3,84	5.02
			Bac2	3.84	6.35	0.70	3.84	1.96	6.35	NA	0.7	NA	0.7	1.96	5.02	NA	NA	2.82	3.84	AN	0.7	NA	AN	1.96	5.02	1.96	2 82	i	i	5.02	6.35
			Bac1	1.96	3.84	3.84	6,35	1.96	3.84	3.84	7.85	0.31	1.96	1.96	3,84	3.84	7.85	1.96	5.02	5.02	6.35	AN	AN	1.96	3.84	1.96	2 82	ł	1	1.96	2.82
annurans	Conc	lm/gu		600	006	600	906	600	006	600	006	600	006	600	006	600	006	600	006	600	006	600	006	600	906	600	00	600	006	600	006
oyloenzo	Yield	(%)		96.4		93.5		95.0		6.79		89.6		95.8		93.0		94.7		96.5		89.0		90.		93.9		93.3		92.9	
ar Disar	MM	Time	(min)	1.5		2.0		2.0		2.5		3.0		2.5		3.0		3.0		3.0		2.5		2.0		1.5		3.0		3.0	
ity or me	Yied	(%)		92.7		85.0		87.3		90.4		84.0		94.0		92.4		90,8		94,5		87.0		87.5		91.2		91.4		906	
I able 2: biological activity of linear pisaroyidenzoditurans	PIC	Rn time	(Hr3.)	6.0		6,5		6.0		6.0		7.0		60		6,5		7.0		7.5		8.0		6.0		6.0		6.0		6.5	
7: 191010	M.P.	(C)		185-7		126-8		158-9		232-5		136-8		157-8		202-3		211-3		177-9		177-9		156-7		177-9		208-9		184-6	
I aDIC	BDF			3a		3b		36		Эd		3e		4a		4b		4		4d		4 e		4f		4g)	Śа		Sb	

Table 2: Biological activity of linear bisaroylbenzodifurans

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Inactive		Inactive		I		I		Inactive		I		pou					
1.25	1.96	1.96	5.02	1.96	3.84	0.7	1.96	1.96	3,84	5.52	8.28	iation metl	noniae	Aureus			
1.25	2 82	1.25	2 82	3.84	5.02	1.25	3.84	5.02	6.35	5.52	8.28	ve irrad	a Pneun	coccus /	iycin		
NA	0.7	AN	NA	1.96	5.02	1.25	2,82	3,84	6,35	3.0	4.5	dicrowave irra	Clebsiell	staphylo	strepton		
0.7	1.95	AN	٩N	1.96	5.02	1.25	2.82	3.84	635	42	6.3	MW N	Bac3: F	Bac6: S	Std: S		
1.96	2.82	٩N	٩N	0.31	1.25	1.96	5.02	5.02	6.35	٩N	NA	P			•		
3.84	5.02	AN	AN	NA	1.96	1.96	3.84	2,82	3.84	6.77	10.1	is metho		SI			
1.96	2.82	٩N	٩N	NA	0.31	5.02	635	1.96	3.84	4.70	7.05	catalys	is	s Citrice	Albus		
1.96	3.84	5.02	6.35	0.70	2.82	3.84	5.02	0.70	2.82	6.77	10.1	Phase Transfer	Vulgar	ococcu	coccus	vity	
600	906	600	906	600	006	600	906	600	906	600	906	Phase 7	Proteus	Staphyl	Strepto	No acti	
94.1		96.2		88.3		89.0		91.0		I		PTC	Bac2:	Bac5:	Bac8:	AN	
3.0		3.0		3.0		2.0		3.0		I		u					and 5d)
93.7		92.6		87.5		87.9		90.0		ł		nzodifura		rinosa			of 3d, 4d and 5d)
6.0		7.0		7.5		60		60		I		aroyl bei	a Coli	nas Aeru	ubtilis	n Time	
		148-9		6- <i>LL</i>		180.2				1		BDF Linear bis aroyl b	Baci: Escherichia Coli	Bac4: Pseudomonas Ae	Bac7: Bacillus Subtilis	Rn Time: Reaction Time	Ar =naphthalene (in case
5c 210-2		5d 14		5e 7.		5f 18		5g 126-8				JF Li	ci: Es	c4: Ps	c7: Ba	Time:	=naph
5		50		5		51		58		Std		BD	Ba	Ba	Ba	Rn	Ar

(iii) The protons of $C_{3',5',3'',5''}$ were observed as doublet at δ 7.3-8.1. Whereas in 3a, 4a and 5a a multiplet was observed at δ 7.4-7.7 for protons of $C_{3',4',5',3'',4'',5''}$

Experimental

Melting points were determined in open capillaries and were uncorrected. The UV spectra were recorded on Shimadzu UV 160A UV-Vis-NIR spectrophotometer, IR spectra on Shimadzu FTIR model 8010 spectrophotometer and the 1H NMR spectra in CDCl₃ on Varion C17-20-ZM-390-200 MHz NMR spectrophotometer using TMS as an internal standard. Mass spectra were recorded on VG 7070H ion analyzer. The C, H and O analysis of the compounds was done on a Carlo Erba Model EA 1108 CHNS-O elemental analyzer.

General Experimental Procedure for Phase transfer catalysis method:

To a magnetically stirred solution of diacyl /aroyl resorcinols (1 i-iii; 0.01 mole) in 30 ml ethylenedichloride, 30 ml of aq. 20% K_2CO_3 solution and 100mg tetrabutyl ammonium hydrogen sulphate were added. The reaction mixture was heated to 50° and the various *p*-substituted phenacyl bromides (2a-g; 0.02 mole) were added drop wise over a period of 30 minutes at 55-60° and the temperature was maintained for 6-8 hours. The organic layer was separated, washed with 5% NaOH solution and then with water. The resulting organic layer was dried over anhydrous sodium sulphate. The excess solvent was removed under reduced pressure and the crude product was crystallized from methanol.

General experimental procedure for microwave irradiation:

Diacyl/ aroyl resorcinols (1i-iii; 0.01 mole) and various substituted α -bromoketones (2a-g; 0.02 mole) were doped with baked K₂CO₃ and were irradiated in microwave oven for 1-3 minutes. The reaction was monitored by TLC. After irradiation, the crude was poured into water to remove the inorganic matter. The resultant product was crystallized from methanol.

Anti-Bacterial Activity

The bisaroylbenzodifurans (3a-e, 4a-g and 5a-g) were tested on four-gram positive bacteria (*Staphylococcus citricus*, *Staphylococcus aureus*, *Bacillus subtilis*, *Streptococcus albus*) and also on four gram negative bacteria (*Escherichia coli*, *Proteus vulgaris*, *Klebsiella pneumoniae*, *Pseudomonas aeruginosa*) at concentrations 600 and 900 μ g/ ml, by adopting Vincent and Vincent filter paper disc method (16). The results were depicted in Table 2.

2,6-diaroyl/naphthoyl-3,5-dimethyl-benzo[1,2-b;5,4-b']difurans (3a-e) have not shown marked activity against the gram negative bacteria-*Escherichia coli*, *Proteus vulgaris* and *Klebsiella pneumoniae* at both 600 and 900 μ g/ ml. Interestingly streptomycin is not active against *Pseudomonus aeruginosa* bacteria, whereas all the benzodifurans 3a-e, 4a-g and 5a-g have shown activity against *Pseudomonus aeruginosa* bacteria. Among all the bis aroylbenzodifurans (3a-e) with methyl substitution at 3 and 5 positions, 2,6-di(6-methoxy 2-naphthoyl)-3,5-dimethyl-benzo[1,2-b; 5,4-b']difuran (3d) has shown the highest percentage of activity against *Pseudomonus aeruginosa* bacteria

All the benzodifurans 3a-e have been observed to be very active against gram positive - bacteria *Staphylococcus citricus*, *Staphylococcus aureus* and *Bacillus subtilis*. Among these bisaroylbenzodifuans, 3c, 3d and 3e have shown high percentage of activity at both concentration levels against *Staphylococcus citricus* bacteria. The compounds 3d and 3e have shown very good activity at both concentration levels against *Staphylococcus aureus*. Among all the bisaroylbenzodifurans (3a-e) with methyl substitution at 3 and 5 positions, 2,6-di(6-methoxy-2-naphthoyl)-3,5-dimethyl-benzo[1,2-b; 5,4-b']difuran (3d) has shown the highest activity against

Staphylococcus citricus bacteria with reference to all the gram positive bacteria. In the 2,6diaroyl/naphthoyl-3,5-diethyl-benzo[1,2-b;5,4-b']difurans (4a-g), only two compounds 4b and 4d have shown marginal activity against the gram negative bacteria at both concentration levels. The remaining compounds were inactive. The bisaroyl benzodifurans, 4a, 4b, 4d and 4f have shown very good activity against *Staphylococcus aureus* bacteria at both concentration levels. Among these, 4b is found to be the highest active compound in the both concentration levels. The bisaroylbenzodifurans (5a-g) were found inactive against gram negative bacteria. Only one compound, 2,6-di(*p*-methoxy benzoyl)-3,5-diphenyl-benzo[1,2-b;5,4-b']difuran (5g) has shown very good activity against *Bacillus* subtilis at both concentrations and the remaining were found inactive against all the gram positive bacteria.

It is evident from the above observations that, most of the aroyl benzodifurans were active against gram-positive bacteria rather than gram-negative bacteria. Among all the active bisaroylbenzodifurans, 2,6-di(6-methoxy-2-naphthoyl)-3,5-dimethyl-benzo[1,2-b;5,4-b'] difuran (3d) has shown very good activity against all the gram positive bacteria. It is generally observed that the percentage of activity is decreased when ethyl/ phenyl group replaces the methyl group at 3,5-bisaroylbenzodifuran positions.

Anti-implantation activity

Linear bisaroylbenzodifurans (3a-e) were reported for anti-implantation activity. By following the same reported procedure⁹, some of the selected compounds in 4a-g & 5a-g series were screened for anti-implantation activity. Only one compound (4c) has shown 67% activity. Remaining compounds were found inactive at 10 mg/kg/rat/day on albino rats. The results were shown in **Table 2**.

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References

- 1. Zawadowski, Kossa kowski J, Rechowicz P; Chem Abstr, 87, 53008, (1977).
- 2. Maziere M, Buu Hoi NP, Dat Xuong N; Bull. Soc. Chim. Fr., 1000 (1963).
- 3. Leonard R. Worden, Albert W. Burgstahler, Kurt. D. Kaufman, James A. Weis and Thomas K. Schaaf; J. Het. Chem., 6, 191 (1969).
- 4. Hismat OH, El Ebrashn NMA, Snalash MR, Ismail I; Chem Abstr, 91, 193211 (1979).
- 5. Dori G, et al; Eur. J. Med. Chem; 1978, 13, 407 & Chem Abstr, 90, 9799 (1979).
- 6. Bharati Sudha KB, Ph. D thesis, Kakatiya Univ., Warangal, Andhra Pradesh, India, (1990).
- 7. Bisagni E, Buu Hoi NP, Royer R; J. Chem. Soc., 3693 (1955).
- 8. Grover PK, Chawla HPS, Anand N, Kamboj VP and Kar AB; J. Med. Chem., 8, 720 (1965).
- 9. K. S. Krishna Murthy, Y. Ratna Kumari, B. Rajitha and M. Kanakalingeswara Rao, (Mrs). J D. Dhar and B. S. Setty., Ind. J. Chem., **38B**, 938 (1999).
- 10. David A. Gordan, Robert L. Hudson, U. S. Pat. 2,794,052 (1957).
- 11. Rather JR., Reid E. M., J. Am. Chem. Soc., 41, 75 (1919).
- 12. Engler, Zielke, Ber., 22 209 (1889).
- 13. Collet M. A., Bull. Soc. Chim. Fr., 21, 68 (1889).
- 14. Langley W. D., Org. Synth. Coll. Vol. 2, 127 (1947).
- 15. Vogel's textbook of practival organic chemistry, fourth edn., 815 (1978).
- 16. Vincent J. C., Vincent H. W., Proc. Soc. Exp. Biol. Medica., 55, 102 (1944).

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