PREPARATION OF 3-SUBSTITUTED 4,6-DINITROANTHRANILS BY THE OXIDATION OF ANIONIC 1,3,5-TRINITROBENZENE σ -COMPLEXES

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We have shown that the oxidation of the anionic σ -complexes of 1,3-dinitrobenzene by a system containing a catalytic amount of CuBr in CCl₄ leads to the formation of substitution products (Ia)-(Ic).

In contrast, anionic σ -complexes of 1,3,5-trinitrobenzene are oxidized by the CuBr/CCl₄ system to give 3-acyl-, 3-carboethoxy-, or 3-phenyl-4,6-dinitroanthranils (IIIa)-(IIIf) in good yields.

$$\begin{array}{c|c} H & CH_2R & NO_2 \\ \hline NO_2N & NO_2 \\ \hline NO_2 & CH_5CN, 4-6 \text{ days} \\ \hline NO_2 & (III) & O_2N \\ \end{array}$$

R = COMe (a), COEt (b), COPh (c), CO- (d), COOEt (e), Ph (f).

The structures of (IIIa)-(IIIf) were confirmed by elemental analysis, PMR and IR spectroscopy, and mass spectrometry.

A mixture of 0.002 mole complex (II) and 0.03 g (0.0002 mole) CuBr in 10 ml 1:1 CH₃CN-CCl₄ was stirred at 20°C for 96-144 h. The products were subjected to chromatography on silica gel and recrystallized from ethanol. Product (IIIa) was obtained in 62% yield, mp 143-144°C. Product (IIIb) was obtained in 70% yield, mp 120-121°C. Product (IIIc) was obtained in 65% yield, mp 163-164°C. Product (IIId) was obtained in 54% yield, mp 110-112°C. Product (IIIe) was obtained in 63% yield, mp 72-73°C. Product (IIIf) was obtained in 30% yield, mp 158-159°C.

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