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THE C-ALKYLATION OF SOME HETEROAROMATIC COMPOUNDS BY THE TRIFLUOROACETYLMINE OF METHYL TRIFLUOROPYRUVATE

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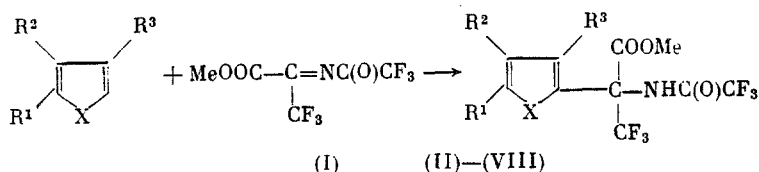
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The trifluoroacetylimine of methyl trifluoropyruvate regioselectively C-alkylates furan, N-methylpyrrole, and thiophenes at the free C² position, indoles at the C³ position, and 1-phenyl-3-methyl-5-pyrazolone at the C⁴ position. The alkylation products were converted to trifluoromethylated amino acids and their ester and amine derivatives.

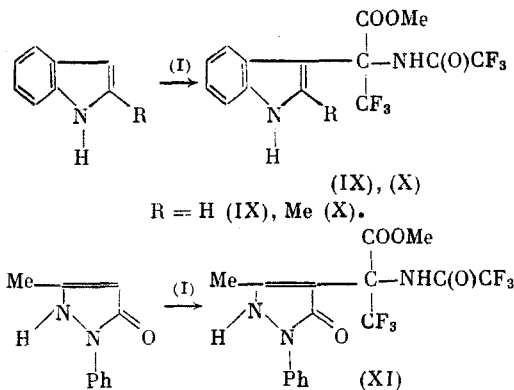
The C-alkylation of indoles and thiophene by the benzenesulfonylimine of methyl trifluoropyruvate was described in our previous work [1]. Details are given below for the reactions of the trifluoroacetylimine of methyl trifluoropyruvate (I) with some five-membered heterocyclic compounds.

Thiophene derivatives are selectively alkylated by imine (I) in CCl₄ at reflux at the free C² position to give thiophenes (II)-(VI) in preparative yields. Furan and N-methylpyrrole undergo this reaction in chloroform from -50 to +20°C to give the C²-alkylation products (VII) and (VIII).



X = S, R¹ = R³ = H, R² = *p*-Tol(II); X = S, R¹ = Ph, R² = H, R³ = Me(III); X = S, R¹ = R² = Ph, R³ = Me(IV); X = S, R¹ = Me, R² = H, R³ = Ph(V); X = S, R¹ = H, R² = Ph, R³ = Me(VI); X = O, R¹ = R² = R³ = H(VII); X = N - Me, R¹ = R² = R³ = H(VIII).

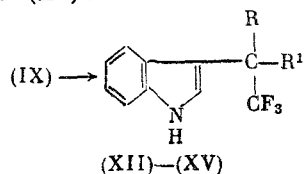
Indole, 2-methylindole, and 1-phenyl-3-methyl-5-pyrazolone react with (I) in chloroform at 20°C to give C³-alkylated indoles (IX) and (X) and C⁴-alkylated pyrazolone (XI).



The structure of indole (IX) was confirmed by the following transformations. The alkaline hydrolysis of (IX) gives acid (XII), which decarboxylates upon sublimation in vacuum to give

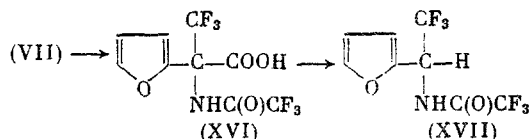
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(XIII). Treatment of substituted indole (IX) with HCl in methanol gave α -amino acid ester (XIV). Alkaline hydrolysis of (XIV) with subsequent neutralization is accompanied by spontaneous decarboxylation to give amine (XV).



$R = \text{COOH}, R^1 = \text{NHC(O)CF}_3$ (XII); $R = \text{H}, R^1 = \text{NHC(O)CF}_3$ (XIII);
 $R = \text{COOMe}, R^1 = \text{NH}_2$ (XIV); $R = \text{H}, R^1 = \text{NH}_2$ (XV).

Furan (VII) undergoes analogous transformations upon alkaline hydrolysis to give (XVI) and (XVII).



The structures of these products were supported by elemental analysis and ^1H and ^{19}F NMR spectroscopy (Tables 1 and 2).

EXPERIMENTAL

The ^1H and ^{19}F NMR spectra were taken on a Bruker WP-200SY spectrometer at 200 and 188 MHz, respectively. The chemical shifts (δ , ppm) were determined relative to HMDS as the internal standard and $\text{CF}_3\text{CO}_2\text{H}$ as the external standard. The R_f values were obtained on Kavalier Silufol plates manufactured in Czechoslovakia.

2-(α -Carbomethoxy- α -trifluoroacetamidotrifluoroethyl)-3-p-tolylthiophene (II). A mixture of 2.0 g imine (I) and 1.3 g 3-p-tolylthiophene in 10 ml CCl_4 was heated at reflux for 4 h and cooled. A yield of 2.04 g (II) was obtained. PMR spectrum in acetone- d_6 : 7.38 s (1H, NH), 7.27 d and 7.02 d (4H, p-MeC₆H₄, $J_{\text{H-H}} = 8.0$ Hz), 7.29 d and 6.98 d (2H, H^{2,4}, $J_{\text{H-H}} = 1.5$ Hz), 3.86 s (3H, OMe), 2.33 s (3H, Me).

Analogous procedures gave 2-(α -carbomethoxy- α -trifluoroacetamidotrifluoroethyl)-3-methyl-5-phenylthiophene (III), 3-methyl-4,5-diphenylthiophene (IV), 3-phenyl-5-methylthiophene (V), and 3-methyl-4-phenylthiophene (VI). PMR spectra in CDCl_3 (δ , ppm): III) 7.51 m and 7.25 m (6H, H⁴, Ph), 3.87 s (3H, OMe), 2.12 s (3H, Me); IV) 7.28 m and 7.15 m (10H, 2Ph), 3.92 s (3H, OMe), 1.80 s (3H, Me); V) 7.72 m and 7.15 m (6H, H⁴, Ph), 3.67 s (3H, OMe), 2.50 s (3H, Me); VI) 7.62 m and 7.33 m (6H, H⁵, Ph), 3.82 s (3H, OMe), 2.10 s (3H, Me).

2-(α -Carbomethoxy- α -trifluoroacetamidotrifluoroethyl)furan (VII). A sample of 1.4 g furan in 5 ml chloroform was added to 5.0 g imine (I) in 5 ml chloroform at -50°C . The mixture was warmed to 20°C . The yield of product (VII) was 4.5 g. PMR spectrum in acetone- d_6 (δ , ppm): 7.65 d.d. (1H, H⁵, $J_{\text{H}^5-\text{H}^4} = 1.9$, $J_{\text{H}^5-\text{H}^3} = 0.8$ Hz), 6.77 d.d. (1H, H³, $J_{\text{H}^3-\text{H}^4} = 3.3$, $J_{\text{H}^3-\text{H}^5} = 0.8$ Hz), 6.5 d.d. (1H, H⁴, $J_{\text{H}^4-\text{H}^3} = 3.3$, $J_{\text{H}^4-\text{H}^5} = 1.9$ Hz), 3.79 s (3H, OMe).

2-(α -Carbomethoxy- α -trifluoroacetamidotrifluoroethyl)-1-methylpyrrole (VIII) was obtained by analogy to (VII). PMR spectrum in acetone- d_6 (δ , ppm): 9.12 br.s. (1H, NH), 6.75 d.d. (1H, H⁵, $J_{\text{H}^5-\text{H}^4} = 2.7$, $J_{\text{H}^5-\text{H}^3} = 1.7$ Hz), 6.37 m (1H, H³), 6.00 d.d. (1H, H⁴, $J_{\text{H}^4-\text{H}^3} = 3.9$, $J_{\text{H}^4-\text{H}^5} = 2.7$ Hz), 3.81 s (3H, OMe), 3.71 s (3H, NMe).

3-(α -Carbomethoxy- α -trifluoroacetamidotrifluoroethyl)indole (IX). A mixture of 1.4 g imine (I) and 0.65 g indole in 10 ml chloroform was maintained for 1 h at 20°C . The yield of (IX) was 1.3 g. An analogous procedure gave 3-(α -carbomethoxy- α -trifluoroacetamidotrifluoroethyl)-2-methyl-indole (X).

4-(α -Carbomethoxy- α -trifluoroacetamidotrifluoroethyl)-3-methyl-1-phenyl-5-pyrazolone (XI). A mixture of 2.0 g imine (I) and 1.3 g phenylpyrazolone in 10 ml chloroform was maintained for 4 h at 25°C to give 3.2 g (XI). PMR spectrum in acetone- d_6 (δ , ppm): 7.50 m (5H, Ph), 3.82 s (3H, OMe), 2.20 s (3H, Me).

3-(α -Carbomethoxy- α -trifluoroacetamidotrifluoroethyl)indole (XII). A mixture of 2.0 g (IX) and 10 ml 5% aq. NaOH was heated for 8 h at 70°C and neutralized with hydrochloric acid. Extraction with ether gave 1.2 g (XII).

3-(α -Trifluoroacetamidotrifluoroethyl)indole (XIII) was obtained upon the sublimation of 1.0 g (XII) at 100°C (1 mm). The product yield was 0.66 g.

3-(α -Carbomethoxy- α -aminotrifluoroethyl)indole (XIV). A sample of 3.2 g (IX) was dissolved in anhydrous methanol and dry HCl was bubbled through for 8 h at 40°C . The solu-

TABLE 1. The Properties of (II)-(XVII)

Compound	Mp, °C	Yield, %	R_f a : b **	Found/Calculated, %			Chemical formula	^{19}F NMR spectrum (δ , ppm)
				C	H	N		
(II)	114	62	0.57 1 : 6	47.94 48.00	2.99 3.05	3.44 3.29	$\text{C}_{17}\text{H}_{13}\text{NO}_3\text{SF}_6$	-6.2 s, -2.3 s
(III)	93-94	87	0.49 1 : 6	47.83 48.00	2.84 3.05	3.36 3.29	$\text{C}_{17}\text{H}_{13}\text{NO}_3\text{SF}_6$	-7.1 s, -2.4 s
(IV)	175-176	91	0.51 1 : 12	54.86 55.08	3.21 3.39	2.58 2.79	$\text{C}_{23}\text{H}_{17}\text{NO}_3\text{SF}_6$	-6.5 s, -1.9 s
(V)	95-97	83	0.49 1 : 12	48.08 48.00	3.25 3.05	3.58 3.29	$\text{C}_{17}\text{H}_{13}\text{NO}_3\text{SF}_6$	-6.9 s, -2.0 s
(VI)	68-69	84	0.48 1 : 12	48.26 48.00	3.21 3.05	3.15 3.29	$\text{C}_{17}\text{H}_{13}\text{NO}_3\text{SF}_6$	-5.9 s, -3.4 s
(VII)	74-76	71	0.38 1 : 12	38.01 37.61	2.51 2.19	4.59 4.38	$\text{C}_{10}\text{H}_7\text{NO}_4\text{F}_6$	-1.3 s, 0.2 s
(VIII)	110-112	68	0.50 1 : 12	40.09 39.75	2.82 3.01	8.93 8.43	$\text{C}_{11}\text{H}_{10}\text{N}_2\text{O}_3\text{F}_6$	-3.2 s, -3.0 s
(IX)	175-176	65	0.32 1 : 3	44.98 45.65	2.30 2.71	7.76 7.60	$\text{C}_{14}\text{H}_{10}\text{N}_2\text{O}_3\text{F}_6$	-6.8 s, -3.0 s
(X)	136-137	81	0.57 1 : 3	47.05 47.12	3.24 3.14	7.51 7.32	$\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}_3\text{F}_6$	-7.9 s, -2.6 s
(XI)	144-146	90	0.68 1 : 1	45.33 45.17	3.11 3.05	9.85 9.98	$\text{C}_{16}\text{H}_{13}\text{N}_3\text{O}_4\text{F}_6$	-6.0 s, -1.5 s
(XII)	145-148 (dec.)	63	0.41 1 : 1	44.38 44.06	2.12 2.25	7.80 7.90	$\text{C}_{13}\text{H}_8\text{N}_2\text{O}_3\text{F}_6$	-6.1 s, -3.3 s
(XIII)	150-153	72	0.55 1 : 3	44.32 44.17	2.80 2.45	8.87 8.58	$\text{C}_{12}\text{H}_8\text{N}_2\text{OF}_6$	-4.6 d, -2.7 s ($J_{\text{H-F}}=8.0$ Hz)
(XIV)	98-100	70	0.42 1 : 3	53.15 52.94	4.17 4.04	10.35 10.29	$\text{C}_{12}\text{H}_{11}\text{N}_2\text{O}_2\text{F}_3$	-3.2 s
(XV)	145-147	61	0.52 1 : 2	56.37 56.07	3.95 4.20	12.83 13.08	$\text{C}_{10}\text{H}_8\text{N}_2\text{F}_3$	-1.5 d ($J_{\text{H-F}}=8.0$ Hz)
(XVI)	98 (dec.)	65	0.42 1 : 6	35.28 35.40	2.01 1.63	4.14 4.59	$\text{C}_8\text{H}_5\text{NO}_4\text{F}_6$	-2.2 s, 0.5 s
(XVII)	48-49	61	0.41 1 : 12	37.15 36.78	2.22 1.91	5.12 5.36	$\text{C}_8\text{H}_5\text{NO}_2\text{F}_6$	-1.3 s, 0.2 d ($J_{\text{H-F}}=7.6$ Hz)

*Hexane served as the crystallization solvent.

**The acetone: CCl_4 ratio in the system for determination of the R_f value.

TABLE 2. PMR Spectra of Synthesized Compounds

Compound	^1H NMR (δ , ppm)						$J_{\text{H-F}}$, Hz
	2	4	5	6	7	$\text{CH}-\text{CF}_3$	
(IX)	7.52	7.65	7.20	7.11	7.47	—	—
(X)	—	7.77	7.16	7.02	7.58	—	—
(XII)	7.41	7.55	7.10	6.95	7.38	—	—
(XIII)	7.53	7.55	7.15	7.13	7.43	6.13	8.0
(XIV)	7.55	7.71	7.15	7.15	7.45	—	—
(XV)	7.45	7.81	7.12	7.17	7.47	5.41	8.0

The PMR spectra of (IX), (X), and (XIV) have signals corresponding to the signals of the OMe-group.

tion was evaporated. The residue was neutralized with 5% aq. NaOH and extracted with ether to give 1.6 g (XIV).

3-(α -Aminotrifluoroethyl)indole (XV) was obtained by analogy to (XII).

2-(α -Carboxy)- (XVI) and (α -hydro)- α -trifluoroacetamidotrifluoroethyl)furans were obtained by analogy to (XII) and (XIII). PMR spectra in acetone- d_6 (δ , ppm): XVI) 7.72 m and 6.63 m (3H, $\text{H}^{3,4,5}$); XVII) 9.91 d (1H, NH, $J_{\text{H-H}} = 7.6$ Hz), 7.78 d.d. (1H, H^5 , $J_{\text{H}^5-\text{H}^4} = 1.9$, $J_{\text{H}^5-\text{H}^3} = 0.8$ Hz), 6.81 d.d. (1H, H^3 , $J_{\text{H}^3-\text{H}^4} = 3.3$, $J_{\text{H}^3-\text{H}^5} = 0.8$ Hz), 6.69 d.d. (1H, H^4 , $J_{\text{H}^4-\text{H}^3} = 3.3$, $J_{\text{H}^4-\text{H}^5} = 1.9$ Hz), 6.15 pent (1H, CH- CF_3).

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