Efficient Synthesis of Clitocine via 1,3-N (endo) to N (exo) Migration: A Revision to Kini's Work

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This paper is dedicated to Prof. James D. White.

Abstract: Efficient synthesis of clitocine has been accomplished via unprecedented 1,3-N (*endo*) to N (*exo*) migration as a key transformation. Incorporation of *p*-chlorobenzoyl (PCB) group as a protecting group led to the easy solidification of the intermediate of **7**, thus making the isolation very facile and to the minimization of the epimerization of the anomeric center at the final deprotection stage.

Key words: 1,3-N (*endo*) to N (*exo*) migration, isomerizations, clitocine

Clitocine [6-amino-5-nitro-4-(β-D-ribofuranosylamino)pyrimidine], isolated from the mushroom Clitocybe inversa by Kubo et al.,¹ has potent insecticidal activity against Pectinophora gossypiella. Later it was found that clitocine has also strong cytostatic effect towards several leukemia cell lines and is an inhibitor of adenosine kinase.² The interesting biological activity together with biogenetically close relationship with adenosine triggered intensive synthetic studies to lead to the completion of two independent total syntheses of **1**. Kamikawa's group³ used the condensation reaction of ribofuranosylamine derivatives with 4,6-dichloro-5-nitropyrimidine as a key step (Scheme 1, route A), while Kini's group² employed a direct coupling of silvlated 4, 6-diamino-5-nitropyrimidine (3) with 1-O-acetyl-2,3,5-tri-O-benzoyl-D-ribofuranose (4) as a critical step (Scheme 1, route B). Compared to the former synthesis, the latter approach is more efficient in terms of chemical yield and stereocontrol, particularly at the anomeric center.

As a part of our interest towards further biological evaluation of **1**, we repeated Kini's synthesis. In the repetition of the coupling reaction of **3**⁴ and **4**, we were surprised by the observation that no desired product was formed on the basis of the ¹H NMR spectrum of the crude mixture: neither the doublet NH proton at $\delta = 9.63$ ppm nor the broad NH₂ protons at $\delta = 8.62$ ppm of the expected product **8** were observed.⁵ However, column chromatography of the crude product provided pure **8** in 28% yield and a 3:1 mixture of the desired **8** and the side product in 23% yield that was later identified as the *endo*-isomer **6**. Hinted by the

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isomerization of the *endo*-product **6** to the *exo*-product **8** during silica gel chromatography, we further treated the 3:1 mixture with silica gel (100% wt/wt to the mixture) for 24 hours in dichloromethane to reach the equilibrated ratio of 93:7. Thus, contrary to the reported result, the *endo*-product **6** was formed exclusively as a kinetic product, which was isomerized to the thermodynamically controlled *exo*-product **8** through 1,3-*N* (*endo*) to *N* (*exo*) migration under acid catalysis. Another drawback of Kini's synthesis is the partial epimerization of the anomeric center at the final deprotection stage: 98:2 of β : α anomer ratio of **8** was deteriorated to 92:8 of β : α anomer ratio of **1** under the deprotection conditions (Scheme 2).

To minimize the partial epimerization of the anomeric center, we changed the benzoyl protecting group to *p*-chlorobenzoyl (PCB) group on the assumption that more electron deficient aromatic ring would facilitate the meth-anolysis reaction. In the same way as with the benzoate derivatives, 1-acetoxy-PCB-protected ribofuranose **5** was prepared without any incidence. Coupling of **3** and **5** under the conditions used by Kini et al., provided exclusively the *endo*-product **7**. Serendipitously, the kinetic *endo*-product **7** was easily solidified in high purity simply by triturating the crude reaction mixture with ethyl acetate–diethyl ether co-solvent to make the isolation of the prod-



Scheme 2 Reagents and conditions: i) HMDS, pyridine, H_2SO_4 , $(NH_4)_2SO_4$; ii) TMSOTf; iii) silica gel or HOAc; iv) NaOMe, MeOH-dioxane.

uct very facile (for the ¹H NMR spectra of **7** and **9**, see Figure 1). Furthermore, X-ray analysis of a single crystal of **7** unambiguously determined the structure of the *endo*-product (Figure 2).⁶ Subsequent isomerization of the *endo*-product **7** by silica gel for 24 hours gave exclusively the *exo*-product **9** in high yield and purity.⁷ Besides silica gel, Brønsted acid such as acetic acid also caused isomerization to reach the *exo:endo* ratio of 96:4 after 24 hours but Lewis acids like lithium chloride or magnesium chloride did not trigger isomerization.

As expected, the deprotection of the PCB group proceeded much faster (in 2 h at 0 °C) than the benzoyl group (in 17 h at 0–8 °C) to result in no epimerization of the



Figure 2 ORTEP plot of 7.

anomeric center. Simple recrystallization of the crude reaction mixture in methanol provided **1** having 98:2 ratio of β/α anomer in 51% yield. One more recrystallization cycle of the filtrate provided additional 32% of **1** having the same purity of the first crop.⁸

In conclusion, we have discovered that the *endo*-product **6** or **7** was kinetically formed in the glycosylation stage, which was isomerized to the thermodynamic *exo*-product **8** or **9**, respectively, under acid catalysis. Fortunately, the incorporation of the PCB group facilitated the solidification of the *endo*-product **7**, thus making the isolation and purification very facile. Isomerization and subsequent deprotection underwent without any sign of the deterioration of the anomeric purity. The established synthesis provides highly pure clitocine (1) without any column separation and delineates the new reaction pathway, 1,3-N (*endo*) to N (*exo*) migration.





Figure 1 ¹H NMR spectra of 7 and 9.



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- (5) We observed peaks at $\delta = 9.61$ (s, 1 H), 9.43 (s, 1 H), and 9.35 (s, 1 H) in DMSO- d_6 , which are typical peaks of the *endo*-product **6**. Since **6** was not solidified under various conditions, it was difficult to purify the product thoroughly.
- (6) X-ray data for the compound **7**: $C_{30}H_{22}Cl_3N_5O_9$, monoclinic, space group P2₁ with cell parameters: a = 14.6041 (11) Å, b = 6.8064 (5) Å, c = 16.1106 (12) Å; $a = 90^\circ$, $\beta = 108.0770$ (10)°, $\gamma = 90^\circ$; V = 1522.4 (2) Å³; $D_c = 1.533$ mg/m³; Z = 2.
- (7) 4-Amino-6-imino-5-nitro-1-N-(2,3,5-tri-O-parachlorobenzoyl-β-D-ribofuranosyl)pyrimidine (7). A suspension of 4, 6-diamino-5-nitropyrimidine (1.0 g, 6.45 mmol) in HMDS (30 mL) was treated with pyridine (4.8 mL), H₂SO₄ (0.2 mL) and ammonium sulfate (60 mg). The mixture was refluxed at 130 °C for 18 h, and HMDS was evaporated in vacuo. The residual solid was dissolved in MeCN (50 mL), cooled to 0 °C, and treated with 1-O-acetyl-2,3,5-tri-O-para-chlorobenzoyl-D-ribofuranose (5, 4.5 g, 7.42 mmol) and TMSOTf (2.0 mL, 11.0 mmol). The mixture was stirred at 0 °C for 23 h, and it turned into dark clear solution. Then, MeCN was evaporated in vacuo, and the residue was treated with sat. NaHCO₃ (100 mL). The aqueous layer was extracted with CH_2Cl_2 (100 mL \times 2), and the combined organic phase was dried over anhyd MgSO₄. After concentration, the residue was triturated with EtOAc (4 mL) and enough amount of Et₂O. The resulting solid was filtered and washed with Et₂O to give the endo-product 7 (2.44 g, 55%) as a yellow powder. ¹H NMR (400 MHz, DMSO- d_6): $\delta = 9.61 (1 \text{ H}, \text{ s}), 9.44 (1 \text{ H}, \text{ s}), 9.36 (1 \text{ H}, \text{ s}), 8.33$ (1 H, s), 7.98 (2 H, dd, J = 6.8, 2.0 Hz), 7.94 (2 H, dd, *J* = 6.8, 2.0 Hz), 7.81 (2 H, dd, *J* = 6.8, 2.0 Hz), 7.57 (4 H, 2 dd, J = 6.8, 2.0 Hz), 7.50 (2 H, dd, J = 6.8, 2.0 Hz), 6.33 (1

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H, d, J = 2.4 Hz), 6.12 (1 H, dd, J = 6.4, 2.4 Hz), 6.06 (1 H, dd, J = 7.2, 6.4 Hz), 4.60–4.80 (3 H, m). ¹³C NMR (100 MHz, DMSO- d_6): $\delta = 164.8$, 164.0, 163.9, 158.5, 152.3, 148.5, 139.0, 138.9, 138.7, 131.4, 131.3, 131.2, 129.2, 129.1, 128.2, 127.7, 127.5, 111.3, 91.2, 78.7, 74.1, 70.4, 63.7. IR: 3424, 3300, 1713, 1632, 1589, 1478, 1396, 1252, 1087, 1009, 752 cm⁻¹. UV/Vis (CH₂Cl₂): $\lambda_{max} = 243$, 374 nm.

6-Amino-5-nitro-4-[(2,3,5-tri-*O-para*-chlorobenzoyl-β-D-ribofuranosyl)amino]pyrimidine (9).

A solution of **7** (2.166 g, 3.08 mmol) in methylene chloride (60 mL) was stirred with silica gel (2.2 g) for 20 h. Silica gel was filtered off and washed with CH₂Cl₂. The filtrate was concentrated to give **9** (2.06 g, 95%) as an yellow foam. ¹H NMR (400 MHz, DMSO-*d*₆): δ = 9.60 (1 H, d, *J* = 8.0 Hz), 8.58 (2 H, br s), 8.05 (1 H, s), 8.00 (2 H, dd, *J* = 6.8, 2.0 Hz), 7.90 (2 H, dd, *J* = 6.8, 2.0 Hz), 7.84 (2 H, dd, *J* = 6.8, 2.0 Hz), 7.55 (6 H, m), 6.35 (1 H, dd, *J* = 8.4, 4.8 Hz), 5.97 (1 H, dd, *J* = 5.6, 4.8 Hz), 5.89 (1 H, dd, *J* = 6.0, 5.2 Hz), 4.65 (2 H, m), 4.55 (1 H, m). ¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.5, 164.8, 164.7, 160.0, 159.4, 157.4, 139.8, 139.7, 139.4, 132.0, 131.9, 129.9, 129.8, 129.7, 128.9, 128.3, 128.2, 113.4, 84.8, 78.7, 75.1, 72.1, 64.8. UV/Vis (CH₂Cl₂): λ_{max} = 240, 331 nm.

(8) Clitocine (1).

6-Amino-5-nitro-4-[(2,3,5-tri-O-para-chlorobenzoyl-β-Dribofuranosyl)amino]pyrimidine (9, 1.93 g, 2.75 mmol) was dissolved in 1,4-dioxane (26 mL) and diluted with MeOH (65 mL). After cooling to 0 °C, the solution was treated with MeONa (0.55 mmol, 0.1 mL of 28% MeOH soln was diluted with 0.4 mL of MeOH) and stirred at 0 $^\circ\mathrm{C}$ for 2 h. The resulting solution was stirred with Amberlite IR-120 (900 mg, 3.96 mol equiv, acidic resin was washed with distilled H₂O and MeOH before use) for 30 min. The resin was filtered off and the filtrate was concentrated in vacuo. The residue was triturated with Et₂O and washed with Et₂O (10 mL \times 3). The solid was recrystallized from hot MeOH (135 mL) to give $\mathbf{1} (\geq 98\% \text{ of } \beta\text{-anomer}, 400 \text{ mg}, 51\%)$ as a slightly yellow solid. More clitocine ($1, \ge 98\%$ of β -anomer, 255 mg, 32%) was obtained from the mother liquor. ¹H NMR (400 MHz, DMSO- d_6): $\delta = 9.28$ (1 H, d, J = 7.6 Hz), 8.56 (2 H, s), 8.00 (1 H, s), 5.79 (1 H, dd, *J* = 7.6, 3.6 Hz), 5.20 (1 H, d, J = 5.2 Hz), 5.06 (1 H, t, J = 4.8 Hz), 4.92 (1 H, d, J = 6.0 Hz), 4.05 (1 H, m), 3.93 (1 H, m), 3.79 (1 H, m), 3.53 (1 H, m), 3.47 (1 H, m). ¹³C NMR (100 MHz, DMSO d_6): $\delta = 160.2, 159.5, 156.8, 112.8, 87.0, 84.8, 75.9, 70.8,$ 61.3.