W. A. M. Duncan, G. J. Durant, J. C. Emmett, C. R. Ganellin, and M. E. Parsons, J. Int. Med. Res., 3, 86 (1975).

- (6) For an earlier study on histamine analogs as potential gastric acid secretion inhibitors, see M. I. Grossman, C. Robertson, and C. E. Rosiere, J. Pharmacol. Exp. Ther., 104, 277 (1952).
- (7) For a general structure-activity review of histamine analogs, see R. G. Jones, Handb. Exp. Pharmacol., 18 (1), 1 (1966).
- (8) F. N. Fastier, Pharmacol. Rev., 14, 37 (1972).
- (9) H. King, E. M. Lourie, and W. York, Ann. Trop. Med. Parasitol., 32, 177 (1938).
- (10) A. Albert, Aust. J. Sci., 6, 137 (1944).
- (11) H. Blashko and R. Duthie, Biochem. J., 39, 347 (1945).
- (12) G. J. Durant, A. M. Roe, and A. L. Green, *Prog. Med. Chem.*, 7, 124 (1970).
- (13) H. Blashko, F. N. Fastier, and J. Wajda, Biochem. J., 49, 250 (1951).
- (14) R. A. Maxwell, R. P. Mull, and A. J. Plummer, *Experientia*, 15, 267 (1959).
- (15) G. J. Durant, A. M. Roe, and A. L. Green, Prog. Med. Chem., 7, 177 (1970).
- (16) C. C. Chang, E. Costa, and B. B. Brodie, J. Pharmacol. Exp. Ther., 147, 303 (1965).
- (17) F. N. Fastier, Pharmacol. Rev., 14, 63 (1972).
- (18) G. J. Durant, A. M. Roe, and A. L. Green, Prog. Med. Chem., 7, 174 (1970).
- (19) M. N. Ghosh and H. O. Schild, Br. J. Pharmacol. Chemother., 13, 54 (1958).
- (20) For comparison purposes it may be noted that by this method burimamide was shown subsequently³ to have an ID₅₀ of 6.1 μmol/kg.
- (21) E. J. Ariens, A. M. Simonis, and J. M. van Rossum in "Molecular Pharmacology", Vol. 1, E. J. Ariens, Ed., Academic Press, London and New York, 1964, p 169.

- (22). For comparison purposes, it may be noted that by this method, burimamide was shown subsequently³ to have a pA_2 of 5.1.
- (23) S. J. Angyal and W. K. Warburton, J. Chem. Soc., 2492 (1951).
- (24) The pK_a was determined potentiometrically at 25° on 0.005 M solutions of N^{α} -guanylhistamine dihydrochloride in 0.1 M KCl by titration against HCl. The measured pK_a (6.78 \pm 0.03) was corrected to 37° by subtracting 0.02 units per degree rise in temperature.²⁵
- (25) T. B. Paiva, M. Tominaga, and A. C. M. Paiva, J. Med. Chem., 13, 689 (1970).
- (26) C. R. Ganellin, J. Pharm. Pharmacol., 25, 787 (1973).
- (27) B. Belleau in "Adrenergic Mechanisms", Ciba Foundation Symposium, Wolstenholme et al., Ed., Churchill, London, 1960, p 223.
- (28) B. M. Bloom and J. M. Goldman, Adv. Drug Res., 3, 121 (1966).
- (29) J. Walker, J. Chem. Soc., 1996 (1949).
- (30) O. Kennard and J. Walker, J. Chem. Soc., 5513 (1963).
- (31) (a) F. A. Cotton, V. W. Day, E. E. Hazen, and S. Larsen, J. Am. Chem. Soc., 95, 4834 (1973); (b) F. A. Cotton, V. W. Day, E. E. Hazen, S. Larsen, and S. T. K. Wong, *ibid.*, 96, 4471 (1974).
- (32) S. Furberg and J. Solbakk, Acta Chem. Scand., 26, 3699 (1972).
- (33) C. Tanford, J. Am. Chem. Soc., 76, 945 (1954).
- (34) R. C. Neuman and V. Jonas, J. Phys. Chem., 75, 3550 (1971).
- (35) D. Ackermann, Naturwissenschaften, 27, 515 (1939).
- (36) P. van der Merwe, Hoppe-Seyler's Z. Physiol. Chem., 177, 301 (1928).
- (37) J. H. Short, U. Biermacher, D. A. Dunnigan, and T. D. Leth, J. Med. Chem., 6, 275 (1963).

Hypolipidemic Imidazoles

Keith H. Baggaley,* Monique Heald, Richard M. Hindley, Brian Morgan, John L. Tee, and Joseph Green

Beecham Pharmaceuticals, Research Division, Nutritional Research Centre, Walton Oaks, Tadworth, Surrey, KT20 7NT, United Kingdom. Received January 14, 1975

A series of analogs of N-benzylimidazole was prepared and tested for hypolipidemic activity. Both plasma cholesterol and triglyceride-lowering activity were found in several members of the series. The most active compounds were N-3-methoxy-, N-4-methoxy-, and N-4-methylbenzylimidazole. Structure-activity relationships are discussed.

In the course of a search for novel hypolipidemic agents, it was discovered that a number of simple imidazole derivatives, namely N-alkylimidazoles of type I, inhibited cholesterol biosynthesis at the 2,3-oxidosqualene cyclase step both in vitro and in vivo and also showed hypocholesterolemic activity in experimental animals.¹

$$N \longrightarrow N \longrightarrow (CH_2)_n CH_2$$

I, $n = 9-13$

Extension of this work revealed that certain other substituted imidazoles also showed both hypocholesterolemic and hypotriglyceridemic activity and we here report the results of these studies.

Chemistry. Compounds 1-27 and 29-33 (see Tables I and II and the Experimental Section) were prepared by the alkylation of imidazole or a substituted imidazole with the corresponding halide, usually the bromide (see the Experimental Section). The amino derivative 28 was obtained by NaBH₄-10% Pd/C reduction of the nitro analog 10.² The known compounds 34-36 were synthesized by literature methods (see Table III).

Structure-Activity Relationships. It was observed that many benzylimidazoles (Table I) possessed hypolipi-

demic properties, the hypocholesterolemic effect being more general than the hypotriglyceridemic property. The derivatives bearing a free 4-hydroxy group 24, esterified hydroxy group 23, or 4-carbethoxy group 22 were inactive. However, related to the hypocholesterolemic activity, an associated increase in liver lipids and liver cholesterol was seen.

Various modifications to the substitution pattern in the benzene ring (Table I) and to the basic arylimidazolylmethane structure (Tables II and III) were investigated in an attempt to obtain a cholesterol-lowering effect without production of fatty liver. The performance of two known hypolipidemic agents and N-dodecylimidazole in our test system is shown in Table IV for comparison.

Substitution in the imidazole ring resulted in loss of activity (29 and 30) as did substitution on the methylene bridge (33). Removal of the methylene bridge as in compounds 34 and 35 preserved the hypocholesterolemic activity but caused an increase in liver lipid levels. Extension of the bridging link to 2 and 3 carbon atoms gave the active compounds 37 and 39. The compounds which exhibited hypocholesterolemic properties yet gave the smallest rise in liver lipids, the nitro derivatives 10, 12, and 37 and the cinnamyl derivative 39, were further evaluated in rats. How-

								ipids a ropria		
							Pla	sma	Liv	ver ^a
				77 2 - 1				Tri-		Cho-
	Ar	Formula ^b			t, Mp or bp (mm), °C [lit.ref]	Crystn solvent		glyc- eride		
1°	Phenyl	C ₁₀ H ₁₀ N ₂	A ₁	46	72 [70-71]		94	59		
2	4-Chlorophenyl	$C_{10}H_9ClN_2$	\mathbf{A}_{1}	32	118-121 (0.6)		83 ^d	84	179	118
3	3-Chlorophenyl	$C_{10}H_9ClN_2$	\mathbf{A}_1	59	130-132 (0.7)		85	62^{d}		
4	2-Chlorophenyl	$C_{10}H_9ClN_2$	A ₁	42	154-156 (1.0)		95	70		
5	2,6-Dichlorophenyl	$C_{10}H_8Cl_2N_2$	A_1	68	226 (HCl salt)	EtOH-H ₂ O	78^d	87	127	110
6	2,5-Dichlorophenyl	$C_{10}H_8Cl_2N_2$	A_1	80	167 (0.2); 59-60 ^e		91	61^d		
7	3,4-Dichlorophenyl	$C_{10}H_8Cl_2N_2$	A_1	55	5051	EtOAc	75ª	70^d	132	119
8	4-Fluorophenyl	$C_{10}H_9FN_2$	\mathbf{A}_1	80	115-117 (0.8)		100	77		
9	3-Trifluoromethyl- phenyl	$C_{11}H_9F_3N_2$	$\mathbf{A}_{1}^{'}$	82	127 (HCl salt)	EtOH-H ₂ O	68 ^d	72 ^d	156	285
10 ^f	4-Nitrophenyl	$C_{10}H_{9}N_{3}O_{2}$	A ₁	57	53 [51]	EtOAc	91 ^d	112	123	100
11	3-Nitrophenyl	$C_{10}H_9N_3O_2$	A_1	60	88–89	EtOAc	83 ^d	89	146	130
12	2-Hydroxy-5- nitrophenyl	$C_{10}H_9N_3O_3$	A ₁	73	217-218	EtOH-H ₂ O	82 ^d	110	122	111
13	3-Methoxyphenyl	$C_{11}H_{12}N_2O$	A_1	26	122 (0.5)		36 ^d	43^d	208	220
14	4-Methoxyphenyl	$C_{11}H_{12}N_2O$	\mathbf{A}_1	52	59	EtOAc-petr ether ^s	48 ^d	41^{d}	120	264
15	2,3-Dimethoxy- phenyl	$\mathbf{C}_{12}\mathbf{H}_{14}\mathbf{N}_{2}\mathbf{O}_{2}$	A_1	28	158 (0.5)		49 ^{<i>d</i>}	70 ^d	230	260
16	3,4-Dimethoxy- phenyl	$C_{12}H_{14}N_2O_2$	A ₁	27	39–41	EtOAc-petr ether ^s	66 ^{<i>d</i>}	57 ^d	225	240
17	3,4-Methylene- dioxyphenyl	$C_{11}H_{10}N_2O_2$	A ₁	51	198 (5.0)		52 ^d	61	370	235
18	2-Chloro-4,5-meth- ylenedioxyphenyl	$C_{11}H_9ClN_2O_2$	A ₁	45	188 (4.0); 86–87 ^e		75 ^d	84	158	152
19 ^{<i>f</i>}	4-Tolyl	$C_{11}H_{12}N_2$	\mathbf{A}_1	30	51 [145-152 (0.3)]		42^d	35 ^d	281	258
20	4-tert-Butylphenyl	$C_{14}H_{18}N_2$	\mathbf{A}_1	35	138 (0.5)		83ď	84	162	138
21	4-Biphenyl	$C_{16}H_{14}N_2$	\mathbf{A}_2	74	129-131	EtOAc	71^d	100	176	168
22	4-Ethoxycarbonyl- phenyl	$\mathbf{C}_{13}\mathbf{H}_{14}\mathbf{N}_{2}\mathbf{O}_{2}$	A ₁	22	68	C_6H_6 -petr ether ^g	118	125 ^d		
23	4-Benzoyloxy- phenyl	$C_{17}H_{14}N_2O_2$	A ₁	95	85	C ₆ H ₆ -petr ether ^e	100	94		
24	3,5-Di- <i>tert</i> -butyl- 4-hydroxyphenyl	$C_{18}H_{26}N_2O$	\mathbf{A}_2	30	160	C ₆ H ₆ -petr ether ^e	96	104		
25	2,4-Dichlorothienyl	$C_8H_6ClN_2S$	A_1	35	68	EtOAc-petr ether ^s	113	75^d		
26	$4 - \alpha - N$ -Imidazolyl- tolyl	$C_{14}H_{14}N_4$	\mathbf{A}_{1}	50	131-132	EtOAc	62 ^d	100	154	143
27	2-α-N-Imidazolyl- methyl-4,5-di- methylphenyl	$C_{16}H_{18}N_4$	\mathbf{A}_3	32	151–153	EtOAc	63 ^d	68 ^d	178	223
28	4-Aminophenyl	$C_{10}H_{14}N_{3}$	h	78	127	EtOAc	76	100	239	197

^aLivers analyzed only if hypocholesterolemic effect noted. ^bKnown compounds (literature location cited) not analyzed. The other compounds give analytical results for C, H, N, Cl, and S, if present, within $\pm 0.4\%$ of theoretical values. ^cO. Wallach, *Ber.*, 16, 539 (1883). ^dSignificantly different from control, p < 0.05. ^eProduct crystallized on standing. [/]British Patent 1148103 (1969); *Chem. Abstr.*, 78, 77228c (1973). ^gBp 40-60°. ^hSee the Experimental Section.

ever, body weight gain was depressed and the animals developed fatty livers. In view of these undesirable effects it was felt that further work was not justified.

After this work had been completed, Topliss described an operational scheme "to maximize the chances of synthesizing the most potent compounds in the series as quickly as possible".³

A few retrospective observations on the hypolipidemic activities of N-benzylimidazoles prepared in this work are of interest. Although the increased activity of the 4-chloro

analog 2 (prepared first in the Topliss scheme) over that of the parent compound 1 suggests that more lipophilic derivatives would be most promising in terms of cholesterol-lowering activity, the substitution patterns of the most active hypocholesterolemic compounds in this series, 13-15 and 19, in fact carried lipophobic groups. The Topliss scheme leads one directly to the most potent hypotriglyceridemic compound 19.

In conclusion, we have demonstrated that a number of analogs of N-benzylimidazole exhibit significant hypocho-

Table II. Hypolipidemic .	Activity of Analogs of N	/-Benzylimidazoles Subst	ituted in Both Rings
----------------------------------	--------------------------	--------------------------	----------------------

				$\begin{array}{c} \text{ArCH} & \text{N} \\ \\ \text{R} \end{array} \\ R \end{array} $					Plasma lipids as % of appro- piate control		
	Ar	R	R'	Formula ^a N		Yield, %	Mp, °C	Crystn solvent	Cho- lesterol	Tri- glyc- eride	
29	4-Nitrophenyl	Н	2-CH ₃	$C_{11}H_{11}N_3O_2$	A ₁	54	96	EtOAc-petr ether ^c	106	155°	
30	3,5-Di- <i>tert</i> -butyl- 4-hydroxyphenyl	Н	2- CH ₃	$\mathbf{C_{19}H_{28}N_{2}O}$	A ₃	67	166	C ₆ H ₆ -petr ether ^c	107	90	
31	4-Tolyl	н	$4(5) - C_6 H_5$	$C_{17}H_{16}N_2$	\mathbf{A}_3	25	94	EtOAc	125°	100	
32	4-Tolyl	н	$4,5-(C_6H_5)_2$	$C_{23}H_{20}N_2$	\mathbf{A}_1	23	114	EtOAc	91	94	
33 ^d	Phenyl	C_6H_5	H	$C_{16}H_{14}N_2$	\mathbf{A}_{1}^{T}	25	86 (lit. 85.5-36.5)	EtOAc-petr ether°	91	89	

^aSee footnote b in Table I. ^bSee footnote d in Table I. ^cSee footnote g in Table I. ^dH. A. Staab and K. Wendel, Justus Liebigs Ann. Chem., 694, 91 (1966).

Table III. Hypolipidemic Activity of Miscellaneous N-Substituted Imidazoles

							Lipids as $\%$ of appropriate control			
							Plasma			
	R	Formula ^b	Meth-Yie od	eld, %	Mp or bp (mm), °C [lit. ref]	Crystn solvent	Cho- les- terol	Tri- glyc- eride	<u>-</u>	Choles- terol
34°	4-Methoxyphenyl	C ₁₀ H ₁₀ N ₂ O	2	22	6667 [6768]	H ₂ O	70 ^đ	80	157	170
35°	4-Acetoxyphenyl	$C_{11}H_{10}N_2O$	1	2	118–119 [118–119]	EtOH-H ₂ O	85 ⁴	62 ^d	158	132
36°	β -4-Chlorophenylethyl	$C_{11}H_{11}ClN_2$	3	33	159–161 (0.1) [159–161 (1.0)]		94	134		
37 38	β -4-Nitrophenylethyl 3-Phenylpropyl	$C_{11}H_{11}N_3O_2 \\ C_{12}H_{14}N_2$		13 36	59-61 142-144 (0.5)	EtOAc	86 ⁴ 90	85 53₫	124	107
39	Cinnamyl	$C_{12}H_{14}N_2$ $C_{12}H_{12}N_2$		59	46-48	Petr ether ^f	72 ^d	82	113	119

^{a,b}See corresponding footnotes in Table I. ^cL. M. Sitkina and A. M. Simonov, *Khim. Geterotsikl. Soedin Akad. Nauk. Latv. SSR*, 1, 143 (1966); *Chem. Abstr.*, **65**, 13686e (1966). ^dSee corresponding footnote in Table I. ^eW. N. Cannon, C. E. Powell, and R. G. Jones, *J. Org. Chem.*, **22**, 1323 (1957). ^fSee footnote g in Table I.

Table IV. Effect of Standard Hypolipidemic Agents in the Test S	vetom Used for N. Benzylimidezeles
Table IV. Effect of Standard Hypothetine Agents in the Test S	stem Osed for A-Denzymmuazoies

	Lipids as $\%$ of appropriate control						
	F	Plasma	Liver				
Compound	Cholesterol	Triglycerides	Total lipid	Cholesterol			
Atromid-S ^a	101	84	99	100			
Atromid-S (0.5% in diet)	112	69 ^{<i>b</i>}	106	87			
Probucol ^c	46 ^b	đ	101	102			
N-Dodecylimidazole	48 ^b	44 ^b	190	180			

^aEthyl *p*-chlorophenoxyisobutyrate. ^bSee footnote *d* in Table I. ^c4,4'-(Isopropylidenedithio)bis(2,6-di-*tert*-butylphenol), DH-581. ^aNot determined.

lesterolemic and hypolipidemic activity in laboratory animals. The order of activity is similar to that shown by Nalkylimidazoles. However, the inhibition of hepatic cholesterolgenesis shown by 1-alkylimidazoles could not be demonstrated using a number of N-benzylimidazoles, indicating that the mode of action differs between the two types of imidazole.

Experimental Section

Satisfactory spectra (ir and NMR) were obtained for all compounds. The following instruments were used: for melting point, Reichert Thermopan apparatus; ir, Perkin-Elmer 257; and NMR, Varian A-60. The compounds were prepared by the alkylation of imidazole with halides which were either commercially available or prepared by standard methods. Method A. Alkylations using benzylic or unsaturated halides (e.g., 39) were carried out in the presence of anhydrous K_2CO_3 and a suitable solvent: A₁, Me₂CO; A₂, DMF; or A₃, MeCOEt. The preparation of 14 below is a typical procedure.

N-(4-Methoxybenzyl)imidazole (14). Imidazole (3.4 g, 0.05 mol), 4-methoxybenzyl bromide (10.05 g, 0.05 mol), and anhydrous K_2CO_3 (14 g, 0.1 mol) in dry Me₂CO (50 ml) were stirred and heated under reflux for 4 hr. The mixture was allowed to cool, the solids were filtered off, and the filtrate was evaporated. The product 14 was isolated by chromatography on silica with CHCl₃-5% MeOH: 4.88 g (52%); mp 59° [EtOAc-petroleum ether (bp 40-60°)].

Method B. N-(3-Phenylpropyl)imidazole (38). To a stirred solution of imidazole (6.8 g, 0.1 mol) and finely powdered NaOH)5.0 g, 0.125 mol) in dry *n*-BuOH (100 ml) at 125°, 1-bromo-3-phenylpropane (19.9 g, 0.1 mol) was added dropwise over 20 min. The reaction mixture was maintained at 125° for a further 20 min, allowed to cool, and diluted with water (200 ml). The mixture was extracted with Et_2O and dried (MgSO₄) and after removal of the solvent, the residue was distilled yielding 38 (6.4 g, 36%), bp 142-144° (0.5 mm).

N-(4-Aminobenzyl)imidazole (28). A solution of N-(4-nitrobenzyl)imidazole (10, 2.03 g, 0.01 mol) in MeOH (25 ml) was added dropwise to a stirred suspension of 10% Pd/C (0.5 g) in water (10 ml) containing NaBH₄ (0.78 g, 0.02 mol) under nitrogen over a period of 5 min. The mixture was stirred for a further 10 min, allowed to cool, filtered, acidified (2 N HCl) to destroy excess NaBH₄, and then basified (1 N NaOH). The mixture was extracted with CHCl₃ and the product isolated by evaporation of the dried (MgSO₄) CHCl₃ solution, followed by chromatography on

silica with CHCl₃-5% MeOH to yield 28 (1.34 g, 78%), mp 127° (EtOAc).

Biological Methods. Male mice of the CFLP strain (Carworth Europe) weighing 20-22 g were allocated to experimental groups of ten animals, so that the mean body weight of each group was the same. The compounds to be tested were added to powdered commercial feed at a level of 0.1% (w/w). This is equivalent to a daily dose of ca. 140 mg/kg; for a typical compound such as 14 this is ca. 750 µmol/kg. Animals were allowed food and water ad libitum for 10 days, after which they were killed and bled and their livers removed and frozen pending analysis.

Plasma cholesterol and triglycerides were measured using a Technicon Autoanalyzer (method N24A for cholesterol and method N78 for triglycerides). Treated groups were compared with controls using Student's t test. Liver lipid and cholesterol were determined on CHCl₃-MeOH extracts of pooled liver samples.⁴

Acknowledgments. The authors are grateful to Mr. J. Jennings for skilled technical assistance.

References and Notes

- S. D. Atkin, B. Morgan, K. H. Baggaley, and J. Green, Biochem. J., 130, 153 (1972); British Patent 1364312 (1974) [Chem. Abstr., 82, 4255c (1975)].
- (2) T. Neilson, H. C. S. Wood, and H. G. Wylie, J. Chem. Soc., 371 (1962).
- (3) J. G. Topliss, J. Med. Chem., 15, 1006 (1972).
- (4) J. Folch, M. Lees, and G. H. S. Stanley, J. Biol. Chem., 226, 497 (1957).

Conformational Effects on the Activity of Drugs. 5.¹ Pharmacological Properties of 2-(*p*-Nitrophenyl)-4-isopropylmorpholine, a Cyclic Analog of INPEA

M. Del Tacca, A. Bertelli, L. Mazzanti, B. Stacchini,

Istituto di Farmacologia dell'Università di Pisa, 56100 Pisa, Italy

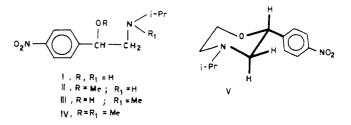
A. Balsamo, P. Crotti, B. Macchia,* and F. Macchia

Istituti di Chimica Farmaceutica e di Chimica Organica dell'Università di Pisa, 56100 Pisa, Italy. Received January 20, 1975

2-(p-Nitrophenyl)-4-isopropylmorpholine (V), an analog of 1-(p-nitrophenyl)-2-isopropylaminoethanol (INPEA, I) in which the OCHCHN chain of I is locked in a morpholine ring, loses the β -receptor blocking activity of I on various isolated preparations. The same ineffectiveness is observed in the O-methyl (II), N-methyl (III), and N,O-dimethyl analog (IV) of I. However, some other properties which are present in I, such as inhibitory effect on acetylcholine or on 5-HT, intrinsic α -sympathomimetic activity, and potentiation of catecholamines, are maintained; this demonstrates a complete dissociation of these effects from β -receptor blockade. The interactions with the α -adrenoceptors and with the uptake mechanism are discussed on the basis of the structure-activity relationship between I and its analogs II-V.

In addition to blockade of β -receptors² INPEA, 1-(*p*-nitrophenyl)-2-isopropylaminoethanol (I), has been shown to exert various effects on the adrenergic effector system. INPEA exhibited sympathomimetic activity on cardiac muscle³ and on smooth muscle strips of taenia coli.⁴ Furthermore, Gulati and others⁵ observed that INPEA competitively antagonized the excitatory effects of catecholamines on α -receptors. On the other hand, INPEA appeared to potentiate the effects of exogenous catecholamines.⁶ Janiec and Chrusciel⁷ observed that INPEA inhibited the uptake of [³H]noradrenaline by the adrenergic nerves of rat heart muscle. Finally, INPEA has been shown to antagonize in various degrees the effects of histamine and 5-HT on rabbit aortic strips.⁵

In a previous paper¹ we observed that 2-(p-nitrophenyl)-4-isopropylmorpholine (V), an analog of INPEA in which the chain OCHCHN is incorporated in a morpholine ring, lost the β -receptor blocking activity on isolated prepara-



tions of cardiac muscle. However, some other properties observed in the parent compound (INPEA) were still present. The aim of this work was to investigate further the pharmacological properties of the INPEA analog (V) and of O-methyl (II), N-methyl (III), and N,O-dimethyl (IV) derivatives of I, in order to compare their pharmacological effects with those of I.

Pharmacology. 1. Methods. (a) Isolated Rat Vas Deferens. Vasa deferentia were obtained by using the method