



Tetrahedron

Tetrahedron 61 (2005) 8499-8504

3'-Selective modification of a 4',5'-didehydro-5'-deoxy-2',3'epoxyuridine using nucleophiles

Hideki Takasu,^a Yoshie Tsuji,^b Hironao Sajiki^{b,*} and Kosaku Hirota^{b,*}

^aMedicinal Chemistry Research Institute, Otsuka Pharmaceutical Co., Ltd, 463-10 Kagasuno, Kawauchi, Tokushima 771-0192, Japan ^bLaboratory of Medicinal Chemistry, Gifu Pharmaceutical University, 5-6-1 Mitahora-higashi, Gifu 502-8585, Japan

Received 18 May 2005; accepted 13 June 2005

Available online 11 July 2005

Abstract—1-(2,3-Anhydro-5-deoxy-4,5-didehydro- α -L-*erythro*-pent-4-enofuranosyl)uracil **4** was obtained by the treatment of 5'-iodo-2',3'epoxyuridine **5** with LiHMDS in excellent yield. The pyrimidine nucleoside **4** possesses quite unique vinyl epoxide moiety within the molecules. The reactions of **4** with a variety of nucleophiles gave 3'-substituted pyrimidine nucleosides without the formation of the corresponding 2'-substituted isomers. In the case of NaN₃ or PhSH, the corresponding 5'-adduct was obtained as a minor product together with the expected 3'-adduct.

© 2005 Elsevier Ltd. All rights reserved.

1. Introduction

Chemical modification of nucleosides has been very important for the synthesis of biologically active compounds such as anti-viral agents¹ and synthetic oligonucleotide probes.² A number of modifications of the sugar moiety in nucleosides have been carried out up to now. Especially, a wide variety of attempts have been made to devise novel methodologies to functionalize the 2'- and 3'-sites of nucleosides in connection with AZT, a potent anti-HIV agent. 2',3'-Anhydro-β-D-lyxofuranosyl pyrimidine nucleosides (1) first synthesized by Fox et al.,³ are useful key intermediates for the 2'- or 3'-modified pyrimidine nucleoside analogues. A large number of reactions of 1 with a variety of nucleophiles have been investigated.^{4,5} However, a limitation to these methods is the poor regio-control of 2'or 3'-addition of nucleophiles. Usually, 3'-adducts (2) were obtained as major products together with minor 2'-adducts (3). Although a few reports indicated that 3'-adducts (2) were obtained selectively⁵, the yields of the 3'-adducts were very low and the existence of unisolable 2'-adducts (3) was suspected. Previously, we have reported the reaction of 1 (R=OH, or OR') with AlMe₃ with a view to the regioselective 2'-attack of the methyl group by the cyclic coordination effect of AlMe₃ between 5'- and epoxideoxygen, while a mixture of the 3'- and 2'-adduct (2 and 3) was obtained⁶ as well as other reported results.⁴

In order to develop an entirely regioselective nucleophilic attack toward the epoxide moiety of 2',3'-anhydro nucleosides, we expected the use of a conjugated epoxide with a double bond such as **4** would lead to a better result. Here we would like to report the synthesis of 1-(2,3-anhydro-5-deoxy-4,5-didehydro- α -L-*erythro*-pent-4-enofuranosyl)uracil (**4**) and its application to the regioselective synthesis of 3'-adducts (**2**).⁷



2. Results and discussion

The synthesis of 4',5'-didehydro-5'-deoxy-2',3'-epoxyuridine (4) was achieved by treatment of 5'-iodo-2',3'epoxyuridine (5)^{3c} with *t*-BuOK, an appropriate nonnucleophilic base, at room temperature (64%), while the use of DBU (1,8-diazabicyclo[5.4.0]undec-7-ene) as a base gave a complex mixture. After optimization of the reaction conditions, we finally found the use of LiHMDS, a bulky and totally non-nucleophilic base, in DMF at 0 °C gave 4 in 92% yield (Scheme 1).

Therefore, the reaction of **4** using a variety of nucleophiles was investigated. We first carried out the reaction of **4** with

Keywords: Nucleophile; Nucleoside; Adduct.

^{*} Corresponding authors. Tel.: +81 58 237 3931; fax: +81 58 237 5979; e-mail: sajiki@gifu-pu.ac.jp

^{0040–4020/\$ -} see front matter @ 2005 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2005.06.053



Scheme 1. Synthesis of 1-(2,3-anhydro-5-deoxy-4,5-didehydro- α -L-*ery-thro*-pent-4-enofuranosyl)uracil **4** from **5**. Reactant and condition: (a) *t*-BuOK, DMF, rt (64%) or (b) LiHMDS, DMF, 0 °C, Ar (92%).

5 equiv of AlMe₃ in CH_2Cl_2 under reflux conditions. The regioselective nucleophilic addition of a methyl group to epoxide proceeded on the 3'-position and the 3'-adduct (6a) was obtained as the sole product in 81% isolated yield and the isomeric 2'-adduct (7a) and other by-products were not detected in the reaction mixture (Scheme 2). To ascertain the generality of this regioselective reaction, treatment of 4 with various nucleophiles was investigated. The results are summarized in Table 1. When MeONa was used as a nucleophile, reflux conditions were required for completion of the reaction. On the other hand, all other additions were achieved at room temperature (entries 3-8). Reactions using BnNH₂, AlMe₃, MeONa and NaN₃ proceeded without an additional base (entries 1-3, and 7). In the case of BzOH, BzSH and PhSH, Et₃N was employed as a base (entries 5, 6, and 9). When NaH was employed as a base instead of Et₃N for the reaction of BzOH, a lowering of the yield of 6e and a remaining of the starting material 4 were observed. The reaction of BzOH was very slow (168 h) and the corresponding arabinofuranosyl derivative as a by-product based upon the hydrolysi the 3'-addition of BzSH was completed w of the high nucleophilicity.



Scheme 2. Reaction of 4 with AlMe₃.

Table 1. Nucleophilic addition to 4

e(6e') was formed	0	or		
s of (6e), although vithin 48 h because		PhSH		
	4			
		R = N ₃		
		r.t.		
		120 °C		

r.t. 80% 90 °C 33%

R = SPh

Next, we investigated the reaction of 4 with NaN₃ in DMF at

room temperature. Similar to the other reactions (entries 1–6), 3'-adduct (**6g**) was afforded as a major product together with 5'-adduct (**8g**),⁸ formed by the nucleophilic addition onto the 5'-position of **4** as a side product (3%)

(Scheme 3 and Table 1, entry 7). When this reaction was

carried out at 120 °C, the 5'-adduct (8g) was afforded as a

major product (28%) along with 3'-adduct (**6g**) as a minor product (4%) (Scheme 3). In addition, a similar tendency was observed in the reaction of **4** with PhSH (Scheme 3

and Table 1, entry 8). In these reactions, formation of a

2'-adduct (7) was never observed as well as in other

reactions of 4 with nucleophiles (entries 1-6). Usually, the

3'-cation is more stable than the 2'-cation owing to the

electron-withdrawing effect of the 1'-uracil moiety of

2',3'-epoxyuridine derivatives (1) and the formation of

3'-adduct (2) was superior to the formation of 2'-adduct (3),

although the effect is not enough to perform the

regioselective 3'-nucleophilic addition. On the other hand,

the 3'-position of **4** is the allylic position and it is obvious

that the 3'-cation is strongly stabilized by the conjugation

with the 4',5'-double bond as an allylic cation (Fig. 1).

Therefore, regioselective nucleophilic addition onto the

3'-position occurred readily and the 3'-adduct (6) was

obtained regioselectively. In the reaction with a compara-

tively soft nucleophile such as NaN₃ or PhSH, 5'-adduct (8) was also formed as a by-product.^{9,10}

(6g)

63%

4%

(6h)

(8g)

3%

28%

(8h)

11%

50%

Scheme 3. Reaction of 4 with NaN₃ or PhSH.

	- -		nucleophil	$\xrightarrow{\circ} \qquad \qquad$					
		4			6	8	6e'		
Entry	Nucleophile Additional Solvent		Tempera- ture	<i>T</i> (h)	R	Yield (%) ^a			
							Product	6	8
1	AlMe ₃	_	CH ₂ Cl ₂	reflux	12	Me	а	81	nd ^b
2	MeONa		MeOH	reflux	6	OMe	b	80	nd ^b
3	BnNH ₂	_	CH_2Cl_2	rt	24	NHBn	с	81	nd ^b
4	$CH_2(CO_2Me)_2$	MeONa	MeOH	rt	12	$CH(CO_2Me)_2$	d	69	nd ^b
5	BzOH	Et ₃ N	CH_2Cl_2	rt	168	OBz	е	61	nd ^b
6	BzSH	Et ₃ N	CH ₂ Cl ₂	rt	48	SBz	f	52	nd ^b
7	NaN ₃	_	DMF	rt	3	N_3	g	63	3
8	PhSH	Et ₃ N	_	rt	1	SPh	ĥ	80	11

^a Isolated yield after chromatographic separation.

^b Not detectable.

8500



Figure 1.

It was noteworthy that the use of Et_2AlCN as a nucleophile provided 3',4'-unsaturated-3'-adduct (9) as the sole product via isomerization of the 3'-adduct (10) because of the efficient activation of 3'-hydrogen of 10 by the strong electron-withdrawing cyano group after regular nucleophilic attack of the cyano anion onto the 3'-position of 4 (Scheme 4).



Scheme 4. Reaction of 4 with Et₂AlCN.

Usually, the 5'-hydroxy group of nucleosides is supposed to be important for biological activity, while 3'-adducts (6) do not possess the 5'-hydroxy group. So we investigated the conversion method of the olefin of 6 to a hydroxymethyl group. The hydroboration reaction of **6b** should be applicable to the introduction of the 5'-hydroxy group. The reaction of **6b** with BH₃–THF at room temperature, followed by H₂O₂-NaOH treatment, gave only undesirable α -isomer (11). When the reaction was heated at reflux temperature, the desired β -isomer (12) was afforded as a minor product (17%) together with corresponding α -isomer (11) (53%) (Scheme 5). As the reason for this isomer ratio, it was indicated that the steric hindrance effect of the 3'-methoxy group was more effective than that of the uracil ring. At present, we are unable to determine the underlying cause of this stereoselectivity.



Scheme 5. 5'-Hydroxylation reaction of 6b: (i) BF_3 -THF, THF, reflux (ii) H_2O_2 , NaOH.

3. Conclusion

We have shown a simple synthetic method of 1-(2,3anhydro-5-deoxy-4,5-didehydro- β -D-*erythro*-pent-4-enofuranosyl)uracil (4) and mild and efficient regioselective method for the conversion of 4 to 3'-substituted pyrimidine nucleosides (6), (9), (11) and (12) without the formation of the corresponding 2'-substituted derivatives. The reaction is general for a variety of nucleophiles and the simplicity of this method makes it an attractive new tool for synthesis of various sugar modified pyrimidine nucleosides.

4. Experimental

4.1. General

¹H and ¹³C NMR spectra were recorded on a JEOL EX 400 spectrometer or a JEOL GX 270 spectrometer (¹H: 400 or 270 MHz, ¹³C: 100 MHz). Chemical shifts (δ) are given in ppm relative to residual solvent or tetramethylsilane as an internal standard. Low and high-resolution mass spectra were taken on a JEOL JMS-SX 102 or JMS-D300 machine. Melting points were determined on a Yanagimoto micro-melting-point apparatus and were corrected. IR spectra were recorded on a Perkin Elmer model 1600 FT-IR spectrophotometer. UV spectra were obtained from EtOH solution on a Shimazu UV-260 spectrophotometer. All reagents were commercially available and used without further purification. Compounds known in the literature were characterized by comparison of their ¹H NMR data with the previously reported data.

4.1.1. 1-(2,3-Anhydro-5-deoxy-4,5-didehydro- α -L-*ery-thro*-pent-4-enofuranosyl)uracil (4). To a stirred solution of 5^{3c} (336 mg, 1.0 mmol) in dry DMF (20 ml) was added 1.0 M THF solution of LiN[(CH₃)₃Si]₂ (2.2 ml, 2.2 mmol) at 0 °C under argon atmosphere. The reaction mixture was stirred for 4 h at 0 °C and the mixture was evaporated in vacuo at room temperature. The residue was diluted with water and neutralized with saturated NH₄Cl and extracted with AcOEt. The organic solution was evaporated in vacuo and the residue was subjected to silica gel column chromatography (benzene/AcOEt 3:1) to afford 4 (192 mg, 92%) as a white foam.

MASS *m/z* (relative intensity): 208 (M⁺, 43%), 112 (B⁺ + 1, 18%), 97 (S⁺, 100%). ¹H NMR (CDCl₃) δ : 8.71 (1H, br s, N³–H), 7.50 (1H, d, *J*=8.3 Hz, 6-H), 6.39 (1H, s, 1'-H), 5. 76 (1H, d, *J*=2.4 Hz, 5'-H), 4.74 (1H, d, *J*=2.4 Hz, 5'-Ha), 4.59 (1H, d, *J*=2.4 Hz, 5'-Hb), 4.23 (1H, d, *J*=2.9 Hz, 2'-H), 4.05 (1H, d, *J*=2.9 Hz, 3'-H). ¹³C NMR (CDCl₃) δ : 162.84 (4-C), 154.34 (4'-C), 150.30 (2-C), 140.75 (6-C), 102.80 (5-C), 95.95 (5'-C), 81.79 (1'-C), 56.52 (2'-C), 54.56 (3'-C). HRMS *m/z* calcd for C₉H₈N₂O₄: 208.0484. Found: 208.0473.

4.1.2. 1-(3,5-Dideoxy-3-methyl- β -D-*threo*-pent-4-enofuranosyl)uracil (6a). To a stirred solution of 4 (48 mg, 0.23 mmol) in dry CH₂Cl₂ (10 ml) was added 1.0 M hexane solution of AlMe₃ (1.20 ml, 1.20 mmol) at room temperature under argon atmosphere. The reaction mixture was refluxed for 12 h and the mixture was partitioned between CHCl₃ (30 ml) and water (30 ml). The aqueous layer was filtered using a Celite cake and the filtrate was concentrated in vacuo. The residue was subjected to silica gel column chromatography (CHCl₃/MeOH 10:1) to afford **6a** (42 mg, 81%) as a white foam.

MASS *m*/*z* (relative intensity): 224 (M⁺, 2%), 113 (S⁺ and B⁺+2, 14%), 112 (B⁺+1, 14%), 95 (100%). ¹H NMR (DMSO-*d*₆) δ : 11.35 (1H, br s, N³-H), 7.31 (1H, d, *J*=8.

3 Hz, 6-H), 6.26 (1H, d, J=4.9 Hz, 1'-H), 5.77 (1H, d, J=4. 9 Hz, 2'-OH), 5.58 (1H, d, J=8.3 Hz, 5-H), 4.32 (1H, d, J= 1.5 Hz, 5'-Ha), 4.03 (1H, t, J=4.9 Hz, 2'-H), 4.01 (1H, d, J=1.5 Hz, 5'-Hb), 2.68 (1H, q, J=6.8 Hz, 3'-H), 1.15 (3H, d, J=6.8 Hz, 3'-CH₃). ¹³C NMR (DMSO- d_6) δ : 163.93 (4-C), 163.15 (4'-C), 150.47 (2-C), 141.47 (6-C), 100.86 (5-C), 85.17 (5'-C), 82.40 (1'-C), 74.36 (2'-C), 41.99 (3'-C), 16.60 (3'-CH₃). Anal. Calcd for C₁₀H₁₂N₂O₄ (M_w =224.21): C, 53.57; H, 5.39; N, 12.49. Found: C, 53.30; H, 5.42; N, 12.23.

4.1.3. 1-(5-Deoxy-3-*O***-methyl-** β **-D-***threo***-pent-4-enofuranosyl)uracil (6b).** To a stirred solution of **4** (13 mg, 0.06 mmol) in dry MeOH (5 ml) was added 28% MeOH solution of sodium methoxide (0.06 ml, 0.30 mmol) at room temperature under argon atmosphere. The reaction mixture was refluxed for 6 h and the mixture was neutralized with AcOH. The solvent was evaporated in vacuo. The residue was subjected to silica gel column chromatography (CHCl₃/ MeOH 20:1) to afford **6b** (12 mg, 80%). Residue was recrystallized from MeOH (colorless solid).

Mp: 166–167 °C. MASS *m/z* (relative intensity): 240 (M⁺, 1%), 111 (B⁺, 100%). ¹H NMR (DMSO-*d*₆) δ : 11.47 (1H, br s, N³–H), 7.41 (1H, d, *J*=8.0 Hz, 6-H), 6.29 (1H, d, *J*=3. 4 Hz, 1'-H), 6.07 (1H, d, *J*=2.9 Hz, 2'-OH), 5.74 (1H, d, *J*=8.0 Hz, 5-H), 4.67 (1H, s, 5'-Ha), 4.43 (1H, s, 5'-Hb), 4. 21 (1H, br s, 2'-H), 4.13 (1H, br s, 3'-H), 3.42 (3H, s, 3'-OCH₃). ¹³C NMR (DMSO-*d*₆) δ : 163.16 (4-C), 157.49 (4'-C), 150.31 (2-C), 141.60 (6-C), 100.62 (5-C), 88.07 (5'-C), 86.70 (1'-C), 83.96 (3'-C), 72.15 (2'-C), 55.76 (3'-OCH₃). Anal. Calcd for C₁₀H₁₂N₂O₅ (*M*_w=240.21): C, 50.00; H, 5.04; N, 11.66. Found: C, 50.07; H, 5.04; N, 11.52.

4.1.4. 1-(3-Benzylamino-3,5-dideoxy-\beta-D-*threo***-pent-4enofuranosyl)uracil (6c). To a stirred solution of 4** (137 mg, 0.66 mmol) in dry CH₂Cl₂ (5 ml) was added benzylamine (3.60 ml, 32.9 mmol) at room temperature under argon atmosphere. The reaction mixture was stirred for 24 h and the mixture was concentrated in vacuo. The residue was subjected to silica gel column chromatography (CHCl₃/MeOH 30:1) to afford **6c** (169 mg, 81%). Residue was recrystallized from AcOEt (colorless solid).

Mp: 153–155 °C. MASS *m/z* (relative intensity): 315 (M⁺, 5%), 297 (M⁺ – H₂O, 4%), 224 (M⁺ – CH₂Ph, 6%), 203 (S⁺ + 1, 12%), 186 (S⁺ – H₂O, 58%), 91 (PhCH₂⁺, 100%). IR (KBr) cm⁻¹: 3395.9, 1685.7. ¹H NMR (DMSO-*d*₆) δ : 11. 41 (1H, br s, N³–H), 7.27–7.42 (6H, m, Ph-H and 6-H), 6.48 (1H, d, *J*=3.4 Hz, 1'-H), 5.80 (1H, d, *J*=4.3 Hz, 2'-OH), 5. 65 (1H, d, *J*=7.8 Hz, 5-H), 4.52 (1H, s, 5'-Ha), 4.21 (1H, br s, 2'-H), 4.19 (1H, s, 5'-Hb), 3.88 (1H, d, *J*=10 Hz, 3'-NHCH₂), 3.78 (1H, d, *J*=10 Hz, 3'-NHCH₂), 3.54 (1H, br s, 3'-NH). Anal. Calcd for C₁₆H₁₇N₃O₄ (*M*_w=315.33): C, 60.94; H, 5.43; N, 13.33. Found: C, 60.94; H, 5.46; N, 13.17.

4.1.5. 1-(3,5-Dideoxy-3-di(methoxycarbonyl)methyl- β -*b*-*threo*-pent-4-enofuranosyl)uracil (6d). To a stirred solution of dimethyl malonate (1.28 ml, 11.19 mmol) in dry MeOH (5 ml) was added 28% MeOH solution of sodium methoxide (1.08 ml, 5.60 mmol) at room temperature under argon atmosphere. The mixture was added to a stirred solution of 4 (291 mg, 1.40 mmol) in dry MeOH (5 ml) at

room temperature under argon atmosphere. The reaction mixture was stirred for 12 h and the mixture was neutralized with AcOH. The solvent was evaporated in vacuo and the residue was diluted with water and extracted with AcOEt. The organic solution was dried over MgSO₄ and the solvent was evaporated in vacuo and residue was subjected to silica gel column chromatography (CHCl₃/MeOH 30:1) to afford **6d** (331 mg, 69%) as a white foam.

MASS m/z (relative intensity): 340 (M⁺, 5%), 322 (M⁺ – H₂O, 100%), 228 (S⁺ – 1, 38%), 211 (M⁺ – CH(CO₂Me)₂ + 2, 80%). ¹H NMR (CDCl₃) δ : 9.01 (1H, br s, N³–H), 7.33 (1H, d, J=8.3 Hz, 6-H), 6.39 (1H, d, J=4.8 Hz, 1'-H), 5.69 (1H, d, J=8.3 Hz, 5-H), 4.86 (1H, d, J=4.8 Hz, 2'-H), 4.60 (1H, d, J=2.4 Hz, 5'-Ha), 4.10 (1H, d, J=2.4 Hz, 5'-Hb), 3. 79 (6H, s, CH₃×2), 3.74 (1H, br s, 3'-H), 3.71 (1H, br s, 2'-OH), 3.35–3.43 (1H, m, 3'-CH(CO₂Me)₂). HRMS m/z calcd for C₁₄H₁₆N₂O₈: 340.0897. Found: 340.0907.

4.1.6. 1-(3-*O*-Benzoyl-5-deoxy- β -D-*threo*-pent-4-enofuranosyl)uracil (6e). To a stirred solution of **4** (107 mg, 0.51 mmol) in dry CH₂Cl₂ (5 ml) was added benzoic acid (188 mg, 1.54 mmol) and Et₃N (0.21 ml, 1.54 mmol) at room temperature under argon atmosphere. The reaction mixture was stirred for 168 h at room temperature and the solvent was evaporated in vacuo. The residue was diluted with water and extracted with AcOEt and the organic solution was dried over MgSO₄ and the solvent was evaporated in vacuo. The residue was subjected to silica gel column chromatography (CHCl₃/MeOH 50:1) to afford **6e** (103 mg, 61%) (white foam) and **6e**' (12 mg, 10%) (white foam), respectively.

Compound **6e**. MASS m/z (relative intensity): 330 (M⁺, 0.7%), 312 (M⁺ - H₂O, 13%), 218 (S⁺ - 1, 18%), 208 (M⁺ - OCOPh + 1, 19%), 201 (S⁺ - H₂O, 7%), 105 (PhCO, 100%). IR (KBr) cm⁻¹: 3424.3, 1686.7. ¹H NMR (CDCl₃) δ : 10.46 (1H, br s, N³-H), 7.98 (2H, d, J=7.3 Hz, Ph-o), 7.32–7.51 (4H, m, Ph-m and p and 6-H), 6.51 (1H, d, J=3.4 Hz, 1'-H), 5.75 (1H, br s, 3'-H), 5.62 (1H, d, J= 8.3 Hz, 5-H), 5.17 (1H, br s, 2'-OH), 4.80 (2H, br s, 2'-H and 5'-Ha), 4.66 (1H, d, J=2.4 Hz, 5'-Hb). HRMS m/z calcd for C₁₆H₁₄N₂O₆: 330.0852. Found: 330.0843.

Compound **6e**'. MASS m/z (relative intensity): 226 (M⁺, 51%), 208 (M⁺ - H₂O, 13%), 115 (S⁺, 100%). IR (KBr) cm⁻¹: 3383.9, 1686.8. ¹H NMR (CDCl₃) δ : 10.83 (1H, br s, N³-H), 7.98 (1H, d, J=8.1 Hz, 6-H), 6.62 (1H, d, J= 2.9 Hz, 1'-H), 5.58 (1H, d, J=8.1 Hz, 5-H), 4.81 (1H, br s, 2'-H), 4.72 (1H, d, J=2.7 Hz, 5'-Ha), 4.63 (1H, br s, 3'-H), 4.55 (1H, d, J=2.7 Hz, 5'-Hb). FABHRMS m/z calcd for C₉H₁₁N₂O₅: 227.0668. Found: 227.0664.

4.1.7. 1-(3-Benzoylthio-3,5-dideoxy-\beta-D-threo-pent-4-enofuranosyl)uracil (6f). To a stirred solution of **4** (169 mg, 0.81 mmol) in dry CH₂Cl₂ (5 ml) was added thiobenzoic acid (0.29 ml, 2.44 mmol) and Et₃N (0.34 ml, 2.44 mmol) at room temperature under argon atmosphere. The reaction mixture was stirred for 2 days at room temperature and the solvent was evaporated in vacuo. The residue was subjected to silica gel column chromatography (CHCl₃/MeOH 45:1) to afford **6f** (145 mg, 52%) as a white foam.

MASS m/z (relative intensity): 346 (M⁺, 4%), 328 (M⁺ – H₂O, 3%), 209 (M⁺ – SCH₂Ph, 5%), 105 (PhCO, 100%), 77 (Ph, 48%). IR (KBr) cm⁻¹: 3425.2, 1683.6. ¹H NMR (CDCl₃) δ : 9.64 (1H, br s, N³–H), 7.93 (2H, d, J=7.3 Hz, Ph-o), 7.58–7.64 (1H, m, Ph-p), 7.44–7.49 (3H, m, Ph-m and 6-H), 6.40 (1H, d, J=3.4 Hz, 1'-H), 5.65 (1H, d, J=7.8 Hz, 5-H), 4.71 (1H, d, J=1.9 Hz, 5'-Ha), 4.60–4.67 (3H, m, 3'-H, 2'-H and 2'-OH), 4.44 (1H, d, J=2.4 Hz, 5'-Hb). HRMS m/z calcd for C₁₆H₁₄N₂O₅S₁: 346.0623. Found: 346. 0610.

4.1.8. 1-(3-Azido-3,5-dideoxy- β -D-threo-pent-4-enofuranosyl)uracil (6g) and 1-(5-azido-3,5-dideoxy- β -D-glyceropent-4-enofuranosyl)uracil (8g). To a stirred solution of 4 (122 mg, 0.59 mmol) in dry DMF (10 ml) was added sodium azido (57 mg, 0.88 mmol) at room temperature under argon atmosphere. The reaction mixture was stirred for 3 h at room temperature and the mixture was concentrated in vacuo. The residue was subjected to silica gel column chromatography (benzene/AcOEt 2:1) to give **6g** (93 mg, 63%) as the first fraction (white foam) and **8g** (5 mg, 3%) as the second fraction (white foam).

Compound **6g**. IR (KBr) cm⁻¹: 2104 (N₃). MASS m/z (relative intensity): 251 (M⁺, 2%), 209 (M⁺ - N₃, 14%), 167 (100%), 112 (B⁺ + 1, 39%). ¹H NMR (DMSO- d_6) δ : 11.44 (1H, br s, N³-H), 7.36 (1H, d, J=8.3 Hz, 6-H), 6.29 (1H, d, J=4.4 Hz, 2'-OH), 6.24 (1H, d, J=4.4 Hz, 1'-H), 5.61 (1H, d, J=8.3 Hz, 5-H), 4.69 (1H, d, J=2.4 Hz, 5'-Ha), 4.63 (1H, br s, 3'-H) 4.41 (1H, d, J=2.4 Hz, 5'-Ha), 4.63 (1H, br s, 2'-H). Anal. Calcd for C₉H₉N₅O₄ (M_w = 251.20): C, 43.03; H, 3.61; N, 27.88. Found: C, 43.13; H, 3.67; N, 27.64.

Compound **8g**. IR (KBr) cm⁻¹: 2105 (N₃). MASS m/z (relative intensity): 251 (M⁺, 12%), 209 (M⁺ - N₃, 4%), 167 (100%), 113 (B⁺ + 2, 44%). ¹H NMR (DMSO- d_6) δ : 11.42 (1H, br s, N³-H), 7.32 (1H, d, J=8.0 Hz, 6-H), 6.42 (1H, d, J=6.8 Hz, 1'-H), 5.62 (1H, d, J=8.0 Hz, 5-H), 5.57 (1H, d, J=4.9 Hz, 2'-OH), 5.29 (1H, d, J=1.5 Hz, 3'-H), 4.88–4.91 (1H, m, 2'-H), 4.09 (2H, s, 5'-H×2). ¹³C NMR (DMSO- d_6) δ : 163.32 (4-C), 154.29 (4'-C), 150.53 (2-C), 142.37 (6-C), 103.89 (5'-C), 100.82 (5-C), 86.02 (1'-C), 70.68 (3'-C), 46.45 (2'-C). HRMS m/z calcd for C₉H₉N₅O₄: 251.0662. Found: 251.0655.

4.1.9. 1-(3,5-Dideoxy-3-phenylthio- β -D-threo-pent-4-enofuranosyl)uracil (6h) and 1-(3,5-dideoxy-5-phenylthio- β -D-glycero-pent-4-enofuranosyl)uracil (8h). To a stirred solution of 4 (110 mg, 0.53 mmol) in Et₃N (10 ml) was added thiophenol (82 µl, 0.80 mmol) at room temperature under argon atmosphere. The reaction mixture was stirred for 1 h at room temperature and the solvent was evaporated in vacuo and a small amount of the remaining Et₃N was removed as the toluene azetrope. The residue was subjected to silica gel column chromatography (CHCl₃/MeOH 150:1) to give 6h (134 mg, 80%) as the first fraction (white foam) and 8h (19 mg, 11%) as the second fraction (white foam).

Compound **6h**. MASS *m*/*z* (relative intensity): 318 (M⁺, 58%), 206 (S⁺ - 1, 100%), 112 (B⁺ + 1, 16%). ¹H NMR (CDCl₃) δ : 11.04 (1H, br s, N³-H), 7.23–7.50 (6H, m, 6-H and 3'-SPh), 6.37 (1H, d, *J*=2.9 Hz, 1'-H), 5.46 (1H, d, *J*=

8.3 Hz, 5-H), 5.38 (1H, d, J=4.4 Hz, 2'-OH), 4.80–4.82 (1H, m, 2'-H), 4.61 (1H, d, J=2.0 Hz, 5'-Ha), 4.23 (1H, br s, 3'-H), 4.21 (1H, d, J=2.0 Hz, 5'-Hb). ¹³C NMR (CDCl₃) δ : 165.99 (4-C), 157.96 (4'-C), 149.91 (2-C), 142.95 (6-C), 132.63 (SPh-C), 129.25 (SPh-C), 128.17 (SPh-C), 100.41 (5-C), 88.06 (1'-C), 87.41 (5'-C), 74.24 (2'-C), 53.50 (3'-C). Anal. Calcd for C₁₅H₁₄N₂O₄S₁ (M_w =318.35): C, 56.59; H, 4.43; N, 8.80. Found: C, 56.34; H, 4.49; N, 8.78.

Compound **8h.** MASS m/z (relative intensity): 318 (M⁺, 33%), 209 (M⁺ – SPh, 50%), 206 (S⁺ – 1, 7%), 123 (PhSCH₂⁺, 100%), 112 (B⁺ + 1, 9%). ¹H NMR (CDCl₃) δ : 10.75 (1H, br s, N³–H), 7.23–7.47 (6H, m, 6-H and 3'-SPh), 6.46 (1H, d, J=6.4 Hz, 1'-H), 5.51 (1H, d, J=8.3 Hz, 5-H), 5.16 (2H, br s, 2'-H and 3'-H), 4.58 (1H, d, J=3.9 Hz, 2'-OH), 3.66 (2H, s, 5'-H×2). ¹³C NMR (CDCl₃) δ : 165.22 (4-C), 156.88 (4'-C), 150.48 (2-C), 143.00 (6-C), 134.83 (SPh-C), 130.53 (SPh-C), 129.07 (SPh-C), 127.15 (SPh-C), 102.27 (3'-C), 100.74 (5-C), 86.91 (1'-C), 71.88 (2'-C), 31.46 (5'-C). Anal. Calcd for C₁₅H₁₄N₂O₄S₁ (M_w = 318.35): C, 56.59; H, 4.43; N, 8.80. Found: C, 56.30; H, 4.40; N, 8.72.

4.1.10. 1-(3-Cyano-3,5-dideoxy-β-D-*glycero***-pent-3-eno-furanosyl)uracil (9).** To a stirred solution of **4** (296 mg, 1.42 mmol) in dry THF (10 ml) was added 1.0 M toluene solution of Et₂AlCN (14 ml, 14 mmol) under argon atmosphere. The reaction mixture was refluxed for 2 h. EtOH was added and the mixture was stirred for 30 min and the solvent was evaporated in vacuo. The residue was diluted with water and extracted with AcOEt and the organic solution was dried over MgSO₄ and the solvent was evaporated in vacuo. The residue gel column chromatography (CHCl₃/MeOH 30:1) to afford **9** (193 mg, 58%). Residue was recrystallized from acetone and AcOEt (colorless solid).

Mp: 219–222 °C. MASS m/z (relative intensity): 235 (M⁺, 29%), 217 (M⁺ – H₂O, 9%), 192 (M⁺ – H₂O–CN+1, 6%), 123 (S⁺ – 1, 38%), 113 (B⁺ + 2, 100%). IR (KBr) cm⁻¹: 3426.6, 3196.6, 2218.0 (CN), 1697.9, 1671.7, 1653.5. ¹H NMR (DMSO- d_6) δ : 11.53 (1H, br s, N³–H), 7.44 (1H, d, J=7.8 Hz, 6-H), 6.66 (1H, d, J=7.3 Hz, 1'-H), 6.23 (1H, d, J=5.9 Hz, 2'-OH), 5.65 (1H, d, J=7.8 Hz, 5-H), 5.07–5.11 (1H, m, 2'-H), 2.15 (3H, s, 5'-CH₃). Anal. Calcd for C₁₀H₉N₃O₄ (M_w =235.20): C, 51.07; H, 3.86; N, 17.87. Found: C, 51.27; H, 3.93; N, 17.68.

4.1.11. 1-(3-O-Methyl-\alpha-L-xylo-furanosyl)uracil (11) and 1-(3-O-methyl-\beta-D-arabino-furanosyl)uracil (12). To a stirred solution of **6b** (30 mg, 0.125 mmol) in dry THF (10 ml) was added 1 M THF solution of BH₃–THF (0.625 ml, 0.625 mmol) at 0 °C under argon atmosphere and the reaction mixture was stirred for 1 h at 0 °C. One molar THF solution of BH₃–THF (1.625 ml, 1.625 mmol) was added to the reaction mixture and the reaction mixture was refluxed for 17 h. After cooling to room temperature, H₂O and 3 N NaOH (0.125 ml, 0.375 mmol) and 30% H₂O₂ (0.12 ml, 1.25 mmol) was added to the reaction mixture. The reaction mixture was stirred for 6 h at room temperature. After acidification with AcOH, the reaction mixture was washed with saturated NaCl and the aqueous layer was extracted with THF. The organic solution was combined and dried over Na₂SO₄. The solvent was removed in vacuo and the mixture was analyzed by the ¹H NMR (17% of the **12** was detected). The crude mixture was purified by silica gel column chromatography (CHCl₃/ MeOH 15:1) to give **11** (53%).

Compound **11**. FABMASS m/z (relative intensity): 259 (M⁺ +1, 7%), 154 (100%). ¹H NMR (DMSO- d_6) δ : 11.25 (1H, br s, N³–H), 7.41 (d, J=8.2 Hz, 6-H), 5.92 (1H, d, J= 3.6 Hz, 1'-H), 5.76 (1H, d, J=4.5 Hz, 2'-OH), 5.55 (1H, d, J=8.2 Hz, 5-H), 4.72 (1H, t, J=5.5 Hz, 5'-OH), 4.28–4.34 (1H, m, 3'-H), 4.21–4.25 (1H, m, 2'-H), 3.74–3.76 (1H, m, 4'-H), 3.48–3.63 (2H, m, 5'-H×2), 3.36 (3H, s, 3'-OCH₃). NOE experiment, 4'-H/3'-H (7.3%), 4'-H/2'-OH (1.2%), 3'-H/6-H (6.2%) and 1'-H/2'-H (6.4%). FABHRMS m/z calcd for C₁₀H₁₅N₂O₆: 259.0930. Found: 259.0938.

Compound **12**. ¹H NMR (DMSO- d_6) δ : 11.28 (1H, br s, N³–H), 7.60 (d, J=8.2 Hz, 6-H), 5.90 (1H, d, J=4.5 Hz, 1'-H), 5.76 (1H, d, J=5.1 Hz, 2'-OH), 5.57 (1H, d, J=8.2 Hz, 5-H), 5.08 (1H, t, J=5.7 Hz, 5'-OH), 4.12–4.16 (1H, m, 2'-H), 3.79–3.84 (1H, m, 3'-H), 3.66–3.68 (1H, m, 4'-H), 3.58–3.61 (2H, m, 5'-H×2), 3.34 (3H, s, 3'-OCH₃). FABHRMS *m*/*z* calcd for C₁₀H₁₅N₂O₆: 259.0930. Found: 259.0943.



References and notes

- Townsend, L. R. Chemistry of Nucleosides and Nucleotides; Plenum: New York, 1988. Matsuda, A. J. Synth. Org. Chem. Jpn. 1990, 48, 907. Perigaud, C.; Gosselin, G.; Imbach, J.-L. Nucleosides Nucleotides 1992, 11, 903. Maag, H.; Rydzewski, R. M.; McRobert, M. J.; Crawford-Ruth, D.; Verheyden, J. P. H.; Prisbe, E. J. J. Med. Chem. 1992, 35, 1440. Camarasa, M.-J.; Perez-Perez, M.-J.; San-Felix, A.; Balzarini, J.; De Clercq, E. J. Med. Chem. 1992, 35, 2721. Perez-Perez, M.-J.; San-Felix, A.; Balzarini, J.; De Clercq, E.; Camarasa, M.-J. J. Med. Chem. 1992, 35, 2988.
- Ruth, T. L. Oligonucleotides and Their Analogues; IRL: London, 1991. Giese, B.; Imwinkelried, P.; Petretta, M. Synlett 1994, 1003 and references therein.
- (a) Codington, J. F.; Fecher, R.; Fox, J. J. J. Am. Chem. Soc. 1960, 82, 2794. (b) Chang, N. C.; Burchannel, J. H.; Fecher, R.; Duschinsky, R.; Fox, J. J. J. Am. Chem. Soc. 1961, 83, 4060. (c) Codington, J. F.; Fecher, R.; Fox, J. J. J. Org. Chem. 1962, 27, 163.

- 4. (a) Doerr, I. L.; Codington, J. F.; Fox, J. J. J. Org. Chem. 1965, 30, 467. (b) Kowollik, G.; Langen, P. Z. Chem. 1975, 15, 147. (c) Hummel, C. F.; Carty, R. P. Nucleosides Nucleotides 1983, 2, 249. (d) Misra, H. K.; Gati, W. P.; Knaus, E. E.; Wiebe, L. I. J. Heterocycl. Chem. 1984, 21, 773. (e) Häbich, D.; Barth, W. Synthesis 1988, 943. (f) Wu, J.-C.; Pathak, T.; Tong, W.; Vial, J.-M.; Remaud, G.; Chattopadhyaya, J. Tetrahedron 1988, 44, 6705. (g) Herdewijn, P.; Van Aerschot, A.; Kerremans, L. Nucleosides Nucleotides 1989, 8, 65. (h) Perlman, M. E.; Watanabe, K. A. Nucleosides Nucleotides 1989, 8, 145. (i) Wu, J.-C.; Chattopadhyaya, J. Tetrahedron 1989, 45, 450. (j) Wigerinck, P.; Van Aerschot, A.; Janssen, G.; Claes, P.; Balzarini, J.; De Clercq, E.; Herdewijn, P. J. Med. Chem. 1990, 33, 868. (k) Haung, J.-T.; Chen, L.-C.; Wang, L.; Kim, M.-H.; Warshaw, J. A.; Armstrong, D.; Zhu, Q.-Y.; Chou, T.-C.; Watanabe, K. A.; Matulic-Adamic, J.; Su, T.-L.; Fox, J. J.; Polsky, B.; Baron, P. A.; Gold, J. W. M.; Hardy, W. D.; Zuckerman, E. J. Med. Chem. 1991, 34, 1640. (1) Haraguchi, K.; Tanaka, H.; Maeda, H.; Itoh, Y.; Saito, S.; Miyasaka, T. J. Org. Chem. 1991, 56, 5401. (m) Xi, Z.; Agback, P.; Plavec, J.; Sandström, A.; Chattopadhyaya, J. Tetrahedron 1992, 48, 349. (n) Ariza, X.; Garces, J.; Vilarrasa, J. Tetrahedron Lett. 1992, 33, 4069. (o) Minamoto, K.; Hamano, Y.; Matsuoka, Y.; Watanabe, K.; Hirota, T.; Eguchi, S. Nucleosides Nucleotides 1992, 11, 457. (p) Watanabe, K.; Yanagihara, K.; Minamoto, K.; Iwasaki, H. Nucleosides Nucleotides 1993, 12, 1061. (q) Johnson, R.; Joshi, B. V.; Reese, C. B. J. Chem. Soc., Chem. Commun. 1994, 133. (r) Bera, S.; Pathak, T.; Langley, G. J. Tetrahedron 1995, 51, 1459. (s) Roussev, C. D.; Simeonov, M. F.; Petkov, D. D. J. Org. Chem. 1997, 62, 5238. (t) Chen, B.-C.; Quinlan, S. L.; Reid, J. G.; Spector, R. H. Tetrahedron Lett. 1998, 39, 729. (u) Liu, X.; Reese, C. B. J. Chem. Soc., Perkin Trans. 1 2000, 2227. (v) Haraguchi, K.; Itoh, Y.; Takeda, S.; Honma, Y.; Tanaka, H.; Nitanda, T.; Baba, M.; Dutschman, G. E.; Cheng, Y.-C. Nucleosides Nucleotides Nucleic Acids 2004, 23, 647.
- (a) Horwitz, J. P.; Chua, J.; Da Rooge, M. A.; Noel, M.; Klundt, I. L. J. Org. Chem. 1966, 31, 205. (b) Sasaki, T.; Minamoto, K.; Hattori, K. Tetrahedron 1974, 30, 2689. (c) Reichman, U.; Hollenberg, D. H.; Chu, C. K.; Watanabe, K. A.; Fox, J. J. J. Org. Chem. 1976, 41, 2042. (d) Hollenberg, D. H.; Watanabe, K. A.; Fox, J. J. J. Med. Chem. 1977, 20, 113. (e) Mete, A.; Hobbs, J. B.; Scopes, D. I. C.; Newton, R. F. Tetrahedron Lett. 1985, 26, 97. (f) Ashwell, M.; Jones, A. S.; Walker, R. T. Nucleic Acids Res. 1987, 15, 2157. (g) Webb, T. R.; Mitsuya, H.; Broder, S. J. Med. Chem. 1988, 31, 1475. (h) Matsuda, A.; Satoh, M.; Nakashima, H.; Yamamoto, N.; Ueda, T. Heterocycles 1988, 27, 2545. (i) Herdewijn, P.; Balzarini, J.; Baba, M.; Pauwels, R.; Van Aerschot, A.; Janssen, G.; De Clercq, E. J. Med. Chem. 1988, 31, 2040. (j) Bamford, M. J.; Coe, P. L.; Walker, R. T. J. Med. Chem. 1990, 33, 2494.
- 6. Hirota, K.; Takasu, H.; Sajiki, H. Heterocycles 2000, 52, 1329.
- Hirota, K.; Takasu, H.; Tsuji, Y.; Sajiki, H. Chem. Commun. 1999, 1827.
- 8. Dalla, V.; Pale, P. Tetrahedron Lett. 1996, 37, 2781.
- 9. Marshall, J. A. Chem. Rev. 1989, 89, 1503.
- 10. Detailed results for the reaction mechanism will be reported in due course.