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Synthesis and Enantioselective Rearrangement of 4-Amino-substituted Cyclopentene Oxides

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Abstract: Several N-mono- and diprotected alkenes have been prepared and the stereoselectivity of their epoxidation has been investigated: N-monoprotected alkenes give *cis* epoxides preferentially (due to hydrogen bonding directed epoxidations) whereas N-diprotected alkenes produce *trans* epoxides exclusively (due to steric effects). Chiral lithium amide base-mediated rearrangement of a *cis*-monoprotected epoxide generated the corresponding amino-cyclopentenol in good yield and with an enantiomeric excess of 60%. © 1998 Elsevier Science Ltd. All rights reserved.

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The conversion of *meso* 4-substituted cyclopentene oxides into enantiomerically enriched allylic alcohols using chiral lithium amide bases has received considerable attention over the last few years.¹⁻³ Despite this, the enantioselective rearrangement of 4-amino-substituted epoxides *trans*- and *cis*-2 has not been reported. Amino alcohols 3 and 4, the products of such reactions, are useful intermediates for the synthesis of antiviral carbocyclic nucleoside analogues⁴ and 4-aminocyclopent-2-ene-1-one.⁵ Thus, we decided to investigate the stereoselective synthesis of each of the epoxides *trans*- and *cis*-2 and to attempt their chiral base-mediated rearrangement. Our results in both of these areas are described in this paper.



Amine hydrochloride salt 5 is a known compound that has been synthesised using (i) Curtius rearrangement of acid $6^{6,7}$ or (ii) Mitsunobu displacement (HN₃) of alcohol 7 with concomitant azide reduction^{8,9} or (iii) hydroboration-amination of cyclopentadiene.¹⁰ Each of these routes is unsatisfactory – either the route is many steps or low yielding or both. There are only a few reported examples of epoxides $2^{8,11-14}$ Of these, epoxide *cis*-9^{8,11} was prepared by a highly stereoselective amide-directed^{15,16} epoxidation of alkene 8 and epoxide *trans*-11 was synthesised by epoxidation of alkene 10 (*via* a sterically controlled

process). We envisaged making use of directed epoxidations on N-monoprotected amines 1 ($R^2 = H$) as a route to a range of epoxides *cis*-2 and, in order to further probe the effect of N-protecting groups on the stereoselectivity of epoxidation, we also wanted to prepare some N-diprotected amines 1.¹⁷

As a starting point, we developed a new approach to amine hydrochloride 5 which is as good if not better than previously published routes.⁶⁻¹⁰ Thus, as shown below, known¹⁸ alcohol 7 was converted into hydrochloride salt 5 (*via* mesylate 12 and volatile azide 13) in 69% yield over the three steps; for the first two steps, the reactions were worked-up but 12 and 13 were not purified. Standard *N*-monoprotection generated benzamide 8,¹¹ sulfonamides 14 and 15 and carbamates 16^{14} and 17^7 in good yields.¹⁹



In order to prepare representative N-diprotected alkenes, we decided to use a Mitsunobu approach with alcohol 7. Of the known methods for Mitsunobu reactions with nitrogen, 2^{0-22} we selected Weinreb's TsNHBoc reagent (18)^{20,23} as it generally gives high yielding Mitsunobu reactions. We also developed the novel Weinreb-Fukuyama hybrid reagent (19)^{20,21,23} as this would produce an alkene with orthogonal and easily removed N-protecting groups.²⁴ It was satisfying that combination of alcohol 7 with each of 18 and 19 under normal Mitsunobu conditions furnished good yields of N-diprotected alkenes 20 and 21.

With a range of N-mono- and diprotected alkenes in hand, we were ready to study the stereoselectivity of their epoxidation. All of the epoxidations were carried out under standard conditions (m-CPBA, NaHCO₃, CH₂Cl₂, room temperature, overnight) and the crude products were analysed by ¹H NMR spectroscopy to determine the stereoselectivity (Table). The major products of epoxidation of the N-monoprotected alkenes (Entries 1-5) were assigned as having *cis* stereochemistry by comparison with the known^{8,11} epoxidation of alkene **8**; our assignments are also consistent with a ¹H NMR spectroscopy correlation method.²⁵ In contrast, epoxidation of the N-diprotected alkenes (Entries 6 and 7) was completely *trans* selective. The *trans* selectivity was expected;¹³ it was established by synthesising N-diprotected epoxide *cis*-26 by Boc-protection (Et₃N, DMAP, Boc₂O, CH₂Cl₂) of the 84:16 mixture of N-monoprotected alkenes, steric factors result in *trans* selectivity. However, with N-monoprotected alkenes, hydrogen bonding to m-CPBA leads to *cis* selectivity; amides and sterically small sulfonamides ($R^1 = Ms$) and carbamates ($R^1 = Cbz$) give the largest proportions of *cis* epoxides. Notably, the use of an epoxidation system that *cannot* participate in hydrogen bonding [*in situ* generated methyl(trifluoromethyl)dioxirane²⁶] with N-monoprotected alkene **15** gave a 60:40 mixture of epoxides *trans-* and *cis-23*. To summarise the epoxidation results, we have found N-protecting groups that allow preparation of a *cis* epoxide (*eg cis-9*, 79% isolated yield) or a *trans* epoxide (*eg trans-26*, 88% isolated yield) in diastereomerically pure form.

Table: Stereoselective Epoxidation of N-Mono- and Diprotected 4-Amino-substituted Cyclopentenes



Entry	Alkene	\mathbb{R}^1	R ²	Epoxide	cis : trans ^a
1	8	Bz	Н	9	97 : 3 ^b
2	14	Ms	Н	22	98 : 2 ^c
3	15	Ts	Н	23	84 : 16
4	16	Cbz	Н	24	82: 18
5	17	Boc	н	25	75 : 25
6	20	Ts	Boc	26	2 : 98 ^{c,d}
7	21	Nse	Boc	27	2:98 ^{c,f}

^a The ratio of *cis* and *trans* epoxides was determined from the ¹H NMR spectrum of the crude product mixtures; ^b Epoxides *cis*-9 (79%) and *trans*-9 (3%) were isolated after chromatography; ^c Only one diastereosiomer was visible in the ¹H NMR spectrum of the crude product mixture; ^d Epoxide *trans*-26 (88%) was isolated after chromatography; ^e Ns = 2-NO₂C₆H₄SO₂-; ^f Epoxide *trans*-27 (84%) was isolated after chromatography.

Based on our^{27} experience with chiral base-mediated epoxide rearrangement reactions, we tried to rearrange *N*-diprotected epoxides *trans*-26 and *trans*-27 with two equivalents of Singh's²⁸ chiral lithium amide base *rac*-28. Unfortunately, no allylic alcohol could be detected in the ¹H NMR spectrum of the crude product

mixture and in the case of *trans*-27, a 65% yield of recovered starting epoxide was isolated. We had more success with the rearrangement of epoxide *cis*-9. In this case, three equivalents of Singh's chiral base (R)-28 were used (because of the presence of an acidic amide NH) and smooth rearrangement occurred to generate allylic alcohol (1*S*,4*R*)-29 in 73% yield and with 60% ee



(as shown by chiral HPLC). The absolute stereochemistry was established by formation of the Mosher's esters and analysis of the resulting ¹H NMR spectrum (Kakisawa's method).^{27,29} Thus, the sense of induction was the same as we had observed previously using chiral base (R)-28 with meso-cyclohexene oxides²⁷ and as Singh had observed using similar chiral bases with meso-cyclopentene oxides.²⁸

In summary, we have described methods for the stereoselective synthesis of the previously scarse epoxides trans- and cis-2 and have reported the first ever enantioselective rearrangement of a 4-aminosubstituted cyclopentene oxide. Our preliminary conclusion on the chiral base reaction is that a deprotonated amide cis to the epoxide is important for facilitating the rearrangement process.

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References and Notes

- O'Brien, P. J. Chem. Soc., Perkin Trans. 1 1998, 1439-1457. 1.
- 2.
- 3.
- Hodgson, D. M.; Gibbs, A. R.; Lee, G. P. Tetrahedron **1996**, 52, 14361-14384. Hodgson, D. M.; Gibbs, A. R. Tetrahedron Lett. **1997**, 38, 8907-8910. Crimmins, M. T. Tetrahedron **1998**, 54, 9229-9272; Zhang, D.; Miller, M. J. J. Org. Chem. **1998**, 63, 4. 755-759.
- 5. Ramesh, N. G.; Klunder, A. J. H.; Zwanenburg, B. Tetrahedron Lett. 1998, 39, 1429-1432.
- 6.
- Murdock, K. C.; Angier, R. B. J. Org. Chem. 1962, 27, 2395-2398. Hodgson, D. M.; Thompson, A. J.; Wadman, S. Tetrahedron Lett. 1998, 39, 3357-3358. 7.
- Elliott, R. D.; Rener, G. A.; Riordan, J. M.; Secrist, J. A.; Bennett, L. L.; Parker, W. B.; Montgomery, J. A. J. Med. Chem. 1994, 37, 739-744. 8.

- Fabiano, E.; Golding, B. T.; Sadeghi, M. M. Synthesis 1987, 190-192.
 Curran, D. P.; Gothe, S. A.; Choi, S-M. Heterocycles 1993, 35, 1371-1395.
 Patil, S. D.; Koga, M.; Schneller, S. W. J. Med. Chem. 1992, 35, 2191-2195; Patil, S. D.; Koga, M.; Schneller, S. W. Tetrahedron Lett. 1990, 31, 5861-5864. 12. Murdock, K. C.; Angier, R. B. J. Am. Chem. Soc. 1962, 84, 3748-3758.
- 13. Legraverend, M.; Huel, C.; Guilhem, J. Bisagni, E. Carbohydr. Res. 1992, 228, 21-27.

- Lai, Y-S.; Mendoza, J. S.; Jagdmann, G. E.; Menaldino, D. S.; Biggers, C. K.; Heerding, J. M.; Wilson, J. W.; Hall, S. E.; Jiang, J. B.; Janzeng, W. P.; Ballas, L. M. J. Med. Chem. 1997, 40, 226-235.
 For a review, see: Hoveyda, A. H.; Evans, D. A.; Fu, G. C. Chem. Rev. 1993, 93, 1307-1370.
 P. Kocovsky and I. Stary, J. Org. Chem. 1990, 55, 3236-3243; Jenmalm, A.; Berts, W.; Luthman, K.; Oriert, M. J. Med. Med. 1020 Csöregh, I.; Hacksell, U. J. Org. Chem. 1995, 60, 1026-1032.
- 17. To the best of our knowledge, only three diprotected amines 1 have previously been prepared and in two cases the nitrogen was part of a heterocyclic base. See: Zhou, J.; Bouhadir, K.; Webb, T. R.; Shevlin, P. B. Tetrahedron Lett. 1997, 38, 4037-4038 and references 12 and 13 above.
- 18. Crandall, J. K.; Banks, D. B.; Colyer, R. A.; Watkins, R. J.; Arrington, J. P. J. Org. Chem. 1968, 33, 423-425
- 19. All novel compounds gave satisfactory microanalysis and/or high resolution mass spectrometry data.
- 20. Henry, J. R.; Marcin, L. R.; McIntosh, M. C.; Scola, P. M.; Harris, G. D.; Weinreb, S. M. Tetrahedron Lett. 1989, 30, 5709-5712.
- 21. Fukuyama, T.; Jow, C-K.; Cheung, M. Tetrahedron Lett. 1995, 36, 6373-6374; Fukuyama, T.; Cheung, M.; Jow, C-K.; Hidai, Y.; Kan, T. Tetrahedron Lett. 1997, 38, 5831-5834.
- 22. Bell, K. E.; Knight, D. W.; Gravestock, M. B. Tetrahedron Lett. 1995, 36, 8681-8684; Decicco, C. P.; Grover, P. Synlett 1997, 529-530; Campbell, J. A.; Hart, D. J. J. Org. Chem. 1993, 58, 2900-2903.
- N-Boc protected sulfonamides 18 (93% yield) and 19 (86% yield) were prepared from the corresponding sulfonamides according to a literature procedure: Neustadt, B. R. *Tetrahedron Lett.* 1994, 35, 379-380.
 Deprotection of either protecting group in N-diprotected alkene 21 was easy. Boc deprotection using TFA
- or sulfonamide deprotection using mercaptoacetic acid (according to Fukuyama's procedure reported in reference 18) gave the N-monoprotected alkenes (each in quantitative yield).
- 25. Previous studies on epoxide 9 (and a related compound) indicated that, in the ¹H NMR spectra, the $\delta_{\rm H}$ (CHO) signal for *cis* epoxides was more downfield than that for *trans* epoxides. We have used this correlation to support our assignments of epoxide stereochemistry for N-monoprotected epoxides $-\delta_{\rm H}$ (CHO) values: 3.57 for cis-9, 3.54 for trans-9; 3.49 for cis-23, 3.40 for trans-23; 3.47 for cis-24, 3.40 for trans-24; 3.58 for cis-25, 3.54 for trans-25. However, this correlation did not apply to N-diprotected epoxides $-\delta_{\rm H}$ (CHO) values: 3.47 for *cis*-26, 3.57 for *trans*-26. 26. Yang, D.; Wong, M-K.; Yip, Y-C. J. Org. Chem. 1995, 60, 3887-3889. 27. O'Brien, P.; Poumellec, P. Tetrahedron Lett. 1996, 37, 8057-8058; O'Brien, P.; Poumellec, P. J. Chem.
- Soc., Perkin Trans. 1 1998, 2435-2441.
- 28. Bhuniya, D.; DattaGupta, A.; Singh, V. K. J. Org. Chem. 1996, 61, 6108-6113; Saravanan, P.; Singh, V. K. Tetrahedron Lett. 1998, 39, 167-170.
- 29. Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. J. Am. Chem. Soc. 1991, 113, 4092-4096.