A LIGNAN FROM ACTINODAPHNE LONGIFOLIA

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Abstract—A new lignan, actifolin, was isolated from the leaves of *Actinodaphne longifolia* and the structure was established on the basis of chemical and spectroscopic evidence.

INTRODUCTION

Previously, Hayashi and co-workers [1] reported the isolation of two furanosesquiterpenes (sesquirosefuran and longifolin) from the leaves of *Actinodaphne longifolia* (Blume) Nakai (Lauraceae), which grows in the southern part of Japan. In the course of our further investigation of the fresh leaves of this plant, three lignans (1, 8, and 9) were isolated and one of them was a new compound. We now wish to describe the isolation and characterization of this compound (1).

RESULTS AND DISCUSSION

The new compound, actifolin (1), $C_{22}H_{24}O_7$, gave a brown colour with ethanolic ferric chloride and exhibited the presence of hydroxy group (3550 cm^{-1}) and ester group (1735 cm⁻¹) in the IR spectrum. Acetylation of actifolin with acetic anhydride and pyridine gave a diacetate (2). The ¹H NMR spectrum of actifolin displayed the presence of six aromatic protons in a complicated pattern at δ 6.64–6.85 (6H, m), a methylenedioxy group at δ 5.94 (2H, s), a hydroxy group at δ 5.51 (1H, s), a methoxy group at δ 3.88 (3H, s), and an acetyl group at δ 2.03 (3H, s). The remaining nine protons were found in the region between $\delta 4.8$ and 2.5 (see Experimental). In the ¹³C NMR spectra, all the signals for compound 1 were very similar to those of piperitol (8), isolated from the same plant, except for the signals of C-8, C-9, C-7', and C-8' (Table 1). These data suggested that actifolin would be a seco-derivative of 8. As the mass spectrum of actifolin showed the greater abundance of the fragmentation peak at m/z 137, the structure of this compound must be 1 and therefore the alternative structure 4 was ruled out [2].

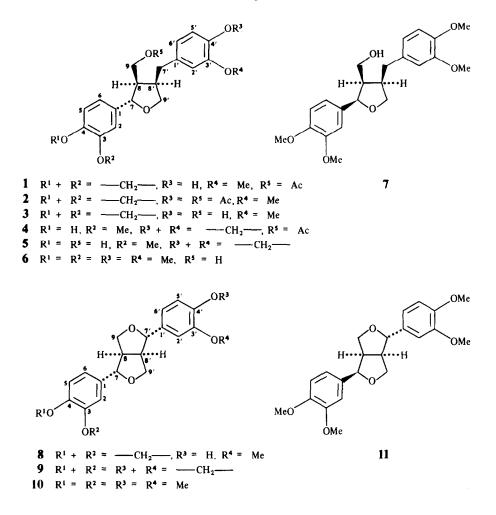
The stereochemistry of 1 was determined as follows. Two compounds, 6 (*trans*-orientation between 7-H and 8-H) and 7 (*cis*-orientation between 7-H and 8-H), were prepared by hydrogenolysis of eudesmin (10) [3] and epieudesmin (11) [3], respectively. In the ¹H NMR spectra, the signal of 7-H in 6 (*trans*-isomer) was observed at δ 4.82, against that in 7 (*cis*-isomer) at δ 5.13. The corresponding signal in the spectrum of 1 was observed at δ 4.76, which agrees well with the assignment of the *trans*-orientation between 7-H and 8-H.

In order to confirm the structure of 1, a chemical study was also carried out. Hydrolysis of 1 afforded an alcohol (3). Hydrogenolysis of piperitol (8), with known stereochemistry, gave a mixture of alcohols, 3 and 5. Separation of the mixture by HPLC afforded 3, which was identical with the hydrolysis product of 1. Thus, the structure of actifolin is established to be 1.

In addition to the new compound, actifolin (1), two known lignans piperitol (8) [4] and sesamin (9) were isolated and identified from this plant. This is the first report of lignans isolated from the genus *Actinodaphne*.

С	1	8
1	136.5 s	135.1 s
2	106.2 d	106.5 d
3	147.8 s	148.0 s
4	146.5 s	146.8 s
5	108.1 d	108.2 d
6	119.2 d	119.3 d
7	83.1 d	85.9 d
8	49.1 d	54.3 a
)	62.7 t	71.7 t
ľ	131.8 s	132.9 s
2'	111.2 d	108.7 a
3'	147.0 s	147.1 s
1′	144.1 s	145.3 s
5'	114.5 d	114.4 d
6'	121.1 d	119.0 d
7'	33.2 t	85.9 a
R'	42.5 d	54:2 a
θ΄	72.8 t	71.7 t
OMe	55.9 q	56.0 g
OCH ₂ O	101.0 t	101.1 t
C = O	170.9 s	
Me	20.1 q	

Table. 1. ¹³C NMR spectral data of compounds 1 and 8



EXPERIMENTAL

CC was run on Merck silica gel 60 (230–400 mesh) and florisil (100–200 mesh). TLC was performed on glass plates precoated with Kieselgel 60 F_{254} (Merck). ¹H NMR (270 MHz) and ¹³C NMR (25 MHz) spectra were determined in CDCl₃. Chemical shifts are in ppm (δ). HPLC was conducted on a Develosil pack (ODS-10) column (20 × 250 mm) with UV detector (254 nm).

Extraction and separation of compounds. The MeOH extract of the fresh leaves (6.0 kg) of Actinodaphne longifolia collected in Kagoshima prefecture in August 1987 was divided into the *n*hexane soluble (150 g) and CHCl₃ soluble fractions (32 g). The *n*hexane soluble fraction was chromatographed on florisil with C_6H_6 as an eluent to give sesquirosefuran (1.3 g) and longifolin (7.7 g). The CHCl₃ soluble fraction was chromatographed on florisil. Elution with CHCl₃ afforded an oil (3.7 g), a part of which (0.7 g) was chromatographed on silica gel to yield sesamin (9) (8 mg). Further elution with CHCl₃-Me₂CO (9:1) afforded an oil (1.8 g) which was subjected by CC on silica gel (CHCl₃-Me₂CO, 19:1) and successively by prep. TLC (CHCl₃-Me₂CO, 9:1) providing piperitol (8) (37 mg) and actifolin (1) (24 mg).

Actifolin (1). Colourless oil. $[\alpha]_D + 13.3^{\circ}$ (CHCl₃; c 0.48). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3550, 1735, 1610, 1515. UV $\lambda_{\text{max}}^{\text{EIOH}}$ nm: 231. MS m/z: 400 [M]⁺, 357, 340, 217, 203, 190, 176, 164, 149, 137. HRMS m/z: 400.1516 (M⁺, calcd for C₂₂H₂₄O₇: 400.1522). ¹H NMR (CDCl₃): δ 2.03 (3H, s, COMe), 2.51 (1H, dd, J = 13.1, 9.8 Hz, 7'-H), 2.51 (1H, d, J = 6.7 Hz, 8-H), 2.67–2.76 (1H, m, 8'-H), 2.82 (1H, dd, J = 13.1, 5.0 Hz, 7'-H), 3.72 (1H, dd, J = 8.7, 6.7 Hz, 9'-H), 3.88 (3H, s, OMe), 4.05 (1H, dd, J = 8.7, 6.7 Hz, 9'-H), 4.16 (1H, dd, J = 11.4, 7.7 Hz, 9-H), 4.34 (1H, dd, J = 11.4, 7.1 Hz, 9-H), 4.76 (1H, d, J = 6.1 Hz, 7-H) 5.51 (1H, s, OH), 5.94 (2H, s, OCH₂O), 6.64–6.85 (6H, m, arom. H).

Acetylation of actifolin. A mixture of 1 (2 mg), Ac₂O (0.5 ml), and pyridine (0.5 ml) was stirred overnight at room temp. and worked up in the usual way to afford a colourless oil (2, 2 mg). $[\alpha]_{D} + 23.0^{\circ}$ (CHCl₃; c 0.10). IR $v_{mxx}^{CHCl_3}$ cm⁻¹: 1765, 1740, 1605, 1510. UV λ_{mxx}^{EiOH} nm: 279, 227. MS m/z: 442 [M]⁺, 400, 382, 340, 203, 149, 137. HRMS m/z: 442.1647 (M⁺, calcd for C₂₄H₂₆O₈: 442.1626). ¹H NMR (CDCl₃): δ 2.03 (3H, s, COMe), 2.31 (3H, s, COMe), 2.52 (1H, m, 8-H), 2.56 (1H, dd, J = 13.1, 10.4 Hz, 7'-H), 2.68–2.81 (1H, m, 8'-H), 2.87 (1H, dd, 'J = 13.1, 5.0 Hz, 7'-H), 3.74 (1H, dd, J = 8.4, 6.4 Hz, 9'-H), 3.82 (3H, s, OMe), 4.08 (1H, dd, J = 8.4, 6.4 Hz, 9'-H), 4.17 (1H, dd, J = 11.4, 7.4 Hz, 9-H), 4.34 (1H, dd, J = 11.4, 7.4 Hz, 9-H), 4.77 (1H, d, J = 6.4 Hz, 7-H), 5.95 (2H, s, OCH₂O), 6.72–6.97 (6H, m, arom. H).

Hydrogenolysis of eudesmin. A suspension of eudesmin (10) (12 mg) and 5% Pd-C (12 mg) in MeOH (1 ml) was stirred under a hydrogen atmosphere at room temp. for 7.5 hr. The reaction mixture was filtered and the filtrate was evapd. The residue was purified by prep. TLC (CHCl₃-Me₂CO, 9:1) to afford a colourless oil (6, 3.1 mg). $[\alpha]_{\rm D}$ + 14.7° (CHCl₃; c 0.08). IR $v_{\rm CHCl_3}^{\rm CHCl_3}$ cm⁻¹: 1515, 1465. MS m/z: 388 [M]⁺, 151. ¹H NMR (CDCl₃): δ 2.43

(1H, quint., J = 6.7 Hz, 8-H), 2.58 (1H, dd, J = 13.5, 10.6 Hz, 7'-H), 2.72–2.80 (1H, m, 8'-H), 2.94 (1H, dd, J = 13.5, 5.2 Hz, 7'-H), 3.76 (1H, dd, J = 8.7, 6.7 Hz, 9'-H), 3.80–3.89 (1H, dd, J = 10.6, 6.7 Hz, 9-H), 3.86, 3.87, 3.89 (12H, $3 \times s$, $4 \times OMe$), 3.94 (1H, dd, J = 10.6, 6.7 Hz, 9-H), 4.07 (1H, dd, J = 8.7, 6.4 Hz, 9'-H), 4.82 (1H, d, J = 6.7 Hz, 7-H), 6.71–6.89 (6H, m, arom. H).

Hydrogenolysis of epieudesmin. Epieudesmin (11) (11 mg) in MeOH (1 ml) and 5% Pd-C (11 mg) was hydrogenated at room temp. for 13.5 hr according to the method described above. Purification of the reaction product by prep. TLC (CHCl₃-Me₂CO, 9:1)afforded a colourless oil (7, 1.7 mg). $[\alpha]_D$ + 45.7° (CHCl₃, c 0.04). IR v^{CHCl3}_{max} cm⁻¹: 1515, 1465. MS m/z: 388 [M]⁺, 151. ¹H NMR (CDCl₃): δ 2.40–2.48 (1H, m, 8-H), 2.78 (1H, dd, J = 11.1, 4.6 Hz, 7'-H), 2.90–2.99 (2H, m, 7'-H and 8'-H), 3.54 (1H, dd, J = 12.1, 4.0 Hz, 9-H), 3.63 (1H, dd, J = 12.1, 5.4 Hz, 9-H), 3.87, 3.89 (12H, 2 × s, 4 × OMe), 3.83–3.91 (1H, m, 9'-H), 4.03 (1H, t, J = 8.1 Hz, 9'-H), 5.13 (1H, d, J = 5.4 Hz, 7-H), 6.76–6.95 (6H, m, arom. H).

Hydrolysis of actifolin. A soln of 1 (2.3 mg) and 3% aq NaOH (3 drops) in MeOH (0.5 ml) was stirred at room temp. for 10 min. After removal of solvent, the residue was diluted with H₂O, neutralized with dil. HCl, and extracted with CHCl₃. The organic layer was dried over Na₂SO₄ and evapd to dryness to give a colourless oil (3, 2.0 mg). $[\alpha]_D + 11.1^{\circ}$ (CHCl₃, c 0.09). IR $\nu_{max}^{CHCl_3}$ cm⁻¹: 3530, 1610, 1510. UV λ_{max}^{EuOH} nm: 283, 230. MS m/z: 358 [M]⁺, 137. HRMS m/z: 358.1389 (M⁺, calcd for C₂₀H₂₂O₆: 358.1415). ¹H NMR (CDCl₃): δ 2.37 (1H, quint., J = 6.7 Hz, 8-H), 2.54 (1H, dd, J = 13.1, 10.7 Hz, 7'-H), 2.72 (1H, m, 8'-H), 2.90 (1H, dd, J = 13.1, 5.0 Hz, 7'-H), 3.75 (1H, dd, J = 8.7, 6.1 Hz, 9'-H), 3.77 (1H, dd, J = 10.7, 6.7 Hz, 9-H), 4.05 (1H, dd, J = 8.7, 0Me), 3.92 (1H, dd, J = 10.7, 6.7 Hz, 9-H), 4.05 (1H, dd, J = 8.7, 0Me), 3.92 (1H, dd, J = 10.7, 6.7 Hz, 9-H), 4.05 (1H, dd, J = 8.7)

6.7 Hz, 9'-H), 4.78 (1H, d, J = 6.7 Hz, 7-H), 5.49 (1H, s, OH), 5.95 (2H, s, OCH₂O), 6.68–6.86 (6H, m, arom. H).

Hydrogenolysis of piperitol. Piperitol (8) (10 mg) in MeOH (1 ml) and 5% Pd-C (10 mg) was hydrogenated at room temp. for 30 hr as already described above. Treatment of the reaction mixture in the usual way afforded a mixture of alcohols (3 and 5). The mixture was separated by HPLC (MeOH-H₂O, 1:1; flow rate 9.0 ml) to give 3 (0.6 mg) at R_t 31 min. and 5 (1.5 mg) at R_t 37 min. The alcohol (3) was identical with the product obtained by hydrolysis of 1. Compound 5 was a colourless oil. $[\alpha]_{D} + 6.0^{\circ}$ (CHCl₃; c 0.06). IR $\nu_{max}^{CHCl_3}$ cm⁻¹: 3530, 1600, 1510, 1500. UV λ_{max}^{EtOH} nm: 283, 230. MS m/z: 358 [M]⁺, 135. HRMS m/z: 358.1427 (M⁺, calcd for $C_{20}H_{22}O_6$: 358.1415). ¹H NMR $(CDCl_3)$: δ 2.40 (1H, quint., J = 6.7 Hz, 8-H), 2.55 (1H, dd, J = 13.4, 10.4 Hz, 7'-H), 2.65-2.76 (1H, m, 8'-H), 2.90 (1H, dd, J = 13.4, 5.0 Hz, 7'-H), 3.72 (1H, dd, J = 8.7, 6.4 Hz, 9'-H), 3.77 (1H, dd, J = 8dd, J = 10.7, 6.7 Hz, 9-H), 3.90 (3H, s, OMe), 3.91 (1H, dd, J = 10.7, 6.7 Hz, 9-H), 4.06 (1H, dd, J = 8.7, 6.4 Hz, 9'-H), 4.79 (1H, d, J = 6.7 Hz, 7-H), 5.51 (1H, br s, OH), 5.93 (2H, s, OCH₂O), 6.62-6.89 (6H, m, arom. H).

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