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Synthesis of Novel Chiral Semicrown Ether-Like Bis(oxazoline) Ligands and Application in Enantioselective Cyclopropanation of Styrene

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Synthesis of Novel Chiral Semicrown Ether-Like Bis(oxazoline) Ligands and Application in Enantioselective Cyclopropanation of Styrene

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Abstract: A series of chiral C_2 -symmetric bis(oxazoline) ligands containing semicrown ether unit were synthesized. The copper complexes prepared in situ from copper(I)-triflate, and the new chiral oxazoline ligands were assessed as chiral catalysts in the enantioselective cyclopropanation of styrene with diazoacetates. Enantioselectivities up to 84% and up to 65%, respectively, for *trans*- and *cis*-2-phenylcyclopropanecarboxylate were observed.

Keywords: Bis(oxazoline), synthesis, enantioselective cyclopropanation, styrene, asymmetric catalysis

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INTRODUCTION

Cyclopropane is found as a basic structural unit in a wide variety of synthetic and naturally occurring compounds^[1-5] A number of catalyst systems have been developed for the asymmetric cyclopropanation of olefins.^[6-10] Many of these are based on the decomposition of diazo compounds, and the corresponding reaction mechanism is believed to proceed through metal carbenoid intermediate. These systems include copper complexes of semicorrin, bis(oxazoline), chiral Schiff base and bipyridine as well as rhodium carboxamides and porphyrins, ruthenium pybox, and cobalt and ruthenium salen complexes. In recent years, C₂-symmetric bis(oxazoline) are versatile chiral ligands, and their metal complexes have been recognized as an effective class of chiral catalyst in a variety of transition metal catalyzed asymmetric reactions.^[11-16] Thus, the design and synthesis of new chiral bis(oxazoline) ligands has inspired many chemists to work with great efforts. The search for chiral ligands with broad applicability is a particularly interesting aspect in the field of chemical research on enantioselective reactions promoted by chiral catalysts. Continuing on our ongoing project on the synthesis, structure and catalytic enantioselective reaction of novel chiral bis(oxazoline) ligands, [17-23] in this article, we report the synthesis and the primary catalytic enantioselective cyclopropanation of styrene by copper(I) complex of a series of new bis(oxazoline) ligands 4 and 5 containing semicrown ether unit. Crown ether has inclusion ability to a variety of ions and neutral molecules; we expect the combination of semicrown ether and chiral oxazoline unit in new ligands may result in unique characteristics for catalytic reaction.

RESULTS AND DISCUSSION

Many methods have been reported for the synthesis of oxazoline ligands.^[24–27] The usual method for preparation of chiral bis(oxazoline) is the reaction of diacid derivatives with chiral β -amino alcohols. The enantiomerically pure β -amino alcohols are either commercially available or easily obtained by reduction of the corresponding α -amino acids.^[28–31] The diacid **1** were synthesized according to literature procedure from corresponding dibromide and methyl salicylate.^[32] The diacid **1** was refluxed with thionyl chloride to afford the diacyl chloride. The diacyl chloride was treated with β -amino alcohols and triethylamine to afford the corresponding chiral intermediate dihydroxy diamides **2** and **3** in 60–97% yields. The procedure for synthesize the desired chiral bis(oxazoline) ligands **4** and **5**. The dihydroxy diamides were treated with methanesulfonyl chloride (MsCl) (2.3 eq.) and Et₃N (5 eq.) in dichloromethane to afford the corresponding bismesylates,



Scheme 1. Synthesis of new bis(oxazoline) 4 and 5.

which were treated with an aqueous methanolic solution of NaOH to furnish the bis(oxazoline) **4** and **5** in good yields (53-85%) (Scheme 1). The structures of these new chiral bis(oxazoline) ligands were characterized by ¹H NMR, MS, IR, and elemental analysis or HRMS.

With the new ligands in hand, we try to evaluate the efficiency of these bis(oxazoline) ligands in the copper (I)-catalyzed asymmetric cyclopropanation of olefins. We first carried out the model reaction of styrene with ethyl diazoacetate to give the *trans-* and *cis-*cyclopropanes **7** and **8** (Scheme 2). The asymmetric cyclopropanation was carried out in dichloromethane in the presence of 2% mol of copper(I) catalyst generated in situ by mixing Cu(OTf).1/2C₆H₆ and the bis(oxazoline). The reaction was first performed at room temperature by slow addition of ethyl diazoacetate to a solution of styrene in dichloromethane containing the copper (I)-bis(oxazoline) catalyst. The ratio of *trans-* and *cis-*isomers were determined by ¹H NMR analysis,



Scheme 2. Cyclopropanation reaction of styrene with diazoacetates.

Entry	Ligand	Diazoacetate	Temp. (°C)	$\operatorname{Yield}^{a}(\%)$	$Trans/cis^b$	% ee ^c (trans)	Config. ^d	$\% ee^c (cis)$	Config. ^d
1	4a	6a	rt	74	70:30	24	1 <i>S</i> , 2 <i>S</i>	32	1S, 2R
2	4b	6a	ц	68	75:25	7	1R, 2R	10	1 <i>R</i> , 2 <i>S</i>
З	4c	6a	ц	71	77:23	32	1R, 2R	21	1 <i>R</i> , 2 <i>S</i>
4	4d	6a	ц	64	76:24	19	1R, 2R	14	1 <i>R</i> , 2 <i>S</i>
5	5a	6a	ц	83	75:25	45	1S, 2S	43	1 <i>S</i> , 2 <i>R</i>
9	5b	6a	ц	63	60:40	8	1R, 2R	S	1 <i>R</i> , 2 <i>S</i>
7	5c	6a	ц	85	74:26	35	1R, 2R	20	1 <i>R</i> , 2 <i>S</i>
8	5d	6a	ц	76	77:23	34	1R, 2R	22	1 <i>R</i> , 2 <i>S</i>
6	5a	6a	0	79	76:24	68	1S, 2S	61	1 <i>S</i> , 2 <i>R</i>
10	5c	6a	0	78	78:22	56	1R, 2R	41	1 <i>R</i> , 2 <i>S</i>
11	5a	6b	0	80	82:18	84	1S, 2S	65	1S, 2R
^a Isola	ted yield, ba	sed on ethyl diazc	pacetate for the r	nixture of trans	s and cis produc	ts.			

Table 1. Asymmetric cyclopropanation of styrene with diazoacetate catalyzed by copper(I)-bis(oxazoline)

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^bDetermined by ¹H NMR of the crude product. ^cDetermined by chiral HPLC using Chiralcel OI column. ^dBased on the sign of optical rotation.^[34] and the enantiomeric excesses were determined by HPLC analysis using a chiral column (Chiracel OJ). The results are summarized in Table 1.

Among the bis(oxazoline) ligands tested, 5a containing phenyl substituent on its oxazolines gave the highest enantiomeric excess for trans-7 in 45% ee and cis-8 in 43% ee (entry 5), the ligands 4b and 5b containing benzyl group afforded comparably lower enantioselectivity (trans-7% ee and 8% ee, respectively, entries 2 and 6). In addition, the ligands 4c and 5c with iso-butyl substituent afford moderate enantioselectivity for cyclopropanation in 32% ee and 35% ee (entries 3 and 7). It is of interest that the phenyl substituent on the oxazoline **5a** provides the best enantiomeric excess of trans-7 and cis-8. However, in the same reaction, the introduction of a phenyl substituent at the 4-position of the oxazoline has been previouly reported to have a pronounced effect on enantioselectivity.^[17] This phenomenon may be ascribed to the flexibility of 5a, resulting in little torsional strain between the two oxazoline units in corresponding copper(I) complex. Meanwhile, we examined the effect of temperature for cyclopropanation of the ligands 5a and 5c as shown in Table 1. The reactions were carried out under the same condition as above, except addition of diazoacetate for 4 hr and the reaction was proceeding for 18 hr at 0° C, the ee value of copper(I)catalyzed cyclopropanation rise to 68% and 56% ee, respectively (entries 9 and 10). As expected, when the steric bulkier diazoacetate ester t-butyl diazoacetate was used instead of ethyl diazoacetate, the enantioselectivity was improved. The reaction of t-butyl diazoacetate with styrene in the presence of Cu(I)-ligand 5a catalyst gave trans-t-butyl cyclopropanecarboxylates up to 84% ee and *cis* isomer in 65% ee (entry 8).

The following conclusion can also be made from the results gathered in Table 1. (1) The diastereoselectivity of the reaction is in favor of *trans*-selectivity. Preference for the *trans*-isomer has been previously observed for the same reaction using alternative bis(oxazoline) ligands.^[35-41] (2) The enantioselectivity of *cis*-isomer is almost the same level as that of *trans*-isomer. (3) The enantioselectivity was improved when temperature was reduced with a slight lower chemical yield (entries 9 and 10 *vs* entries 5 and 7). (4) The enantioselectivity of **5a-d** was better than **4a-d**. (5) The predominant formation of (1*R*, 2*R*)-*trans*-cyclopropane **7** and (1*R*, 2*S*)-*cis*-cyclopropane **8** were observed irrespective of the reaction temperature (room temperature or 0°C) with S-bis(oxazoline). As expected, the preferred absolute configurations of *trans*-cyclopropane **7** and *cis*-cyclopropane **8** were antipodal when use *R*-bis(oxazoline) **4a** and **5a**.

Although the enantioselectivity is moderate for these new ligands, one advantage is that the moderate enantioselectivity can be obtained by using the very cheap ligand **5a** or **5c** synthesized from cheap β -amino alcohol (*R*-phenylglycinol or *S*-leucinol). In contrast, very expensive ligands derivated from *L*-*t*-leucinol were usually used in literature.^[42-44] Furthermore, we obtained moderate enantioselectivity of the *cis*-isomer **8** and did

not need to use the bulky adamantyl substituted bis(oxazoline). There still exists potential to optimize the enantioselectivity; the enantioselectivity of the metal catalyzed cyclopropanation is usually higher when the steric bulky ester is used.^[45–47] It seem that the combination of semicrown ether and chiral oxazoline unit in these new ligands may result in many coordination sites with copper for formation of suitable chiral complexes, which may explain the moderate enantioselectivity obtained.

In conclusion, an efficient synthetic procedure was developed for synthesis of new chiral semicrown ether-like bis(oxazoline) ligands. Preliminary results in asymmetric cyclopropanation of styrene with ethyl diazoacetate have been obtained. The C_2 -symmetric bis(oxazoline) **4** and **5** derived from *m*-phenylenebis(methyleneoxy) and 2,6-pyridinediylbis(methyleneoxy) units present different reactivities in the asymmetric copper-catalyzed cyclopropanation. The highest enantioselectivity was obtained with phenyl substituted bis(oxazoline) **5a**. Although the *trans*-enantioselectivity is moderate, these primary results suggest that these novel semicrown ether-like bis(oxazolines) may have potential to become catalyst for other asymmetric reactions. The semicrown ether-like chiral bis(oxazolines) have many coordination sites may be suitable for chiral complexes formation with other metals such as zinc, cobalt, and rhodium. Further studies in other asymmetric reactions are in progress in our laboratory.

EXPERIMENTAL

Melting points were measured on an XT-4 melting point apparatus and were uncorrected. ¹H NMR and ¹³C NMR spectra were recorded on a Mercury 300 MHz or 200 MHz spectrometer with tetramethylsilane (TMS) serving as internal standard. Infrared spectra were obtained on a Bruker Vector 22 spectrometer. Mass spectra were obtained on a VG-ZAB-HS mass spectrometer. Optical rotations were measured on a Perkin–Elmer 241 MC spectrometer. Elemental analyses were carried out on an Elementar Vario EL instrument. Solvents used were purified and dried by standard procedures. 2,2'-[1,3-phenylenebis(methyleneoxy)]-bisbenzoic acid **1a** and 2,2'-[2,6-pyridinediylbis(methyleneoxy)]-bisbenzoic acid **1b** were synthesized according to literature procedure (Elwahy and Abbas, 2000).

General Procedure A

1,3-Bis[2-[N-(2-hydroxy-(1R)-1-phenylethyl)carbamoyl] phenoxymethyl]benzene (**2a**)

A solution of diacid **1a** (3.0 g, 8 mmol), $SOCl_2$ (50 mL) and 3 drops of DMF was refluxed for 12 hr. The excess $SOCl_2$ was removed under reduced

pressure, and then anhydrous benzene (20 mL) was added and the solvent was removed again to dryness to remove the trace of SOCl₂ and afford the diacyl dichloride. The above diacyl dichloride in CH2Cl2 (150 mL) was added dropwise to a solution of R-phenylglycinol (2.19 g, 16 mmol) and Et₃N (10 mL) in CH₂Cl₂ (50 mL) at 0°C and stirred at room temperature for 24 hr. The reaction mixture was washed with 2 N HCl, 10% NaHCO₃, and H₂O. The organic layer was dried over anhydrous Na₂SO₄ and concentrated to give crude product. Purification by silica gel column chromatography (petroleum ether-ethyl acetate-methanol 5:3:1) afforded the dihydroxy diamide **2a** as colorless solid (3.92 g, 79.5%). m.p. 78-80°C. $[\alpha]_{D}^{20} =$ $+91.7^{\circ}$ (c = 0.38, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 3.07 (2H, s, OH), 3.54-3.68 (4H, m, CH₂), 5.09 (4H, s, OCH₂), 5.16-5.22 (2H, m, CH), 6.99-7.14 (14H, m, ArH), 7.37-7.45 (5H, m, ArH), 7.60 (1H, s, ArH), 8.14–8.17 (2H, s, ArH), 8.54 (2H, d, J = 7.8 Hz, NH). ¹³C NMR (CDCl₃): δ 55.64, 66.29, 70.84, 112.37, 121.40, 121.59, 126.49, 127.27, 128.25, 128.46, 128.65, 129.27, 132.33, 132.93, 136.15, 139.19, 156.62, 165.29. IR(KBr): v 3384, 3061, 3030, 2932, 1708,1638, 1599, 1530, 1482, 1450, 1384, 1300, 1223, 1162, 1105, 996, 755, 700 cm⁻¹. ESIMS: m/z 617 (M+H, 48), 639 (M+Na, 100), 360 (20), 223 (18). Anal. Calcd. for C₃₈H₃₆N₂O₆ · 0.5H₂O: C, 72.94; H, 5.91; N, 4.47. Found: C, 72.89; H, 5.64; N. 4.35.

1,3-Bis[2-[N-((1S)-1-benzyl-2-hydroxyethyl)carbamoyl] phenoxymethyl]benzene (2b)

Following general procedure A, from diacid **1a** (2.64 g, 7 mmol), SOCl₂ (45 mL), *S*-Phenylalanol (2.12 g, 14 mmol) and Et₃N (10 mL) to give a colorless solid **2b** in 71.8% yield (3.24 g). m.p. 59–60°C. $[\alpha]_D^{20} = -63.5^{\circ}$ (c = 0.38, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 2.72 (4H, d, *J* = 7.2 Hz, CH₂), 3.35–3.59 (6H, m, CH₂ + OH), 4.27–4.31 (2H, m, CH), 5.11 (4H, s, OCH₂), 6.94–7.54 (20H, m, ArH), 8.12–8.18 (4H, m, ArH + NH). ¹³C NMR (CDCl₃): δ 36.89, 52.97, 63.40, 70.78, 112.58, 112.56, 121.61, 126.31, 127.31, 128.11, 128.31, 129.10, 132.13, 132.79, 136.39, 137.82, 156.53, 165.41. IR(KBr): ν 3389, 3072, 2960, 1735, 1637, 1599, 1537, 1479, 1449, 1385, 1298, 1221, 1161, 1101, 993, 754 cm⁻¹. ESIMS: *m*/*z* 645 (M + H, 76), 667 (M + Na, 100), 525 (80), 374 (20). Anal. Calcd. for C₄₀H₄₀N₂O₆ · 2H₂O: C, 70.56; H, 6.51; N, 4.11. Found: C, 70.67; H, 6.39; N, 3.91.

1,3-Bis[2-[N-(2-hydroxy-(1S)-1-isobutylethyl)carbamoyl] phenoxymethyl]benzene (**2c**)

Following general procedure A, from diacid **1a** (2.64 g, 7 mmol), $SOCl_2$ (40 mL), S-leucinol (1.65 g, 14 mmol) and Et_3N (10 mL) to give a colorless

solid **2c** in 91.6% yield (3.7 g). m.p. $60-61^{\circ}$ C. $[\alpha]_{D}^{20} = -65.8^{\circ}$ (c = 0.39, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 0.74 (6H, d, J = 6.8 Hz, CH₃), 0.80 (6H, d, J = 6.4 Hz, CH₃), 0.95–1.28 (4H, m, CH₂), 1.28–1.48 (2H, m, CH), 3.25 (2H, s, OH), 3.37–3.66 (4H, m, CH₂), 4.05–4.16 (2H, m, CH), 5.19 (4H, s, OCH₂), 7.02–7.13 (4H, m, ArH), 7.40–7.62 (6H, m, ArH), 7.95 (2H, d, J = 7.4 Hz, ArH), 8.18 (2H, d, J = 7.8 Hz, NH). ¹³C NMR (CDCl₃): δ 22.07, 22.92, 24.76, 40.08, 50.17, 66.18, 70.91, 112.43, 115.95, 121.58, 121.73, 128.52, 129.50, 132.32, 132.86, 136.38, 156.56, 165.80. IR(KBr): ν 3384, 3073, 2955, 2869, 1739, 1637, 1600, 1537, 1483, 1449, 1384, 1299, 1222, 1160, 1101, 754 cm⁻¹. ESIMS: m/z 577 (M + H, 100), 599 (M + Na, 86), 559 (15), 340 (40), 197 (70). Anal. Calcd. for C₃₄H₄₄N₂O₆ · 0.5H₂O: C, 69.71; H, 7.74; N, 4.78. Found: C, 69.58; H, 7.65; N, 4.63.

1,3-Bis[2-[N-(2-hydroxy-(1S)-1-isopropylethyl)carbamoyl] phenoxymethyl]benzene (2d)

Following general procedure A, from diacid **1a** (2.64 g, 7 mmol), SOCl₂ (40 mL), *S*- leucinol (1.44 g, 14 mmol) and Et₃N (10 mL) to give a colorless oil **2c** in 97.4% yield (3.74 g). $[\alpha]_{D}^{20} = -47.5^{\circ}$ (c = 0.48, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 0.67 (6H, d, J = 7.0 Hz, CH₃), 0.76 (6H, d, J = 6.4 Hz, CH₃), 1.65–1.76 (2H, m, CH), 3.31 (2H, s, OH), 3.45–3.60 (4H, m, CH₂), 3.82–3.90 (2H, m, CH), 5.20 (4H, s, OCH₂), 7.01–7.14 (4H, m, ArH), 7.41–7.49 (5H, m, ArH), 7.64 (1H, s, ArH), 8.02 (2H, d, J = 8.2 Hz, NH), 8.20 (2H, d, J = 7.6 Hz, ArH). ¹³C NMR (CDCl₃): δ 18.33, 19.32, 28.95, 57.35, 64.06, 70.85, 112.30, 121.62, 121.67, 128.16, 128.73, 129.44, 132.42, 132.85, 136.28, 156.55, 166.10. IR(KBr): ν 3391, 3073, 2960, 2873, 1735, 1636, 1599, 1537, 1483, 1449, 1385, 1298, 1223, 1163, 1102, 996, 755 cm⁻¹. ESIMS: m/z 549 (M + H, 8), 571 (M + Na, 100), 197 (5). Anal. Calcd. for C₃₂H₄₀N₂O₆·0.5H₂O: C, 68.92; H, 7.41; N, 5.02. Found: C, 68.70; H, 7.26; N, 4.83.

2,6-Bis[2-[N-(2-hydroxy-(1R)-1-phenylethyl)carbamoyl] phenoxymethyl]pyridine (**3a**)

Following general procedure A, from diacid **1b** (2.28 g, 6 mmol), SOCl₂ (40 mL), *R*-phenylglycinol (1.65 g, 12 mmol) and Et₃N (8 mL) to give a colorless solid **3a** in 76.8% yield (2.85 g). m.p. 118–119°C. $[\alpha]_D^{20} = +81.4^{\circ}$ (c = 0.35, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 3.41 (2H, t, *J* = 6.0 Hz, OH), 3.65–3.80 (4H, m, CH₂), 5.21(4H, s, OCH₂), 5.20–5.32 (2H, m, CH), 6.97–7.19 (14H, m, ArH), 7.27–7.44 (4H, m, ArH), 7.64 (1H, t, *J* = 7.8 Hz, ArH), 8.12–8.18 (2H, dd, *J* = 1.6, 7.6 Hz, ArH), 8.60 (2H, d, *J* = 7.6 Hz, NH). ¹³C NMR (CDCl₃): δ 55.78, 66.25, 71.47, 112.72, 121.97, 122.27, 126.61, 127.40, 128.54, 132.39, 132.95, 138.25, 139.33, 155.31,

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156.21, 165.42. IR(KBr): ν 3381, 3287, 3059, 2921, 1737, 1636, 1599, 1529, 1489, 1439, 1384, 1287, 1244, 1163, 1060, 1026, 752, 702 cm⁻¹. ESIMS: *m/z* 618 (M + H, 80), 640 (M + Na, 100), 600 (10), 498 (15), 481 (25), 343 (25), 225 (10). Anal. Calcd. for C₃₇H₃₅N₃O₆ · 0.5H₂O: C, 70.91; H, 5.79; N, 6.70. Found: C, 71.13; H, 5.85; N, 6.48.

2,6-Bis[2-[N-((1S)-1-benzyl-2-hydroxyethyl)carbamoyl] phenoxymethyl]pyridine (**3b**)

Following general procedure A, from diacid **1b** (2.28 g, 6 mmol), SOCl₂ (40 mL), *S*-phenylalaninol (1.82 g, 12 mmol) and Et₃N (8 mL) to give a colorless solid **3b** in 68.2% yield (2.64 g). m.p. 173–175°C. $[\alpha]_D^{20} = -56.7^{\circ}$ (c = 0.12, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 2.85 (4H, d, *J* = 7.2 Hz, CH₂), 3.34 (2H, *J* = 6.0 Hz, OH), 3.51–3.73 (4H, m, CH₂), 4.34–4.42 (2H, m, CH), 5.28 (4H, s, OCH₂), 6.96–7.29 (14H, m, ArH), 7.35–7.44 (4H, m, ArH), 7.69 (1H, t, *J* = 7.8 Hz, ArH), 8.13–8.18 (4H, dd, *J* = 1.6, 7.8 Hz, ArH + NH). ¹³C NMR (CDCl₃): δ 37.07, 53.30, 63.93, 71.35, 112.99, 122.06, 122.34, 126.49, 128.47, 129.19, 132.27, 132.82, 137.86, 155.60, 156.00, 165.61. IR(KBr): ν 3300, 3027, 2931, 2873, 1637, 1599, 1534, 1489, 1439, 1375, 1239, 1113, 1045, 1027, 752, 700 cm⁻¹. ESIMS: *m/z* 646(M + H, 100), 668 (M + Na, 95), 628 (30), 495 (46), 362 (15), 344 (12), 225 (5). Anal. Calcd. for C₃₉H₃₉N₃O₆ · 0.5H₂O: C, 71.50; H, 6.16; N, 6.41. Found: C, 71.67; H, 6.11; N, 6.24.

2,6-Bis[2-[N-(2-hydroxy-(1S)-1-isobutylethyl)carbamoyl] phenoxymethyl]pyridine (**3c**)

Following general procedure A, from diacid **1b** (2.28 g, 6 mmol), SOCl₂ (40 mL), *S*-leucinol (1.41 g, 12 mmol) and Et₃N (8 mL) to give a colorless solid **3c** in 87.3% yield (3.03 g). m.p. 116–118°C. $[\alpha]_D^{20} = -62.2^{\circ}$ (c = 0.37, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 0.80 (6H, d, *J* = 6.4 Hz, CH₃), 0.86 (6H, d, *J* = 6.4 Hz, CH₃), 1.20–1.39 (4H, m, CH₂), 1.47–1.64 (2H, m, CH), 3.45–3.76 (6H, m, OH + CH₂), 4.10–4.25 (2H, m, CH), 5.32 (4H, s, OCH₂), 7.02–7.13 (4H, m, ArH), 7.38–7.49 (4H, m, ArH), 7.82 (1H, t, *J* = 7.8 Hz, ArH), 7.96 (2H, d, *J* = 7.8 Hz, NH), 8.12–8.17 (2H, dd, *J* = 1.8, 7.6Hz, ArH). ¹³C NMR (CDCl₃): δ 22.16, 23.00, 24.86, 40.27, 50.19, 65.84, 71.44, 112.83, 122.02, 122.19, 122.41, 132.25, 132.73, 138.30, 155.62, 156.00, 165.75. IR(KBr): ν 3383, 3286, 3074, 2953, 2869, 1743, 1637, 1598, 1538, 1490, 1440, 1369, 1288, 1239, 1162, 1111, 1061, 750 cm⁻¹. ESIMS: m/z 578 (M + H, 100), 600 (M + Na, 35), 461 (30), 362 (12), 343(8). Anal. Calcd. for C₃₃H₄₃N₃O₆ · 0.5H₂O: C, 67.55; H, 7.55; N, 7.16. Found: C, 67.90; H, 7.36; N, 6.85.

2,6-Bis[2-[N-(2-hydroxy-(1S)-1-isopropylethyl)carbamoyl] phenoxymethyl]pyridine (**3d**)

Following general procedure A, from diacid **1b** (2.28 g, 6 mmol), SOCl₂ (40 mL), S-leucinol (1.24 g, 12 mmol) and Et₃N (8 mL) to give a colorless solid **3d** in 59.7% yield (1.97 g). m.p. 140–142°C. $[\alpha]_D^{20} = -43.5^{\circ}$ (c = 0.23, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 0.78 (6H, d, J = 6.6 Hz, CH₃), 0.85 (6H, d, J = 6.8 Hz, CH₃), 1.74–1.85 (2H, m, CH), 3.52–3.70 (6H, m, OH + CH₂), 3.89–3.96 (2H, m, CH), 5.32 (4H, s, OCH₂), 7.01–7.13 (4H, m, ArH), 7.37–7.48 (4H, m, ArH), 7.80 (1H, t, J = 7.8 Hz, ArH), 8.04 (2H, d, J = 8.2 Hz, NH), 8.13–8.18 (2H, dd, J = 1.8, 7.6 Hz, ArH). ¹³C NMR (CDCl₃): δ 18.65, 19.48, 29.04, 57.41, 63.69, 71.41, 112.69, 121.95, 122.37, 122.45, 132.32, 132.71, 138.33, 155.51, 156.03, 165.97. IR(KBr): ν 3406, 3311, 3075, 2956, 2870, 1771, 1616, 1580, 1535, 1491, 1441, 1384, 1296, 1234, 1170, 1109, 1078, 754 cm⁻¹. ESIMS: m/z 550 (M + H, 100), 572 (M + Na, 38), 532 (32), 447 (35), 362 (18), 343 (15), 225 (10). Anal. Calcd. for C₃₁H₃₉N₃O₆ · 0.5H₂O: C, 66.65; H, 7.21; N, 7.52. Found: C, 67.01; H, 6.86; N, 7.35.

General Procedure B

1,3-Bis[2-((4R)-4-phenyl-4,5-dihydro-1,3-oxazol-2-yl)phenoxymethyl] benzene (**4a**)

To an ice-cooled solution of the dihydroxy diamide 2a (1.23 g, 2.0 mmol) and Et₃N (1.4 mL, 10 mmol) in CH₂Cl₂ (20 mL) was added, dropwise MsCl (0.36 mL, 4.6 mmol). The mixture was allowed to warm to room temperature and was stirred for 12 hr. The mixture was guenched with water and extracted with CH_2Cl_2 (3 × 50 mL). The organic layer was dried over anhydrous Na₂SO₄ and concentrated to dryness in vacuo to give the crude bismesylate as a yellow oil. The crude bismesylate was dissolved in CH₃OH (30 mL) and NaOH solution (11 mL, 0.25 N), The mixture was refluxed for 3 hr. The methanol was removed in vacuo and the residue was extracted with CH₂Cl₂ $(2 \times 30 \text{ mL})$. The combined organic phase was washed with 1 N HCl, 5% NaHCO₃ and H₂O. The organic layer was dried over anhydrous Na₂SO₄ and concentrated in vacuo to give a pale yellow oil. Purification by silica gel column chromatography (petroleum ether-ethyl acetate 1:1) to afford light yellow oil **4a** 0.85 g (73.3 yield). $[\alpha]_{D}^{20} = +28.5^{\circ}$ (c = 0.33, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 4.18 (2H, t, J = 8.2 Hz, CH₂), 4.71 (2H, dd, $J = 8.2, 9.8 \text{ Hz}, \text{CH}_2$, 5.07 (4H, s, OCH₂), 5.40 (2H, dd, J = 8.2, 9.8 Hz, CH), 6.90–7.43 (19H, m, ArH), 7.71 (1H, s, ArH), 7.85 (2H, dd, J = 1.8, 7.8 Hz, ArH). ¹³C NMR (CDCl₃): δ 70.15, 70.19, 74.19, 113.40, 117.63, 120.53, 125.16, 125.87, 126.62, 127.29, 128.39, 128.50, 131.22, 132.17,

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136.98, 142.53, 157.39, 163.54. IR(KBr): ν 3061, 3029, 2959, 2894, 1643, 1600, 1582, 1494, 1451, 1384, 1355, 1250, 1163, 1034, 949, 752, 699 cm⁻¹. EIMS: m/z 580 (M⁺, 28), 355 (20), 341 (15), 239 (32), 121 (40), 104 (38). HREIMS Calcd. for C₃₈H₃₂N₂O₄: 580.23621. Found: 580.23656.

1,3-Bis[2-((4S)-4-benzyl-4,5-dihydro-1,3-oxazol-2-yl)phenoxymethyl] benzene (**4b**)

Following general procedure B, from dihydroxy diamide **2b** (0.95 g, 1.47 mmol) to afford light yellow oil 0.723 g (76.1% yield). $[\alpha]_{D}^{20} = -4.5^{\circ}$ (c = 0.27, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 2.74 (2H, dd, J = 8.8, 13.6 Hz, CH₂), 3.24 (2H, dd, J = 5.0, 13.6 Hz, CH₂), 4.08 (2H, t, J = 8.2 Hz, CH₂), 4.29 (2H, t, J = 8.2 Hz, CH₂), 4.55–4.67 (2H, m, CH), 5.17 (4H, s, OCH₂), 6.93–7.46 (19H, m, ArH), 7.68 (1H, s, ArH), 7.77 (2H, dd, J = 1.4, 7.6 Hz, ArH). ¹³C NMR (CDCl₃): δ 41.62, 67.75, 70.36, 71.23, 74.19, 113.53, 117.85, 120.59, 124.94, 125.87, 126.24, 128.31, 128.41, 129.08, 131.13, 132.04, 137.05, 137.86, 157.25, 162.81. IR(KBr): ν 3062, 3027, 2922, 1647, 1601, 1581, 1495, 1453, 1384, 1356, 1256, 1164, 1030, 965, 752, 700 cm⁻¹. EIMS: m/z 608 (M⁺, 25), 517 (48), 489 (15), 355 (45), 264 (26), 213 (40), 105 (100). HREIMS Calcd. for C₄₀H₃₆N₂O₄: 608.26751. Found: 608.26829.

1,3-Bis[2-((4S)-4-isobutyl-4,5-dihydro-1,3-oxazol-2-yl) phenoxymethyl]benzene (**4c**)

Following general procedure B, from dihydroxy diamide **2c** (1.5 g, 2.6 mmol) to afford a colorless oil 1.16 g (82.3% yield). $[\alpha]_D^{20} = -76.5^{\circ}$ (c = 0.27, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 0.97 (12H, t, J = 5.8 Hz, CH₃), 1.33–1.46 (2H, m, CH), 1.68–1.89 (4H, m, CH₂), 3.95 (2H, t, J = 7.6 Hz, CH₂), 4.31–4.53 (4H, m, CH + CH₂), 5.20 (4H, s, OCH₂), 6.97–7.03 (4H, m, ArH), 7.33–7.53 (5H, m, ArH), 7.61 (1H, s, ArH), 7.77 (2H, dd, J = 1.8, 8.0 Hz, ArH). ¹³C NMR (CDCl₃): δ 22.70, 22.78, 25.38, 45.58, 65.20, 70.45, 72.63, 113.57, 118.28, 120.70, 124.90, 126.01, 128.48, 131.25, 131.94, 137.15, 157.31, 162.20. IR(KBr): ν 3045, 2956, 2867, 1651, 1637, 1579, 1498, 1443, 1385, 1356, 1256, 1166, 1057, 1029, 792, 750 cm⁻¹. EIMS: m/z 540 (M⁺, 14), 483 (10), 429 (15), 321 (60), 219 (36), 121 (37), 105 (100). HREIMS Calcd. for C₃₄H₄₀N₂O₄: 540.29881. Found: 540.29790.

1,3-Bis[2-((4S)-4-isopropyl-4,5-dihydro-1,3-oxazol-2-yl) phenoxymethyl]benzene (**4d**)

Following general procedure B, from dihydroxy diamide **2d** (1.1 g, 2.0 mmol) to afford a colorless oil 0.665 g (64.6% yield). $[\alpha]_D^{20} = -84.5^\circ$ (c = 0.20,

CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 0.92 (6H, d, J = 6.8 Hz, CH₃), 1.02 (6H, d, J = 6.6 Hz, CH₃), 1.81–1.91 (2H, m, CH), 4.04–4.19 (4H, m, CH₂), 4.31–4.41 (2H, m, CH), 5.19 (4H, s, OCH₂), 6.97–7.03 (4H, m, ArH), 7.33–7.51 (5H, m, ArH), 7.61 (1H, s, ArH), 7.7 (2H, dd, J = 1.8, 7.8 Hz, ArH). ¹³C NMR (CDCl₃): δ 18.16, 18.87, 32.77, 69.72, 70.44, 72.79, 113.43, 118.35, 120.68, 125.16, 126.19, 128.50, 131.24, 131.92, 137.10, 157.31, 162.32. IR(KBr): ν 3069, 2958, 2898, 1647, 1601, 1581, 1496, 1449, 1384, 1354, 1256, 1165, 1042, 752 cm⁻¹. EIMS: m/z 512 (M⁺, 8), 469 (8), 415 (5), 321 (6), 307 (48), 105 (54). HREIMS Calcd. for C₃₂H₃₆N₂O₄: 512.26751. Found: 512.26562.

2,6-Bis[2-((4R)-4-phenyl-4,5-dihydro-1,3-oxazol-2-yl) phenoxymethyl]pyridine (**5a**)

Following general procedure B, from dihydroxy diamide **3a** (1.22 g, 1.97 mmol) to afford a colorless oil 0.963 g (83.7% yield). $[\alpha]_D^{20} = + 43.1^{\circ}$ (c = 0.13, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 4.30 (2H, t, J = 8.2 Hz, CH₂), 4.81 (2H, dd, J = 8.2, 10.0 Hz, CH₂), 5.29 (4H, s, OCH₂), 5.47 (2H, dd, J = 8.2, 10.0 Hz, CH), 7.01–7.09 (5H, m, ArH), 7.29–7.50 (12H, m, ArH), 7.66 (2H, d, J = 7.8 Hz, ArH), 7.90 (2H, dd, J = 1.6, 7.8 Hz, ArH). ¹³C NMR (CDCl₃): δ 70.54, 70.84, 74.07, 112.97, 117.35, 120.02, 120.79, 126.86, 127.49, 128.66, 131.49, 132.55, 142.57, 156.24, 157.19, 163.29. IR(KBr): ν 3062, 3026, 2893, 1640, 1595, 1579, 1494, 1441, 1375, 1356, 1294, 1252, 1126, 1063, 1030, 952, 748, 699 cm⁻¹. EIMS: m/z 581 (M⁺, 32), 550 (8), 431 (10), 342 (44), 238 (100), 121 (73), 104 (49). HREIMS Calcd. for C₃₇H₃₁N₃O₄: 581.23280. Found: 581.23214.

2,6-Bis[2-((4S)-4-benzyl-4,5-dihydro-1,3-oxazol-2-yl) phenoxymethyl]pyridine (**5b**)

Following general procedure B, from dihydroxy diamide **3b** (0.969 g, 1.5 mmol) to afford a colorless oil 0.784 g (85.7% yield). m.p. 117–118 °C. $[\alpha]_D^{20} = -16.4^\circ$ (c = 0.14, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 2.82 (2H, dd, J = 8.8, 13.8 Hz, CH₂), 3.31 (2H, dd, J = 5.2, 13.6 Hz, CH₂), 4.14 (2H, t, J = 8.0 Hz, CH₂), 4.38 (2H, t, J = 8.8 Hz, CH₂), 4.64–4.72 (2H, m, CH), 5.30 (4H, s, OCH₂), 6.99–7.06 (5H, m, ArH), 7.20–7.46 (11H, m, ArH), 7.58–7.84 (5H, m, ArH). ¹³C NMR (CDCl₃): δ 41.84, 68.13, 70.85, 71.24, 113.04, 117.54, 119.91, 120.80, 126.45, 128.53, 129.24, 131.41, 132.38, 137.71, 138.08, 156.43, 157.03, 162.69. IR(KBr): ν 3084, 3025, 2923, 1642, 1625, 1593, 1492, 1441, 1384, 1355, 1294, 1253, 1128, 1064, 1026, 962, 798, 751, 701 cm⁻¹. EIMS: m/z 609 (M⁺, 20), 518 (100), 490 (15), 356 (56), 266 (24), 213 (33), 162 (48). Anal. Calcd. for C₃₉H₃₅N₃O₄ · 0.5H₂O: C, 75.71; H, 5.86; N, 6.79. Found: C, 75.84; H, 5.80; N, 6.56.

2,6-Bis[2-((4S)-4-isobutyl-4,5-dihydro-1,3-oxazol-2-yl) phenoxymethyl]pyridine (**5c**)

Following general procedure B, from dihydroxy diamide **3c** (0.967 g, 1.67 mmol) to afford a colorless solid 0.756 g (83.5% yield). m.p. 97–98°C. $[\alpha]_{D}^{20} = -86.9^{\circ}$ (c = 0.236, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 1.02 (12H, t, J = 6.8 Hz, CH₃), 1.37–1.51 (2H, m, CH), 1.71–2.00 (4H, m, CH₂), 3.97 (2H, t, J = 7.0 Hz, CH₂), 4.38–4.57 (4H, m, CH + CH₂), 5.30 (4H, s, OCH₂), 6.98–7.06 (4H, m, ArH), 7.37–7.46 (2H, m, ArH), 7.74–7.90 (5H, m, ArH). ¹³C NMR (CDCl₃): δ 22.57, 23.01, 25.47, 45.70, 65.48, 70.76, 72.43, 112.88, 117.68, 119.96, 120.77, 131.38, 132.25, 137.52, 156.51, 157.00, 161.85. IR(KBr): ν 3073, 2950, 2868, 1635, 1595, 1580, 1495, 1446, 1384, 1368, 1292, 1256, 1167, 1063, 1028, 799, 749 cm⁻¹. EIMS: m/z 541 (M⁺, 18), 498 (15), 484 (34), 430 (28), 322 (100), 268 (55), 224 (65), 121 (72), 106 (53). Anal. Calcd. for C₃₃H₃₉N₃O₄: C, 73.17; H, 7.26; N, 7.76. Found: C, 72.84; H, 7.26; N, 7.60.

2,6-Bis[2-((4S)-4-isopropyl-4,5-dihydro-1,3-oxazol-2-yl) phenoxymethyl]pyridine (**5d**)

Following general procedure B, from dihydroxy diamide **3d** (1.26 g, 2.29 mmol) to afford a colorless solid 0.625 g (53% yield). m.p. $91-93^{\circ}$ C. $[\alpha]_{D}^{20} = -74.1^{\circ}$ (c = 0.116, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 0.99 (6H, d, J = 6.8 Hz, CH₃), 1.09 (6H, d, J = 6.8 Hz, CH₃), 1.87–1.97 (2H, m, CH), 4.07–4.25 (4H, m, CH₂), 4.36–4.47 (2H, m, CH), 5.29 (4H, s, OCH₂), 6.98–7.05 (5H, m, ArH), 7.37–7.46 (2H, m, ArH), 7.71 (4H, m, ArH). ¹³C NMR (CDCl₃): δ 18.27, 18.99, 32.83, 69.59, 70.73, 73.07, 112.84, 117.73, 120.03, 120.75, 131.37, 132.21, 137.57, 156.45, 156.97, 162.08. IR(KBr): ν 3078, 2958, 2869, 1644, 1595, 1579, 1497, 1443, 1384, 1369, 1291, 1253, 1171, 1064, 748 cm⁻¹. EIMS: m/z 513 (M⁺, 25), 470 (42), 416 (13), 385 (11), 308 (100), 266 (15), 224 (33), 106 (32). HREIMS Calcd. for C₃₁H₃₅N₃O₄: 513.26410. Found: 513.26441.

Typical Procedure

Asymmetric Cyclopropanation

A solution of the ligand (2.4 mol%) in CH₂Cl₂ (2 mL) was added [Cu(OTf)(C₆H₆)_{0.5}] (2 mol%). After being stirred at room temperature for 1.5 hr, styrene (8 mmol) was added, followed by dropwise addition of ethyl diazoacetate (2 mmol) in CH₂Cl₂ (1 mL) solution during 4 hr. The mixture was stirred for another 18 hr and then quenched with 10% NH₄Cl (5 mL). The mixture was extracted with ether (25 mL), and then the organic phase

was separated. The organic phase was washed with water and brine, dried over anhydrous sodium sulfate. The solvent was removed under reduced pressure, the residue was purified by column chromatography on silica gel (petroleum ether/ethyl acetate 15:1) to afford the mixture of *trans*- and *cis*-2-phenylcyclopropane-1-carboxylates as colorless oil. The *trans/cis* ratio of this mixture was analyzed by ¹H NMR (300 MHz) in CDCl₃ using TMS as internal standard. The diastereomeric and enantiomeric excess were determined by HPLC with a chiral column (Daicel Chiralcel OJ; eluent, hexaneisopropyl alcohol 90:10; flow rate, 0.5 mL/min; UV detector, 220 nm).

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REFERENCES

- 1. Donaldson, W. A. Tetrahedron 2001, 57, 8589-8627.
- Doyle, M. P.; McKervey, M. A.; Ye, T. Modern Catalytic Methods for Organic Synthesis with Diazo Compounds; Wiley: New York, 1998.
- 3. Faust, R. Angew. Chem. Int. Ed. 2001, 40, 2251-2253.
- 4. Suckling, C. J. Angew. Chem. Int. Ed. Engl. 1988, 27, 537.
- 5. Ye, T.; McKervey, M. A. Chem. Rev. 1994, 94, 1091-1160.
- Davies, H. M.L.; Antoulinakis, E. G. Organic Reactions; Overman, L. E., Ed.; Wiley: New York, 2001; Vol. 57, pp. 1–326.
- Doyle, M. P. Catalytic Asymmetric Synthesis; Ojima, I., Ed.; Wiley-VCH: New York, 2000; pp. 191–228.
- Lebel, H.; Marcoux, J.-F.; Molinaro, C.; Charette, A. B. Chem. Rev. 2003, 103, 977–1050For a recent reviews, see:.
- Noels, A. F.; Demonceau, A. Applied Homogeneous Catalysis with Organometallic Compounds; Cornils, B., Herrman, W. A., Eds.; Wiley-VCH: Weinheim, 2002; Vol. 2, pp. 793–808.
- Pfaltz, A. Comprehensive Asymmetric Catalysis; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Berlin, 1999a; pp. 513–603.
- 11. Fache, F.; Schulz, E.; Tommasino, M. L.; Lemaire, M. Chem. Rev. 2000, 100, 2159–2231.
- Gosh, A. K.; Mathivanan, P.; Capiello, J. *Tetrahedron Asymmetry* 1998, 9, 1–45For recent reviews, see:.
- 13. Johnson, J. S.; Evans, D. A. Acc. Chem. Res. 2000, 32, 325-335.
- Jørgensen, K. A.; Johannsen, M.; Yao, S. L.; Audrain, H.; Thorhauge, J. Acc. Chem. Res. 1999, 32, 605–613.
- 15. Pfaltz, A. J. Heterocyclic Chem. 1999b, 36, 1437-1451.
- 16. Pfaltz, A. Synlett 1999c, 835-842.

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- 17. Du, D. M.; Wang, Z. Y.; Xu, D. C.; Hua, W. T. Synthesis 2002, 2347–2352.
- 18. Du, D. M.; Fu, B.; Hua, W. T. Tetrahedron 2003, 59, 1933-1938.
- 19. Fu, B.; Du, D. M. Chinese J. Chem. 2003, 21, 597-599.
- 20. Fu, B.; Du, D. M.; Wang, J. B. Tetrahedron. Asymmetry 2004a, 15, 119-126.
- 21. Fu, B.; Du, D. M.; Xia, Q. Synthesis 2004b, 221-226.
- 22. Wang, Z. Y.; Du, D. M.; Wu, D.; Hua, W. T. Synth. Commun. 2003, 33, 1275–1283.
- Xu, D. C.; Du, D. M.; Ji, N.; Wang, Z. Y.; Hua, W. T. Synth. Commun. 2003, 33, 2563–2574.
- 24. Breton, P.; Andre-Barres, C.; Langlois, Y. Synth. Commun. 1992, 22, 2543.
- 25. Gant, T. G.; Meyers, A. I. Tetrahedron 1994, 50, 2297.
- 26. Jones, R. C.F.; Ward, G. J. Tetrahedron Lett. 1988, 29, 3853.
- 27. Reuman, M.; Meyers, A. I. Tetrahedron 1985, 41, 837.
- 28. Giannis, A.; Sandhoff, K. Angew. Chem. Int. Ed. Engl. 1989, 28, 218.
- Mckennon, M. J.; Meyers, A. I.; Drauz, K.; Schwarm, M. J. Org. Chem. 1993, 58, 3568.
- Seki, H.; Koga, K.; Matsuo, H.; Ohki, S.; Matsuo, I. I.; Yamada, S. Chem. Pharm. Bull. 1965, 13, 995.
- 31. Stanfield, C. F.; Parker, J. E.; Kanellis, P. J. Org. Chem. 1981, 46, 4799.
- 32. Elwahy, A. H.M.; Abbas, A. A. Tetrahedron 2000, 56, 885-895.
- Denmark, S. E.; Nakajima, N.; Nicaise, O. J. C.; Faucher, A. M.; Edwards, J. P. J. Org. Chem. 1995, 60, 4884.
- 34. Aratani, T.; Nakanisi, Y.; Nozaki, H. Tetrahedron 1970, 26, 1675.
- 35. Chelucci, G.; Sanna, M. G.; Gladiali, S. Tetrahedron 2000, 56, 2889-2893.
- France, M. B.; Milojevich, A. K.; Stitt, T. A.; Kim, A. J. *Tetrahedron Lett.* 2003, 44, 9287–9290.
- 37. Ito, K.; Katsuki, T. Tetrahedron Lett. 1993, 34, 2661.
- Iwasa, S.; Tsushima, S.; Nishiyama, K.; Tsuchiya, Y.; Takezawa, F.; Nishiyama, H. *Tetrahedron Asymmetry* 2003, 14, 855–865.
- 39. Lyle, M. P.A.; Wilson, P. D. Org. Lett. 2004, 6, 855-857See for examples:.
- Nishiyama, H.; Itoh, Y.; Matsumoto, H.; Park, S.-B.; Itoh, K. J. Am. Chem. Soc. 1994, 116, 2223–2224.
- Schinnerl, M.; Böhm, C.; Seitz, M.; Reiser, O. *Tetrahedron: Asymmetry* 2003, 14, 765–771.
- 42. Bedekar, A. V.; Andersson, P. G. Terahedron Lett. 1996, 37, 4073.
- 43. Bedekar, A. V.; Koroleva, E. B.; Andersson, P. G. J. Org. Chem. 1997, 62, 2518.
- 44. Evans, D. A.; Wocrpel, K. A.; Hinman, M. M. J. Am. Chem. Soc. 1991, 113, 726.
- 45. Alexander, K.; Cook, S.; Gibsona, C. L. Tetrahedron Lett. 2000, 41, 7135.
- Boulch, R.; Scheurer, A.; Mosset, P.; Saalfrank, R. W. Tetrahedron Lett. 2000, 41, 1023.
- 47. Uchida, T.; Irie, R.; Katsuki, T. Tetrahedron 2000, 56, 3501.