# TOTAL SYNTHESIS OF NEOMYCIN B\*†

TAKAYUKI USUI AND SUMIO UMEZAWA

Institute of Bioorganic Chemistry, 1614 Ida, Nakahara-ku, Kawasaki 211 (Japan) (Received June 23rd, 1987; accepted for publication, September 2nd, 1987)

## ABSTRACT

Total synthesis of neomycin B, a pseudo-tetrasaccharide aminoglycoside antibiotic, has been achieved through two key glycosylation reactions. Coupling of 3-O-acetyl-2,6-diazido-4-O-benzyl-2,6-dideoxy-L-idopyranosyl chloride with 5-Obenzoyl-1,2-O-isopropylidene- $\alpha$ -D-ribofuranose under modified Koenigs-Knorr conditions gave 70% of the desired  $\beta$ -L disaccharide (3) corresponding to neobiosamine in structure. After deisopropylidenation of 3 and acetylation, 1,2-di-O-acetyl-3-O-(3-O-acetyl-2,6-diazido-4-O-benzyl-2,6-dideoxy- $\beta$ -L-idopyranosyl)-5-O-benzoyl-D-ribofuranose was coupled to HO-5 of 3,2',6'-tri-N-(benzyloxycarbonyl)-1-N:6-O-carbonyl-3',4'-di-O-(o-methoxybenzoyl)neamine, using trimethylsilyl trifluoromethanesulfonate, to give 60% of the pseudo-tetrasaccharide 19 possessing the framework and masked functionality corresponding to neomycin B. Deblocking and reduction of the azido groups then gave neomycin B.

# INTRODUCTION

Neomycin B is a main component of the neomycin complex which was independently discovered by H. Umezawa *et al.*<sup>3</sup> and Waksman *et al.*<sup>4</sup>. The neomycin group is a representative of pseudo-tetrasaccharide antibiotics, which provide some fascinating problems of synthesis<sup>5,6</sup>. The complete structures and absolute stereochemistry of the neomycins have been elucidated<sup>7</sup> and a total synthesis of neomycin C has been achieved<sup>8</sup>. Neamine (neomycin A), the pseudo-disaccharide portion of neomycin B and C, was prepared<sup>9</sup> by 6'-amination of paromamine and later<sup>10</sup> by a coupling reaction. However, the total synthesis of neomycin B has been hindered<sup>11</sup> by the difficulty associated with glycosylation with 2,6-diamino-2,6-dideoxy-L-idose (neosamine B), which distinguishes neomycin B.

The key glycosyl chloride, 3-O-acetyl-2,6-diazido-4-O-benzyl-2,6-dideoxy-Lidopyranosyl chloride<sup>12</sup> (1) was already known and the following reactions were involved: (1) coupling of 1 to HO-3 of a protected ribose<sup>13</sup> (2) to give a masked

<sup>\*</sup>Dedicated to Professor Hans Paulsen.

<sup>\*</sup>For preliminary reports of part of this work, see refs. 1 and 2.

disaccharide (3) corresponding to neobiosamine B; and (2) coupling of 5, derived from 3, to HO-5 of a selectively protected neamine (18) to give a masked pseudo-tetrasaccharide (19), which was then converted into neomycin B.



## **RESULTS AND DISCUSSION**

Glycosylation of HO-3 of 5-O-benzoyl-1,2-O-isopropylidene- $\alpha$ -D-ribofuranose<sup>13</sup> (2) with 3-O-acetyl-2,6-diazido-4-O-benzyl-2,6-dideoxy-L-idopyranosyl chloride<sup>12</sup> (1) in the presence of mercuric cyanide, mercuric bromide, and molecular sieves (4 Å) gave 70% of the desired 1,2- $cis(\beta-L)$  disaccharide derivative 3 together with 7% of the  $\alpha$ -L anomer 4 after column chromatography. The  $\beta$ -L and  $\alpha$ -L configurations of 3 and 4 assigned on the basis of the  $[\alpha]_{\rm D}$  values (+139° and +28°, respectively) were confirmed by the small  $J_{2',3'}/J_{4',5'}$  values of 3 (Table I), and the long-range coupling  $(J_{2',4'} \ge 1 \text{ Hz})$  indicates the  ${}^{1}C_{4}(L)$  conformation almost exclusively<sup>14,15</sup>. The  $J_{2',3'}$  and  $J_{3',4'}$  values of the  $\alpha$ -L isomer 4 are larger than those of 3, which shows that 4 departs from the pure  ${}^{1}C_{4}$  conformer and some contribution of the  ${}^{4}C_{1}$  conformer is suggested  ${}^{12,14-17}$ . However, anomeric configurations of 3 and 4 were not assigned unequivocally because of the small difference of the  $J_{1',2'}$ values<sup>18</sup>. Therefore, the <sup>1</sup>H-n.m.r. spectra of **3** and **4** were compared with those of other idoside derivatives. The  $J_{1,2}/J_{4,5}$  values for methyl 2,6-diamino-2,6-dideoxy- $\beta$ -L-idopyranoside<sup>12</sup> (8) were close to those of 3, suggesting the  $\beta$  configuration of 3. On the other hand, the  $J_{1',2'}$  value of 4 was essentially identical with that of the corresponding  $\alpha$ -glycosyl acetate 7, but different from that of the  $\alpha$ -glycoside 9. The J values of other related compounds (10-13) prepared in our laboratory<sup>16</sup> are shown in Table I for comparison.

Since the <sup>1</sup>H-n.m.r. data gave no clear-cut conclusion for the anomeric configurations of 3 and 4, the <sup>13</sup>C-n.m.r. spectra were measured (Table II). The magnitude and the sign of the chemical shift difference of C-1/C-6 of the idoside

| $\begin{array}{c ccccccccccccccccccccccccccccccccccc$  | Compound              | Chemic | cal shifts <sup>a</sup> |      |      |      |      |             | First-o            | rder coupl | ling constar     | nts              |                   |                   |                  |                    |
|--|-----------------------|--------|-------------------------|------|------|------|------|-------------|--------------------|------------|------------------|------------------|-------------------|-------------------|------------------|--------------------|
| $3(\beta)$ 5.11       3.59       5.32       3.24       3.91       2.95       3.73       2.0       3.2       3.2       3.8       8.5 $4(\alpha)$ 4.87       3.60       5.12       3.52       4.41       3.27       3.59       3.4       5.5       3.8       8.5 $6^{0}(\beta)$ 7°(\alpha)       8.87       3.60       5.12       3.52       4.41       3.27       3.59       3.4       5.5       3.5       8.5       8.5 $7^{n}(\alpha)$ 8.86       3.52       4.41       3.27       3.59       3.4       5.5       3.5       8.5 |                       | I-H    | Н-2                     | Н-3  | H-4  | Н-5  | H-6a | <i>q9-Н</i> | $\mathbf{J}_{I,2}$ | $J_{2,3}$  | J <sub>3,4</sub> | J <sub>4,5</sub> | J <sub>5,6a</sub> | J <sub>5,6b</sub> | J <sub>2,4</sub> | $\mathbf{J}_{I,3}$ |
|  | 3( <i>b</i> )         | 5.11   | 3.59                    | 5.32 | 3.24 | 3.91 | 2.95 | 3.73        | 2.0                | 3.2        | 3.2              | 2.2              | 3.8               | 8.5               | *                | 0                  |
| $\theta(\theta)$ 2.5       4.5       4.5       3.5       8 $T^{h}(\alpha)$ 3.5       4.5       4.5       3.5       8 $\theta^{h}(\theta)$ 3.5       4.5       4.5       3.5       8 $\theta^{h,d}(\alpha)$ 3.5       3.5       3.5       3.5       8 $\theta^{h,d}(\alpha)$ 4.5       6.5       6       3.5       4.5       9 $\theta^{h,d}(\alpha)$ 1.1       4       4       2.5       4       7.5 $11^{0+d}(\theta)$ 3.5       5       4       2.5       4       7.5 $12^{-d}(\alpha)$ 3.5       5       4       2.5       4       7.5  | <b>4</b> (a)          | 4.87   | 3.60                    | 5.12 | 3.52 | 4.41 | 3.27 | 3.59        | 3.4                | ~5.5       | ~4.5             | ÷                | 5.4               | 8.5               | 0                | 0                  |
| $7^b(\alpha)$ $3.5$ $4.5$ $4.5$ $3.5$ $8.5$ $8^b.4(\beta)$ $2$ $3.5$ $3.5$ $2.5$ $4.5$ $8.5$ $9^b.4(\alpha)$ $4.5$ $6.5$ $6.$ $3.5$ $4.5$ $9.5$ $9^b.4(\alpha)$ $1.$ $4.5$ $6.5$ $6.$ $3.5$ $4.5$ $9.5$ $9^b.4(\alpha)$ $1.$ $4.5$ $6.5$ $6.$ $3.5$ $4.5$ $9.5$ $9^b.4(\alpha)$ $1.$ $4.$ $4.$ $2.5$ $4.$ $7.5$ $10^{b.e}(\beta)$ $1.$ $4.$ $2.5$ $4.$ $7.5$ $12^{c.4}(\beta)$ $1.5$ $4.$ $2.5$ $4.$ $7.5$   | 6 <sup>b</sup> (B)    |        |                         |      |      |      |      |             | 2.5                | 4.5        | 4.5              | ÷                | S                 | 8                 | ≤1′              |                    |
| $\mathfrak{g}^{b,d}(\beta)$ 2 $3.5$ $3.5$ $2.$ $4.5$ $8.5$ $\mathfrak{g}^{b,d}(\alpha)$ $4.5$ $6.5$ $6.$ $3.5$ $4.5$ $9.6$ $\mathfrak{g}^{b,d}(\alpha)$ $1.$ $4.5$ $6.5$ $6.$ $3.5$ $4.5$ $9.6$ $\mathfrak{g}^{b,d}(\alpha)$ $1.$ $4.$ $4.$ $2.5$ $4.$ $7.5$ $\mathfrak{g}^{b,d}(\alpha)$ $1.5$ $5.$ $4.$ $2.5$ $4.$ $7.5$ $\mathfrak{g}^{c,d}(\beta)$ $1.5$ $4.$ $2.5$ $4.$ $7.5$ $\mathfrak{g}^{c,d}(\beta)$ $1.5$ $4.$ $2.5$ $4.$ $2.5$ $4.$ $7.5$ $\mathfrak{g}^{c,d}(\beta)$ $1.5$ $4.$ $2.5$ $4.$ $2.5$ $4.$ $9.5$ $\mathfrak{g}^{c,d}(\beta)$ $1.5$ $4.$ $2.5$ $4.$ $2.5$ $4.$ $9.5$  | $T^{b}(\alpha)$       |        |                         |      |      |      |      |             | 3.5                | 4.5        | 4.5              | ÷                | ŝ                 | 8                 | ≤1'              | ₹¥                 |
| $9^{b,d}(x)$ 4.5       6.5       6       3.5       4.5       9 $10^{b,e}(B)$ 1       4       4       2.5       4       7.5 $11^{c,e}(a)$ 3.5       5       4       2.5       4       9 $12^{c,e}(B)$ 1.5       4       2.5       4       9   | 8 <sup>b, d</sup> (B) |        |                         |      |      |      |      |             | 6                  | 3.5        | 3.5              | 7                | 4.5               | 8.5               | 7                |                    |
| $10^{ke}(\beta)$ 1       4       4       2.5       4       7.5 $11^{ke}(\alpha)$ 3.5       5       4       2.5       4       9 $12^{ke}(\beta)$ 1       5       4       4       2.5       4       9  | $9^{b,d}(\alpha)$     |        |                         |      |      |      |      |             | 4.5                | 6.5        | 9                | 3.5              | 4.5               | 6                 |                  |                    |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$   | 10c.e(B)              |        |                         |      |      |      |      |             | 1                  | 4          | 4                | 2.5              | 4                 | 7.5               |                  |                    |
| 12s-d(B) 15 4 4 2  | $11^{c,e}(\alpha)$    |        |                         |      |      |      |      |             | 3.5                | 5          | 4                | 2.5              | 4                 | 6                 |                  |                    |
|  | $12^{c,d}(\beta)$     |        |                         |      |      |      |      |             | 1.5                | 4          | 4                | 7                |                   |                   | -                |                    |
| $13^{c,d}(\alpha)$ 6 8.5 7.5 4.5   | $13^{c,d}(\alpha)$    |        |                         |      |      |      |      |             | 9                  | 8.5        | 7.5              | 4.5              |                   |                   |                  |                    |

SYNTHESIS OF NEOMYCIN B

TABLE I

135



portion between  $3(\beta)$  and  $4(\alpha)$  showed a similar pattern with that between  $8(\beta)$  and  $9(\alpha)$ . The significant up-field shift (6.6 p.p.m. between 3 and 4, 4.3 p.p.m. between 8 and 9) of the signals for C-5 of the  $\alpha$  anomers compared to those of the  $\beta$  anomers reflect the  $\gamma$ -gauche effect<sup>19</sup> caused by H-5 and the axially oriented aglycon in the  $\alpha$  anomer, which adopts the  ${}^{1}C_{4}(L)$  conformation. The consistent shielding of C-5 of  $\alpha$ -aldohexopyranose derivatives compared to respective  $\beta$  anomers has been documented<sup>20</sup>. These results clearly indicate the anomeric configurations of 3 and 4 are as described.

Treatment of 3 with a mixture of acetic acid and 2M hydrochloric acid followed by conventional acetylation gave an  $\alpha\beta$ -mixture of glycosyl acetates (5, 92%,  $\alpha\beta$ -ratio ~1:3 as estimated from the <sup>1</sup>H-n.m.r. spectrum) which were indistinguishable on chromatography. When deisopropylidenation was carried out with HCl or H<sub>2</sub>SO<sub>4</sub> in aqueous alcohol or 1,4-dioxane, the yields were markedly lower.

A suitably protected neamine derivative with HO-5 free was next sought. When tetra-N-(benzyloxycarbonyl)-5,6-O-cyclohexylideneneamine<sup>21,22</sup> (15) was treated with benzyl bromide and base, a mixture of products resulted. Benzylation under acid catalysis<sup>23</sup> was also unsuccessful. Acetyl and benzoyl protection was in-

|             |                   | <b>x</b> 3, | · · · · | _                     |                     |                     | -                 |
|-------------|-------------------|-------------|---------|-----------------------|---------------------|---------------------|-------------------|
| Carbon      | <b>3</b> <i>a</i> | <b>4</b> ª  | 85      | <b>9</b> <sup>b</sup> | $\Delta\delta(3-4)$ | $\Delta\delta(8-9)$ | 2                 |
| C-1         | 104.7             | 104.2       |         |                       |                     |                     | 104.1             |
| <b>C-2</b>  | 76.7              | 78.8        |         |                       |                     |                     | 78.5 <sup>c</sup> |
| <b>C-3</b>  | 77.0              | 79.5        |         |                       |                     |                     | 72.2              |
| C-4         | 75.2              | 75.9        |         |                       |                     |                     | 78.4 <sup>c</sup> |
| C-5         | 63.2              | 62.7        |         |                       |                     |                     | 63.4              |
| C-1'        | 97.5              | 100.1       | 101.6   | 103.1                 | -2.6                | -2.5                |                   |
| C-2'        | 57.5              | 58.6        | 53.7    | 54.6                  | -1.1                | -0.9                |                   |
| C-3'        | 67.6              | 67.9        | 71.4    | 72.7                  | -0.3                | -1.3                |                   |
| C-4'        | 70.9              | 72.8        | 69.7    | 71.4                  | -1.9                | -1.7                |                   |
| C-5'        | 74.8              | 68.2        | 76.9    | 72.6                  | +6.6                | +4.3                |                   |
| C-6'        | 51.2              | 50.5        | 42.1    | 40.4                  | +0.7                | +1.7                |                   |
| $C(CH_3)_2$ | 113.4             | 113.0       |         |                       |                     |                     | 112.8             |
| $C(CH_3)_2$ | 26.8              | 26.4        |         |                       |                     |                     | 26.5 <sup>d</sup> |
| S/2         | 26.9              | 26.7        |         |                       |                     |                     |                   |
| C=O         | 166.2             | 166.2       |         |                       |                     |                     | 166.5             |
|             | 169.0             | 169.6       |         |                       |                     |                     |                   |

<sup>13</sup>C-N.M.R. CHEMICAL SHIFT DATA (CDCl<sub>1</sub>, 62.9 MHz) FOR **3** AND **4** COMPARED WITH THOSE FOR **2**, **8**, AND **9** 

<sup>a</sup>Determined by the <sup>13</sup>C<sup>-1</sup>H shift-correlated 2D spectrum. <sup>b</sup>Taken from ref. 12, measured in 20% ND<sub>3</sub> in D<sub>2</sub>O. <sup>c</sup>Assignments may be interchanged. <sup>d</sup>Duplicated.

appropriate because of the use of basic conditions in the subsequent formation of the cyclic carbamate.  $\alpha$ -Naphthoyl protection was found to improve the situation, but was still unsatisfactory. Protection by the *o*-methoxybenzoyl group proved to be suitable. Thus, **15** was acylated with *o*-methoxybenzoyl chloride in pyridine to give the 3',4'-di-O-acyl derivative (**16**) in good yield. After decyclohexylidenation ( $\rightarrow$ **17**), HO-6 was protected by cyclic carbamate formation with NH<sub>2</sub>-1 ( $\rightarrow$ **18**), using sodium hydride in *N*,*N*-dimethylformamide. The yield of the carbamate was  $\sim$ 60% when the reaction was conducted in the usual manner, but it was raised to  $\sim$ 80% by the addition of excess benzyltriethylammonium chloride. Without the quaternary ammonium salt, polar and faster-moving by-products appeared (t.1.c.). The <sup>1</sup>H-n.m.r. spectrum of the higher-mobility substance showed it to be benzyl *o*-methoxybenzoate.

Coupling of 5 and 18 in benzene in the presence of trimethylsilyl trifluoromethanesulfonate<sup>24</sup> gave 60% of the desired pseudo-tetrasaccharide derivative 19, having the  $\beta$ -D configuration, as the only isolable product; formation of the  $\alpha$  anomer was not detected. Although the resonances of H-1" and H-1" of 19 were overlapped with other signals, the mechanism of the glycoside formation under Lewis-acid catalysis involving neighbouring-group participation<sup>25</sup> strongly suggested the formation of a  $\beta$ -D(1,2-trans) ribosyl bond, which was eventually established by the small (2.3 Hz)  $J_{1",2"}$  value of fully deblocked neomycin B. The addition of molecular sieves as reported<sup>24</sup> did not improve the yield of 19. BF<sub>3</sub>etherate and SnCl<sub>4</sub> did not promote the glycosylation, but trityl perchlorate<sup>26</sup> showed a result almost identical with that obtained with trimethylsilyl trifluoromethanesulfonate.



Treatment of **19** with sodium benzyloxide in benzyl alcohol gave the 1,3,2',6'-tetra-*N*-(benzyloxycarbonyl) derivative **20**. The *o*-methoxybenzoyl group in **19** was resistant to saponification with aqueous alkali. Catalytic hydrogenolysis of **20** over Pd/C removed both the *N*-benzyloxycarbonyl and *O*-benzyl groups, and the azido groups were converted into amino groups to give 65% of neomycin B. The <sup>1</sup>H-<sup>27</sup> and <sup>13</sup>C-n.m.r.<sup>28</sup> spectra of the free base and the specific rotation<sup>29</sup> of the hydrochloride accorded with the data reported for natural neomycin B. The synthetic and natural specimens showed the same activity<sup>2</sup> against Gram-positive and -negative bacteria.



### **EXPERIMENTAL**

General methods. — Melting points were determined with a Kofler block and are uncorrected. Optical rotations were measured at 20–22° with a Perkin–Elmer 241 polarimeter for solutions in chloroform, unless otherwise stated. T.I.c. was performed on Kieselgel 60  $F_{254}$  (Merck) with detection by u.v. light (254 nm), iodine vapor, and charring with sulfuric acid. Flash column chromatography was performed on Kieselgel 60 (40–63  $\mu$ m) or Wakogel C-300. I.r. spectra were recorded for KBr discs with a JASCO A-202 grating spectrophotometer. <sup>1</sup>H-N.m.r. (250 MHz) and <sup>13</sup>C-n.m.r. (62.9 MHz) spectra were recorded with a Bruker WM-250 spectrometer for solutions in CDCl<sub>3</sub> (internal Me<sub>4</sub>Si) unless otherwise stated.

3-O-(3-O-Acetyl-2,6-diazido-4-O-benzyl-2,6-dideoxy-β- (3) and -α-L-idopyranosyl)-5-O-benzoyl-1,2-O-isopropylidene-β-D-ribofuranose (4). — A mixture of 1 (1.00 g, 2.62 mmol), 2 (398 mg, 1.35 mmol), Hg(CN)<sub>2</sub> (380 mg, 1.50 mmol), HgBr<sub>2</sub> (100 mg, 0.28 mmol), and powdered molecular sieves (4 Å, 2.5 g) in dry CH<sub>2</sub>Cl<sub>2</sub> was stirred for 28 h at room temperature under a gentle stream of N<sub>2</sub>. T.l.c. (toluene–ethyl acetate, 4:1) then revealed no 2 ( $R_F$  0.22), and the presence of 3 (0.31), 4 (0.58), and some minor by-products. The mixture was filtered with CHCl<sub>3</sub> through a pad of Celite, washed successively with saturated aqueous NaHCO<sub>3</sub>, H<sub>2</sub>O, aqueous 10% KI, H<sub>2</sub>O, and aqueous 10% NaCl, dried (Na<sub>2</sub>SO<sub>4</sub>), and concentrated. Column chromatography [silica gel (100 mL); toluene–ethyl acetate, 4:1] of the residue (~1.2 g) gave 3 as a glassy solid (605 mg, 70%), [ $\alpha$ ]<sub>D</sub> +139° (*c* 0.8);  $\nu_{max}$  2110 (N<sub>3</sub>), 1740 and 1720 (C=O), 1270, 1220, 1040 cm<sup>-1</sup>. <sup>1</sup>H-N.m.r. (see Table I also):  $\delta$  5.87 (d, J<sub>1,2</sub> 3.7 Hz, H-1), ~4.75 (H-5b), ~4.73 (H-2), ~4.4 (3 H, H-3,4,5a), 1.40 and 1.60 (2 s, each 3 H, CMe<sub>2</sub>). See Table II for <sup>13</sup>C-n.m.r. data.

Faster moving fractions containing 4 were re-chromatographed using preparative h.p.l.c. (Senshu Pak SSC-Silica-842, 5  $\mu$ m, 30 × 250 mm; toluene–ethyl acetate, 9:1, 13 mL/min) to give 4 (61 mg, 7%),  $[\alpha]_D$  +28° (c 1.1). <sup>1</sup>H-N.m.r. (see Table I also):  $\delta$  5.79 (d,  $J_{1,2}$  3.6 Hz, H-1), 4.73 (dd,  $J_{2,3}$  4.4 Hz, H-2), 4.64 (dd,  $J_{4,5b}$  3 Hz, H-5b), 4.50 (dd,  $J_{5gem}$  12.3,  $J_{4,5a}$  5 Hz, H-5a), 4.29 (ddd,  $J_{3,4}$  9.2 Hz, H-4), 3.84 (dd, H-3), 1.36 and 1.57 (2 s, each 3 H, CMe<sub>2</sub>). See Table II for <sup>13</sup>C-n.m.r. data.

Anal. Calc. for  $C_{30}H_{34}N_6O_{10}$  (638.6): C, 56.42; H, 5.37; N, 13.16. Found: **3**, C, 56.79; H, 5.28; N, 13.12; **4**, C, 56.61; H, 5.49; N, 13.40.

1,2-Di-O-acetyl-3-O-(3-O-acetyl-2,6-diazido-4-O-benzyl-2,6-dideoxy-β-L-idopyranosyl)-5-O-benzoyl-D-ribofuranose (5). — To a solution of **3** in acetic acid (15 mL) was added 2M hydrochloric acid (7 mL), and the solution was left for 2.5 h at room temperature. After concentration to ~5 mL, CHCl<sub>3</sub> (50 mL) was added and the solution was washed with saturated aqueous NaHCO<sub>3</sub> and aqueous 10% NaCl, dried (Na<sub>2</sub>SO<sub>4</sub>), and concentrated to dryness. The residue (810 mg) was acetylated with acetic anhydride (0.8 mL) in pyridine (4 mL) in the usual manner, to give **5** as a glassy solid (910 mg, 92%),  $[\alpha]_D + 82^\circ$  (c 0.9);  $\nu_{max}$  2110, 1740, 1270, 1220 cm<sup>-1</sup>. <sup>1</sup>H-N.m.r.: δ 6.48 (d, J<sub>2.3</sub> 4.5 Hz, H-1α), 6.20 (s, H-1β), 5.32 (d, J<sub>2.3</sub> 4.5 Hz, H-2β), 5.27 (t,  $J_{2',3'} = J_{3',4'} = 2.7$  Hz, H-3' $\beta$ , maybe overlapped with H-3' $\alpha$ ), 5.04 (dd,  $J_{2,3}$ 7 Hz, H-2 $\alpha$ ), 5.00 (d,  $J_{1',2'}$  2 Hz, H-1' $\beta$ ), 4.4–4.9 (m, 6 H, CH<sub>2</sub>Ph and H-3,4,5,5), 3.91 (ddd,  $J_{4',5'}$  2,  $J_{5',6'a}$  4,  $J_{5',6'b}$  8.2 Hz, H-5' $\beta$ ), 3.80 (ddd, H-5' $\alpha$ ), 3.69 (dd,  $J_{6'gem}$ 12.6 Hz, H-6'b $\alpha$ ), 3.66 (dd,  $J_{6'gem}$  13 Hz, H-6'b $\beta$ ), 3.48 (apparent t,  $J_{2',4'} \ge 1$  Hz, H-2' $\alpha$ ), 3.40 (apparent t,  $J_{2',4'} \ge 1$  Hz, H-2' $\beta$ ), 3.20–3.22 (H-4' $\alpha$ , $\beta$ ), 3.12 (dd, H-6'a $\alpha$ ), 3.02 (dd, H-6'a $\beta$ ), 2.16–1.86 (5 s, 9 H, 3 Ac); the  $\alpha\beta$ -ratio varied depending on the signals integrated and was in the range 1:2.7 based on H-6'a to 1:3.7 based on H-1.

*Anal.* Calc. for C<sub>31</sub>H<sub>34</sub>N<sub>6</sub>O<sub>12</sub> (682.6): C, 54.54; H, 5.02; N, 12.31. Found: C, 54.24; H, 5.11; N, 12.46.

1,3-Di-O-acetyl-2,6-diazido-4-O-benzyl-2,6-dideoxy- $\alpha$ -L-idopyranose (7). — The crystalline  $\alpha\beta$ -mixture<sup>12</sup> was repeatedly recrystallised from ether-pentane to give 7 with a constant m.p. 70-71°,  $[\alpha]_D - 8.6^\circ$  (c 0.6). The <sup>1</sup>H-n.m.r. data agreed with those<sup>12</sup> of the major anomer (see Table I).

Anal. Calc. for  $C_{17}H_{20}N_6O_6$  (404.4): C, 50.49; H, 4.99; N, 20.78. Found: C, 50.80; H, 4.93; N, 20.69.

1,3,2',6'-Tetra-N-(benzyloxycarbonyl)-5,6-O-cyclohexylidene-3',4'-di-O-(omethoxybenzoyl)neamine (16). — The described<sup>21</sup> preparation of 15 was improved. To a solution of tetra-N-(benzyloxycarbonyl)neamine<sup>22</sup> (14; 5.20 g, 6.05 mmol) in dry N, N-dimethylformamide (40 mL) containing 1,1-dimethoxycyclohexane (8 mL) and anhydrous toluene-p-sulfonic acid (200 mg) was added dry CH<sub>2</sub>Cl<sub>2</sub> (200 mL), and the cloudy solution was boiled under reflux, using a Soxhlet-type apparatus filled with molecular sieves (5 Å) to trap the methanol liberated. After 6 h, the solution became clear, and it was then washed with saturated aqueous NaHCO<sub>3</sub> and aqueous 10% NaCl, dried (Na<sub>2</sub>SO<sub>4</sub>), and concentrated to dryness. A solution of the residue (6.18 g, almost pure dicyclohexylidene derivative) in ethanol (65 mL) containing pyridinium toluene-p-sulfonate (300 mg) was kept for 16 h at 25°. Ether (~200 mL) was added to complete the precipitation and the bulk of mother liquor was removed by centrifugation. The solid was collected, washed with H<sub>2</sub>O, and dried to give crude 15 (4.88 g, 86%; contaminated with  $\leq 10\%$  of 14), which could be used for the next acylation. A portion (430 mg) of this preparation was purified by acetylation (pyridine and acetic anhydride), chromatography of the resulting diacetate (389 mg, 83%;  $\delta_{\rm H}$  1.82 and 1.90 for 2 Ac), and deacetylation (28% NH<sub>4</sub>OH-methanol, 1:9) to give 15 as needles (69% overall from 14). Recrystallisation from methanol-H<sub>2</sub>O (~6:1) gave material with m.p. 181-182°,  $[\alpha]_{\rm D}$ +6.7° (c 0.8), +29° (c 0.8, 1,4-dioxane); lit.<sup>21</sup> (for amorphous solid) m.p. 185.5-187°,  $[\alpha]_{\rm D}$  +5.3° (c 0.9); lit.<sup>30</sup> +26.67° (1,4-dioxane).

Anal. Calc. for  $C_{50}H_{58}N_4O_{14}$  (939.0): C, 63.95; H, 6.23; N, 5.97. Found: C, 63.87; H, 6.21; N, 6.06.

To an ice-cold solution of crude 15 (1.95 g) in dry pyridine (10 mL) was added a solution of *o*-methoxybenzoyl chloride (1.45 g) in dry  $CH_2Cl_2$  (20 mL) during ~5 min. The mixture was left overnight at room temperature, aqueous 50% pyridine (0.4 mL) was added and, after 0.5 h, the bulk of the solvent was

evaporated. A solution of the residue in CHCl<sub>3</sub> (40 mL) was washed successively with H<sub>2</sub>O, saturated aqueous NaHCO<sub>3</sub>, aqueous 10% KHSO<sub>4</sub>, and H<sub>2</sub>O, dried (Na<sub>2</sub>SO<sub>4</sub>), and concentrated to dryness. The residue (2.77 g) was chromatographed on silica gel (160 mL) with CHCl<sub>3</sub>-ethyl acetate (3:1) to give 16 as foamy solid (2.11 g, 84%; 72% from 14),  $[\alpha]_{\rm D}$  +22.5° (c 0.8);  $\nu_{\rm max}$  1720 (broad), 1520 (Amide II), 1300, 1250, 1020 cm<sup>-1</sup>. <sup>1</sup>H-N.m.r.:  $\delta$  3.72 and 3.70 (2 s, each 3 H, 2 OMe), 2.43 (H-2e), 1.2-1.6 (11 H, cyclohexylidene and H-2a).

Anal. Calc. for  $C_{66}H_{70}N_4O_{18}$  (1207.3): C, 65.66; H, 5.84; N, 4.64. Found: C, 65.95; H, 5.80; N, 4.61.

1,3,2',6' - Tetra-N-(benzyloxycarbonyl)-3',4'-di-O-(o-methoxybenzoyl)neamine (17). — A solution of 16 (2.05 g, 1.70 mmol) in acetic acid-oxolane-H<sub>2</sub>O (3:1:1, 35 mL) was kept for 3 h at 60° and then concentrated to dryness with several additions of toluene to give amorphous 17 (1.88 g, 98%),  $[\alpha]_D$  +59° (c 0.8);  $\nu_{max}$  1720 (broad), 1520, 1250 cm<sup>-1</sup>. <sup>1</sup>H-N.m.r.:  $\delta$  3.68 and 3.62 (2 s, each 3 H, 2 OMe), 2.22 (m, H-2e), 1.43 (m, H-2a).

Anal. Calc. for  $C_{60}H_{62}N_4O_{18}$  (1127.2): C, 63.93; H, 5.54; N, 4.97. Found: C, 63.75; H, 5.57; N, 4.74.

3,2',6'-Tri-N-(benzyloxycarbonyl)-1-N:6-O-carbonyl-3',4'-di-O-(o-methoxybenzoyl)neamine (18). — To an ice-cold solution of 17 (330 mg, 0.29 mmol) in dry N,N-dimethylformamide (5 mL) was added NaH (26 mg, 60% suspension in oil, net ~0.65 mmol), and the mixture was stirred under N<sub>2</sub>. After 20 min, a solution (10 mL) of benzyltriethylammonium chloride (550 mg, 2.42 mmol) in dry N,N-dimethylformamide was added at 0°, and the mixture was stirred for 2 h, then neutralised with acetic acid, and concentrated. A solution of the residue in CHCl<sub>3</sub> (20 mL) was washed successively with H<sub>2</sub>O, saturated aqueous NaHCO<sub>3</sub>, aqueous 10% KHSO<sub>4</sub>, and H<sub>2</sub>O, dried (Na<sub>2</sub>SO<sub>4</sub>), and concentrated to dryness. The residue (350 mg) was chromatographed on silica gel (30 mL) with CHCl<sub>3</sub>-ethanol (25:1) to give amorphous 18 (232 mg, 78%),  $[\alpha]_D + 80^\circ$  (c 1.1);  $\nu_{max}$  1770 (sh, cyclic carbamate), 1720, 1520, 1300, 1250 cm<sup>-1</sup>. <sup>1</sup>H-N.m.r.:  $\delta$  3.64 and 3.68 (2 s, each 3 H, 2 OMe), 2.15 (m, H-2e), 1.55 (q,  $J \sim 12$  Hz, H-2a).

Anal. Calc. for  $C_{53}H_{54}N_4O_{17}$  (1019.0): C, 62.47; H, 5.34; N, 5.50. Found: C, 62.38; H, 5.27; N, 5.48.

5-O-[2-O-Acetyl-3-O-(3-O-acetyl-2,6-diazido-4-O-benzyl-2,6-dideoxy- $\beta$ -Lidopyranosyl)-5-O-benzoyl- $\beta$ -D-ribofuranosyl]-3,2',6'-tri-N-(benzyloxycarbonyl)-1-N:6-O-carbonyl-3',4'-di-O-(o-methoxybenzoyl)neamine (**19**). — To a solution of **18** (320 mg, 0.31 mmol) and **5** (325 mg, 0.48 mmol) in dry benzene (16 mL) was added trimethylsilyl trifluoromethanesulphonate (100  $\mu$ L, ~0.52 mmol) under N<sub>2</sub>, and the solution was left for 4 h at room temperature. T.l.c. (CHCl<sub>3</sub>-acetone, 2:1) then showed spots of **19** ( $R_F$  0.43), **18** (0.25, slight), and **5** (0.92). The mixture was washed with saturated aqueous NaHCO<sub>3</sub> and aqueous 10% NaCl, dried (Na<sub>2</sub>SO<sub>4</sub>), and concentrated to dryness. The residue (640 mg) was chromatographed on silica gel (100 mL) with CHCl<sub>3</sub>-acetone (5:2) to give **19** (310 mg, 60%). An analytical sample, prepared by re-precipitation from 1,4-dioxane-H<sub>2</sub>O, had [ $\alpha$ ]<sub>D</sub> +55° (c 0.6);  $\nu_{\text{max}}$  2110, 1770 (sh), 1720, 1240 cm<sup>-1</sup>. <sup>1</sup>H-N.m.r.:  $\delta$  3.68 (s, 6 H, 2 OMe), ~2.2 (overlapped with Ac, H-2e), 2.13 and 2.11 (2 s, 6 H, 2 Ac), 1.40 (unresolved q, H-2a).

*Anal.* Calc. for C<sub>82</sub>H<sub>84</sub>N<sub>10</sub>O<sub>27</sub> (1641.6): C, 59.99; H, 5.16; N, 8.53. Found: C, 59.98; H, 5.01; N, 8.33.

Elution with  $CHCl_3$ -acetone (1:1) gave 18 (60 mg, 19%).

5-O-[3-O-(2,6-Diazido-4-O-benzyl-2,6-dideoxy-β-L-idopyranosyl)-β-D-ribofuranosyl]-1,3,2',6'-tetra-N-(benzyloxycarbonyl)neamine (20). — To a solution of 19 (150 mg, 91 µmol) in dry N,N-dimethylformamide (0.25 mL) and dry benzyl alcohol (1.9 mL) was added NaH (11 mg, 60% suspension in oil) under N<sub>2</sub>, and the solution was left for 4 h at room temperature. After neutralisation with solid CO<sub>2</sub>, the solution was diluted with CHCl<sub>3</sub> (15 mL), washed with aqueous 10% NaCl, dried (Na<sub>2</sub>SO<sub>4</sub>), and concentrated to dryness. The residue (140 mg) was chromatographed on silica gel (15 mL). Elution of the faster moving by-products with 20:1 CHCl<sub>3</sub>-ethanol was followed by elution with 10:1 CHCl<sub>3</sub>-ethanol to give amorphous 20 (100 mg, 85%),  $[\alpha]_D + 40^\circ$  (c 1);  $\nu_{max}$  3400 (broad), 2100, 1720 (sh), 1700 (broad), 1260, 1040 cm<sup>-1</sup>. <sup>1</sup>H-N.m.r. (CDCl<sub>3</sub> + D<sub>2</sub>O): δ 7.25 and 7.28 (2 s, 25 H, 5 Ph), 1.95 (H-2e), 1.25 (H-2a).

Anal. Calc. for  $C_{62}H_{72}N_{10}O_{21}$  (1293.3): C, 57.58; H, 5.61; N, 10.83. Found: C, 57.23; H, 5.61; N, 10.62.

Neomycin B. — A solution of 20 (80 mg, 62  $\mu$ mol) in 1,4-dioxane-H<sub>2</sub>Oacetic acid (2:2:1, 7.5 mL) was hydrogenated over 10% Pd/C (150 mg) under 4 kg.cm<sup>-1</sup> H<sub>2</sub>. After 26 h, t.l.c. of the mixture showed a ninhydrin-positive major spot having the same  $R_{\rm F}$  (0.18; CHCl<sub>3</sub>-methanol-17% NH<sub>4</sub>OH, 1:4:3) as natural neomycin B. The catalyst was removed (Celite), and the filtrate was concentrated to dryness with several additions of toluene. A solution ( $\sim 5 \text{ mL}$ ) of the residue in 5mm NH<sub>4</sub>OH was applied to a column (15 mL) of CM-Sephadex C-25 (NH<sup>+</sup><sub>4</sub> form, equilibrated with 5mM NH<sub>4</sub>OH) and, after washing with 5mM NH<sub>4</sub>OH (~30 mL), elution was continued with a gradient of NH<sub>4</sub>OH (5mM to 0.4M) to give neomycin B (25 mg, 65%). N.m.r. spectra were measured on the sample after passing through a short bed of Dowex 1-X2 (HO<sup>-</sup>) resin with decarbonated water and lyophilising the eluate. <sup>1</sup>H-N.m.r. (D<sub>2</sub>O, internal Me<sub>4</sub>Si) and <sup>13</sup>C-n.m.r. (D<sub>2</sub>O, corrected to Me<sub>4</sub>Si by the signal of 1,4-dioxane at 67.4 p.p.m.) spectra were essentially identical with those reported<sup>27,28</sup>. Typical values are given below, with the respective reported values<sup>27,28</sup> in brackets: <sup>1</sup>H, δ 5.45 (d, J 3.5 Hz, H-1') [5.45, J 3.9 Hz], 5.38 (d, J 2.3 Hz, H-1") [5.37, 2.7 Hz], 4.97 (slightly broadened s, H-1") [4.96, J 1.8 Hz], 1.94 (dt,  $J \sim 4$ ,  $\sim 4$ , and  $\geq 12$  Hz, H-2e) [1.95, J 4.1, 4.1, and 12.4 Hz], 1.20 (q, J≥12 Hz, H-2a) [1.20, J 12.5 Hz]; <sup>13</sup>C,  $\delta$  109.1 (C-1") [109.2], 100.3 (C-1') [100.3], 99.8 (C-1") [99.8], 51.1 (duplicated, C-1 and C-3) [51.2], 36.4 (C-2) [36.5].

The hydrochloride was prepared by dissolving the free base (30 mg) in a minimum amount of  $H_2O$ , acidifying to pH ~2 with M hydrochloric acid, and trituration with ethanol (~1 mL) and acetone (~5 mL). The resulting solid was centrifuged and washed with acetone, and the solvent was evaporated. An aqueous

solution was lyophilised to give the amorphous hydrochloride (32 mg),  $[\alpha]_D$  +54° (c 1, water); lit.<sup>29</sup> +54°.

*Anal.* Calc. for C<sub>23</sub>H<sub>46</sub>N<sub>6</sub>O<sub>13</sub>·6 HCl·H<sub>2</sub>O (851.4): C, 32.44; H, 6.39; N, 9.87; Cl, 24.98. Found: C, 32.48; H, 6.44; N, 9.56; Cl, 24.41.

# ACKNOWLEDGMENTS

We thank Dr. Tsutomu Tsuchiya of this Institute for helpful discussions, Ms. Yoshiko Koyama for the n.m.r. spectra, and Ms. Hiroko Hino (Institute of Applied Microbiology) for the microanalyses.

### REFERENCES

- 1 T. USUI AND S. UMEZAWA, National Meeting of the Chemical Society of Japan, 54th, Tokyo, April, 1987, Abstr. 4IIIL16.
- 2 T. USUI AND S. UMEZAWA, J. Antibiot., 40 (1987) 1464-1467.
- 3 H. UMEZAWA, S. HAYANO, AND Y. OGATA, Jap. Med. J., 1 (1948) 504-511.
- 4 S. A. WAKSMAN AND H. A. LECHEVALIER, Science, 109 (1949) 305-307.
- 5 S. UMEZAWA AND T. TSUCHIYA, in H. UMEZAWA AND I. R. HOOPER (Eds.), Handbook of Experimental Pharmacology, Vol. 62, Springer-Verlag, Berlin, 1982, ch. 2.
- 6 S. UMEZAWA, S. KONDO, AND Y. ITO, in H. PAPE AND H.-J. REHM (Eds.), *Biotechnology*, Vol. 4, VCH, Weinheim, 1986, ch. 11.
- 7 M. HICHENS AND K. L. RINEHART, JR., J. Am. Chem. Soc., 85 (1963) 1547-1548.
- 8 S. UMEZAWA AND Y. NISHIMURA, J. Antibiot., 30 (1977) 189–191; S. UMEZAWA, A. HARAYAMA, AND Y. NISHIMURA, Bull. Chem. Soc. Jpn., 53 (1980) 3259–3262.
- 9 S. UMEZAWA, K. TATSUTA, T. TSUCHIYA, AND E. KITAZAWA, J. Antibiot., Ser. A, 20 (1967) 53-54.
- 10 A. HARAYAMA, T. TSUCHIYA, AND S. UMEZAWA, Bull. Chem. Soc. Jpn., 52 (1979) 3626-3628.
- 11 S. UMEZAWA, in K. L. RINEHART, JR. AND T. SUAMI (Eds.), ACS Symp. Ser., 125 (1980) 20-26.
- 12 T. USUI, Y. TAKAGI, T. TSUCHIYA, AND S. UMEZAWA, Carbohydr. Res., 130 (1984) 165-177.
- 13 (a) I. WATANABE, T. TSUCHIYA, T. TAKASE, S. UMEZAWA, AND H. UMEZAWA, Bull. Chem. Soc. Jpn., 50 (1977) 2369–2374; (b) K. OKA AND H. WADA, Yakugaku Zasshi, 83 (1963) 890–891.
- 14 H. PAULSEN AND M. FRIEDMANN, Chem. Ber., 105 (1972) 705-717.
- 15 N. S. BHACCA, D. HORTON, AND H. PAULSEN, J. Org. Chem., 33 (1968) 2484-2487.
- 16 K. MATSUDA, T. TSUCHIYA, T. TORII, AND S. UMEZAWA, Bull. Chem. Soc. Jpn., 59 (1986) 1397– 1401.
- 17 A. S. PERLIN, B. CASU, G. R. SANDERSON, AND J. TSE, Carbohydr. Res., 21 (1972) 123-132.
- 18 S. J. ANGYAL, Angew. Chem., Int. Ed. Engl., 8 (1969) 175-226.
- 19 A. S. PERLIN, in E. BUNCEL AND C. C. LEE (Eds.), *Isotopes in Organic Chemistry*, Vol. 3, Elsevier, Amsterdam, 1977, ch. 4.
- 20 (a) K. BOCK AND C. PEDERSEN, Adv. Carbohydr. Chem. Biochem., 41 (1983) 27-66; (b) E. BREIT-MAIER AND V. VOELTER, Carbon-13 NMR Spectroscopy, 3rd edn., VCH, Weinheim, 1987, pp. 379-401.
- 21 Y. TAKAGI, D. IKEDA, T. TSUCHIYA, S. UMEZAWA, AND H. UMEZAWA, Bull. Chem. Soc. Jpn., 47 (1974) 3139-3141.
- 22 S. UMEZAWA, S. KOTO, K. TATSUTA, AND H. HINENO, Bull. Chem. Soc. Jpn., 42 (1969) 537-541.
- 23 H.-P. WESSEL, T. IVERSON, AND D. R. BUNDLE, J. Chem. Soc., Perkin Trans. 1, (1985) 2247-2250.
- 24 T. OGAWA, K. BEPPU, AND S. NAKABAYASHI, Carbohydr. Res., 93 (1981) c6-c9.
- 25 S. HANESSIAN AND J. BANOUB, Carbohydr. Res., 59 (1977) 261-267.
- 26 T. MUKAIYAMA, S. KOBAYASHI, AND S. SHODA, Chem. Lett., (1984) 907-910.
- 27 R. E. BOTTO AND B. COXON, J. Carbohydr. Chem., 3 (1984) 545-563.
- 28 S. HANESSIAN, T. TAKAMOTO, R. MASSÉ, AND G. PATIL, Can. J. Chem., 56 (1978) 1482-1491.
- 29 J. D. DUTCHER, N. HOSANSKY, M. N. DONIN, AND O. WINTERSTEINER, J. Am. Chem. Soc., 73 (1951) 1384–1385.
- 30 D. H. R. BARTON, D.-K. ZHENG, AND S. GERO, J. Carbohydr. Chem., 1 (1982) 105-118.