# LABDANE DITERPENOIDS FROM HALIMIUM VISCOSUM AND H. VERTICILATUM

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Key Word Index—Halimium viscosum; H. verticilatum; Cistaceae; labdane diterpenoid acids.

Abstract—In addition to three known acids, six new diterpenoid acids with a labdane skeleton, were isolated from the acid fraction of *Halimium viscosum*. Four of them were isolated from the acid fraction of *H. verticilatum*, together with another, as their methyl derivatives.

# INTRODUCTION

We have previously described the isolation from H. viscosum of three new diterpenoid acids with a labdane skeleton containing a C-17 carboxyl group and having two double bonds at C-7 and C-13 [1]. In the present report we describe the isolation from H. viscosum of six new diterpenoid acids, two with an unsaturated chain and four of them with a saturated chain. From H. verticilatum, Corroios (Portugal) has been isolated five diterpenoid acids, four of them also present in the acid fraction of H. viscosum.

### **RESULTS AND DISCUSSION**

The sodium hydroxide soluble part of the hexane extract of H. viscosum, Valparaiso (Zamora, Spain) was separated by column chromatography into three major fractions, I (24%), II (8%) and III (49%). A portion of fraction I, was treated with diazomethane and the resulting mixture of methyl esters was chromatographed on silica gel, to yield three mixtures (85/15) of two components 1h/8 (5%), 1i/9 (5%) and 1d/4 (85%). Column chromatography of the methyl esters corresponding to fraction II gave 1e as the principal component (85%) [1]. Fraction III contained a mixture (85/15) of hydroxy acids 1b/3. This mixture can be obtained from 1h/8, 1i/9 and 1d/4 by alkaline hydrolysis. Only a small quantity of 2 was isolated from fraction I by column chromatography. Using the same method it was not possible to separate the methyl esters mixtures 1d/4 and 1f/5. The hydroxyester mixture 1f/5 was resolved by transformation into the tetrahydropyranyl derivatives 1g and 6 that can be separated by column chromatography. The acid hydrolysis of 6 gave 5, which by acetylation yields 4, which can also be obtained by treatment of 2 with CH<sub>2</sub>N<sub>2</sub>.

Specimens of *H. verticilatum* were collected at Corroios (Portugal). The sodium hydroxide-soluble part of the hexane extract, after methylation, was separated into three fractions from which five methyl esters (4, 5, 7, 8, 9) were obtained by column chromatography.

Compounds 8 and 9 were also isolated from fraction I of *H. viscosum* as mixtures 1h/8 and 1i/9, which could not

be separated by column chromatography. Compounds 1h and 8 were obtained from 1f and 5 respectively, by treatment with isobutyryl chloride [2]. Treatment of 5 with formic acid [3] gave 9.

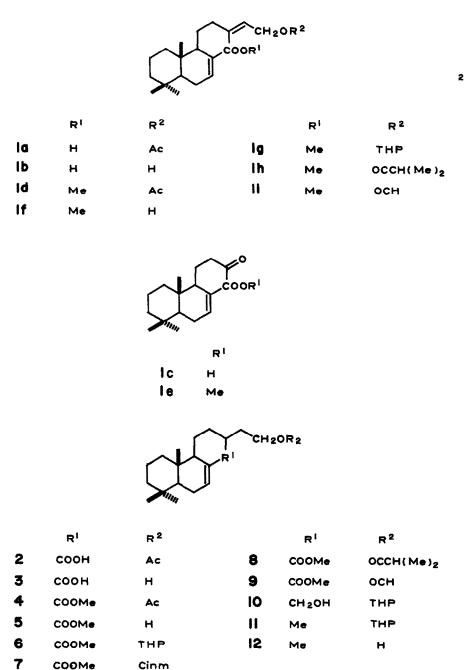
Compound 5 was an  $\alpha,\beta$ -unsaturated hydroxyester (IR: 3420, 1725 and 1640 cm<sup>-1</sup>). Its <sup>13</sup>C NMR spectrum showed signals for 21 carbons: five Me, eight CH<sub>2</sub>, four CH and four completely substituted carbons. Its <sup>1</sup>H NMR spectrum showed signals corresponding to the following groups: -CH=CCOOMe ( $\delta$ 6.64, 1H, m; 3.70, 3H, s), -CH<sub>2</sub>CH<sub>2</sub>OH (3.69, 2H, m); Me-CH(0.89, 3H, d) and three MeC (0.91, 0.87 and 0.82).

The mass spectrum of 5 exhibited the molecular ion at m/z 336 (C<sub>21</sub>H<sub>36</sub>O<sub>3</sub>) corresponding to a bicyclic diterpene with one double bond, a hydroxyl group and a meth-oxycarbonyl group. The ion of m/z 235 was formed by loss of a side chain. The methoxycarbonyl function must be situated at C-8 conjugated with a trisubstituted double bond at C-7 [1]. With respect to the natural compounds 2 and 3 they have the structure of 15-hydroxy-7-labden-17-oic acid and 15-acetoxy-7-labden-17-oic acid, respectively, which is shown upon transforming their carboxylic function into a methyl group, thereby obtaining cativol (12) [4].

Treatment of 3 with diacetyl ether-pyridine and diazomethane gave 2 and 5, respectively. Compound 6 was transformed into 10 by reduction with lithium aluminium hydride. Treatment of 10 with CIMs and reduction of the reaction product gave 11 which on acid hydrolysis produced 12 [5].

Compound 7 was an  $\alpha,\beta$ -unsaturated diester with an aromatic ring (IR: 1710, 1640 and 1500 cm<sup>-1</sup>). Its <sup>13</sup>C NMR spectrum showed signals for 30 carbons: five Me, eight CH<sub>2</sub>, eleven CH (eight sp<sup>2</sup>), and six completely substituted carbons. Its <sup>1</sup>H NMR spectrum showed signals corresponding to the following groups: -CH = CCOOMe ( $\delta 6.62$ , 1H, m; 3.70, 3H, s), -CH<sub>2</sub>CH<sub>2</sub>OOC (4.24, 2H, m) and -OOC-CH=CH-Ph (7.67, 1H, d; 6.44, 1H, d).

Alkaline hydrolysis of 7 and treatment with diazomethane of the hydrolysis product yielded 5 and methyl cinnamate (authentic sample); therefore 7 had the structure methyl 15-cinnamoyloxy-7-labden-17-oate.



# EXPERIMENTAL

Mps: uncorr, <sup>1</sup>H NMR: 200 MHz, CDCl<sub>3</sub>, TMS as int. standard; <sup>13</sup>C NMR: 50.3 MHz.

Extraction and isolation. The aerial part (0.86 kg) of H. verticilatum collected in Corroios (Portugal) was dried and extracted with n-hexane in a Soxhlet for 24 hr. The extract (21.2 g) was dewaxed with MeOH (21.1 %) and then extracted with 10% Na<sub>2</sub>CO<sub>3</sub> (12%) and 4% NaOH (43%). The neutral fraction represented 22% of the original extract. One portion of the NaOH-soluble acid fraction (2.0 g) of the hexane extract of H. verticilatum, after treatment with CH<sub>2</sub>N<sub>2</sub> was separated by CC into three fractions, I (n-hexane-Et<sub>2</sub>O, 8:2) (10%), II (n-hexane-Et<sub>2</sub>O, 1:1) (50%) and III (n-hexane-Et<sub>2</sub>O, 3:7) (40%).

Fraction I (0.2 g) was resolved by CC. Elution with *n*-hexane-Et<sub>2</sub>O (9:1) gave 4 (12 mg), 7 (20 mg) and 8 (10 mg). Fraction II (1.0 g) with the same method gave 5 (400 mg) and 9 (20 mg).

The extraction of the aerial part (1 kg) of *H. viscosum* collected in Valparaiso (Zamora, Spain) has been described elsewhere [1].

15-Acetoxy-7-labden-17-oic acid (2). Fraction I (1 g) of H. viscosum was resolved by CC. Elution with n-hexane-Et<sub>2</sub>O (3:2) gave 2 (30 mg). Colourless oil.  $[\alpha]_{22}^{22} - 36.6^{\circ}$  (CHCl<sub>3</sub>, c 1.10); IR  $v_{\text{max}}^{\text{film}}$  cm<sup>-1</sup>: 3500-2500, 1750, 1700, 1650, 1240, 1050,

	Labdane	diterpenoids	from	Halimium	species
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С	1 <b>h</b>	2	3	4	5	6	7	8	9	10	11	12
1	39.52	39.59	39.61	39.72	39.55	39.70	39.73	39.72	39.52	39.20	39.98	39.81
2	18.60	18.35	18.61	18.62	18.54	18.60	18.63	18.63	18.60	18.87	18.92	18.92
3	42.13	42.05	42.16	42.15	42.08	42.16	42.16	42.16	42.13	42.40	42.48	42.25
ţ	32.85	32.76	32.84	32.85	32.69	32.78	32.84	32.84	32.85	33.00	33.02	33.02
5	49.56	49.44	49.56	49.62	49.53	49.64	49.63	49.64	49.56	50.09	50.36	50.41
5	24.01	24.11	24.23	23.96	23.88	23.90	23.96	23.97	24.01	23.76	23.92	23.93
7	137.56	139.38	139.20	136.76	136.59	136.25	136.66	136.79	137.05	125.16	122.05	122.17
3	135.40	134.79	134.50	135.62	135.55	135.76	135.63	135.58	135.40	139.63	135.65	135.92
)	50.80	50.94	51.35	51.18	51.10	51.04	51.22	51.22	50.80	52.77	55.55	55.62
10	36.99	36.92	36.99	37.01	36.88	36.98	37.02	37.03	36.99	36.87	37.00	37.04
1	26.89	25.39	25.96	25.60	25.61	25.53	25.62	25.62	26.89	24.37	24.44	24.97
2	41.09	38.17	38.59	38.25	38.33	38.38	38.26	38.22	38.29	38.57	39.31	39.37
13	143.25	30.82	30.68	30.68	30.56	30.80	31.04	30.86	<b>29</b> .71	30.99	31.02	30.62
14	118.36	35.37	39.61	35.41	39.68	36.59	35.57	35.32	35.51	36.53	36.58	40.02
15	62.13	63.04	60.97	63.14	60.62	62.17	63.19	62.60	62.13	62.87	62.32	61.35
16	16.45	19.46	19.94	19.52	19.70	19.71	19.61	19.41	19.55	20.04	20.00	19.84
17	169.70	173.68	173.50	169.82	169.68	169.66	169.73	169.75	169.70	65.72	22.18	22.10
18	33.16	33.08	33.16	33.20	33.09	33.15	33.20	33.19	33.16	33.17	33.23	33.20
19	21.96	21.88	21.26	22.00	21.88	21.95	22.00	21.99	21.96	21.92	21.90	21.87
20	14.46	14.41	14.43	14.47	14.32	14.40	14.48	14.45	14.46	13.68	13.63	13.65
COOMe	51.28			51.25	51.25	51.26	51.22	51.37	51.28			
Me-COO		20.78		21.05								
Me-COO		171.03		171.16								
Me₂CH-												
coo	51.23								51.23			
Me <sub>2</sub> CH-												
coo	19.04								19.04			
Me <sub>2</sub> CH-												
coo	169.70								169.70			
HCOO								161.11				
l'							169.73					
2'						98.72	136.66			99.40	99.02	
- V						31.06	144.49			33.17	30.88	
¥′						25.58	134.66			25.58	25.61	
5′						19.06	128.88			19.79	19.73	
5'						66.02	128.06			65.97	66.13	
, 7'						00.04	130.17			00.07	00.15	
, 3'							128.06					
9′							128.88					

Table 1. <sup>13</sup>C NMR data of compounds 1h and 2-12\* (50.3 MHz, CDCl<sub>3</sub>, TMS as internal standard)

\*Assignments based on DEPT experiments and, particularly in the case of 5, on C/H(HCCORR) two dimensional correlations.

870; <sup>1</sup>H NMR:  $\delta 6.80$  (1H, m, H-7), 3.90 (2H, m, H-15), 1.90 (3H, s) 0.83 (3H, d, J = 7.0 Hz, Me-16), 0.85, 0.80, 0.79 (3H, each s, Me-19, Me-18 and Me-20, respectively). Treatment of 2 with CH<sub>2</sub>N<sub>2</sub> gave 4 and the alkaline hydrolysis (30 mg, 2 ml NaOH-MeOH 10%) yielded 3.

Methyl 15-acetoxy-7-labden-17-oate (4). Colourless oil.  $[\alpha]_{D}^{22}$  - 84.5° (CHCl<sub>3</sub>, c 1.18). IR v<sup>film</sup><sub>max</sub> 1750, 1730, 1650, 1250; <sup>1</sup>H NMR:  $\delta 6.62$  (1H, m, H-7), 4.09 (2H, m, H-14), 3.70 (3H, s), 2.04 (3H, s), 0.91, 0.87, 0.82 (3H, each s, Me-19, Me-18 and Me-20, respectively), 0.88 (3H, d, J = 7.7 Hz, Me-16).

15-Hydroxy-7-labden-17-oic acid (3). Colourless oil.  $[\alpha]_{D^2}^{2D}$ - 24.1° (CHCl<sub>3</sub>, c 1.31); IR ν<sup>fimat</sup><sub>max</sub> cm<sup>-1</sup>: 3500-2600, 1700, 1260; <sup>1</sup>H NMR: δ6.82 (1H, m, H-7), 3.67 (2H, m, H-15), 0.88 (3H, d, J = 7.3 Hz, Me-16), 0.90, 0.87, 0.82 (3H, each s, Me-19 Me-18 and Me-20, respectively)

Methyl 15-hydroxy-7-labden-17-oate (5). To 5.0 g of methyl

esters mixture 1f/5 dissolved in 15 ml dry C<sub>6</sub>H<sub>6</sub> were added 2 ml dihydropyrane and 0.18 g TsOH. After shaking the mixture at room temp. 140 mg K<sub>2</sub>CO<sub>3</sub> were added and after 30 min the mixture was filtered and evapd to obtain 5.9 g of reaction product which on CC gave 4.2 g (*n*-hexane-Et<sub>2</sub>O, 9:1) 1g and 0.5 g of 6. Colourless oil. IR  $v_{\text{max}}^{\text{film}}$  cm<sup>-1</sup>: 1730, 1655, 1270, 1250, 1080, 980, 910, 870; <sup>1</sup>H NMR:  $\delta 6.57$  (1H, *m*, H-7), 4.53 (1H, *m*, H-2'), 3.77 (2H, *m*), 3.66 (3H, s), 3.40 (2H, *m*), 0.85 (3H, d, J = 7.7 Hz, Me-16), 0.87, 0.83, 0.78 (3H, each s, Me-19, Me-18 and Me-20, respectively).

*Hydrolysis of compound* 6. To 50 mg of 6, dissolved in 3 ml MeOH, were added 2.5 mg of TsOH and the mixture was shaken at room temp for 1 hr. Following this, the solvent was evapd, Et<sub>2</sub>O was added and the product washed with NaHCO<sub>3</sub> and H<sub>2</sub>O. After evaporating off the solvent, 35 mg of 5 was obtained. Colourless oil.  $[\alpha]_{D}^{22} - 51.0^{\circ}$  (CHCl<sub>3</sub>, c1.27); UV $\lambda_{max}^{EEOH}$  nm (log

ε): 2.16 (2.00). IR  $v_{max}^{flin}$  cm<sup>-1</sup>: 3420, 1720, 1640, 1260; <sup>1</sup>H NMR: δ6.63 (1H, m, H-7), 3.71 (3H, s), 3.69 (2H, m, H-15), 0.89 (3H, d, J = 7.5 Hz, Me-16), 0.91, 0.87, 0.83 (3H, each s, Me-19, Me-18 and Me-20, respectively). EIMS 70 eV, m/z (rel. int): 336 [M]<sup>+</sup> (2), 305 (25), 290 (8), 235 (18), 176 (35), 124 (11), 109 (100), 105 (17), 43 (21), 41 (18).

Reduction of 6 with LiAlH<sub>4</sub>. To 320 mg of 6 dissolved in 3 ml Et<sub>2</sub>O, were added 15 mg LiAlH<sub>4</sub>. The reaction mixture was kept at room temp for 1 hr. The usual procedure gave 283 mg of reduction product which on silica gel CC (*n*-hexane-Et<sub>2</sub>O, 9:1) yielded 10 (212 mg). Colourless oil. IR  $v_{max}^{flim}$  cm<sup>-1</sup>: 3420, 1330, 1210, 1140, 1040, 910, 870. <sup>1</sup>H NMR:  $\delta$ 5.74 (1H, m, H-7), 4.54 (1H, m, H-2'), 4.02 (2H, m), 3.85 (2H, m), 3.50 (2H, m), 0.91 (3H, d, J = 6.8 Hz, Me-16), 0.87, 0.86, 0.72 (3H, each s, Me-19, Me-18 and Me-20, respectively).

Preparation and reduction of the mesylate. Et<sub>3</sub>N (0.11 ml) was added to 212 mg 10 in 3 ml CH<sub>2</sub>Cl<sub>2</sub> and the mixture left to cool between  $-10^{\circ}$  and  $0^{\circ}$ . MeSO<sub>2</sub>Cl (0.06 ml) was added and the product kept at room temp. for 1.5 hr. It was then washed with ice-water, 2 M HCl and then again with H<sub>2</sub>O. The reaction product (272 mg) was dissolved in Et<sub>2</sub>O (3 ml) and treated with 10 mg LiAlH<sub>4</sub>. The mixture was shaken for 1 hr. The usual procedure gave 200 mg of reduction production which on silica gel CC (*n*-hexane-Et<sub>2</sub>O, 49:1) gave 60 mg 11. Colourless oil. IR  $v_{max}^{\rm imm}$  cm<sup>-1</sup>: 1330, 1210, 1160, 1140, 1110, 940, 840; <sup>1</sup>H NMR:  $\delta$ 5.35 (1H, *m*, H-7), 4.57 (1H, *m*, H-2'), 3.82 (2H, *m*), 3.44 (2H, *m*), 1.66 (3H, *s*, Me-17), 0.92 (3H, *d*, *J* = 6.2 Hz, Me-16), 0.87, 0.85, 0.75 (3H, each *s*, Me-19, Me-18 and Me-20, respectively).

Hydrolysis of compound 11. p-Toluensulphonic acid (2.5 mg) was added to 41 mg 11 dissolved in 2.5 ml MeOH and the mixture shaken for 4 hr at room temp. The usual procedure gave 20 mg 12. Colourless oil.  $[\alpha]_D^{22} + 1.4^\circ$  (CHCl<sub>3</sub>, c 0.85); IR  $v_{max}^{\dim} \text{cm}^{-1}$ : 3360, 1080, 980, 840. <sup>1</sup>H NMR:  $\delta$ 5.39 (1H, m, H-7), 3.68 (2H, m, H-15), 1.67 (3H, s, Me-17), 0.93 (3H, d, J = 6.2 Hz, Me-16), 0.88, 0.85, 0.76 (3H, each s, Me-19, Me-18 and Me-20, respectively).

Methyl 15-cinnamoyloxy-7-labden-17-oate (7). Colourless oil.  $[\alpha]_D^{22} - 33.5^{\circ}$  (CHCl<sub>3</sub>, c 1.30); IR  $\nu_{max}^{film}$  cm<sup>-1</sup>: 1710, 1640, 1500, 1310, 1240, 1160, 980. <sup>1</sup>H NMR:  $\delta$ 7.67 (1H, d, J = 16 Hz, H-3'), 7.52 (2H, m), 7.36 (3H, m), 6.40 (1H, d, J = 16 Hz, H-2'), 6.62 (1H, m, H-7), 4.24 (2H, m, H-15), 3.70 (3H, s), 0.92 (3H, d, J = 6.4 Hz, Me-16), 0.88, 0.84, 0.80 (3H, each s, Me-19, Me-18 and Me-20, respectively). Methyl 15-isobutyloxy-7-labden-17-oate (8). Colourless oil.  $[\alpha]_{D}^{22}-28.4^{\circ}$  (CHCl<sub>3</sub>, c 0.74); IR v  $\frac{\text{finar}}{\text{max}}$  cm<sup>-1</sup>: 1750, 1740, 1660, 1250, 1080; <sup>1</sup>H NMR:  $\delta 6.61$  (1H, m, H-7), 4.07 (2H, m, H-15), 3.67 (3H, s), 2.50 (1H, h, J = 6.8 Hz, H-2'), 1.11 (6H, d, J = 6.8 Hz, H-3'), 0.92 (3H, d, J = 6.4 Hz, Me-16), 0.88, 0.84, 0.80 (3H, each s Me-19, Me-18 and Me-20, respectively).

Compound 5 (90 mg) dissolved in 0.5 ml dry pyridine were cooled to 0°. Isobutyryl chloride (0.2 ml) was added and after 1 hr the mixture was kept at room temp for 8 hr. Et<sub>2</sub>O was added and it was then washed with 2 M HCl, 5% NaCO<sub>3</sub>H and H<sub>2</sub>O. Evapn of the solvent after drying over Na<sub>2</sub>SO<sub>4</sub> gave 110 mg 8.

*Methyl* 15-isobutyloxy-7,13E-labdadien-17-oate (1h). Treatment of 1.0 g of fraction I of *H. viscosum*, with CH<sub>2</sub>N<sub>2</sub>, gave 1.0 g of a mixture of Me esters, which was resolved by CC. Elution with *n*-hexane-Et<sub>2</sub>O (9:1), gave 1h/8 (50 mg), 1i/9 (30 mg) and 1d/4 (800 mg). By a similar treatment as described before 100 mg of 1f [1] gave 1h (100 mg). IR  $v_{\text{film}}^{\text{film}}$  cm<sup>-1</sup>: 1750, 1740, 1660,1250, 1080; <sup>1</sup>H NMR:  $\delta 6.61$  (1H, *m*, H-7), 5.29 (1H, *t*, *J* = 7.3 Hz, H-14), 4.54 (2H, *d*, *J* = 7.3 Hz, H-15), 3.68 (3H, s), 2.52 (1H, *h*, *J* = 7.1 Hz, H-2'), 1.65 (3H, s, Me-16), 1.15 (6H, *d*, *J* = 7.1 Hz, CH(Me)<sub>2</sub>), 0.88, 0.84, 0.80, (3H, each s, Me-19, Me-18 and Me-20, respectively).

Methyl 15-formyloxy-7-labden-17-oate (9). Colourless oil.  $[\alpha]_D^{22} - 56.5^{\circ}$  (CHCl<sub>3</sub>, c 1.21); IR  $v_{\rm max}^{\rm film}$  cm<sup>-1</sup>: 1745, 1650, 1260, 1200, 1090, 800. <sup>1</sup>H NMR:  $\delta$ 8.04 (1H, s, OOCH), 6.63 (1H, m, H-7), 4.19 (2H, m, H-15), 3.70 (3H, s), 0.85 (3H, d, J = 6.7 Hz, Me-16), 0.90, 0.86, 0.82 (3H, each s Me-19, Me-18 and Me-20, respectively).

Formylation of Compound 5. Compound 5 (50 mg) was dissolved in HCOOH (0.5 ml) and kept for 1 min, following this it was extracted with  $Et_2O$ , then washed with  $H_2O$  and dried over  $Na_2SO_4$ . Evaporation of the solvent gave 9 (50 mg).

#### REFERENCES

- De Pascual Teresa, J., Urones, J. G., Marcos, I. S., Diez Martin, D. and Alvarez, V. M. (1986) *Phytochemistry* 25, 711.
- 2. Ross, M. J. and Rickborn, B. (1971) Org. Synthesis 51, 11.
- 3. Cortese, F. and Bauman, L. (1935) J. Am. Chem. Soc. 57, 1393.
- 4. Zeiss, H. H. and Grant, F. W. (1957) J. Am. Chem. Soc. 79,
- 1201.
  Corey, E. J., Niwa, H. and Knolle, J. (1978), J. Am. Chem. Soc 100, 1942.