This article was downloaded by: [The University of Manchester Library] On: 20 November 2014, At: 13:02 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

New Facile Preparation of Dihydro Spiro-1,5benzodiazepines

P. Shanthan Rao^a & R. V. Venkataratnam^a ^a Indian Institute of Chemical Technology, Hyderabad, 500 007, India Published online: 24 Sep 2006.

To cite this article: P. Shanthan Rao & R. V. Venkataratnam (1991) New Facile Preparation of Dihydro Spiro-1,5-benzodiazepines, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 21:17, 1811-1817, DOI: <u>10.1080/00397919108021581</u>

To link to this article: <u>http://dx.doi.org/10.1080/00397919108021581</u>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

NEW FACILE PREPARATION OF DIHYDRO SPIRO-1,5-BENZODIAZEPINES ⁺

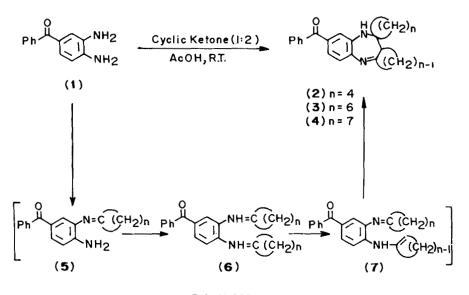
P. Shanthan Rao and R.V. Venkataratnam[®] Indian Institute of Chemical Technology, Hyderabad 500 007, India.

Abstract: 2,3-Cycloalkano-3,4-dihydro(5H)-7-benzoyl-1,5-benzodiazepine -4-spirocycloalkanes have been prepared from 4-benzoyl-o-phenylenediamine and cycloalkanones in presence of acetic acid. Only in the case of cyclohexanones, dihydrospiro benzimidazoles are formed as coproducts which are absent in others.

Following the earliest synthesis of 2,3-cyclopentano-3,4-dihydro(5H)-1,5-benzodiazepine-4-spiro cyclopentane, we desired to prepare dihydro spiro benzodiazepines derived from other cyclic ketones under the same procedure¹. The expected products were not obtained even after a reaction time of 24 h. Addition of glacial acetic acid to provide a mild acidic medium did not improve the reaction. This experience induced us to differentiate the relative reactivity of the amino groups in o-phenylenediamine, in order to make the reaction to be of any significant preparative value. This was achieved by introducing a benzoyl group in the 4-position of o-phenylenediamine, as in (1) and carrying out the reaction in acetic acid medium. Yields of dihydro spiro benzodiazepines (2, 3 & 4) were more than 70%. The function of the 4-benzoyl group is to facilitate the stepwise formation of the

[°] To whom correspondence should be addressed.

⁺ IICT Communication No.2825.



SCHEME-I

intermediates (5 & 6) and to selectively drive (6) into the enamine form (7) in presence of acetic acid for facile cyclisation. Dihydro spiro benzimidazoles arising out of (5) are not formed in this reaction. Absence of acetic acid inhibits the formation of diazepine products, thus establishing its requirement for the conversion of (6) to (7). Synthesis of similar spiro diazepines have been reported by the reductive condensation of o-phenylenediamine and ketones in the presence of sodium borohydride², but the products are preferentially tetrahydro diazepines in admixture with dihydrodiazepines. The present procedure affords dihydrodizepines exclusively, even with different mole ratios of reactants.

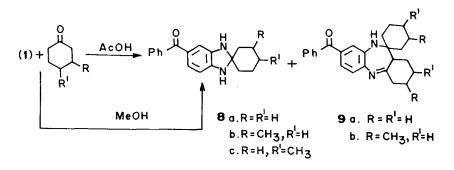
Comp. No.	m₊p₊ °C	Yield [*] %	IR(KBr) Ƴcm ⁻¹	¹ HNMR	MS
2	85	70	3320, 1650, 1620	0.88-2.88, m(15H) 4.44-4.93, b(1H) 6.44-8.20 m (8H)	344,236 105,77
3	173	75	3250,1640 1620	1.10-2.88, m(23H) 4.00-4.66, b(1H) ⁺ 6.44-8.00, m(8H)	400,249 236,105 77
4	165	77	3250, 1645 1635	0.62-3.00, m(27H) 4.30-4.75, b(1H) ⁺ 6.37-7.90, m (8H)	428, 357 249, 236 105, 77

Table 1

^{*} yields based on isolated and crystallised product.

⁺ Exchangeable with D₂O

The structures of the dihydro benzo diazepines follow from the mode of formation from the most probable intermediate (7). The spectral data given in the Table-1 conform to the structures assigned. The isomeric structures to the dihydro benzo diazepines have been ruled out because of the improbability of the relevant enamine intermediate isomeric to (7). The influence of benzoyl group on the para imino group provides the driving force for the facile formation of (7) in acetic acid medium and brings about the cyclisation to the diazepines. The mass spectra of the products besides the moelcular ion, are characteristic in that they show three common fragment ions viz., m/z 236, 105 and 77. While m/z 105 and 77 are well known, the



SCHEME-2

structure of m/z 236 as dihydroquinoxaline rather than 2-methylenebenzimidazoline is under investigation.

same reaction with cyclohexanone and its substituted The derivatives show variations with respect to the degree of ring flexibility. Trans-4-tert-butyl cyclohexanone with rigid conformation produces 4-methylcyclohexanone with a slight degree of no product. The conformational freedom produces only the dihydro spiro benzimidazole Cyclohexanone and 3-methylcyclohexanone produce mixtures (8c). of dihydro spiro benzimidazoles (8a,b) and dihydro spiro benzodiazepines (9a,b) in which the spiro benzimidazoles are predominating. The results and spectral data are summarised in Table 2. The spiro benzimidazoles (8a-c) have also been independently synthesised from (1) and the corresponding cyclohexanone in methanol medium. It is evident that the best configurational requirements for the formation of dihydro benzo diazepines are attained in the case of cyclopentanone, cycloheptanone and cyclooctanone.

Comp. No.	m₊p₊ °C	Yield [*] %	IR(KBr) cm ⁻¹	¹ h nmr	MS
8a	210	64	3250-3350 1620	1.11-2.44, m (10H) 3.50-4.33, b (2H) ⁺ 6.22-7.78, m (8H)	292, 249 236, 105 77
9a	87	16	3350, 1650 1625	1.12-2.66, m (19H) 3.88-4.55, b (1H) ⁺ 6.44-8.00, m (8H)	372, 249 236, 105 77
8b	152	66	3200-3400 1620	0.62-2.35, m (12H) 3.10-4.00, b (2H) ⁺ 6.22-7.80, m (8H)	306, 263 236, 105 77
9Ь	98	21	3350, 1650 1630	0.62-2.44, m (22H) 3.55-4.20, b (1H) ⁺ 6.44-7.90, m (8H)	400, 357 236, 105 77
8 C	208	77	3350, 1640 1620	0.80-2.22, m (12H) 3.77-4.44, b (2H) ⁺ 6.22-7.78, m (8H)	306, 249 236, 105 77

Table 2

* Yield based on isolated and crystallised product.

⁺ Exchangeable with D_2O_1 .

EXPERIMENTAL

Melting points are uncorrected. Infrared spectra were recorded on a Perkin Elmer 810 model, ¹H NMR and ¹³C NMR spectra were recorded on Jeol FX90Q, FT NMR and Varian Gemini-200 spectrometers respectively using tetramethylsilane as internal standard and the chemical shifts are reported in ppm ($\boldsymbol{6}$ values). Mass spectra were recorded onVG Micromass 7070H instrument. The spectral data (IR, ¹H NMR) of the compounds are summarised in Tables 1 and 2.

2,3-Cycloalkano-3,4-dihydro(5H)-7-Benzoyl-1,5-benzodiazepine-4-spirocyclo alkane(2-4) General Procedure:

To a solution of 1 (0.5 g; 0.0023 mole) in acetic acid (10 g) appropriate cycloalkanone (0.0046 mole) was added with constant stirring and kept at room temperature for 8 h. The mixture was poured into ice water, neutralised with amonia solution (20%) and the resultant solids were recrystallised from benzene to give the respective dihydro benzodiazepines.

Compound (2): 2,3-Cyclopentano-3,4-dihydro (5H)-7-benzoyl-1,5-benzodiazepine-4-spiro cyclopentane.

Compound (3): 2,3-Cycloheptano-3,4-dihydro (5H)-7-benzoyl-1,5-benzodiazepine-4-spiro cycloheptane. 13 C NMR (CDCl₃) : 175.2 (C=N), 194.8 (CO), 54.0 (C tert), 65.5 (C), 42.3, 38.7, 38.4, 26.0, 22.0, 21.7,

21.2, 29.8, 29.6, 28.4, 26.8 (C_{aliph}), 142.6, 138.4, 135.5, 132.0, 130.9, 129.1, 128.2, 127.7, 127.4, 127.2, 127.0, 118.6 (C_{arom}).

Compound (4): 2,3-Cyclooctano-3,4-dihydro (5H)-7-benzoyl-1,5-benzodiazepine-4-spirocyclooctane.

Condensation of (1) with cyclohexanones : Products 8a-c and 9a,b:

The diamine **1** (0.5 g; 0.0023 mole) and appropriate cyclohexanone (0.0046 mole) dissolved in acetic acid (10 g) and the mixture was kept at room temperature with constant stirring for 8 h. It was diluted with water, neutralised with ammonia solution (20%) and the resultant gummy material was subjected to column chromatography to give compounds dihydro spiro benzodiazepines and dihydro spiro benzimidazoles respectively. When 4-methylcyclohexanone was reacted with (1) exclusively **8c** was obtained.

Compound (8a): 2,3-Dihydro-5(or 6) benzoyl benzimidazole-2-spiro cyclohexane.

Compound (8b): 2,3-Dihydro-5(or 6) benzoyl benzimidazole-2-spiro-3'-methyl cyclohexane.

Compound (8c): 2,3-Dihydro-5(or 6) benzoyl benzimidazole-2-spiro-4'-methyl cyclohexane.

Compound (9a): 2,3-Cyclohexano-3,4-dihydro (5H)-7-benzoyl-1,5-benzodiazepine-4-spiro cyclohexane. ¹³C NMR (CDCl₃) : 175.6 (C=N), 195.0 (CO), 54.2 (C_{tert}), 67.0 (C_{quart}), 42.0, 34.3, 21.7, 21.6, 21.2, 29.0, 28.0, 25.0, 25.4 (C_{aliph}), 144.0, 139.0, 135.8, 135.4, 134.0, 131.4, 129.6, 129.1, 128.7, 128.1, 127.0, 119.6 (C_{arom}).

Compound (9b): 2,3-(3'-Methylcyclohexano)-3,4-dihydro (5H)-7-benzoyl-1,5-benzodiazepine-4-spiro-3'-methylcyclohexane.

Reaction of (1) with cyclohexanones in methanol : General Procedure: To a solution of **1** (0.5 g; 0.0023 mole) in methanol (25 ml) appropriate cyclohexanone (0.0023 mole) was added and the mixture was stirred at room temperature for 5 h. The solvent was concentrated to half of its volume and the resultant solid was filtered, dried and characterised as dihydro spiro benzimidazole (**8a-c**).

Acknowledgement: The authors are thankful to Dr.A.V. Rama Rao, Director, IICT for constant encouragement.

References:

- Reddi, P.S.N., Ratnam, C.V. and Subba Rao, N.V., Indian J.Chem., 1972, 10, 982.
- Hilda, R.M., Arturo, B. and Rosalinda, C., Heterocycles, 1986, 24, 135.

(Received in UK 7 May, 1991)