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# New Synthesis of 10-Alkoxy-5H-dibenz[b,f]azepines

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#### SYNTHETIC COMMUNICATIONS, 24(5), 683-687 (1994)

#### NEW SYNTHESIS OF 10-ALKOXY-5H-DIBENZ[b,f]AZEPINES

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<u>Abstract</u>-The reaction of 5-acetyl-5*H*-dibenz[b,f]azepines with sodiumhypochlorite led to the 5-acetyl-10,11-epoxy-10,11-dihydro-5*H*-dibenz[b,f] azepine (1). The lithium iodide induced rearrangement of 1 gave the keton 2 which was reacted with trialkyl-orthoformates leading to the vinyl ethers 3a,b.

The title compounds play important rule in the preparation of oxcarbazepine(5-carboxamide-10-oxo-10,11-dihydro-5*H*-dibenz[b,f]azepine) one of the recently introduced antiepileptic drugs<sup>1</sup>.

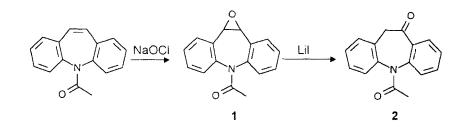
The preparation of these compounds usually occurs *via* the synthesis of the 10,11-dibromo derivative followed by a reaction with sodium alcoholates<sup>2</sup>.

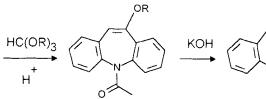
Now we report here a new and simple synthesis using the oxirane 1 and the 5-acetyl-10-oxo-10,11-dihydro-5H-dibenz[b,f]azepine (2) as key intermediates.

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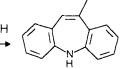
The epoxidation of electrophilic alkenes proceeds well with sodiumhypochlorite in the presence of alumina<sup>3</sup> or silica gel<sup>4</sup>. This phenomenon was utilized when the 5-acetyl-5*H*-dibenz[b,f]azepine was reacted with sodium-hypochlorite in the presence of silica gel yielding the oxirane 1(Scheme). Our previous results<sup>5</sup> showed that one of the best reagents for the rearrangements of 1 to the keto compound 2 is the lithium iodide. The acid-catalysed reactions of 2 with trialkyl orthoformates smoothly gave the vinyl ethers 3a,b which were reacted with potassium-hydroxide in ethylene glycol yielding the 10-alkoxy-5*H*-dibenz[b,f]azepines (4a,b).

The preparation of **4a**,**b** was performed either with or without the isolation and purification of **3a**,**b** and **1**.





3a,b



4a,b

OR

R: a=Me, b=Et H<sup>+</sup>=HCI/i-PrOH, BF<sub>3</sub>·Et<sub>2</sub>O, AlCl<sub>3</sub>, TsOH, NH<sub>4</sub>CI

Scheme

#### EXPERIMENTAL

Melting points are uncorrected. <sup>1</sup>H-NMR spectra were taken on a VARIAN GEMINI-200 instrument (Chemical shifts are in ppm, the spectra were recorded in CDCl<sub>3</sub>. TMS was used as internal standard) MS spectra were determined with a VG TRIO-2 quadropol mass spectrometer EI 70 eV instrument.

A) 5-Acetyl-10,11-epoxy-10,11-dihydro-5H-dibenz[b,f]azepine, 1

5-Acetyl-5*H*-dibenz[b,f]azepine (20.00g ,85mmol) , 40 cm<sup>3</sup> acetonitrile, 300 cm<sup>3</sup> sodium-hypochlorite (4M/dm<sup>3</sup>) and 30 g silica gel (0.06-0.2 mm) were stirred for 1.5 hrs at reflux temperature. The silica gel was filtered at room temperature and was washed with 100 cm<sup>3</sup> CHCl<sub>3</sub>.

The two phases were separated and the aqueous layer was extracted with 100 cm<sup>3</sup> CHCl<sub>3</sub>. The collected organic phase was washed with 100 cm<sup>3</sup> water and dried (Na<sub>2</sub>SO<sub>4</sub>).

This solution was reacted with Lil without further purification (B) or the solvent was evaporated and the crude product was recrystallized from EtOH (yield: 75,1 %, mp: 151-152<sup>0</sup>).

<sup>1</sup>H-NMR:1.95(s,3H), 4.28(s,2H), 7.2-7.5(m,8H)

B) 5-Acetyl-10-oxo-10,11-dihydro-5H-dibenz[b,f]azepine, 2

The oxirane 1 (20.00g, 79 mmol) was solved in 180 cm<sup>3</sup> CHCl<sub>3</sub>. LiI (9.05g 80 mmol) was added and after 20 minutes stirring at reflux temperature the solution was cooled to  $20^{0}$  and was washed with 10 % Na<sub>2</sub>SO<sub>3</sub> (50 cm<sup>3</sup>) and water (100 cm<sup>3</sup>).

The organic phase was dried and the  $CHCl_3$  was evaporated. The crude product was recrystallyzed from EtOH yielding 2 in 85 %. The reaction

performed without isolation of 1 yielded 2 in 67 % based on 5-Acetyl-5*H*-dibenz[b,f]azepine.

mp: 142-143 (ethanol)

<sup>1</sup>H-NMR:2.13(s,3H), 3.89(d,1H,15Hz), 4.36(d,1H,15Hz), 7.2-8.2(m,8H) MS(%):251(20), 209(100), 195(30), 180(55), 152(20)

C) 5-Acetyl-10-alkoxy-5H-dibenz[b,f]azepine, 3a,b

The keto compound 2 (10.00 g, 40 mmol) was solved in the appropriate alcohol (90 cm<sup>3</sup>). To this solution were added the trialkyl-orthoformates (52mmol)and the catalytic amount of the acid (for example 0.1 cm<sup>3</sup> of 14 % HCl in i-PrOH).

After two hours stirring at reflux temperature the solution was cooled to  $20^{\circ}$ , and water was added ( $200 \text{ cm}^3$ ). The resulted oil was extracted with CHCl<sub>3</sub> ( $3x100 \text{ cm}^3$ ). The organic layer was washed with saturated Na<sub>2</sub>CO<sub>3</sub> solution and water and dried (Na<sub>2</sub>SO<sub>4</sub>).

The crude products formed after evaporation of the solvent was recrystallyzed from ethanol.

3a: yield: 90 %, mp:124-250. 1H-NMR:1.91(s,3H), 3.92(virt.d.,4.5Hz,3H)

6.15(virt.d., 13.5Hz, 1H), 7.2-7.8(m,8H)

MS(%):265(50), 222(100), 180(80), 152(20)

**3b**: yield: 82 %, mp:156-57<sup>0</sup>, <sup>1</sup>H-NMR: 1.42(t, 8.0 Hz, 3H), 4.1(m, 2H), 6.35(s, 1H), 7.2-7.8(m, 8H)

MS(%):279(50), 236(60), 208(45), 180(100), 152(25)

D) 10-alkoxy-5H-dibenz[b,f]azepines, 4a,b

The compounds **3** (12mmol) ethylene glycol (4cm<sup>3</sup>) and KOH (0.678g, 12mmol) were stirred for 5hrs at reflux temperature. The crude products **4a,b** were filtered off after cooling to  $0-5^{0}$  for two hours.

The recrystallization from ethanol yielded 4a in 80 % and 4b in 75 %.

**4a**, mp:123-24<sup>0</sup>. <sup>1</sup>H-NMR:3.93(s,3H), 5.15(s,br,1H), 5.86(s,1H)

**4b**, mp:132-33<sup>0</sup>, <sup>1</sup>H-NMR:1.48(t,6.9Hz,3H), **4.07(q,6.9Hz,2H)**,

5.15(s,br,1H), 5.89(s,1H), 6.6-7.6(m,H)

This reaction(D) was performed with the concentrated reaction mixture of C) without the isolation of compound **3** yielding the compounds **4a,b** in 72% and 65 % calculated to the keton **2**.

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