

A SHORT SYNTHESIS OF THE ANTIFUNGAL PRINCIPLE OF *SAPIUM JAPONICUM*

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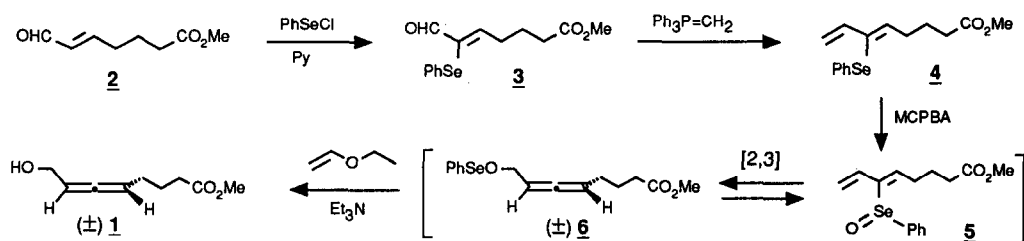
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Abstract: Methyl 8-hydroxy-5,6-octadienoate (**1**), the antifungal principle of *Sapium japonicum*, was prepared via the oxidation and [2,3] sigmatropic rearrangement of a dienyl selenide precursor.

Sapium japonicum, a deciduous tree that is common in parts of Japan, is resistant to certain types of fungal infections. This has been attributed to the presence of methyl 8-hydroxy-5,6-octadienoate (**1**), a strong antifungal agent, in its leaves.¹ Several recent syntheses of allenic alcohol **1** have appeared,² and we now report a concise, new approach to this compound, which is shown in Scheme 1.

The α,β -unsaturated aldehyde **2** is readily available from δ -valerolactone.³ α -Selenenylation of **2** with benzeneselenenyl chloride in pyridine^{4a} provided **3** in 70% yield,^{4b} followed by a Wittig reaction to afford 46% of the dienyl selenide **4**. Oxidation of **4** to the corresponding selenoxide **5** and subsequent [2,3] sigmatropic rearrangement^{5,6} was expected to provide the desired product **1** after hydrolysis of the selenenic ester **6**. However, pyrolysis of the relatively stable selenoxide **5** under a variety of conditions produced complex mixtures of selenenylated products containing little or none of the allenic alcohol **1**. Enol ethers have been previously employed to scavenge selenenic electrophiles formed during selenoxide eliminations.^{7,8} In the present case, transselenenylation of an added enol ether with **6** would drive the rearrangement to completion and prevent competing additions of the selenenic ester, or related Se (II) electrophiles, to the allene moiety. Thus, oxidation of **4** with *m*-chloroperbenzoic acid (MCPBA) and treatment of the crude selenoxide **5** with triethylamine and ethyl vinyl ether afforded 47% of **1**.⁹ Scheme 1 therefore provides a simple, three-step synthesis of the racemic¹⁰ allenic alcohol **1** from aldehyde **2**.

Scheme 1



Acknowledgement

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References and Footnotes

1. Ohigashi, H.; Kawazu, K.; Egawa, H.; Mitsui, T. *Agr. Biol. Chem.* **1972**, *36*, 1399.
2. a) Rodini, D.J.; Snider, B.B. *Tetrahedron Lett.* **1980**, *21*, 3857. b) Huguet, J.; Reyes, M. del C. *Tetrahedron Lett.* **1990**, *31*, 4279. c) Gooding, O.W.; Beard, C.C.; Jackson, D.Y.; Wren, D.L.; Cooper, G.F. *J. Org. Chem.* **1991**, *56*, 1083.
3. Gleason, J.G.; Bryan, D.B.; Kinzig, C.M. *Tetrahedron Lett.* **1980**, *21*, 1129.
4. a) Zima, G.; Liotta, D. *Synth. Commun.* **1979**, *9*, 697. b) Compound **3** was isolated as a pure geometric isomer with the (Z)-configuration, as determined by NOE between the aldehyde proton and the olefinic proton.
5. For a review of [2,3] sigmatropic rearrangements of selenium compounds, see: Reich, H.J. *Organoselenium Chemistry*; Liotta, D., Ed., Wiley: New York, 1987, Ch. 8.
6. Only a few previous examples of allenic alcohol formation by the rearrangement of dienyl selenoxides are known: a) Lerouge, P.; Paulmier, C. *Tetrahedron Lett.* **1984**, *25*, 1987. b) Back, T.G.; Lai, E.K.Y.; Muralidharan, K.R. *J. Org. Chem.* **1990**, *55*, 4595.
7. a) Jaurand, G.; Beau, J.-M.; Sinay, P. *J. Chem. Soc., Chem. Commun.* **1982**, 701. b) Trost, B.M.; Balkovec, J.M.; Mao, M.K.-T. *J. Am. Chem. Soc.* **1983**, *105*, 6755. c) Eaton, P.E.; Andrews, G.D.; Krebs, E.-P.; Kunai, A. *J. Org. Chem.* **1979**, *44*, 2824.
8. The use of either secondary or tertiary amines in the absence of ethyl vinyl ether gave very low yields of **1**. For secondary amines as scavengers of Se (II) electrophiles, see: Reich, H.J.; Wollowitz, S.; Trend, J.E.; Chow, F.; Wendelborn, D.F. *J. Org. Chem.* **1978**, *43*, 1697.
9. Selenide **4** was treated with excess MCPBA in chloroform for 15 min, followed by washing with aqueous NaHCO₃, drying and concentration of the mixture. The crude selenoxide **5** was stirred in ethyl vinyl ether containing ca. 12 eq. of triethylamine at room temperature for 12 h, the solution was evaporated, and the product **1** isolated by flash chromatography over silica-gel. Its identity was confirmed by its IR, ¹H-NMR and high and low resolution mass spectra.
10. An attempted enantioselective preparation of **1** using chiral oxaziridines as oxidants instead of MCPBA resulted in low (< 10%) enantiomeric excesses. This procedure was previously employed for the enantioselective synthesis of allylic alcohols from allylic selenides: Davis, F.A.; Stringer, O.D.; McCauley Jr., J.P. *Tetrahedron* **1985**, *41*, 4747.