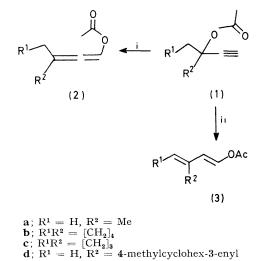
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Isomerisation of Prop-2-ynylic Esters into 1,2- and 1,3-Dienyl Esterst

By RICHARD C. COOKSON,* MICHAEL C. CRAMP, and PHILIP J. PARSONS* (Chemistry Department, University of Southampton, Southampton SO9 5NH)

Summary While the prop-2-ynyl acetates (1) rearrange to the allenyl acetates (2) in boiling benzene containing copper(I) chloride, in the presence of silver trifluoroacetate the butadienyl acetates (3), useful in Diels-Alder reactions, are formed in high yield.

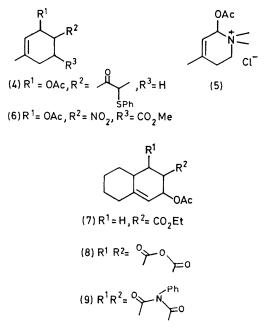
ALLENYL (2), and particularly 1,3-dienyl (3), acetates are valuable synthetic intermediates. We report improved and specific ways of making them from prop-2-ynyl acetates (1).^{1,2}



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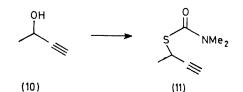
Reagents: i, ${\rm Cu_2Cl_2},$ boiling benzene; ii, ${\rm AgOCOCF_3},$ in boiling benzene.

When the acetate (1a) was heated in benzene containing a catalytic quantity of silver tetrafluoroborate, the isomer (2a) was formed in 60% yield, as reported.² A better procedure, however, was to boil (1a) in benzene in the presence of Cu₂Cl₂, which gave (2a) quantitatively. Heating (1a) in benzene (or toluene) with silver trifluoroacetate surprisingly produced the diene (3a) as the sole product $(7:1 \ trans: cis)$. These conditions also transformed the presumed intermediate (2a) into (3a). The same result was obtained when platinum chloride was used as catalyst in place of silver trifluoroacetate.



The dienes (3) underwent Diels-Alder reactions with suitable electron-poor olefins (Table).

Thus (1a) (3.4 g) in benzene (60 ml) containing silver trifluoroacetate (0.3 g) was heated under reflux for 6 h, cooled, and poured into water, and the organic layer was washed with aqueous NaHCO₃. Evaporation of the solvent gave (3a) (3.1 g, 91%), b.p. 62 °C at 15 mmHg (lit.,³ 66—68 °C at 29 mmHg).



Diene	Dienophile	Solvent	Product ^a	Yield b/%
(3a)	CH ₂ =CHC(:O)CH(SPh)Me	Boiling THF ^c	(4)	52
(3a)	Me ₂ N ⁺ =CH ₂ Cl ⁻	MeCNa	(5)	64
(3a)	O,NCH=CHCO,Me ^e	Benzened	(6)	63
(3b)	Ethyl acrylate	Boiling THF	(7)	64
(3b)	Maleic anhydride	Boiling THF	(8)	72
(3b)	N-Phenylmaleimide	Boiling benzene	(9)	91

TABLE

^a Assumed *endo*-addition, partly confirmed by n.m.r. spectroscopy. ^b Pure isolated product. ^c THF = tetrahydroluran. ¹ At room temperature. ^e *trans*.

† No reprints available.

In contrast, treatment of the prop-2-ynyl alcohol (10) with sodium hydride in dimethylformamide at 5 °C, followed by dimethylthiocarbamoyl chloride gave the product (11) and none of the allene

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¹ H Schlossarczyk W Sieber, M Hesse, H J Hanson, and H Schmid, *Helv Chim Acta*, 1973, 56, 875 ² D G Oelberg and M D Schiavelli, *J Org Chem*, 1977, 42, 1804 ³ J F W Keana, J S Bland, P E Eckler, and V Nelson, *J Org Chem*, 1976, 41, 2124