

Convenient Procedures for the Asymmetric Reduction of 1,4-Diphenylbutane-1,4-dione and Synthesis of 2,5-Diphenylpyrrolidine Derivatives

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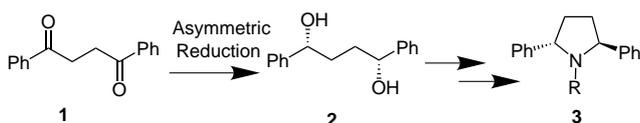
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Abstract: Asymmetric reduction of 1,4-diphenylbutane-1,4-dione (**1**) was carried out using the reducing agents NaBH_4 , $\text{BH}_3\cdot\text{THF}$, and $\text{PhNEt}_2\cdot\text{BH}_3$ in combination with the chiral reagents (*S*)-(-)- α,α -diphenyl-2-pyrrolidinemethanol (**4**) or (*S*)-proline (**5**), in the presence of TMSCl or $\text{B}(\text{OMe})_3$ under various conditions to obtain the corresponding 1,4-diol **2** in 52% to 97% ee. The chiral 1,4-diol **2** was converted to various C_2 -symmetric (*2S,5S*)-2,5-diphenylpyrrolidine derivatives **3a–e** (45 to 75% yield) via the corresponding dimesylate prepared using MsCl and Et_3N .

Key words: asymmetric reduction, chiral 1,4-diols, chiral pyrrolidine derivatives, 1,4-dione

Asymmetric synthesis is one of the major expanding areas of research in organic chemistry. The use of C_2 -symmetric chiral auxiliaries has gained considerable importance in current asymmetric syntheses.¹ Among the various chiral auxiliaries, the 2,5-disubstituted pyrrolidine system is an important class of reagents. The *trans*-2,5-dimethylpyrrolidine was first developed as a useful chiral auxiliary for enantioselective alkylation of enamines derived from it.^{2–4} In this case, the recovery and recycling of chiral auxiliary are somewhat difficult. Therefore, several groups pursued the synthesis of the corresponding diphenyl analogue, a nonvolatile, stable, crystalline compound. The preparation of this system involves the enantioselective reduction of 1,4-diphenylbutane-1,4-dione (**1**) and cyclization of the resultant diol derivative **2** (Scheme 1).



Scheme 1 Synthesis of 2,5-Diphenyl pyrrolidine derivatives

The asymmetric reduction of the diferrocenyl-1,4-diketone to obtain the corresponding diol samples of >98% ee has been reported⁵ using the CBS oxazaborolidine catalyst, which requires considerable care for preparation to achieve good selectivities.⁶ In these reductions, the borane reagents such as $\text{BH}_3\cdot\text{THF}$, and $\text{BH}_3\cdot\text{SMe}_2$, have been used. In recent years, several simplified procedures have appeared^{7,8} for the oxazaborolidine catalysed asymmetric

reduction. For example, an extremely effective oxazaborolidine catalyst can be easily and rapidly prepared in situ from the amino alcohol **4** and trimethyl borate (1 h at r.t.).⁹ Also, it has been reported¹⁰ that (*S*)-proline (**5**) (Figure 1) and $\text{BH}_3\cdot\text{THF}$ reagent combination was effective in enantioselective reduction of acetophenone at 110 °C in 10 minutes. Accordingly, it is desirable to develop simplified, convenient procedures for the asymmetric reduction of the 1,4-diphenylbutane-1,4-dione (**1**). We wish to report here the results of detailed studies on the preparation and use of the chiral 1,4-diol **2** for the synthesis of several chiral 2,5-diphenylpyrrolidine derivatives.

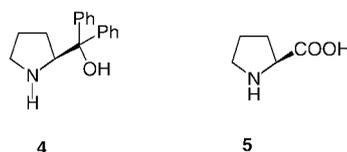
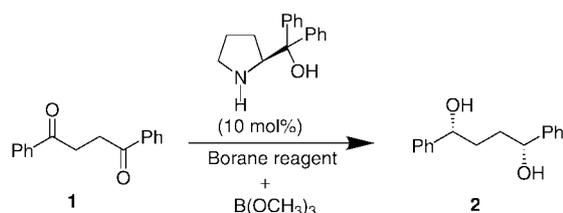


Figure 1 Structures of chiral alcohols **4** and **5**

The required diphenylbutane-1,4-dione (**1**) is easily prepared by following an established protocol via the Friedel–Craft acylation of benzene with fumaryl chloride and subsequent reduction with SnCl_2/HCl .^{11,12}

The borane reagents like $\text{BH}_3\cdot\text{SMe}_2$, $\text{BH}_3\cdot\text{THF}$, diborane, and catecholborane suffer from drawbacks such as thermal decomposition, low concentration, noxious odor or expense. Accordingly, we have examined the asymmetric reduction of 1,4-dione **1** using borane reagents generated in situ using NaBH_4 and amine-boranes in combination with (*S*)-(-)- α,α -diphenyl-2-pyrrolidinemethanol (**4**) and (*S*)-proline (**5**). The results are summarised in Tables 1 and 2. When (*S*)-(-)- α,α -diphenyl-2-pyrrolidinemethanol (**4**) (10 mol%) was used in combination with $\text{NaBH}_4/\text{Me}_3\text{SiCl}$ reagent, the 1,4-diol **2** was obtained in 70% yield, with *dl/meso* ratio 75:25 and in 52% ee (Table 1, entry 1). The results were better when $\text{B}(\text{OMe})_3$ was used in combination with $\text{NaBH}_4/\text{Me}_3\text{SiCl}$, $\text{BH}_3\cdot\text{THF}$ and $\text{PhNEt}_2\cdot\text{BH}_3$ as hydride sources, (Table 1, entries 2, 3 and 4) (Scheme 2).

The chiral (*S*)-(-)- α,α -diphenyl-2-pyrrolidinemethanol (**4**) was prepared from (*S*)-proline (**5**) in several steps.¹⁴ Although good enantioselectivities (Table 1) were achieved using this reagent, we have examined the development of a practical method for the reduction using (*S*)-proline (**5**) itself. Buono et al.¹⁰ have reported that (*S*)-proline (**5**) gives good results in the asymmetric reduction of



Scheme 2 Asymmetric reduction using (*S*)-(-)- α,α -diphenyl-2-pyrrolidinemethanol (**4**) in combination with borane reagent and $B(OMe)_3$ (*dl/*meso = 75:25 to 93:7; 52–97% ee, 70–85% yield)

Table 1 Asymmetric Reduction of 1,4-Dione **1** to 1,4-Diol **2** Using (*S*)-(-)- α,α -Diphenyl-2-pyrrolidinemethanol (**4**)

Entry ^a	Borane Reagent	Yield (%) ^b	<i>dl/</i> meso ratio ^c	ee (%) ^d
1 ^e	$NaBH_4/TMSCl$	70	75:25	52 (1 <i>R</i> ,4 <i>R</i>)
2	$NaBH_4/TMSCl$	70	93:7	93 (1 <i>R</i> ,4 <i>R</i>)
3	$BH_3 \cdot THF$	85	88:12	97 (1 <i>R</i> ,4 <i>R</i>)
4	$PhNEt_2 \cdot BH_3$	70	90:10	97 (1 <i>R</i> ,4 <i>R</i>)

^a All the reactions were carried out using 10 mol% of catalyst **4** and 5 mmol of 1,4-dione **1**.

^b Yields are of 1,4-diol **2** isolated by column chromatography on silica gel using hexane–EtOAc as eluent.

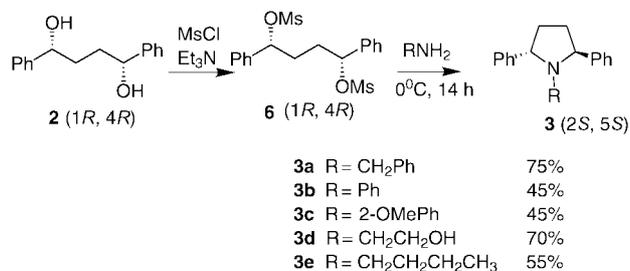
^c The *dl/*meso ratios were calculated from ¹³C NMR data.

^d All ee values reported here are based on maximum $[\alpha]_D^{21} - 58.5$ ($c = 1.01$, $CHCl_3$, >98% ee) for (1*S*,4*S*)-(-)-**2**¹³ and the % of ee was calculated based on amount of *dl* (i.e. *R,R*- and *S,S*-isomers) present in the mixture.

^e Reaction was performed without using $B(OMe)_3$.

acetophenone at refluxing conditions. Accordingly, we have examined the asymmetric reduction of the 1,4-dione **1** using (*S*)-proline **5** along with various hydride sources. The results are summarised in Table 2.

Unfortunately, the 1,4-diol **2** was obtained only with lower selectivities using (*S*)-proline (**5**). However, the mixture of nonracemic and *meso* diastereomers of **2** has been readily purified to obtain the samples of >95 ee using (*S*)-proline (**5**) and $B(OH)_3$.¹⁵ The resultant 1,4-diol **2** (>95% ee) was readily cyclised using primary amines to obtain the corresponding 2,5-diphenylpyrrolidine derivatives **3a–e** via the dimesylate **6** (Scheme 3).



Scheme 3 Synthesis of (2*S*,5*S*)-*N*-alkyl and *N*-aryldiphenylpyrrolidine derivatives

Table 2 Asymmetric Reduction of 1,4-Dione **1** to 1,4-Diol **2** Using (*S*)-Proline (**5**)

Entry	Borane Reagent	Yield (%) ^a	<i>dl/</i> meso Ratio ^b	ee (%) ^c
1 ^{d,e}	$NaBH_4$	60	22:78	64 (1 <i>S</i> ,4 <i>S</i>)
2 ^{d,f}	$BH_3 \cdot THF^g$	60	62:38	81 (1 <i>R</i> ,4 <i>R</i>)
3 ^{f,h}	$BH_3 \cdot THF^i$	60	76:24	53 (1 <i>R</i> ,4 <i>R</i>)
4 ^{f,h}	$PhNEt_2 \cdot BH_3$	75	73:27	85 (1 <i>R</i> ,4 <i>R</i>)

^a Yields are of 1,4-diol **2** isolated by column chromatography on silica gel using hexane–EtOAc as eluent.

^b The *dl/*meso ratios were calculated from ¹³C NMR data.

^c All ee values reported here are based on maximum $[\alpha]_D^{21} - 58.5$ ($c = 1.01$, $CHCl_3$, >98% ee) for (1*S*,4*S*)-(-)-**2**¹³ and the % of ee was calculated based on amount of *dl* (i.e. *R,R*- and *S,S*-isomers) present in the mixture.

^d 100 mol% of **5** was used.

^e 2.5 mmol of the 1,4-dione **1** was used.

^f 2 mmol of the 1,4-dione **1** was used.

^g (*S*)-Proline (**5**) was added to $BH_3 \cdot THF$ prepared in situ using $NaBH_4/I_2$.

^h 20 mol% of **5** was used.

ⁱ $BH_3 \cdot THF$ prepared in situ using $NaBH_4/I_2$ was added to (*S*)-proline (**5**).

In conclusion, the enantioselective reduction of 1,4-diphenylbutane-1,4-dione (**1**) to the corresponding 1,4-diol **2** is conveniently carried out using (*S*)-(-)- α,α -diphenyl-2-pyrrolidinemethanol (**4**) and (*S*)-proline (**5**) and easy to handle borane reagents. The convenient synthetic procedures described here for the preparation of the chiral 2,5-diphenylpyrrolidine system **3** should facilitate the syntheses and application of these derivatives.

Reduction of 1,4-Diphenylbutane-1,4-dione (**1**) Using $BH_3 \cdot THF$ Complex/(*S*)-(-)- α,α -Diphenyl-2-pyrrolidinemethanol (**4**)/ $B(OMe)_3$ (10 mol%) Reagent Combination (Entry 3, Table 1)

$NaBH_4$ (0.76 g, 20 mmol) was suspended in anhyd THF (40 mL) under N_2 . The reaction mixture was cooled to 0 °C and a solution of I_2 (2.1 g, 8.3 mmol) in THF (25 mL) was added dropwise during 2.5 h at 0 °C using a pressure equalizing dropping funnel. A solution of **4** (0.8 mmol) and $B(OMe)_3$ (1 mmol) in THF (10 mL) was added and the mixture was stirred for 10 min. To this mixture, was added slowly 1,2-dibenzoylthane (1 g, 4.6 mmol) dissolved in THF (15 mL) with a pressure equalizing dropping funnel during 1 h at 10 °C and the mixture was further stirred at 25 °C for 1 h. The mixture was hydrolysed using 2 N HCl (15 mL) and the organic layer was separated. The aqueous layer was extracted with Et_2O . The combined organic extracts were washed with brine (10 mL) and dried ($MgSO_4$). After evaporation, the crude product was purified on a silica gel column using hexane–EtOAc (80:20) as eluent to obtain the (+)-1,4-diol **2** in 97% ee; yield: 1.02 g (85%); mp 63–65 °C; *dl/*meso ratio = 88:12; $[\alpha]_D^{21} + 49.8$ ($c = 0.542$, $CHCl_3$) [Lit.¹³ $[\alpha]_D^{21} - 58.5$ ($c = 1.01$, $CHCl_3$, >98% ee) for (1*S*,2*S*)-(-)-**2**].

¹H NMR (200 MHz, $CDCl_3$): $\delta = 1.2$ –1.9 (m, 4 H), 2.8 (s, 2 H), 4.6–4.7 (m, 2 H), 7.2–7.4 (m, 10 H).

¹³C NMR (50 MHz, $CDCl_3$): $\delta = 35.1$ (*meso*), 36.0 (*dl*), 73.9 (*meso*), 74.4 (*dl*), 126.0, 127.4, 128.4, 144.8.

Reduction of 1,4-Diphenylbutane-1,4-dione (1) with NaBH₄/(S)-Proline (5) Complex (Entry 1, Table 2)

NaBH₄ (0.19 g, 5 mmol) and (S)-proline (**5**; 0.58 g, 5 mmol) were taken in THF (10 mL) and stirred for 2 h at r.t. 1,2-Dibenzoyl ethane (0.59 g, 2.5 mmol) in THF (10 mL) was added to this suspension of sodium L-proline borane complex and the mixture was stirred at 40 °C for 4 d. The excess reagent was decomposed with H₂O and the mixture was concentrated under reduced pressure and extracted with Et₂O. The organic extracts were washed with 10% HCl (10 mL), aq sat. NaHCO₃ solution (10 mL), brine (10 mL), and dried (MgSO₄). The solvent was evaporated and the crude product was purified on a silica gel column using hexane–EtOAc (80:20) as eluent to obtain the (–)-1,4-diol (–)-**2** in 64% ee, yield: 0.36 g (60%); *dl/meso* ratio = 22:78; [α]_D²¹ –8.0 (*c* = 0.25, CHCl₃) {Lit.¹³ [α]_D²¹ –58.5 (*c* = 1.01, CHCl₃), >98% ee for (1*S*,4*S*)-(–)-**2**}.

¹H NMR (200 MHz, CDCl₃): δ = 1.7–2.0 (m, 4 H), 2.4 (s, 2 H), 4.6–4.8 (m, 2 H), 7.2–7.4 (m, 10 H).

¹³C NMR (50 MHz, CDCl₃): δ = 35.0 (*meso*), 36.0 (*dl*), 74.1 (*meso*), 74.5 (*dl*), 125.9, 127.4, 128.4, 144.6.

Reduction of 1,4-Diphenylbutane-1,4-dione (1) using (S)-Proline (5) (20 mol%) and *N,N*-Diethylaniline-BH₃ Complex (Entry 4, Table 2)

To a stirred suspension of (S)-proline (**5**; 0.095 g, 0.83 mmol) in toluene (7 mL) was added a 1 M toluene solution of *N,N*-diethylaniline-BH₃ (0.83 mL, 0.83 mmol) at 25 °C. After stirring for a further 10 min, the reaction mixture was heated to reflux (110 °C). 1,2-Dibenzoyl ethane (0.476 g, 2 mmol) in THF (10 mL) was added, followed by dropwise addition of a 1 M toluene solution of *N,N*-diethylaniline-BH₃ (4 mL, 4 mmol) over 15 min. The mixture was further stirred for 0.5 h. After cooling to 25 °C, Et₂O (20 mL) was added. After work-up and purification, the (+)-1,4-diol (+)-**2** was obtained in 85% ee; yield: 0.36 g (75%); *dl/meso* ratio = 73:27, [α]_D²¹ +36.2 (*c* = 0.276, CHCl₃) {Lit.¹³ [α]_D²¹ –58.5 (*c* = 1.01, CHCl₃), >98% ee for (1*S*,2*S*)-(–)-**2**}.

¹H NMR (200 MHz, CDCl₃): δ = 1.3–1.9 (m, 4 H), 2.6 (s, 2 H), 4.6–4.8 (m, 2 H), 7.2–7.4 (m, 10 H).

¹³C NMR (50 MHz, CDCl₃): δ = 35.0 (*meso*), 35.9 (*dl*), 74.0 (*meso*), 74.4 (*dl*), 126.0, 127.4, 128.4, 144.7.

(1*R*,4*R*)-1,4-Bis(methanesulfonyloxy)-1,4-diphenylbutane (6)¹³

To methanesulfonyl chloride (0.4 mL, 5.3 mmol) in CH₂Cl₂ (20 mL) at –20 °C was added a solution of (1*R*,4*R*)-1,4-diphenylbutane-1,4-diol (**2**; 0.5 g, 2.06 mmol, 96% ee) and Et₃N (0.87 mL, 6.2 mmol) in CH₂Cl₂ (20 mL). The reaction mixture was stirred for 1.45 h at –20 °C and then quenched with aq sat. NH₄Cl (2 mL). It was brought to 25 °C and concentrated to approximately 17 mL. The solution was then diluted with EtOAc (80 mL) and washed successively with a mixture of H₂O–brine–aq sat. Na₂CO₃ (1:2:1, 2 × 20 mL), and aq sat. NaHCO₃ (2 × 20 mL). The organic layer was dried (MgSO₄), filtered through Celite and concentrated to approximately 8 mL. The solution was then cooled to 0 °C. The crude dimesylate was precipitated out by dropwise addition of hexane (80 mL). The resulting solid was filtered, dried and immediately used for the next reaction without further purification.

***N*-Substituted 2,5-Diphenylpyrrolidines; (2*S*,5*S*)-*N*-Benzyl-2,5-diphenylpyrrolidine (3a); Typical Procedure**

Benzylamine (21 mL, 196 mmol) was added at 0 °C to (1*R*,4*R*)-1,4-bis(methanesulfonyloxy)-1,4-diphenylbutane (**6**; 0.32 g, 1 mmol) and the mixture was stirred at 0 °C for 14 h. After warming to 25 °C, the excess benzylamine was evaporated and the residue was dissolved in Et₂O (25 mL). The contents were successively washed with aq sat. NaHCO₃ (10 mL), brine (10 mL), dried (MgSO₄), and concentrated to afford the crude product as a gum. The crude prod-

uct was purified on a silica gel column using hexane as eluent; yield: 0.234 g (75%); [α]_D²¹ –126 (*c* = 0.435, CHCl₃).

IR (neat): 3061, 1602 cm^{–1}.

¹H NMR (200 MHz, CDCl₃): δ = 1.98–2.2 (m, 2 H), 2.4–2.8 (m, 2 H), 3.15 (d, *J* = 14 Hz, 1 H), 3.65 (d, *J* = 14 Hz, 1 H), 4.28 (m, 2 H), 7.1–7.3 (m, 15 H).

¹³C NMR (50 MHz, CDCl₃): δ = 33.4, 51.1, 65.5, 126.4, 127.0, 128.0, 128.2, 128.3, 128.4, 140.2, 144.1.

Anal. Calcd for C₂₃H₂₃N: C, 88.25; H, 7.40; N, 4.47. Found: C, 88.50; H, 7.65; N, 4.65.

(2*S*,5*S*)-*N*-Phenyl-2,5-diphenylpyrrolidine (3b)

Yield: 0.134 g (45%); mp 190–194 °C; [α]_D²¹ –31 (*c* = 0.236, CHCl₃).

IR (KBr): 3020, 2935, 1597, 1502, 1359, 748, 698 cm^{–1}.

¹H NMR (200 MHz, CDCl₃): δ = 1.7–1.9 (m, 2 H), 2.5–2.7 (m, 2 H), 5.2–5.4 (m, 2 H), 6.4–6.7 (m, 3 H), 7.0–7.6 (m, 12 H).

¹³C NMR (50 MHz, CDCl₃): δ = 32.5, 63.3, 114.0, 115.7, 126.2, 126.7, 128.6, 128.8, 144.0, 145.1.

MS (EI): *m/z* = 299 (M⁺).

(2*S*,5*S*)-*N*-(2-Methoxyphenyl)-2,5-diphenylpyrrolidine (3c)

Yield: 0.148 g (45%); [α]_D²¹ –15 (*c* = 0.100, CHCl₃).

IR (KBr): 3052, 2965, 1596, 740, 700 cm^{–1}.

¹H NMR (200 MHz, CDCl₃): δ = 1.9–2.1 (m, 2 H), 2.3–2.5 (m, 2 H), 3.4 (s, 3 H), 4.8–4.9 (t, 2 H, *J* = ? Hz), 6.6–6.9 (m, 4 H), 7.2–7.6 (m, 10 H).

¹³C NMR (50 MHz, CDCl₃): δ = 34.5, 55.2, 67.8, 112.6, 120.8, 121.8, 122.0, 126.1, 126.5, 128.0, 139.0, 146.2, 153.1.

MS (EI): *m/z* = 329 (M⁺).

(2*S*,5*S*)-*N*-(2-Hydroxyethyl)-2,5-diphenylpyrrolidine (3d)

Yield: 0.186 g (70%); [α]_D²¹ –123 (*c* = 0.155, CHCl₃).

IR (neat): 3375, 1602 cm^{–1}.

¹H NMR (200 MHz, CDCl₃): δ = 1.8–2.1 (m, 2 H), 2.3–2.8 (m, 5 H), 3.0–3.3 (m, 1 H), 3.35–3.6 (m, 1 H), 4.2–4.5 (m, 2 H), 7.1–7.6 (m, 10 H).

¹³C NMR (50 MHz, CDCl₃): δ = 33.5, 49.1, 59.4, 66.5, 127.2, 127.7, 128.6, 144.1.

Anal. Calcd for C₁₈H₂₁NO: C, 80.97; H, 7.92; N, 5.24. Found: C, 81.35; H, 7.98; N, 5.50.

(2*S*,5*S*)-*N*-Butyl-2,5-diphenylpyrrolidine (3e)

Yield: 0.153 g (55%); [α]_D²¹ –90 (*c* = 0.140, CHCl₃).

IR (neat): 3063, 2959, 1602 cm^{–1}.

¹H NMR (200 MHz, CDCl₃): δ = 0.5–0.71 (t, 3 H), 0.8–1.2 (m, 4 H), 1.7–2.0 (m, 2 H), 2.1–2.3 (m, 2 H), 2.4–2.6 (t, 2 H, *J* = ? Hz), 3.7–3.9 (t, 2 H, *J* = ? Hz), 7.2–7.6 (m, 10 H)

¹³C NMR (50 MHz, CDCl₃): δ = 13.7, 20.5, 29.3, 34.9, 53.0, 69.4, 126.6, 127.2, 128.2, 146.2

MS (EI): *m/z* = 279 (M⁺).

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