

Simple Generation of a Reactive Glycosyl-lithium Derivative

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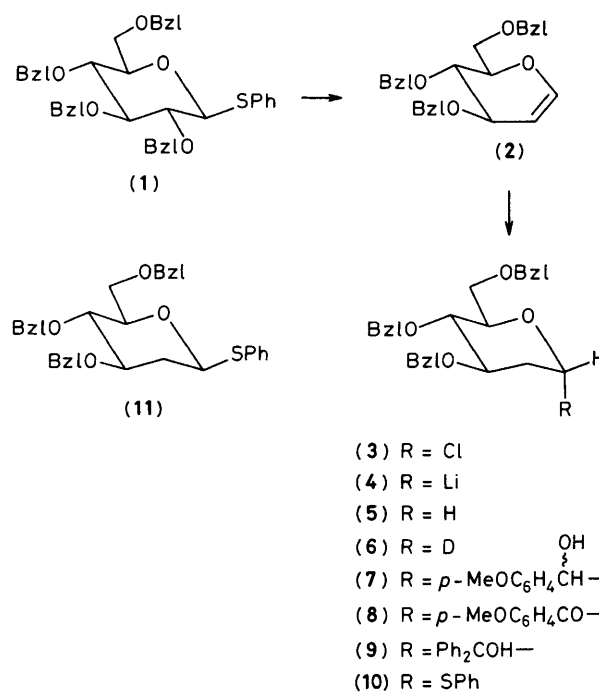
Two-step hydrolithiation of 3,4,6-tri-*O*-benzyl- β -glucal (hydrochlorination and lithium naphthalenide reductive lithiation) gives a reactive glycosyl-lithium derivative which is shown to be a precursor of *C*-glycosides.

A number of natural products regarded as *C*-glycopyranosyl derivatives have, because of their challenging structures,¹ caused intense interest in the field of carbohydrate chemistry. Most of the syntheses of *C*-glycosides developed so far² rely on the electrophilic character of the anomeric centre of a carbohydrate. Less conventional recent approaches derive either from the generation and trapping of a glucosyl radical³ or from the use of 1-deoxy-1-nitro sugars⁴ where the nitro group acts as an anion stabilizing substituent for the generation of a reactive anomeric anion in mild conditions.⁵ We report herein on the generation and reactivity of 3,4,6-tri-*O*-benzyl-2-deoxy- α -D-*arabino*-hexopyranosyl-lithium (4), the first example to-date of a *C*-glycopyranosyl-lithium reagent.

Treatment of phenyl 2,3,4,6-tetra-*O*-benzyl-1-thio- β -D-glucopyranoside (1)⁶ with 2 equiv. of lithium naphthalenide (LN) [tetrahydrofuran (THF), -78°C , 15 min] resulted in quantitative formation of 3,4,6-tri-*O*-benzyl-D-glucal (2).⁷ This result indicates that highly selective reductive lithiation⁸ occurred at the anomeric centre of (1), followed by a fast β -elimination.⁹

Hydrochlorination of (2) (toluene, HCl gas, 0°C , 10 min) gave a quantitative yield of 3,4,6-tri-*O*-benzyl-2-deoxy- α -D-*arabino*-hexopyranosyl chloride (3) which, upon reductive lithiation (2 equiv. LN, THF, -78°C , 3 min) and quenching with D_2O at -78°C , provided selectively the deuteriated derivative (6)[†] [80% from (2)], $[\alpha]_{\text{D}} +19^{\circ}$, most probably from the glycopyranosyl lithium (4).¹⁰ The axial addition of deuterium was shown by ^1H n.m.r. spectroscopy (90 MHz, CDCl_3): δ 3.92 (1 H, dd, $J_{1,2ax}$ 5.0, $J_{1,2eq}$ 2 Hz, 1-Heq).

Treatment of (4) with *p*-anisaldehyde gave a 3 : 1 diastereoisomeric mixture (7) (65%), from which the two pure diastereoisomers were separated on silica gel (toluene-ethyl acetate, 85 : 15 v/v): major product, $[\alpha]_{\text{D}} +16^{\circ}$; minor product, m.p. 65°C (from diethyl ether-hexane), $[\alpha]_{\text{D}} +9^{\circ}$. Oxidation (pyridinium chlorochromate, CH_2Cl_2 , room temperature, 5 h) of the mixture (7) gave a single product (8) (75%), m.p. 74 – 75°C (from hexane), $[\alpha]_{\text{D}} +89^{\circ}$, ^1H n.m.r. (CDCl_3): δ 5.10 (1 H, dd, $J_{1,2ax}$ 6, $J_{1,2eq}$ 2 Hz, 1-H). Treatment of (4)[‡] with



Bzl = CH_2Ph

benzophenone gave (9) (30%), $[\alpha]_{\text{D}} +15^{\circ}$, ^1H n.m.r. (CDCl_3): δ 1.68 (1 H, m, $J_{1,2}$ 4, $J_{2,3}$ 11, $J_{2eq,2ax}$ 13.5 Hz, 2-H_{ax}). In this case, a large amount (30%) of (5) was also isolated. Similar results were obtained when phenyl 3,4,6-tri-*O*-benzyl-2-deoxy-1-thio- α -D-*arabino*-hexopyranoside (10), $[\alpha]_{\text{D}} +188^{\circ}$, or its β anomer (11), $[\alpha]_{\text{D}} -36^{\circ}$, were reductively lithiated¹¹ (2 equiv. LN, THF, -78°C , 45 min). Therefore the reductively generated glycosyl-lithium (4) appears to couple at -78°C so that electrophiles are introduced in the axial position.

In conclusion, the two-step hydrolithiation of tri-*O*-benzyl-D-glucal reverses the characteristic electrophilicity of the

[†] All new compounds gave satisfactory microanalytical and spectral data. Optical rotations were measured for solutions in chloroform at 20°C . The yields of the reactions have not been optimized.

[‡] Reaction of (4) with chlorotrimethylsilane gave a single Si-glycoside (36%), $[\alpha]_{\text{D}} +15^{\circ}$, whose anomers has not been ascertained.

§ (10) and (11) were prepared in four steps from (3) [(a) acetone- H_2O , Ag_2CO_3 ; (b) Ac_2O -pyridine; (c) PhSH , $\text{BF}_3 \cdot \text{Et}_2\text{O}$, CH_2Cl_2 , room temperature, 15 min; (d) silica gel column (CH_2Cl_2 -hexane, 7 : 3, v/v)].

anomeric centre and thus forms a novel route to C-glycosides which complements existing procedures. The use of a phenyl thioglycoside as an excellent precursor to a glycal should also find useful applications.

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