

www.elsevier.nl/locate/carres

CARBOHYDRATE RESEARCH

Carbohydrate Research 319 (1999) 192-198

Note

Cyclic sulfates in the regioselective synthesis of 5- and 6-amino and 5- and 6-fluorohexofuranoses

José Fuentes *, Manuel Angulo, M. Angeles Pradera

Departamento de Química Orgánica, Facultad de Química, Universidad de Sevilla, Apartado de Correos No 553, E-41071 Seville, Spain

Received 18 March 1999; revised 11 May 1999; accepted 20 May 1999

Abstract

Cyclic sulfates of 5,6- and 3,5-D-glucofuranose were used to prepare 6- and 5-azido (amino) and 6- and 5-fluoro derivatives of 1,2-*O*-isopropylidenehexofuranoses (D-*gluco* and L-*ido* configurations). The reactions were completely regioselective and, in the case of stereogenic centers (substitution at C-5), completely stereoselective. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: Cyclic sulfates; Amino sugars; Azido sugars; Fluoro sugars

The cyclic sulfates of *vic*-diols are synthetic equivalents of epoxides and cyclic sulfites which have been widely used in organic synthesis [1,2]. However, their applications in carbohydrate chemistry are limited [1,3–11].

In a recent letter [12], we have communicated the preliminary results on the use of glucofuranose cyclic sulfates in the preparation of 6- and 5-azido(fluoro) aldofuranose derivatives. In this paper, we go further into the preparation of 5,6- and 3,5-glucofuranose cyclic sulfates and the nucleophilic opening of the ester rings with sodium azide and different fluoronucleophiles.

The previously described [3] D-glucofuranose (1, 2) and D-allofuranose (4) 5,6-cyclic sulfates and the new D-glucofuranose 5,6- (3)and 3,5- (5) cyclic sulfates were used as starting materials. Compounds 3 and 5 were prepared from the corresponding diols [13,14] through the cyclic sulfites, which were obtained as mixtures of diastereomers that were directly oxidized [2] with the reagent sodium periodate-ruthenium trichloride hydrate.



The nucleophilic opening of the sulfate ring of compounds 1-4 with sodium azide produced the 6-azido-6-deoxy-D-glucofuranose derivatives 6-9 in high yields in a completely regioselective manner. In the case of the 3-*O*mesyl derivative **8**, no displacement of the mesyloxy group at C-3 was observed under the conditions used for the ring opening of the cyclic sulfate **3**; this mesyloxy group was stable even on treatment of **3** or **8** with sodium azide at 105 °C in DMF for 6 h. The nucle-

^{*} Corresponding author. Fax: + 34-95-462-4960.

E-mail address: jfuentes@cica.es (J. Fuentes)

ophilic sulfate ring opening of the non-vicinal cyclic sulfate 5, using the same work-up as for the preparation of 6-9, produced exclusively the 5-azido-5-deoxy-L-idofuranose derivative 11, as seen from the downfield shift of the H-5 and C-5 resonances.



Conventional Zemplén deacetylation of 6 and 11 afforded the known azido sugar derivatives 10 and 12, respectively [15].

Catalytic (10% Pd–C) hydrogenation of the azidoglycofuranose derivatives 8, 9, and 12 yielded quantitatively the 6-amino-D-gluco-furanose (13), the 3,6-diamino-D-allofuranose (14), and the 5-amino-L-idofuranose (15) derivatives, respectively.

This method to introduce an azido or amino function into a sugar molecule through a cyclic sulfate is a valuable option for the synthesis of 6-amino sugars. It is particularly interesting in the case of 5-amino-5-deoxyfuranoses because there are no problems of mixtures of epimers, as in the reduction of oxime derivatives [16], or of competition between elimination and substitution, as in methods based on inversion of the C-5 configuration [15], and the number of steps from the diol derivative is only three, with high overall yields.

The 6-fluoro-6-deoxy-D-gluco-furanose (16-18) and D-allo-furanose (19) were prepared analogously by sulfate ring opening of 1-4 using tetraethylammonium fluoride dihydrate as a nucleophile. Subsequent reduction of the 3-azido-6-fluoro-D-allofuranoside 19 gave the amino-fluoro derivative 20 in a high yield.

The reaction of the cyclic sulfate 5 with tetraethylammonium fluoride dihydrate led to complex mixtures of decomposition products. In the case of anhydrous tetrabutylammonium fluoride, the reaction was strongly dependent on the nucleophile/starting material ratio, on temperature, and on reaction time. Both ¹H and ¹³C NMR spectra showed the formation of 21 as a minor product, and a major elimination compound for which the ¹H NMR spectrum [6.04 (d, 1 H, J_{1.2} 3.4 Hz, H-1), 5.42 (s, 1 H, H-3), 5.17 (dd, 1 H, J_{5.6a} 9.7, J_{5.6b} 6.5, H-5), 5.05 (d, 1 H, H-2), 4.81 (dd, 1 H, J_{6a.6b} 12.5, H-6a), 4.70 (dd, 1 H, H-6b)] and reported data on related eliminations [10] suggest structure 23. The best ratio (23:21, 5:3) of substitution product was obtained when the reaction was conducted in DMF at 0 °C for 40 h. When tris(dimethylamino)sulfur(trimethylsilyl)difluoride (TAS-F) was used as a nucleophile, practically only the 5-fluoroderivative 21 was formed (23:21 ratio, measured by digital integration of the ¹H NMR signals, \approx 1:100). Treatment of **21** with sulfuric acid in THF produced quantitatively the 5-fluoro-Lidofuranose derivative 22.

The NMR spectra of the 6-fluoroglycofuranose derivatives **16–20** had the signals for H-6a and H-6b at 4.51–4.71 ppm (${}^{2}J_{\rm H,F} \approx 47$ Hz) and for C-6 at 83.2–85.0 ppm (${}^{1}J_{\rm C,F} \approx 169$ Hz) in agreement with the presence of a fluorine atom. In the case of the 5-fluoro-L-idofuranoses **21** and **22**, the resonances for H-5 and C-5 appeared at low field (4.90 and 90.3 ppm, respectively) and the values for ${}^{2}J_{\rm H5,F}$ and ${}^{1}J_{\rm C5,F}$ were ≈ 90.4 and ≈ 175 Hz, respectively, also supporting the introduction of the fluorine atom.

In conclusion, we describe a new, short and inexpensive method for the regio- and stereoselective preparation of 6- and 5-azido, 6- and 5-amino, and 6- and 5-fluoro glycofuranoses. The method is especially useful in the case of the 5-substituted-L-*ido* derivatives.

1. Experimental

General methods.—Melting points are uncorrected. Optical rotations were measured at 24 ± 4 °C in 1 cm cells. FTIR spectra were recorded for KBr discs. ¹H and ¹³C NMR spectra were obtained for solutions in CDCl₃ or MeOH- d_4 . Assignments were confirmed by homonuclear 2D COSY and heteronuclear 2D correlated experiments. FAB mass spectra were recorded with a Kratos MS-80RFA instrument with a resolution of 1000 (10% valley definition). HREIMS (70 eV), HRCIMS (150 eV) experiments were performed with a Micromass AutoSpecQ instrument with a resolution of 10,000 (5% valley definition). Ions were produced by a beam of xenon atoms (6-7 keV) using a matrix consisting of thioglycerol or 3-nitrobenzyl alcohol and NaI as salt. TLC was performed on Silica Gel HF_{254} , with detection by UV light or charring with H_2SO_4 . Silica Gel 60 (E. Merck, 230–400 used for preparative chromesh) was matography.

Preparation of cyclic sulfates 3 and 5.—To an ice-cooled and magnetically stirred solution of the corresponding diol [11,12] (1 mmol) and pyridine (4.08 mmol) in dry CH₂Cl₂ (5 mL), a solution of thionyl chloride (2 mmol) in CH₂Cl₂ (0.4 mL) was added dropwise over a period of 10 min. Stirring was continued at 0 °C until TLC analysis showed complete conversion of the starting material into cyclic sulfite. The mixture was diluted with CH₂Cl₂ (15 mL), washed three times with water, dried $(MgSO_4)$, and concentrated. To a solution (7 mL) of the residue in (2:2:3) CH₂Cl₂-MeCNwater, NaIO₄ (2 equiv) and RuCl₃·H₂O (2 mg) were added. The resulting mixture was stirred for 20-60 min at rt. After dilution with CH_2Cl_2 (15 mL), the organic layer was separated and washed with water, dried (MgSO₄), and concentrated. The corresponding cyclic sulfates were purified bv column chromatography.

1,2-O-Isopropylidene-3-O-mesyl-α-D-glucofuranose-5,6-cyclic sulfate (3). Column chromatography (1:1 Et₂O-light petroleum ether) of the residue gave **3** as a solid (252 mg, 70%) which crystallized from Et₂O-light petroleum ether: mp 154–157 °C; $[\alpha]_D^{21} = -40^\circ$ (*c* 1.0, CH₂Cl₂); IR v_{max} 1385, 1215, 1175, 1094, 1026 and 980 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 5.99 (d, 1 H, $J_{1,2}$ 3.4 Hz, H-1), 5.09 (d, 1 H, $J_{3,4}$ 3.1 Hz, H-3), 5.06 (m, 1 H, H-5), 4.78 (d, 1 H, H-2), 4.85 (dd, 1 H, $J_{5,6a}$ 6.3, $J_{6a,6b}$ 9.3 Hz, H-6a), 4.78 (dd, 1 H, $J_{5,6b}$ 5.5 Hz, H-6b), 4.58 (dd, 1 H, $J_{4,5}$ 8.4 Hz, H-4), 3.12 (s, 3 H, Ms), 1.53, 1.34 (2 s, each 3 H, CMe_2) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 113.5 (CMe₂), 105.3 (C-1), 83.4 (C-2), 80.8 (C-3), 77.1 (C-4), 76.2 (C-5), 70.3 (C-6), 38.2 (OMs), 26.6, 26.0 (CMe₂) ppm; FABMS m/z 383 ([M + Na]⁺). Anal. Calcd for C₁₀H₁₆O₁₀S₂: C, 33.33; H, 4.47. Found: C, 33.22; H, 4.39.

6-O-Acetyl-1,2-O-isopropylidene-α-D-glucofuranose 3.5-cvclic sulfate (5). Column chromatography (3:1 Et₂O-light petroleum ether) of the residue gave 5 as a solid (292 mg, 90%), which crystallized from Et₂O-light petroleum ether: mp 116–118 °C (dec); $[\alpha]_D^{24} = +38^\circ$ (c 1.0, CH₂Cl₂); IR v_{max} 1753, 1398, 1254, 1204, 1094, 1042 and 866 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 6.06 (d, 1 H, $J_{1,2}$ 3.7 Hz, H-1), 5.19 (d, 1 H, $J_{3,4}$ 2.5 Hz, H-3), 5.00 (t d, 1 H, $J_{4,5}$ 2.5, $J_{5,6a} = J_{5,6b}$ 6.4 Hz, H-5), 4.79 (d, 1 H, H-2), 4.57 (m, 1 H, H-6a), 4.54 (m, 1 H, H-6b), 4.35 (t, 1 H, H-4), 2.13 (s, 3 H, COCH₃), 1.53, 1.36 (2 s, each 3 H, CMe₂) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 169.9 (CO), 113.2 (CMe₂), 104.8 (C-1), 85.1 (C-3), 82.9 (C-5), 82.5 (C-2), 71.7 (C-4), 61.5 (C-6), 26.5, 26.0, (CMe₂), 20.4 (COCH₃) ppm; FABMS: m/z 347 ([M + Na]⁺). Anal. Calcd for C₁₁H₁₆O₉S: C, 40.74; H, 4.97. Found: C, 40.78; H, 5.09.

General procedure for the preparation of azido derivatives 6-12.—A solution of the corresponding cyclic sulfate 1-5 (1 mmol) and sodium azide (2 mmol) in DMF (5 mL) at 50 °C was stirred for 10-60 min until total consumption of the starting material, monitoring by TLC, then the mixture was concentrated and dissolved in THF (5 mL). Concd H_2SO_4 (50 µL) and water (18 µL) were added and the mixture was stirred for 20-60 min at rt. The reaction mixture was diluted with EtOAc (15 mL) and washed with sat aq NaHCO₃. The organic layer was separated, washed three times with water, dried $(MgSO_4)$, and concentrated to dryness. The residue was purified by column chromatography (1:2 Et₂O-light petroleum ether) to give 6-9 or 11 (1:1 to 3:1 Et₂O-light petroleum ether), respectively. Zemplén deacetylation of 9 and 11 yielded 10 and 12.

3-O-Acetyl-6-azido-6-deoxy-1,2-O-isopropy*lidene-* α -D-*glucofuranose* (6).—Syrup (238 mg, 83%); $[\alpha]_D^{23} + 9^\circ$ (c 0.8, CH₂Cl₂); IR v_{max} 3472, 2104, 1742, 1375, 1228, 1084, 1024 and 864 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 5.89 (d, 1 H, J_{1.2} 3.6 Hz, H-1), 5.27 (d, 1 H, J_{3.4} 2.6 Hz, H-3), 4.58 (d, 1 H, H-2), 4.17 (dd, 1 H, J_{4.5} 9.2 Hz, H-4), 3.72 (ddd, 1 H, J_{5.6a} 2.7, J_{5.6b} 6.4 Hz, H-5), 3.58 (dd, 1 H, $J_{6a,6b}$ 12.7 Hz, H-6a), 3.47 (dd, 1 H, H-6b), 2.16 (s, 3 H, $COCH_3$), 1.53, 1.32 (2 s, each 3 H, CMe_2) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 171.2 (CO), 112.4 (CMe₂), 104.7 (C-1), 83.0 (C-2), 79.3 (C-4), 76.4 (C-3), 67.5 (C-5), 53.9 (C-6), 26.5, 26.1 (CMe_2), 20.6 ($COCH_3$) ppm; FABMS: m/z 310 ([M + Na]⁺). Anal. Calcd for C₁₁H₁₇N₃O₆: C, 45.99; H, 5.96; N, 14.63. Found: C, 46.24; H, 5.58; N, 14.28.

6-Azido-3-O-benzyl-6-deoxy-1,2-O-isopropy*lidene-α-D-glucofuranose* (7).—Syrup (265 mg, 79%); $[\alpha]_D^{28} - 54^\circ$ (c 1.1, CH₂Cl₂); IR v_{max} 3493, 2104, 1080 and 1022 cm⁻¹; lit. [17] 3500(OH) and 2100(N₃); ¹H NMR (500 MHz, CDCl₃): δ 7.39–7.33 (m, 5 H, Ar), 5.93 (d, 1 H, $J_{1,2}$ 3.7 Hz, H-1), 4.75 (d, 1 H, $J_{H,H}$ 11.8 Hz, CHHPh), 4.64 (d, 1 H, H-2), 4.53 (d, 1 H, CHHPh), 4.09 (m, 3 H, H-3, H-4, and H-5), 3.54 (m, 1 H, H-6a), 3.42 (dd, 1 H, $J_{5.6b}$ 5.4, $J_{6a,6b}$ 12.5 Hz, H-6b), 2.28 (d, 1 H, $J_{5,OH}$ 4.2 Hz, OH), 1.49, 1.33 (2 s, each 3 H, CMe₂) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 136.8-127.9 (6 C, Ar), 111.9 (CMe₂), 105.1 $(C-1), 81.9 (C-2), 81.4 (C-3^*), 79.8 (C-4^*),$ 71.9 (CH₂Ph) 68.3 (C-5), 54.3 (C-6), 26.7, 26.2 (CMe₂) ppm; HREIMS m/z obsd 320.1243, calcd for $C_{15}H_{18}N_3O_5$ 320.1246 ($[M - CH_3]^+$).

6-Azido-6-deoxy-1,2-O-isopropylidene-3-O-(8). *methanesulfonyl-* α -**D**-*glucofuranose* Syrup (284 mg, 88%); $[\alpha]_{\rm D}^{21} - 27^{\circ}$ (c 1.1, CH₂Cl₂) IR v_{max} 3528, 2106, 1360, 1171, 1088, 1020, 943 and 851 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 5.93 (d, 1 H, J_{1,2} 3.6 Hz, H-1), 5.14 (d, 1 H, J₃₄ 2.7 Hz, H-3), 4.75 (d, 1 H, H-2), 4.22 (dd, 1 H, J₄₅ 9.3 Hz, H-4), 3.97 (ddd, 1 H, J_{5,6a} 2.8, J_{5,6b} 6.0 Hz, H-5), 3.64 (dd, 1 H, J_{6a.6b} 12.7 Hz, H-6a), 3.52 (dd, 1 H, H-6b), 3.15 (s, 3 H, Ms), 1.52, 1.33 (2 s, each 3 H, CMe_2) ppm; ¹³C NMR (75.4 MHz, CDCl₃): δ 112.8 (CMe₂), 104.9 (C-1), 83.1 (C-2), 81.1 (C-3), 78.7 (C-4), 67.3 (C-5), 54.0 (C-6), 38.0 (OMs), 26.5, 26.1 (CMe₂) ppm; FABMS: m/z

346 ($[M + Na]^+$). HREIMS: m/z obsd 308.0548, calcd for C₉H₁₄N₃O₇S 308.0552 ($[M - CH_3]^+$). Anal. Calcd for C₁₀H₁₇O₇N₃S: C, 37.14; H, 5.30; N, 13.00; S, 9.92. Found: C, 37.32; H, 5.23; N, 12.66; S, 10.14.

3,6-Diazido-3,6-dideoxy-1,2-O-isopropylidene- α -D-allofuranose (9).—Amorphous solid (216 mg, 80%); $[\alpha]_{\rm D}^{26} + 128^{\circ}$ (c 1.0, CH₂Cl₂); IR v_{max} 2108, 1379, 1262, 1107, 1026, and 872 cm^{-1} ; ¹H NMR (500 MHz, CDCl₃): δ 5.80 (d, 1 H, $J_{1.2}$ 3.6 Hz, H-1), 4.77 (dd, 1 H, $J_{2.3}$ 4.7 Hz, H-2), 4.08-4.05 (m, 1 H, H-4), 4.07-4.03 (m, 1 H, H-5), 3.58 (dd, 1 H, J_{3.4} 8.9 Hz, H-3), 3.54 (dd, 1 H, J_{5,6a} 7.6, J_{6a,6b} 12.7 Hz, H-6a), 3.45 (dd, 1 H, J_{5,6b} 3.7 Hz, H-6b), 2.37 (d, 1 H, J_{5.0H} 3.5 Hz, OH), 1.59, 1.38 (2 s, each 3 H, *CMe*₂) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 113.4 (CMe₂), 104.1 (C-1), 80.6 (C-2), 77.7 (C-4), 70.3 (C-5), 60.2 (C-3), 52.7 (C-6), 26.4 (2C, CMe_2) ppm; FABMS: m/z293 ($[M + Na]^+$). Anal. Calcd for C₉H₁₄N₆O₄: C, 40.00; H, 5.22; N, 31.10. Found: C, 40.18; H, 4.93; N, 30.75.

6-*Azido*-6-*deoxy*-1,2-O-*isopropylidene*-α-D*glucofuranose* (10).— $[\alpha]_{D}^{27}$ +1° (*c* 0.6, Me₂CO); lit. [15] $[\alpha]_{D}^{27}$ +2.48° (Me₂CO).

6-O-Acetyl-5-azido-5-deoxy-1,2-O-isopropy*lidene-* β -L-*idofuranose* (11).—Solid (201 mg, 70%); mp 72–74 °C (Et₂O–light petroleum) ether); $[\alpha]_{D}^{26} = -57^{\circ}$ (c 1.0, CH₂Cl₂); IR v_{max} 3443, 2104, 1740, 1373, 1260, 1082 and 1020 cm^{-1} ; ¹H NMR (500 MHz, CDCl₃): δ 5.98 (d, 1 H, J_{1.2} 3.7 Hz, H-1), 4.53 (d, 1 H, H-2), 4.32 (dd, 1 H, J_{6a.6b} 11.7 Hz, H-6a), 4.22 (m, 1 H, H-3), 4.14 (dd, 1 H, H-6b), 4.13 (dd, 1 H, J_{3.4} 2.5 Hz, H-4), 3.94 (t d, 1 H, J_{4.5} 7.6, J_{5.6a} 3.4, J_{5.6b} 7.6, Hz H-5), 2.66 (d, 1 H, OH), 2.13 (s, 3 H, $COCH_3$), 1.50, 1.32 (2 s, each 3 H, CMe_2); ¹³C NMR (125.7 MHz, CDCl₃): δ 170.9 (CO), 112.0 (CMe₂), 104.5 (C-1), 85.2 (C-2), 79.6 (C-4), 75.2 (C-3), 63.8 (C-6), 59.8 (C-5), 26.7, 26.1 (CMe_2) , 20.6 $(COCH_3)$; FABMS: m/z 310 ([M + Na]⁺). Anal. Calcd for C₁₁H₁₇N₃O₆: C, 45.99; H, 5.96; N, 14.63. Found: C, 46.04; H, 5.94; N, 14.77.

5-*Azido*-5-*deoxy*-1,2-O-*isopropylidene*- β -L*idofuranose* (12).— $[\alpha]_{D}^{21}$ -77° (*c* 1.2, MeOH); lit. [15] $[\alpha]_{D}^{20}$ -77.0 (*c* 2.6, MeOH).

General procedure for the preparation of amino derivatives 13–15.—A mixture of the

corresponding azidoderivative 8, 9 or 12 (1 mmol) and 10% Pd–C (20% for 9) in MeOH (26 mL) was hydrogenated at rt for 45–90 min followed by filtration and evaporation to dryness to give 13-15, respectively.

6-Amino-6-deoxy-1,2-O-isopropylidene-3-Omethanesulfonyl- α -D-glucofuranose (13). Isolated as an amorphous solid (297 mg, 100%) after column chromatography (4:1 CH₂Cl₂-MeOH); $[\alpha]_{D}^{24} - 44^{\circ}$ (c 1.0, CH₂Cl₂); IR v_{max} 3368, 3304, 1348, 1171, 1086, 1018, and 959 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 5.87 (d, 1 H, J_{1.2} 3.7, Hz, H-1), 5.02 (d, 1 H, J_{3.4} 2.8 Hz, H-3), 4.70 (d, 1 H, H-2), 4.02 (dd, 1 H, J_{45} 9.3 Hz, H-4), 3.72 (m, 1 H, H-5), 3.15 (s, 3 H, Ms), 3.05 (bd, 1 H, J_{6a.6b} 12.7 Hz, H-6a), 2.84 (dd, 1 H, J_{5,6b} 6.0 Hz, H-6b), 1.50, 1.32 (2 s, each 3 H, CMe₂) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 112.5 (CMe₂), 105.0 (C-1), 83.5 (C-2), 82.3 (C-3), 79.6 (C-4), 67.4 (C-5), 43.8 (C-6), 38.0 (1C, OMs), 26.5, 26.1 (CMe_2) ppm; FABMS: m/z298 $([M + H]^+);$ HRCIMS: m/z obsd 298.095089; calcd for $C_{10}H_{20}NO_7S$ 298.096049 ([M + H]⁺).

3,6-Diamino-3,6-dideoxy-1,2-O-isopropylidene- α -D-allofuranose (14). Isolated as an amorphous solid (218 mg, 100%). $[\alpha]_{D}^{24} + 50^{\circ}$ (c 1.0, MeOH); IR v_{max} 3372, 3304, 1589, 1377, 1217, 1098, 1017 and 874 cm⁻¹; ¹H NMR (500 MHz, MeOD): δ 5.75 (d, 1 H, $J_{1,2}$ 3.7 Hz, H-1), 4.56 (dd, 1 H, J₂, 4.8 Hz, H-2), 3.65-3.62 (m, 1 H, H-5), 3.59 (dd, 1 H, J_{34} 9.0, J₄₅ 6.1 Hz, H-4), 3.10 (dd, 1 H, H-3), 2.84 (dd, 1[°]H, J_{5.6a} 3.7, J_{6a.6b} 13.3 Hz, H-6a), 2.69 (dd, 1 H, $J_{5,6b}$ 6.8 Hz, H-6b), 1.51, 1.32 (2 s, each 3 H, CMe₂) ppm; ¹³C NMR (125.7 MHz, MeOD): δ 113.2 (CMe₂), 105.7 (C-1), 82.2 (C-2), 83.2 (C-4), 74.7 (C-5), 58.2 (C-3), 45.6 (C-6), 26.9, 26.6 (CMe₂) ppm; FABMS: m/z $([M + H]^+);$ HREIMS: 219 m/zobsd 203.1030, calcd for $C_8H_{15}N_2O_4$ 203.1032 $([M - CH_3]^+)$; HRCIMS: m/z obsd 219.1345; calcd for $C_{0}H_{10}N_{2}O_{4}$ 219.1345 ([M + H]⁺).

5-*Amino*-5-*deoxy*-1,2-O-*isopropylidene*- β -L*idofuranose* (**15**). Crystallized from EtOH: mp 179–183 °C, $[\alpha]_{D}^{24}$ – 6° (*c* 0.6, MeOH); lit. [17] mp 184–185 °C, $[\alpha]_{D}^{20}$ = – 3.7° (*c* 1.1, MeOH); lit. [18] mp 178 °C, $[\alpha]_{D}^{25}$ = – 3.0° (*c* 0.9, MeOH); lit. [19] mp 176–179 °C, $[\alpha]_{D}^{20}$ = – 3.4° (*c* 1.0, MeOH);.¹H NMR (500 MHz, MeOD): δ 5.89 (d, 1 H, $J_{1,2}$ 3.7 Hz, H-1), 4.46 (d, 1 H, H-2), 4.12 (d, 1 H, $J_{3,4}$ 2.9 Hz, H-3), 4.05 (dd, 1 H, $J_{4,5}$ 6.2 Hz, H-4), 3.62 (dd, 1 H, $J_{5,6a}$ 5.3, $J_{6a,6b}$ 10.9 Hz, H-6a), 3.52 (dd, 1 H, $J_{5,6b}$ 6.0 Hz, H-6b), 3.14 (m, 1 H, H-5), 1.43, 1.28 (2 s, each 3 H, CMe_2) ppm; ¹³C NMR (125.7 MHz, MeOD): δ 115.1 (CMe_2), 108.4 (C-1), 89.6 (C-2), 83.8 (C-4), 79.2 (C-3), 67.0 (C-6), 55.8 (C-5), 29.5, 28.9 (CMe_2) ppm; FABMS: m/z 220 ($[M + H]^+$).

General procedure for the preparation of fluoro derivatives 16–19.—To a solution of the corresponding cyclic sulfate 1-4 (1 mmol) in Me₂CO (5 mL), tetraethylammonium fluoride dihydrate (1.5 mmol) was added, and the mixture was stirred at rt until the TLC analysis (3:1 Et₂O-light petroleum ether) showed complete conversion of the cyclic sulfate into baseline material. The mixture was concentrated, and dissolved in THF (5 mL). Concd H_2SO_4 (50 μ L) and water (18 μ L) were added and the mixture was stirred for 15-45 min at rt. The reaction mixture was diluted with EtOAc (15 mL) and washed with sat aq NaHCO₃. The organic layer was separated, washed three times with water, dried (MgSO₄), and concentrated to dryness. The residue was purified by column chromatography.

3-O-Acetyl-6-deoxy-6-fluoro-1,2-O-isopropy*lidene-* α -D-*glucofuranose* (16). Isolated as a syrup (227 mg, 86%) after column chromatography (1:1 Et₂O-light petroleum ether); $[\alpha]_{D}^{23} + 11^{\circ} (c \ 1.1, CH_{2}Cl_{2}); IR v_{max} 3489, 1742,$ 1460, 1425, 1379, 1088 and 1022 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 5.90 (d, 1 H, $J_{1,2}$ 3.6 Hz, H-1), 5.31 (d, 1 H, J_{3.4} 2.6 Hz, H-3), 4.67 (ddd, 1 H, $J_{5,6a}$ 2.2, $J_{6a,6b}$ 9.6, $J_{6a,F}$ 47.4 Hz, H-6a), 4.57 (ddd, 1 H, J_{5.6b} 5.3, J_{6b.F} 47.3 Hz, H-6b), 4.56 (d, 1 H, H-2), 4.22 (dd, 1 H, J_{4.5} 9.3 Hz, H-4), 3.85 (m, 1 H, H-5), 2.15 (s, 3 H, COCH₃), 1.52, 1.32 (2 s, each 3 H, CMe₂) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 170.6 (CO), 112.4 (CMe₂), 104.8 (C-1), 84.5 (d, J_{CF} 169.7 Hz, C-6), 83.0 (C-2), 77.8 (d, J_{CF} 7.5 Hz, C-4), 76.3 (C-3), 67.5 (d, J_{CF} 18.8 Hz, C-5), 26.5, 26.1 (CMe₂), 20.6 (COCH₃) ppm; FABMS: m/z 287 $([M + Na]^+).$ HREIMS: m/z obsd 249.0780, calcd for $C_{10}H_{14}FO_6$ 249.0774 ([M – CH₃]⁺). Anal. Calcd for $C_{11}H_{17}FO_6$: C, 50.00; H, 6.48. Found: C, 49.62; H, 6.38.

3-O-Benzyl-6-deoxy-6-fluoro-1,2-O-iso*propylidene-* α -D-*glucofuranose* (17). Isolated as a syrup (218 mg, 70%) after column chromatography (1:2 Et₂O–light petroleum ether); $[\alpha]_{D}^{20} - 38^{\circ}$ (c 0.9, $CH_{2}Cl_{2}$); IR v_{max} 3474, 1456, 1379, 1217, 1082 and 1020 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 7.38–7.34 (m, 5 H, Ar), 5.95 (d, 1 H, J_{1.2} 3.7 Hz, H-1), 4.73 (d, 1 H, J_{H.H} 11.6 Hz, CHHPh), 4.71–4.57 (m, 1 H, H-6a), 4.64 (d, 1 H, H-2), 4.58 (d, 1 H, CHHPh), 4.60–4.48 (m, 1 H, H-6b), 4.19 (m, 1 H, H-5), 4.17 (m, 1 H, H-4), 4.14 (m, 1 H, H-3), 1.50, 1.33 (2 s, each 3 H, CMe₂) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 137.0– 127.8 (6C, Ar), 111.9 (CMe₂), 105.1 (C-1), 85.0 (d, J_{CF} 168.4 Hz, C-6), 82.0, 81.9 (C-2,3), 78.6 (d, J_{C,F} 6.3 Hz, C-4), 72.2 (CH₂Ph), 68.3 (d, J_{C.F} 17.6 Hz, C-5), 26.7, 26.2 (CMe₂) ppm; FABMS: m/z 335 ([M + Na]⁺). HREIMS: m/z obsd 312.1362, calcd for C₁₆H₂₁FO₅ 312.1373. Anal. Calcd for $C_{16}H_{21}FO_5$: C, 61.53; H, 6.78. Found: C, 61.26; H, 6.79.

6-Fluoro-6-deoxy-1,2-O-isopropylidene-3-O*mesyl*- α -D-glucofuranose (18). Isolated as a syrup (258 mg, 86%) after column chromatography (2:1 Et₂O-light petroleum ether); $[\alpha]_{D}^{21} - 28^{\circ} (c \ 1.0, \ CH_{2}Cl_{2}); \ IR \ \nu_{max} \ 3528, \ 1362, \ 1173, \ 1090, \ 1026 \ and \ 955 \ cm^{-1}; \ ^{1}H \ NMR$ (500 MHz, CDCl₃): δ 5.95 (d, 1 H, $J_{1,2}$ 3.6 Hz, H-1), 5.13 (d, 1 H, J_{3.4} 2.7 Hz, H-3), 4.78 (d, 1 H, H-2), 4.67 (ddd, 1 H, $J_{5,6a}$ 2.1, $J_{6a,6b}$ 9.8, $J_{6a,F}$ 47.5 Hz, H-6a), 4.59 (ddd, 1 H, $J_{5,6b}$ 4.7, J_{6b,F} 47.2 Hz, H-6b), 4.28 (dd, 1 H, J_{4,5} 9.5 Hz, H-4), 4.01 (dddd, 1 H, J_{5.F} 21 Hz, H-5), 3.15 (s, 3 H, Ms), 1.51, 1.33 (2 s, each 3 H, CMe₂) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 112.8 (CMe₂), 105.0 (C-1), 84.4 (d, J_{C,F} 168.4 Hz, C-6), 83.2 (C-2), 81.7 (C-3), 77.3 (d, J_{C,F} 7.0 Hz, C-4), 67.4 (d, J_{C.F} 18.8 Hz, C-5), 38.0 (1C, OMs), 26.5, 26.1 (CMe_2) ppm; FABMS: m/z $([M + Na]^+)$. HREIMS: m/z obsd 323 285.0438, calcd for $C_9H_{14}FO_7S$ 285.0444 $([M - CH_3]^+)$. Anal. Calcd for $C_{10}H_{17}FO_7S$: C, 40.00; H, 5.71; S, 10.68. Found: C, 39.99; H, 6.11; S, 11.01.

3- Azido - 3,6- dideoxy - 6- fluoro - 1,2- O-isopropylidene -α-D-allofuranose (19). Isolated as a syrup (198 mg, 80%) after column chromatography (2:1 Et₂O-light petroleum ether); $[\alpha]_D^{25}$ + 107° (*c* 1.1, CH₂Cl₂); IR v_{max} 3472, 2112, 1379, 1258, 1111, 1026 and 874 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 5.82 (d, 1 H, $J_{1,2}$ 3.6 Hz, H-1), 4.77 (dd, 1 H, $J_{2,3}$ 4.8 Hz, H-2), 4.58 (ddd, 1 H, $J_{5,6a}$ 4.0, $J_{6a,6b}$ 9.7, $J_{6a,F}$ 46.6 Hz, H-6a), 4.54 (ddd, 1 H, $J_{5,6b}$ 6.3, $J_{6b,F}$ 47.4 Hz, H-6b), 4.17 (m, 1 H, H-5), 4.12 (dd, 1 H, $J_{4,5}$ 4.6 Hz, H-4), 3.63 (dd, 1 H, $J_{3,4}$ 9.2 Hz, H-3), 2.38 (d, 1 H, $J_{5,OH}$ 4.3 Hz, OH), 1.58, 1.37 (2 s, each 3 H, CMe_2) ppm; ¹³C NMR (125.7 MHz, CDCl₃): δ 113.4 (CMe_2), 104.1 (C-1), 83.2 (d, $J_{C,F}$ 169.7 Hz, C-6), 80.5 (C-2), 76.8 (d, $J_{C,F}$ 7.3 Hz, C-4), 70.1 (d, $J_{C,F}$ 20.1 Hz, C-5), 60.6 (C-3), 26.4 (2C, CMe_2) ppm; FABMS: m/z 270 ([M + Na]⁺). Anal. Calcd for C₉H₁₄FN₃O₄: C, 43.72; H, 5.71; N, 17.00. Found: C, 44.10; H, 5.44; N, 16.81.

3- Amino - 3,6- dideoxy - 6- fluoro - 1,2-O-isopropylidene- α -D-allofuranose (20).—Prepared following the general procedure described above for 13–15. Column chromatography $(20:1 \text{ CH}_2\text{Cl}_2\text{-MeOH})$ of the residue gave 20 as a syrup (221 mg, 83%); $[\alpha]_D^{25} + 53^\circ$ (c 1.3, CH₂Cl₂); IR v_{max} 3381, 3312, 1377, 1219, 1099, 1013 and 874 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 5.78 (d, 1 H, J_{1.2} 3.7 Hz, H-1), 4.59 (ddd, 1 H, $J_{5,6a}$ 2.6, $J_{6a,6b}$ 9.9, $J_{6a,F}$ 47.4 Hz, H-6a), 4.51 (ddd, 1 H, $J_{5,6b}$ 4.9, $J_{6b,F}$ 47.6 Hz, H-6b), 4.46 (dd, 1 H, $J_{2,3}$ 4.6 Hz, H-2), 3.95 (dddd, 1 H, J_{5,F} 23.2 Hz, H-5), 3.65 (dd, 1 H, J_{4.5} 7.9 Hz, H-4), 3.20 (dd, 1 H, J_{3.4} 9.5 Hz, H-3), 2.35 (bs, 1 H, OH), 1.55, 1.35 (2 s, each 3 H, CMe₂) ppm; ¹³C NMR (75.4 MHz, CDCl₃): δ 112.3 (CMe₂), 104.5 (C-1), 84.1 (d, J_{C.F} 170.4 Hz, C-6), 80.1 (C-2), 76.1 (d, J_{C.F} 7.5 Hz, C-4), 73.2 (d, J_{C.F} 18.1, C-5), 58.6 (C-3), 26.4, 26.3 (CM e_2) ppm; FABMS: m/z222 $([M + Na]^+).$ Anal. Calcd for C₉H₁₆FNO₄: C, 48.86; H, 7.29; N, 6.33. Found: C, 49.16; H, 7.14; N, 6.18.

6-O-Acetyl-5-fluoro-5-deoxy-1,2-O-isopropylidene-β-L-idofuranose (22).—A solution of TAS-F (100 mg) in CH₂Cl₂ (1.5 mL) was added dropwise, at -10 °C, to a stirred solution of the cyclic sulfate 5 (35 mg). After 24 h, the reaction mixture was warmed to 35 °C, and kept at this temperature for a further 72 h. The solvent was evaporated under diminished pressure, and the crude was purified by column chromatography using 2:1 CH₂Cl₂– Me₂CO as eluent to give tris(dimethylamino)sulfonium 6-O-acetyl-5-fluoro-5-deoxy-1,2-O-isopropylidene-β-L-idofuranose-3sulfate (21) (29 mg, 53%) as a syrup. ¹H NMR (500 MHz, CDCl₃): δ 5.98 (d, 1 H, J_{1,2} 3.6,

H-1), 5.01 (m, 1 H, H-2), 5.02-4.89 (m, 1 H, H-5), 4.80 (d, 1 H, J_{34} 2.8 Hz, H-3), 4.58 (m, 1 H, H-6a), 4.48–4.43 (m, 1 H, H-4), 4.48– 4.39 (m, 1 H, H-6b), 2.97 (s, 18 H, NMe₂), 2.10 (s, 3 H, COCH₃), 1.49, 1.30 (2 s, each 3 H, CMe_2); ¹³C NMR (125.7 MHz, CDCl₃): δ 171.4 (CO), 112.0 (CMe₂), 105.0 (C-1), 90.3 (d, J_{C F} 174.7 Hz, C-5), 83.5 (C-2), 79.3 (d, J_{C F} 7.5 Hz, C-3), 78.2 (d, J_{C,F} 20.1 Hz, C-4), 63.2 (d, $J_{\rm CF}$ 20.1 Hz, C-6), 38.6 (NMe₂), 26.8, 26.3 (CMe_2) , 20.7 (COCH₃). Compound **21** (29 mg) was dissolved in THF (0.4 mL) and a catalytic amount of H₂SO₄ was added. The mixture was stirred for 30 min at rt and then diluted with EtOAc (5 mL) and washed with sat aq NaCO₃H. The organic layer was separated, washed three times with water, dried ($MgSO_4$), and concentrated to dryness to yield 22 (15 mg, 100%) as a syrup. $[\alpha]_{\rm D}^{24} - 28^{\circ}$ (c 0.8, CH₂Cl₂); IR v_{max} 3455, 1742, 1377, 1260, 1084, 1020 and 799 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 5.94 (d, 1 H, $J_{1,2}$ 3.7 Hz, H-1), 4.90 (dtd, 1 H, *J*_{5,6a} 2.6, *J*_{5,6b} 6.3, *J*_{5,F} 49.4 Hz, H-5), 4.46 (m, 1 H, H-2), 4.37 (ddd, 1 H, J_{6a,6b} 12.7, J_{6a,F} 27.0 Hz, H-6a), 4.21 (m, 1 H, H-3), 4.22 (ddd, $J_{3,4}$ 2.9, $J_{4,5}$ 6.3, $J_{4,F}$ 17.1 Hz, 1 H, H-4), 4.29 (ddd, 1 H, J_{6b,F} 23.6 Hz, H-6b), 2.06 (s, 3 H, COC H_3), 1.43, 1.26 (2 s, each 3 H, C Me_2); ¹³C NMR (125.7 MHz, CDCl₃): δ 170.9 (CO), 112.1 (CMe₂), 104.8 (C-1), 90.5 (d, J_{C.F} 176.0 Hz, C-5), 85.3 (C-2), 78.6 (d, J_{C,F} 17.6 Hz, C-4), 75.5 (d, J_{C,F} 5.7 Hz, C-3), 63.6 (d, J_{C,F} 22.6 Hz, C-6), 26.7, 26.2 (CMe₂), 20.6 (COCH₃); FABMS: m/z 287 ([M + Na]⁺). HRCIMS: m/z obsd 265.1073, calcd for $C_{11}H_{18}FO_6$ 265.1087 ([M + H]⁺).

Acknowledgements

We thank the Dirección General de Investigación Científica y Técnica of Spain for financial support (grant numbers PB94/1440-C02-01 and PB97/0730) and for the award of a grant to M.A.

References

- (a) H.C. Kolb, M.S. VanNieuwenhze, K.B. Sharpless, *Chem. Rev.*, 94 (1994) 2483–2547. (b) B.B. Lohray, *Synthesis*, (1992) 1035–1052.
- [2] Y. Gao, K.B. Sharpless, J. Am. Chem. Soc., 110 (1988) 7538–7539.
- [3] F.G. Calvo-Flores, P. García-Mendoza, F. Hernández-Mateo, J. Isac-García, F. Santoyo-González, J. Org. Chem., 67 (1997) 3944–3961.
- [4] P.M. Vanhessche, K.B. Sharpless, Chem. Eur. J., 3 (1997) 517–522.
- [5] M.S. Berridge, M.P. Franceschini, E. Rosenfeld, T.J. Tewson, J. Org. Chem., 55 (1990) 1211–1217.
- [6] R. Oi, K.B. Sharpless, *Tetrahedron Lett.*, 32 (1991) 999–1002.
- [7] (a) P.A.M. Van der Klein, W. Filemon, G.H. Veeman, G.A. Van der Marel, J.H. Van Boom, J. Carbohydr. Chem., 11 (1992) 837–848. (b) P.A.M. Van der Klein, J.H. Van Boom, Carbohydr. Res., 224 (1992) 193–200. (c) A.M. Gómez, S. Valverde, B. Fraser Reid, J. Chem. Soc., Chem. Commun., (1991) 1207–1208.
- [8] K. Vanhessche, E. Van der Eycken, M. Vandewalle, H. Roper, *Tetrahedron Lett.*, 31 (1990) 2337–2340.
- [9] T. Tsuchiya, Adv. Carbohydr. Chem. Biochem., 48 (1990) 91–277.
- [10] (a) Y. Mori, K. Harada, N. Morishima, S. Zen, *Chem. Pharm. Bull.*, 41 (1993) 755–757. (b) Y. Mori, N. Morishima, *Bull. Chem. Soc. Jpn.*, 67 (1994) 236–241.
- [11] R. Albert, K. Dax, S. Seidl, H. Sterk, A.E. Stütz, J. Carbohydr. Chem., 4 (1985) 513–520.
- [12] J. Fuentes, M. Angulo, M.A. Pradera, *Tetrahedron Lett.*, 39 (1998) 7149–7152.
- [13] T. Netscher, I. Gautshi, *Liebig Ann. Chem.*, (1992) 543-546.
- [14] W. Szeja, Carbohydr. Res., 158 (1986) 245-248.
- [15] K. Dax, B. Gaigg, V. Grassberger, B. Koblinger, A.E. Stütz, J. Carbohydr. Chem., 9 (1990) 479–499.
- [16] G. Legler, E. Julich, Carbohydr. Res., 128 (1984) 61– 72.
- [17] H. Saeki, E. Ohki, Chem. Pharm. Bull., 16 (1968) 2471– 2476.
- [18] R.E. Gramera, R.M. Bruce, S. Hirase, R.L. Whistler, J. Org. Chem., 38 (1963) 1401–1403.
- [19] H. Paulsen, K. Todt, Chem. Ber., (1966) 3450–3460.