

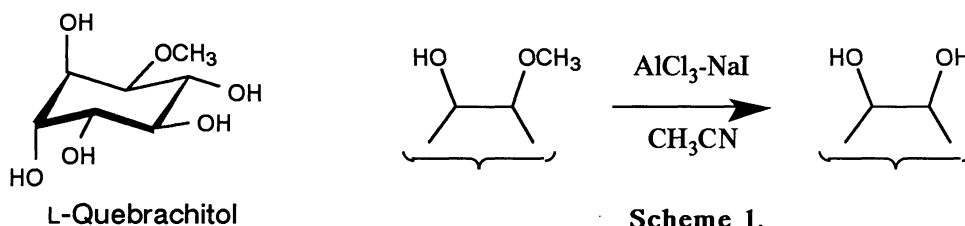
Anchimerically Assisted Demethylation of Methyl Ethers
in Inositol Derivatives with an AlCl_3 -NaI System

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Investigated are AlCl_3 and NaI catalyzed chemoselective demethylation reactions of methyl ethers of partly protected inositol derivatives, reactions which are greatly promoted by vicinal OH group and proceed in preference to the cleavage of the *cis* cyclohexylidene moiety.

L-Quebrachitol (1L-(-)-2-*O*-methyl-*chiro*-inositol), a naturally occurring optically active inositol, has recently stimulated much attention as a chiral building block^{1,2)} and several optically active *myo*-inositol derivatives have been synthesized starting from it.¹⁾ Demethylation of ring substituted *O*-methyl ether in cyclitol system is a crucial problem in manipulating it and traditionally drastic conditions such as HI,²⁾ and BCl_3 ,^{1a)} which cleave usual protecting groups altogether, have been used. Recently Ley³⁾ reported a $\text{BF}_3 \cdot \text{OEt}_2$ - Bu_4NI catalyzed demethylation reaction of *O*-methyl-*O*-pentabenzoyl-*chiro*-inositol which proceeded in modest yield, but the demethylation reaction of cyclitol *O*-methyl ether which was achieved under mild conditions had not been appeared. In a previous paper,^{1d)} we reported that AlCl_3 and NaI system⁴⁾ selectively cleaved the methyl ether of inositol benzoate in preference to the *cis* cyclohexylidene group; this method was used in a chiral synthesis of D-*myo*-inositol 1-phosphate starting from L-quebrachitol. In this paper, we focussed attention on the demethylation reactions of methyl ethers in several inositols derived from L-quebrachitol with an AlCl_3 -NaI system and found that the presence of OH group, vicinall to methyl ether, was essential in the cleavage reaction.



In the first place, we examined the demethylation reactions of inositol benzoates by the combined use of AlCl_3 -NaI in CH_3CN and the results are shown in Table 1. Treatment of **1a** with a mixture of 10 equiv. of AlCl_3 and 10 equiv. of NaI in CH_3CN for 20 min gave an *O*-demethylated product **2a** in 95% yield. (Run 1) Demethylation of **1b** needed much longer time, but underwent smoothly to obtain **2b** in 92% yield. (Run 2) On the other hand, **1c** resisted demethylation under the same conditions and an *O*-demethylated product **2c** was

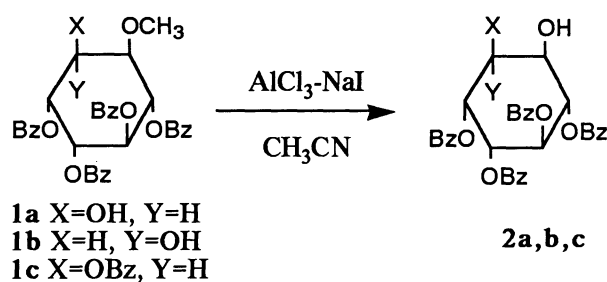
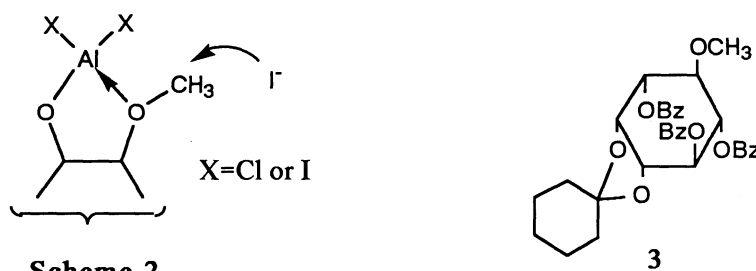


Table 1. Demethylation of inositol benzoate 1

Run	Starting material	$\text{AlCl}_3/\text{equiv.}$	$\text{NaI}/\text{equiv.}$	Conditions	Yield of 2 /%
1	1a	10	10	r.t. 20 min	95
2	1b	10	10	r.t. 15 h	91
3	1c	20	20	r.t. 7 d	27 ^{a)}

a) 73% of the starting material was recovered.

obtained in a poor yield after 7 days with a large amount of the starting material recovered. Based on these results, we considered that vicinal OH group assisted the cleavage of methyl ether by forming the 5-membered ring by intramolecular coordination of aluminum to ether oxygen as shown in Scheme 2. The ease of demethylation of 1a, bearing *cis*OH group to the methyl ether, in comparison to 1b, with *trans*OH group, can be interpreted by considering the stability of the 5-membered rings formed. Pivotal role of vicinal OH group was also observed in the attempted ether cleavage reaction of 3, in which case the *cis* cyclohexylidene moiety was cleaved preferentially and the methyl ether was partially cleaved under the forcing conditions (reflux 24 h) to give a demethylated compound in less than 25% yield with concomitant migration of the benzoyl group.



Scheme 2.

Next we investigated the vicinal OH group promoted demethylation reactions of cyclohexylidene protected inositol derivatives (Table 2). Treatment of a *myo*-inositol benzoate 4a with $\text{AlCl}_3\text{-NaI}$ in CH_3CN at room temperature gave a triol 6a in 83% yield. (Run 1) When the reaction was carried out at 0°C for 10 min, only *trans*cyclohexylidene moiety was cleaved and a diol 5a was obtained exclusively (Run 2), indicating that the methyl ether was cleaved subsequent to the *trans* cyclohexylidene group and OH group, thus emerged, promoted the demethylation reaction. The demethylation reaction of 5a actually proceeded similarly and 6a was obtained. (Run 3) A *chiro*-Inositol benzoate 4b also afforded the corresponding triol 6b, accompanied by a diol 5b at room temperature for 3 h (Run 4), though longer reaction time decreased the yield of 6b by further

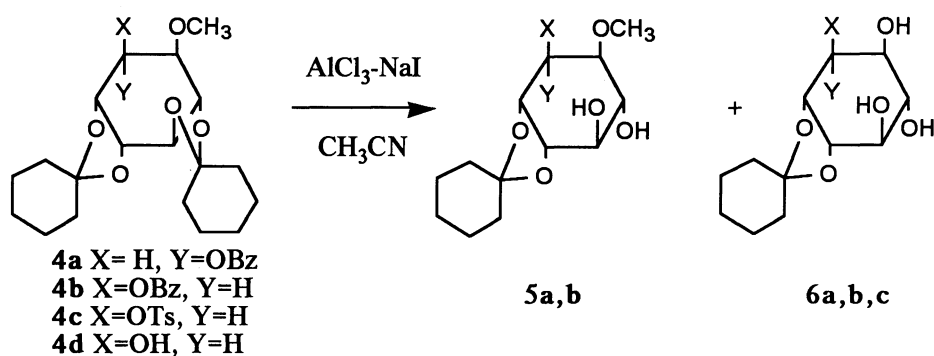
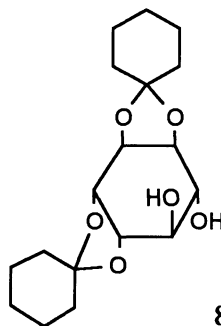
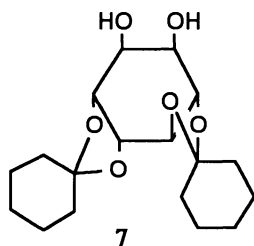


Table 2. Reaction of cyclohexylidene inositol derivatives

Run	Starting Material	AlCl ₃ /equiv.	NaI/equiv.	Conditions	Products / %
1	4a	10	10	r.t. overnight	6a ; 83 ^{a)}
2	4a	5	5	0 °C, 10 min	5a ; 77
3	5a	10	10	r.t. overnight	6a ; 58
4	4b	10	10	r.t. 3 h	5b ; 22 6b ; 52
5	4b	5	5	0 °C, 1 h	5b ; 89
6	4c	10	10	r.t. overnight	6c ; 75
7	4d	4	4	r.t. 30 min	8 ; 49
8	4d	10	10	r.t. 23 h ^{b)}	7 ; 39

a) Reported in Ref. 1d.

b) 10 equiv. of pyridine was added.



cleavage of the *cis*-cyclohexylidene moiety. The demethylation reaction of a tosylate **4c** also proceeded in a good yield. (Run 6) Finally, demethylation of **4d**, bearing *cis* OH group to the methyl ether, was studied. When **4d** was treated with AlCl₃ and NaI at room temperature, a sole product was not **7** but **8**, (Run 7) whose structure was identified with the authentic material synthesized from *chiro*-inositol by comparison with the 270 MHz NMR spectrum and TLC analysis. **7** was obtained exclusively by treatment of **4d** with AlCl₃ and NaI in the presence of 10 equiv. of pyridine. (Run 8) Although clear mechanism of the formation of **8** has not been clarified, **8** was considered to be produced by an acid catalyzed rearrangement of **7** in situ. This reaction is

noteworthy in terms of the chemical selectivity; the methyl ether was cleaved faster than the labile *trans* cyclohexylidene moiety, and the strong accelerating effect caused by the vicinally *cis* substituted OH group was thus demonstrated.

Aluminum trichloride⁵⁾ or boron trihalide⁶⁾ catalyzed regioselective demethylation of aryl methyl ether promoted by the anchimerical assistance are reported. But similar demethylation of alkyl methyl ether usable in cyclitol system has not been studied.

In conclusion, we have demonstrated chemoselective demethylation promoted by vicinal OH group by the combined use of AlCl_3 and NaI, in which the accelerating effect of *cis* OH group is much stronger than that of *trans* OH group. Both benzoyl and *cis* cyclohexylidene moieties are compatible with this demethylation reaction and hence the present method will be a useful tool in manipulating naturally occurring cyclitols bearing methyl ether such as L-quebrachitol and pinitol.^{1c)}

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- 7) A typical experiment procedure is as follows: To a solution of **1a** (110 mg, 0.180 mmol) in CH_3CN solution (2.5 ml) was added AlCl_3 (239 mg, 1.79 mmol) and NaI (270 mg, 1.80 mmol) at room temperature. Stirring was continued for 15 min. The reaction mixture was quenched by addition of water (10 ml) and the aqueous layer was extracted with CH_2Cl_2 (3 x 7 ml). The combined organic layers were successively washed with 10% Na_2SO_3 solution and brine, dried over anhydrous Na_2SO_4 , and the solvent was removed under reduced pressure. The remaining residue was purified by preparative TLC (hexane:ethyl acetate (v/v) = 5:4) to afford **2a** (102 mg, 0.171 mmol, 95%).

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