

Tetrahedron Letters 40 (1999) 5235-5238

TETRAHEDRON LETTERS

## An Efficient Synthesis of Optically Active Metabolites of Platelet Adhesion Inhibitor OPC-29030 by Lipase-Catalyzed Enantioselective Transesterification

Kazuyoshi Kitano,\* Jun Matsubara, Tadaaki Ohtani, Kenji Otsubo, Yoshikazu Kawano, Seiji Morita, and Minoru Uchida

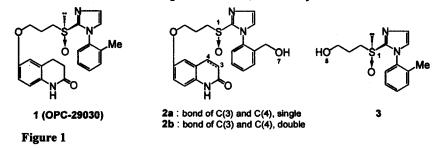
Tokushima Research Institute, Otsuka Pharmaceutical Co., Ltd., Kagasuno 463-10, Kawauchi-cho, Tokushima 771-0192, Japan

Received 15 April 1999; accepted 7 May 1999

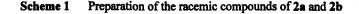
Abstract: The optically active metabolites of (S)-3,4-dihydro-6-[3-(1-o-tolyl-2-imidazolyl)sulfinylpropoxy]-2(1H)-quinolinone (OPC-29030, 1), which is a new anti-platelet agent (platelet adhesion inhibitor), were effectively synthesized by the enzyme-catalyzed enantioselective transesterification of the racemic sulfinyl metabolites. The enzymes can recognize a stereogenic sulfur atom remote from the reaction site. © 1999 Elsevier Science Ltd. All rights reserved.

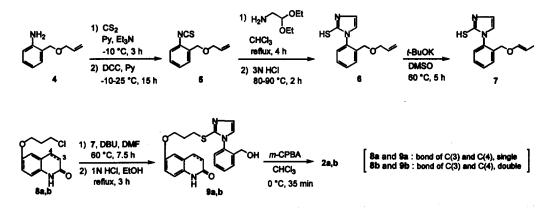
(S)-3,4-Dihydro-6-[3-(1-o-tolyl-2-imidazolyl)sulfinylpropoxy]-2(1*H*)-quinolinone (OPC-29030, 1)<sup>1</sup> is a new platelet adhesion inhibitor which suppresses the production of 12(S)-hydroxyeicosatetraenoic acid (12-HETE) in platelets and is now under clinical trials (Figure 1). As a part of the drug development study, the metabolism of 1 has been investigated in rats, dogs and humans. The structures of the metabolites (2a and 2b) were proposed on the basis of NMR and MS spectral analyses to be the compounds hydroxylated at the tolyl group. These metabolites have a chiral sulfinyl group as well as 1. The pharmacological activity of the enantio-isomeric metabolites and the metabolic chiral inversion of the sulfinyl group pose an interesting problem.

On the other hand, the synthetic method for optically active compounds using lipase-catalysis in an organic solvent has recently been generally accepted. However, substrates of the lipase-catalyzed asymmetric reactions were limited to the compounds whose stereogenic carbon atoms were in the neighborhood of the reaction site. There have been some investigations on substrates whose stereogenic carbon atoms are remote from the reacting site.<sup>2</sup> The application of enzymatic methods for the preparation of chiral sulfoxide has been reported by Nagao et al.<sup>3</sup> and other groups.<sup>4</sup> Usually, the enantioselectivity was low in lipase-catalyzed optical resolution which used a primary alcohol. In a previous paper,<sup>5</sup> we also reported the asymmetric synthesis of the key intermediate (3) of OPC-29030 from a racemic substrate using the lipase catalyst. However, the low enantioselectivity (E < 9)<sup>8</sup> was observed in the reaction of 3 included the flexible primary alcohol, and there was no effect by the added organic solvent. In the course of our investigation we have noted the relationship between the primary hydroxyl- and chiral sulfinyl-group in the enzyme-catalyzed reaction. Herein we wish to describe the synthesis of the optical isomers of the metabolites (2a and 2b), using the lipase-catalyzed reaction of them in which the chiral center and reacting site are distant, when compared with 3.



The racemic compounds (2a and 2b) were prepared as shown in Scheme 1. The reaction of 2-(allyloxymethyl)aniline (4)<sup>6</sup> with carbon disulfide in pyridine-triethylamine followed by treatment with DCC gave the isothiocyanate (5). Condensation of 5 with 2,2-diethoxyethylamine afforded the thiourea, which was cyclized in the presence of 3N HCl to give the 2-mercaptoimidazole (6). Treatment of this 2-propenyl compound (6) with *t*-BuOK afforded the 1-propenyl compound (7). The reaction of 7 with the chloride (8a)<sup>7</sup> in the presence of DBU gave the sulfide, which was treated with 1N HCl to give the hydroxy compound (9a). The desired compound (2a) was prepared by the oxidation of 9a with *m*-CPBA. Another sulfoxide (2b) was synthesized from 8b<sup>7</sup> and 7 by a procedure similar to that used for 2a.





The optical isomers 2a and 2b were synthesized as follows. We examined the lipase-catalyzed enantioselective transesterification of the sulfoxide  $(\pm)$ -2a. As shown in Table 1, Novozym 435 from *Candida antarctica* (Novo Nordisk) was selected after screening several commercially available lipases in the reaction with  $(\pm)$ -2a (Entries 4). Table 2 summarizes the results of the experiments using Novozym 435 and vinyl acetate in various organic solvents. The reaction was usually conducted at 40 °C or room temperature and monitored by HPLC. Under these conditions, it was apparent that the most suitable solvents were dichloromethane and chloroform; good reaction rates and enantioselectivity were achieved at room temperature. Next, a preparative scale reaction was performed. Stopping the reaction at around a 40 % conversion gave (+)-10a and (-)-2a in high enantiomeric purities (95 % ee and 67 % ee, E = 79).<sup>8</sup> The alcohol (+)-2a prepared by hydrolysis of (+)-10a (>99 % ee) was obtained. The target compound (+)-2a (>99 % ee) was synthesized by the hydrolysis of (+)-10a (>99 % ee) with K<sub>2</sub>CO<sub>3</sub>. The antipodal enantiomer (-)-2a (>99 % ee) was similarly obtained as the unreacted alcohol by lipase-catalyzed transesterification<sup>9</sup> of the sample with low optical purity (67 % ee). Compounds (+)- and (-)-2b<sup>10</sup> were also synthesized by a procedure similar to that used for (+)- and (-)-2a.

The absolute configurations of the above optically active compounds were determined as follows (Scheme 2). The benzyl alcohol (+)-2a synthesized from ( $\pm$ )-2a via transesterification was treated with mesyl chloride in the presence of Et<sub>3</sub>N to give the mesylate, which was converted to the tolyl compound (+)-1. The HPLC retention time of this compound was in good agreement with that of the mother compound [1, (S)-(+)]. The compound (+)-11 derived from (+)-10b agreed with the dehydro compound (S)-11 prepared from (S)-(+)-12. Accordingly, the absolute configuration for the isomers of (+)-2a, (+)-2b, (+)-10a and (+)-10b was assigned S.

$(\pm)-2a$											
			Acetate		Unreacted alcohol						
Entry	Lipase <sup>b</sup>	Time (min)	Yield (%) <sup>c</sup>	% ee <sup>d</sup>	Yield (%) <sup>c</sup>	% ee <sup>d</sup>	Ee				
1	Lipase PL	660	83	21 (S)	17	99 (R)	6				
2	Lipase MY	660	54	13 <i>(R</i> )	46	15 (S)	1				
3	Lipase OF	30	46	17 ( <i>S</i> )	54	14 <i>(R</i> )	2				
4	Novozym 435	30	56	65 (S)	44	83 <i>(R</i> )	12				
5	Lipozyme IM	660	63	28 (S)	37	48 <i>(R</i> )	3				
6	SP 524	660	63	31 ( <i>S</i> )	37	53 <i>(R</i> )	3				
7	Toyozyme	120	7 <del>9</del>	19 (S)	21	70 <i>(R</i> )	3				

Table 1. Lipase-catalyzed enantioselective transesterification of  $(\pm)$ -2a<sup>a</sup>

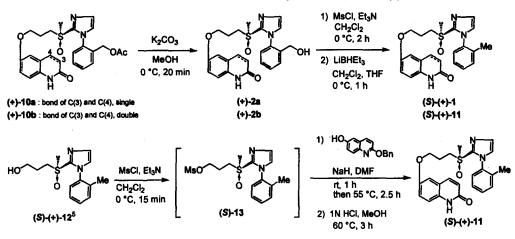
a. All reactions were carried out by stirring a mixture of  $(\pm)$ -2a (10 mg), lipase (10 mg) and vinyl acetate (1 ml) at 40°C. b. PL (Meito, Alcaligenes sp.), MY (Meito, Candida cyclindracea), OF (Meito, Candida cyclindracea), Novozym 435 (Novo Nordisk, Aspergillus oryzae), Lipozyme IM (Novo Nordisk, Mucor miehei), SP 524 (Novo Nordisk, Aspergillus orizae), Toyozyme (Toyo Boseki, Pseudomonas sp.). c. HPLC yield. d. Enantiomeric purities were determined by HPLC analyses using a column packed with ULTRON ES-OVM (solvent: acetonitrile : 20 mM KH<sub>2</sub>PO<sub>4</sub> = 7 : 93). e. The E value is the ratio of the specificity constant of two enantiomers caluculated according to ref. 8.

Entry	Solvent	Temp (°C)	Time (min)	Acetate Yield (%) <sup>b</sup> % ee <sup>c</sup>		Unreacted alcohol Yield (%) <sup>b</sup> % ee <sup>c</sup>		Ee
1		rt	30	38	84	62	51	19
2	i-Pr <sub>2</sub> O	40	5	21	68	79	18	6
3	CH <sub>2</sub> Cl <sub>2</sub>	40	5	37	91	63	54	36
4	CH <sub>2</sub> Cl <sub>2</sub>	rt	30	47	93	53	82	70
5	CHCl <sub>3</sub>	rt	30	44	94	56	72	70
6	CCl <sub>4</sub>	rt	30	34	85	66	45	19
7	toluene	rt	30	31	92	69	41	36
8	AcOEt	rt	30	50	84	50	85	31
9	THF	rt	30	52	87	48	95	53

Table 2. Lipase-catalyzed enantioselective transesterification of  $(\pm)$ -2a in organic solvents<sup>a</sup>

a. All reaction were carried out by stirring a mixture of ( $\pm$ )-2a (10 mg), Novozym 435 (10 mg) and vinyl acetate (0.5 ml) in organic solvent (0.5 ml) at room temperature or 40°C. b. HPLC yield. c. Enantiomeric purities were determined by HPLC analyses using a column packed with ULTRON ES-OVM (solvent; acetonitrile : 20 mM KH<sub>2</sub>PO<sub>4</sub> = 7 : 93). e. see ref. 8.

## Scheme 2 Determination of the absolute configuration of (+)-10a and (+)-10b



In conclusion, we have established the efficient synthesis of optically active metabolites (2a and 2b) of OPC-29030 using lipase-catalyzed kinetic resolution. The metabolites (2a and 2b) were resolved with excellent enantioselectivity by using Novozym 435 in spite of the reaction site being fairly remote from the stereogenic sulfur atom, in contrast to 3. There is also significance in addition to expansion of the utility of the lipase-catalyzed transesterification when we consider the reaction mechanism.

## **References and Notes**

- 1. Uno, T.; Ozeki, Y.; Chu, G.-N.; Okada, M.; Tamura, K.; Igawa, T.; Unemi, F.; Kido, M.; Nishi, T. Chem. Pharm. Bull. 1995, 43, 1724.
- (a) Nagai, H.; Shiozawa, T.; Achiwa, K.; Terao, Y. Chem. Pharm. Bull. 1992, 40, 2227. (b) Murata, M.; Uchida, H.; Achiwa, K. Chem. Pharm. Bull. 1992, 40, 2605. (c) Ebiike, H.; Maruyama, K.; Achiwa, K. Tetrahedron: Asymmetry 1992, 3, 1153. (d) Mizuguchi, E.; Achiwa, K. Tetrahedron: Asymmetry 1993, 4, 2303.
- Tamai, S.; Miyaguchi, S.; Morizane, C.; Miyagi, K.; Shimizu, H.; Kume, M.; Sano, S.; Shiro, M.; Nagao Y. Chem. Lett. 1994, 2381.
- (a) Burgess, K.; Henderson, I.; Tetrahedron Lett. 1989, 30, 3633. (b) Burgess, K.; Henderson, I.; Ho, K-K. J. Org. Chem. 1992, 57, 1290. (c) Mikolajczyk, M.; Kielbasinski, P.; Zurawinski, R.; Wieczorek, M. W.; Blaszczyk, J. Synlett 1994, 127.
- 5. Morita, S.; Matsubara, J.; Otsubo, K.; Kitano, K.; Ohtani, T.; Kawano, Y.; Uchida, M. Tetrahedron: Asymmetry 1997, 8, 3707.
- 6. Garanti, L.; Zeechi, G.; Bruche, L. J. Heterocyclic. Chem. 1993, 30, 559.
- Banno, K.; Fujioka, T.; Kikuchi, T.; Oshiro, Y.; Hiyama, T.; Nakagawa, K. Chem. Pharm. Bull. 1988, 36, 4377.
- 8. Chen, C.-S.; Fujimoto, Y.; Girdaukas, G.; Sih, C. J. J. Am. Chem. Soc. 1982, 104, 7294.
- 9. The reactions were carried out by stirring a mixture of substrates, Novozym 435 and vinyl acetate in CH<sub>2</sub>Cl<sub>2</sub>.
- 10. A mixture of (±)-2b (100 mg, 0.24 mmol), vinyl acetate (3.0 ml) and Novozym 435 (100 mg) in CH<sub>2</sub>Cl<sub>2</sub> (10 ml) was stirred at room temperature for 4 h. The enzyme was filtered off and washed with CH<sub>2</sub>Cl<sub>2</sub>. The combined solution was concentrated *in vacuo* and the residue was purified by column chromatography on silica gel to give the acetate (45 mg, 40%, 97 % ee) and the unreacted alcohol (48 g, 46%, 89 % ee, E = 198).<sup>8</sup>