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Stereoselective Synthesis of Highly Functionalized Cyclobutenes. A Facile Route to Electron-Deficient 1,3-Dienes

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Abstract: Acetoacetanilide undergoes a smooth reaction with triphenylphosphine and dialkyl acetylenedicarboxylates to produce dialkyl 2-(acetoacetanilide-2-yl)-3-(triphenylphosphoranilidene)butandioates, which undergo intramolecular Wittig reaction to produce 2-methyl-3-(*N*-phenylcarbonyl)-1,4-dialkoxycarbonylcyclobutenes. These cyclobutene derivatives undergo electrocyclic ring-opening reactions in boiling toluene to produce highly electron-deficient 1,3-dienes. © 1999 Published by Elsevier Science Ltd. All rights reserved.

INTRODUCTION

Although the common 5-, 6- and 7-membered ring cycloalkenes are produced fairly easily by intramolecular Wittig reaction, the formation of cyclopropenes and cyclobutenes have not received much attention [1]. Cyclobutenes are important intermediates in organic synthesis [2,3], and their synthetic study continues to attract much attention. During the last few decades several methods have been developed for the preparation of polysubstituted cyclobutenes [2-4]. We have recently [5] described the first synthesis of stereoselective intramolecular Wittig reaction of a derivatives from cyclobutene 1 the vinyltriphenylphosphonium salt with ethyl 4-aryl-2,4-dioxobutanoates. Cyclobutenes 1 undergo electrocyclic ring-opening reaction in boiling toluene to produce highly electron-deficient 1,3-dienes 2.



As part of our current studies on the development of new routes to heterocyclic and carbocyclic systems, we now report a facile synthesis of stabilized ylides 4, which are converted to functionalized cyclobutenes 5 via an intramolecular Wittig reaction. Compounds 5a-c undergo electrocyclic ring-opening reactions to produce electron-deficient 1,3-dienes 6a-c in fairly high yields. Thus, reaction of acetoacetanilide with dialkyl *Fax: (98)21-8006544

0040-4020/99/\$ - see front matter © 1999 Published by Elsevier Science Ltd. All rights reserved. *PII:* S0040-4020(99)00671-7 acetylenedicarboxylates 3 in presence of triphenylphosphine leads to the corresponding stabilized phosphorus ylides 4, which are converted to cyclobutene derivatives 5.



RESULTS AND DISCUSSION

On the basis of the chemistry of trivalent phosphorus nucleophiles [6,7], it is reasonable to assume that cyclobutene 5 results from initial addition of triphenylphosphine to the acetylenic ester and subsequent protonation of the reactive 1:1 adduct by acetoacetanilide. Then the positively charged ion is attacked by the enolate anion of acetoacetanilide to form the stable ylide 4, which undergoes intramolecular Wittig reaction in boiling benzene to produce the cyclobutene derivatives **5a-c**. Compounds **5a-c** undergo electrocyclic ring-opening reaction in boiling toluene to produce electron deficient 1,3-dienes **6a-c** in fairly good yields.

The ¹H NMR spectra of stable ylides 4a-c exhibited a singlet at about δ =2.3 for the CH₃-CO group, a double doublet (³J_{HP} 17.6 Hz, ³J_{HH} 10.4 Hz) at about δ =3.3 for CHCO₂R, alonge with a characteristic doublet (³J_{HH} 10.4 Hz) at about δ =5.2 for CH(CO)₂ moiety. The ¹³C NMR spectra of 4a-c displayed a doublet (¹J_{PC} 124 Hz) at about δ =41 for the P=C group and a doublet (³J_{PC} 12.8-14.6 Hz) at about δ =42 for the CHCO₂R moieties. The ³¹P NMR spectrum of ylides 4a-c exhibited, in each case, only one signal at about δ =25 (downfield from 85% H₃PO₄). These shifts are similar to those observed for alkyl triphenylphosphonium iodide [8]. Although compound 4 possesses two stereogenic centers, and two diastereoisomers are expected, only one diastereoisomer is isolated from the reaction mixture.

The ¹H NMR spectra of the cyclobutene derivatives **5a-c** displayed signals at about δ =3.5-3.6 and δ =3.8-3.9 for the two methine groups, which appear as doublets (³J_{HH} 1 Hz), in agreement with the *trans* geometry The ¹H NMR spectra of the butadiene derivatives **6a-c** exhibited two signals at about δ =6.1 and δ =6.2 for the two olefinic protons. The ¹³C NMR spectra of **6a-c** displayed four signals in the olefinic region. Although, we have not proved the stereochemistry of dienes **6a-c**, the geometry shown in Scheme 1 is the most reasonable on steric ground and on the basis of conrotatory cyclobutene opening.

The structure assignments made on the basis of the NMR spectra for compounds 4-6 were supported by measurements of their IR spectra. Of special interest are the strong carbonyl absorption bands at 1645-1712 cm⁻¹ for these compounds and a fairly broad NH peak at about 3300-3420 cm⁻¹ for the NH group.

In conclusion, we have found that the reaction of acetoacetanilide with dialkyl acetylenedicarboxylates in the presence of triphenylphosphine leads to a facile synthesis of highly functionalized cyclobutenes, which are converted to electron-deficient 1,3-dienes.

EXPERIMENTAL SECTION

Dialkyl acetylenedicarboxylates, triphenylphosphine and acetoacetanilide were obtained from Fluka (Buchs, Switzerland) and were used without further purification. Melting points were measured on an Electrothermal 9100 apparatus and are uncorrected. Elemental analyses were performed using a Heraeus CHN-O-Rapid analyzer. ¹H , ¹³C and ³¹P NMR spectra were measured with a BRUKER DRX-500 AVANCE spectrometer at 500, 125.8 and 202.5 MHz, respectively. Mass spectra were recorded on a Finnigan-Matt 8430 mass spectrometer operating at an ionization potential of 70 eV. IR spectra were recorded on a Shimadzu IR-470 spectrometer.

Preparation of dimethyl 2-(acetoacetanilide-2-yl)-3-(triphenylphosphoranylidene)-butanedioate (4a). General procedure

To a magnetically stirred solution of acetoacetanilide (0.35 g, 2 mmol) and triphenylphosphine (0.52 g, 2 mmol) in CH₂Cl₂ (10 ml) was added, dropwise, a mixture of dimethyl acetylenedicarboxylate (0.28 g, 2 mmol) in CH₂Cl₂ (4 ml) at -10 °C over 10 min. The mixture was allowed to stand at 5 °C for 24 hr. The solvent was removed under reduced pressure and the residue was purified by silica gel (Merck silica gel, 230-400 mesh) column chromatography using hexane:ethyl acetate (1:5) as eluent. The solvent was removed under reduced pressure and ylide **4a** (1.05 g, m.p. 131- 133 °C, yield 90%) was obtained as a white powder. IR (KBr) (v_{max} , cm⁻¹): 3400 (NH), 1713, 1669 (C=O); ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H}$ 2.3 (3 H, s, CH₃), 3.1 and 3.6 (6 H, 2 s, 2 OMe), 3.3 (1 H dd, ³J_{HP} 17.4 Hz and ³J_{HH} 10.4 Hz, CH), 5.2 (1 H, d, ³J_{HH} 10.4 Hz, CH), 7.2-7.6 [20 H, m,

P(C₆H₅)₃ and NC₆H₅], 9.1 (1 H, br. s, NH); ¹³C NMR (125.8 MHz, CDCl₃): δ_{C} 29.3 (CH₃), 41.2 (d, ¹J_{PC} 123.4 Hz, P=C), 41.8 (d, ²J_{PC} 13.7 Hz, P=C-CH), 49.0 and 51.6 (2 OCH₃), 62.0 [d, ³J_{PC} 5.3 Hz, CH(CO)₂], 119.3, 123.8, 129.1 and 138.7 (Cortho, C_{para}, C_{meta} and C_{ipso} of NC₆H₅, respectively), 126.2 (d, ¹J_{PC} 90.6 Hz, C_{ipso}), 128.4 (d, ³J_{PC} 12.2 Hz, C_{meta}), 131.9 (C_{para}), 133.8 (d, ²J_{PC} 9.1 Hz, C_{ortho}), 167.4 (C=O amide), 170.9 (d, ²J_{PC} 12.8 Hz, C=O ester), 174.6 (d, ³J_{PC} 3.1 Hz, C=O ester), 202.4 (C=O ketone). ³¹P NMR (202.5 MHz, CDCl₃): δ_{P} 25.6; MS, *m*/*z* (%): 581 (1), 262 (100), 183 (82), 108 (43); Anal. Calcd. for C₃₄H₃₂NO₆P (581.6) C, 70.21; H, 5.55; N, 2.41 %. Found: C, 70.0; H, 5.7; N, 2.7%.

Diethyl 2-(acetoacetanilide-2-yl)-3-(triphenylphosphoranilylidene)-butanedioate (4b)

White powder, mp 190-192 °C, yield 92%; IR (KBr) (v_{max} , cm⁻¹): 3400 (NH), 1712, 1664 (C=O); ¹H NMR (90 MHz, CDCl₃): δ_{H} 0.4 and 1.2 (6 H, 2 t, ³ J_{HH} =7.5 Hz, 2 CH₃), 2.3 (3 H, s, CH₃), 3.3 (1 H dd, ³ J_{HP} =17.6 Hz and ³ J_{HH} =10.4 Hz, CH), 3.7 and 4.1 (4 H, 2 ABX₃ system, 2 OCH₂), 5.2 (1 H, d, ³ J_{HH} 10.8 Hz, CH), 7-8 [20 H, m, P(C₆H₅)₃ and NC₆H₅], 9 (1 H, br. s, NH); ¹³C NMR (125.8.4 MHz, CDCl₃): δ_{C} 13.6 and 13.8 (2 CH₃), 29.0 (CH₃), 40.5 (d, ¹ J_{PC} 123.4 Hz, P=C), 41.8 (d, ² J_{PC} 13.7 Hz, P=C-CH), 57.3 and 60.3 (2 OCH₂), 61.9 [d, ³ J_{PC} =6.6 Hz, CH(CO)₂], 119.1, 124.2, 128.8 and 138.6 (Cortho, Cparo, Cmeta and Cipso of NC₆H₅, respectively), 121.3 (d, ¹ J_{PC} 99.7 Hz, Cpso), 128.0 (d, ³ J_{PC} 12.8 Hz, Cmeta), 131.7 (Cparo), 133.7 (d, ² J_{PC} 10.1 Hz, Cortho), 167.3 (C=O amide), 170.3 (d, ² J_{PC} 13.7 Hz, C=O ester), 174.0 (C=O ester), 202.1 (C=O, ketone). ³¹P NMR (202.5 MHz, CDCl₃): δ_{P} 25.5; MS, *m*/*z* (%): M⁺, 609 (1), 536 (6), 262 (100), 183 (57), 93 (39); Anal. Calcd. for C₃₆H₃₆NO₆P (609.6) C, 70.92; H, 5.95; N, 2.30 %. Found: C, 70.7; H, 5.8; N, 2.1 %.

Di-tert-butyl 2-(acetoacetanilide-2-yl)-3-(triphenylphosphoranilylidene)-butanedioate (4c)

white powder m.p. 154-155 °C, yield 87%; IR (KBr) (v_{max} , cm⁻¹): 3410 (NH), 1710, 1672 (C=O); ¹H NMR (500 MHz, CDCl₃): δ_{H} 0.9 and 1.4 (18 H, 2 s, 2 CMe₃), 2.2 (3 H, s, CH₃), 3.2 (1 H, dd, ³*J*_{HP} 18 Hz and ³*J*_{HH} 10.7 Hz, CH), 5.1 (1 H, d, ³*J*_{HH} 10.7 Hz, CH), 7.2-7.8 [20 H, m, P(C₆H₅)₃ and NC₆H₅], 9 (1 H, br. s, NH); ¹³C NMR (125.8 MHz, CDCl₃): δ_{C} 28.0 and 28.4 (2 CMe₃), 29.2 (CH₃), 40.3 (d, ¹*J*_{PC} 124.3 Hz, P=C), 43.0 (d, ²*J*_{PC} 14.6 Hz, P=C-CH), 62.4 [d, ³*J*_{PC} 5.5 Hz, CH(CO)₂], 77.3 and 80.1 (2 OCMe₃), 119.1, 124.9, 128.9 and 138.9 (Cortho, C_{para}, C_{meta} and C_{ipso} Of NC₆H₅, respectively), 121.3 (d, ¹*J*_{PC} 100.6 Hz, C_{ipso}), 128.1 (d, ³*J*_{PC} 11.9 Hz, C_{meta}), 131.7 (C_{para}), 134.1 (d, ²*J*_{PC} 9.1 Hz, Cortho), 167.8 (C=O amide), 170.5 (d, ²*J*_{PC} 12.8 Hz, C=O ester), 173.4 (C=O ester), 202.3 (C=O, ketone). ³¹P NMR (202.5 Mhz, CDCl₃): δ_{P} 5.2; MS, *m*/*z* (%): M⁺+1, 662 (2), 564 (8), 508 (7), 262 (88), 93 (98), 57 (100); Anal. Calcd. for C₄₀H₄₄NO₆P (665.7) C, 72.16; H, 6.66; N, 2.10 %.

Preparation of dimethyl 3-(anilinocarbonyl)-2-methyl-1-cyclobutene-1,4-dicarboxylate (5a)

Compound 4a was refluxed in benzene for 3 hr. The solvent was removed under reduced pressure and the residue was purified by silica gel (Merck silica gel, 230-400 mesh) column chromatography using hexane:ethyl acetate (8:1) as eluent. The solvent was removed under reduced pressure and 5a was obtained as white powder,

m.p. 88-89 °C, yield 55%; IR (KBr) (v_{max} , cm⁻¹): 3400 (NH), 1702 and 1660 (C=O); ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H} 2.2$ (3 H, s, CH₃), 3.75 and 3.75 (6 H, s, 2 OMe), 3.6 and 3.9 (2 H, s, 2 CH), 7-7.6 (5 H, m, C₆H₅), 8.2 (1 H, br. s, NH); ¹³C NMR (125.8 MHz, CDCl₃): $\delta_{\rm C} 15.3$ (CH₃), 45.2 and 51.4 (2 CH), 51.1 and 52.4 (2 OCH₃), 120.2, 124.6, 129.1 and 137.6 (C_{ortho}, C_{para}, C_{meta} and C_{ipso} of NC₆H₅, respectively), 130.8 (C=C-CO₂Me), 158.9 (C=C-CH₃), 162.0 (C=O amide), 167.1 and 172.1 (C=O ester); MS, *m/z* (%): M⁺, 303 (29), 212 (20), 184 (37), 93 (100), 65 (57); Anal. Calcd. for C₁₆H₁₇NO₅ (303.3) C, 63.36; H, 5.65; N, 4.62 %. Found: C, 63.5; H, 5.4; N, 4.5 %.

Diethyl 3-(anilinocarbonyl)-2-methyl-1-cyclobutene-1,4-dicarboxylate (5b)

White powder, mp 115-117°C, yield 53%; IR (KBr) (v_{max} , cm⁻¹): 3230 (NH), 1715 and 1666 (C=O); ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H}$ 1.1 and 1.3 (6 H, 2 t, ³J_{HH} 6.6 Hz, 2 CH₃), 1.7 (3 H, s, CH₃), 3.5 and 3.8 (2 H, s, 2 CH), 4.1 (4 H, 2 ABX₃ system, 2 OCH₂), 6.8-7.6 (5 H, m, NC₆H₅), 8.5 (1 H, br. s, NH); ¹³C NMR (125.8 MHz, CDCl₃): $\delta_{\rm C}$ 14.5 and 14.6 (2 CH₃), 15.6 (CH₃), 45.8 and 51.4 (2 CH), 60.8 and 61.7 (2 OCH₂), 120.4, 125.0, 129.3 and 138.1 (Cortho, C_{para}, C_{meta} and C_{ipso} of NC₆H₅, respectively), 131.6 (C=C-CO₂Et), 159.2 (C=C-CH₃), 162.1 (C=O amide), 167.7 and 172.2 (2 C=O ester); MS, *m*/*z* (%): M⁺, 331 (39), 212 (35), 184 (41), 93 (100), 65 (51); Anal. Calcd. for C₁₈H₂₁NO₅ (331.4) C, 65.24; H, 6.39; N, 4.23 %. Found: C, 65.1; H, 6.1; N, 4.1 %.

Di-tert-butyl 3-(anilinocarbonyl)-2-methyl-1-cyclobutene-1,4-dicarboxylate (5c)

White powder, mp 97-99 °C yield 50%; IR (KBr) (v_{max} , cm⁻¹): 3420 (NH), 1697 and 1645 (C=O); ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H}$ 1.53 and 1.54 (18 H, 2 s, 2 CMe₃), 2.1 (3 H, s, CH₃), 3.5 and 3.8 (2 H, s, 2 CH), 7-7.8 (5 H, m, C₆H₅), 8.2 (1 H, br. s, NH); ¹³C NMR (125.8 MHz, CDCl₃): $\delta_{\rm C}$ 15.5 (CH₃), 28.5 and 28.7 (2 CMe₃), 47.4 and 51.1 (2 CH), 81.4 and 82.0 (2 OCMe₃), 120.3, 124.9, 128.9 and 138.1 (Cortho, Cpara, Cmeta and Cipso of NC₆H₅, respectively), 133.2 (C=C-CO₂Me), 157.5 (C=C-CH₃), 161.4 (C=O amide), 168.1 and 171.3 (C=O ester); MS, *m*/*z* (%): 331 (35), 212 (31), 184 (39), 93 (100), 65 (41); Anal. Calcd. for C₂₂H₂₉NO₅ (387.4) C, 68.19; H, 7.54; N, 3.62 %. Found: C, 68.5; H, 7.7; N, 3.5 %.

Preparation of dimethyl (Z)-2-[(E)-3-anilino-1-methyl-3-oxo-1-propenyl]-2-butenedioate(6a)

Compound 5a was refluxed in toluene for 24 hr. The solvent was removed under reduced pressure and the residue was purified by silica gel (Merck silica gel, 230-400 mesh) column chromatography using hexane:ethyl acetate (10:1) as eluent. The solvent was removed under reduced pressure and 6a was obtained as yellow viscous oil, (yield 45%); IR (KBr) (ν_{max} , cm⁻¹): 3315 (NH), 1712 and 1668 (C=O); ¹H NMR (500 MHz, CDCl₃): δ_{H} 2.3 (3 H, s, CH₃), 3.7 and 3.9 (6 H, s, 2 OMe), 6.06 and 6.11 (2 H, s, 2 CH), 7-7.8 (5 H, m, C₆H₅), 8 (1 H, br. s, NH); ¹³C NMR (125.8 MHz, CDCl₃): δ_{C} 14.3 (CH₃), 52.2 and 53.0 (2 OCH₃), 120.2, 126.8, 143.5 and 150.2 (olefinic carbons), 119.8, 124.5, 128.9 and 137.8 (Cortho, Cpara, Cmeta and Cipso of NC₆H₅, respectively), 163.4 (C=O amide), 165.2 and 168.4 (2 C=O ester); MS, *m/z* (%): M⁺, 303 (29), 244 (31), 212

(25), 183 (53), 93 (100), 65 (53); Anal. Calcd. for C₁₆H₁₇NO₅ (303.3) C, 63.36; H, 5.65; N, 4.62 %. Found: C, 63.6; H, 5.8; N, 4.4 %.

Diethyl (Z)-2-{(E)-3-anilino-1-methyl-3-oxo-1-propenyl]-2-butenedioate (6b)

Yellow viscous oil, yield 43%; IR (KBr) (ν_{max} , cm⁻¹): 3315 (NH), 1712 and 1669 (C=O); ¹H NMR (500 MHz, CDCl₃): δ_{H} 1.2 and 1.3 (6 H, 2 t, ³ J_{HH} 6.6 Hz, 2 CH₃), 2.3 (3 H, s, CH₃), 4.1 and 4.3 (4 H, m, 2 OCH₂), 6.1 (2 H, s, 2 CH), 7-7.6 (5 H, m, C₆H₅), 7.8 (1 H, br.s, NH); ¹³C NMR (125.8 MHz, CDCl₃): δ_{C} 13.9 (3 CH₃), 61.2 and 62.1 (2 OCH₂), 119.8, 124.5, 129.0 and 137.8 (Cortho, Cpara, Cmeta and Ctpto of NC₆H₅, respectively), 120.6, 126.3, 143.5 and 150.1 (olefinic carbons), 163.3 (C=O amide), 164.7 and 167.5 (2 C=O ester); MS, m/z (%): M⁺, 331 (4), 258 (27), 212 (20), 93 (100), 65 (39); Anal. Calcd. for C₁₈H₂₁NO₅ (331.4) C, 65.24; H, 6.39; N, 4.23 %. Found: C, 64.6; H, 6.0; N, 4.1 %.

Di-tert-butyl (Z)-2-[(E)-3-anilino-1-methyl-3-oxo-1-propenyl]-2-butenedioate (6c)

Yellow viscous oil, yield 40%; IR (KBr) (v_{max} , cm⁻¹): 3300 (NH), 1704, 1671 (C=O); ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H}$ 1.4 and 1.6 (18 H, 2 s, 2 CMe₃), 2.3 (3 H, s, CH₃), 6.05 and 6.1(2 H, s, 2 CH), 7-7.6 (5 H, m, C₆H₅), 7.6 (1 H, br, NH); ¹³C NMR (125.8 MHz, CDCl₃): $\delta_{\rm C}$ 15.6(CH₃), 28.21 and 28.42 (2 CMe₃), 120.29, 125.02, 130.95 and 138.06 (C_{ortho}, C_{para}, C_{meta} and C_{ipso} of NC₆H₅, respectively), 122.42, 125.48, 145.03 and 149.54 (olefinic carbons), 164.05 (C=O amide), 164.35 and 166.97 (2 C=O ester); MS, *m*/*z* (%): M⁺, 387 (10), 212 (6), 184 (12), 93 (100), 57 (100); Anal. Calcd. for C₂₂H₂₉NO₅ (387.4) C, 68.19; H, 7.54; N, 3.62 %. Found: C, 68.0; H, 7.5; N, 3.4 %.

REFERENCES

- [1] Becker, K. B. Tetrahedron, 1980, 36, 1717; Zbiral, E. Synthesis, 1974, 775.
- [2] Durst, T. and Breau, L. in Comprehensive Organic Synthesis, Selectivity, Strategy and Efficiency in Modern Organic Chemistry, Trost, B. M. (editor-in-Chief), vol. 5, Paquette, L. (volume Editor), Pergamon, 1991, pp. 675-697.
- [3] Moore, H. W. and Decker, O. H. W., Chem. Rev., 1986, 86, 821.
- [4] Pollart, D. J. and Moore, H. W., J. Org. Chem., 1989, 54, 5444.
- [5] Yavari, I. and Samzadeh-Kermani, A. R. Tetrahedron Lett., 1998, 39, 6343.
- [6] Corbridge, D. E. C. Phosphorus an Outline of its Chemistry, Biochenistry and Technology, 1995, Elsevier, pp. 42-47.
- [7] Kolodiazhynyi, O. I. Russ. Chem. Rev., 1997, 66, 225; Cherkasov, R. A. and Pudovik, M. A. Russ. Chem. Rev., 1994, 63, 1019; Pietrusiewiz, K. M. and Zablocka, M. Chem. Rev. 1983, 109, 83.
- [8] Vedejs, E. and Snoble, K. A. J., J. Am. Chem. Soc., 1973, 95, 5778; Allen, D. W. and Ward, H. Tetrahedron Lett, 1979, 2707.
- [9] Gunther H. NMR Spectroscopy, 2nd ed., 1995, Wiley, New York, Chapter 4.