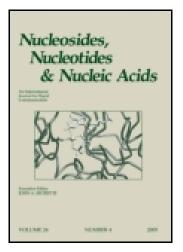
This article was downloaded by: [Chulalongkorn University] On: 10 January 2015, At: 12:03 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Nucleosides and Nucleotides

Publication details, including instructions for authors and subscription information: <u>http://www.tandfonline.com/loi/lncn19</u>

New Phosphitylating Reagents Containing P-F Bond and Their Application in the Synthesis of P-Fluoro Nucleosidyl Phosphates, Thiophosphates and Selenophosphates

W. Dąbkowski $^{\rm a}$, I. Tworowska $^{\rm a}$, E. Poniatowska $^{\rm a}$, J. Michalski $^{\rm a}$ & F. Cramer $^{\rm b}$

^a Centre of Molecular and Macromolecular Studies, Polish Academy of Sciences , 90-363 Łódż, Sienkiewicza, 112, Poland

^b Max-Planck-Institut für experimentelle Medizin , 3400 Göttingen, Hermann-Rein Strasse, 3, Germany Published online: 04 Oct 2006.

To cite this article: W. Dąbkowski, I. Tworowska, E. Poniatowska, J. Michalski & F. Cramer (1999) New Phosphitylating Reagents Containing P-F Bond and Their Application in the Synthesis of P-Fluoro Nucleosidyl Phosphates, Thiophosphates and Selenophosphates, Nucleosides and Nucleotides, 18:6-7, 1247-1248, DOI: <u>10.1080/07328319908044679</u>

To link to this article: http://dx.doi.org/10.1080/07328319908044679

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms &

Conditions of access and use can be found at <u>http://www.tandfonline.com/page/terms-and-conditions</u>

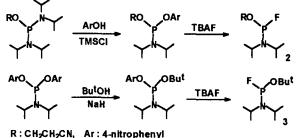
NEW PHOSPHITYLATING REAGENTS CONTAINING P-F BOND AND THEIR APPLICATION IN THE SYNTHESIS OF P-FLUORO NUCLEOSIDYL PHOSPHATES, THIOPHOSPHATES AND SELENOPHOSPHATES

W. Dąbkowski, ^a I. Tworowska, ^a E. Poniatowska, ^a J. Michalski^{a*} and F. Cramer^b ^a Centre of Molecular and Macromolecular Studies, Polish Academy of Sciences, 90-363 Łódź, Sienkiewicza 112, Poland ^b Max-Planck-Institut für experimentelle Medizin, 3400 Göttingen, Hermann-Rein Strasse 3, Germany

Abstract: An efficient and general synthesis of phosphorofluoridates RO-P(X)(OH)F(X=O) and their analogues (X=S, Se) based on two new phosphitylating reagents: 2-cyanoethyl-N,N-diisopropylfluorophosphoroamidite $F-P(NPr_2^i)OCH_2CH_2CN$ and tert-butyl-N,N-diisopropylfluorophosphoroamidite $F-P(NPr_2^i)OBu^t$ is described.

Compounds of the general formula R-O-P(X)(F)OH 1 where X=O (R=nucleosid-3'yl or nucleosid-5'-yl) have been first obtained by Wittmann ¹. Wittmann method and procedures described afterward are not fully satisfactory. We searched for more general synthesis of 1 especially providing approach towards thio- and selenofluorophosphates ².

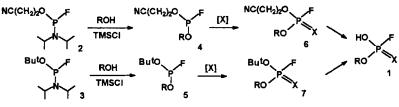
In this communication we present an expedient general method based on a phosphoroamide approach. New phosphitylating reagents containing P-F bond 2 and 3 were synthesized and applied in the synthesis of compounds 1 (X = O, S, Se). The synthetic strategy for the preparation of the phosphitylating reagents 2 and 3 is delineated below.



1247

Preparation of amidites 2 and 3 can be performed in one-flask procedure in over 90% yield. Both compounds are stable colourless liquids which can be purified by destillation *in vacuo* or by silica gel column chromatography and stored at room temperature.

Phosphitylating reagents 2 and 3 react with alcohols in the presence of activators such as tetrazole, benzoyl chloride or trimethylsilyl chloride (TMSCl) at room temperature to give phosphorofluorides 4 and 5 in very high yield. TMSCl proved to be superior to others.



R: a) 2'-deoxynucleosid-3'-yl b) 2'-deoxynucleosid-5'-yl c) citronellyl d) cholesteryl

Oxidation of phosphorofluoridites 4 and 5 by hydrogen tert-butyl peroxide or addition of elemental sulfur in the presence of triethylamine gave the corresponding phosphorofluoridates 6 and 7 which undergo thermal elimination of 2-methyl-1-propen (80°C, 2h) or β -elimination of vinyl cyanide [pyridine-triethylamine (3:1 v/v), r.t., 10 min] to form the final compounds 1 (X=O, S) in excellent yield. All these reactions are best performed as one-flask procedures. It is noteworthy that in the coupling procedure (3 \rightarrow 5) activated by TMSCl the tert-butyl group in unaffected. Preliminary experiments showed that this procedure is applicable for the synthesis of 1 (X=Se).

Examination of the phosphitylation procedure by ${}^{31}P$ and ${}^{19}F$ NMR spectroscopy using phosphorofluoridite 2 clearly shows intermediate formation of fluorophosphorotetrazolidite FP(OCH₂CH₂CN)CHN₄ 8. Formation of fluorochlorophosphine FP(OCH₂CH₂CN)Cl 9 was observed when amidite 2 was allowed to react with benzoyl chloride or TMSCl. It is most likely that the compounds 8 and 9 act as phosphitylating species towards alcohols.

Acknowledgment. This work was supported by the State Committee for Scientific Research, Poland (No 3 T09A 155 10), Stiftung für Deutsch-Polnische Zusammenarbeit (2789/96/LN) and German-Polish project (POL-211-96).

REFERENCES

- 1. Wittmann, R. Chem. Ber., 1963, 96, 771-779.
- 2. Dabkowski, W.; Tworowska, I. Chemistry Letters, 1995, 727-728 and references therein.