## Bioorganic & Medicinal Chemistry Letters 23 (2013) 3013-3017

Contents lists available at SciVerse ScienceDirect



**Bioorganic & Medicinal Chemistry Letters** 



journal homepage: www.elsevier.com/locate/bmcl

# Synthesis and evaluation of 2,3-dinorprostaglandins: Dinor-PGD<sub>1</sub> and 13-*epi*-dinor-PGD<sub>1</sub> are peroxisome proliferator-activated receptor $\alpha/\gamma$ dual agonists

Ayato Sato<sup>a,b</sup>, Kosuke Dodo<sup>a,b</sup>, Makoto Makishima<sup>c</sup>, Yuichi Hashimoto<sup>d</sup>, Mikiko Sodeoka<sup>a,b,\*</sup>

<sup>a</sup> Sodeoka Live Cell Chemistry Project, ERATO, JST, 2-1 Hirosawa, Wako-shi, Saitama 351-0198, Japan

<sup>b</sup> RIKEN Advanced Science Institute, 2-1 Hirosawa, Wako-shi, Saitama 351-0198, Japan

<sup>c</sup> Division of Biochemistry, Department of Biomedical Sciences, Nihon University School of Medicine, 30-1 Oyaguchi-kamicho, Itabashi-ku, Tokyo 173-8610, Japan <sup>d</sup> Institute of Molecular and Cellular Biosciences, The University of Tokyo, 1-1-1 Yayoi, Bunkyo-ku, Tokyo 113-0032, Japan

#### ARTICLE INFO

Article history: Received 17 November 2012 Revised 27 February 2013 Accepted 7 March 2013 Available online 21 March 2013

Keywords: 2,3-Dinorprostaglandin Prostaglandin Fatty acid PPAR γ-Linolenic acid

# ABSTRACT

2,3-Dinorprostaglandins (dinor-PGs) have been regarded as  $\beta$ -oxidation products of arachidonic-acidderived prostaglandins, but their biological activities in mammalian cells remain unclear. On the other hand, C18 polyunsaturated fatty acids (PUFAs), such as  $\gamma$ -linolenic acid (GLA), have various biological activities, and dinor-PGs are speculated to be biosynthesized from GLA. Here, we synthesized dinor-PGs that may possibly be derived from GLA and examined their activities towards peroxisome proliferator-activated receptors (PPARs). Dinor-PGD<sub>1</sub> (1) and its epimer 13-*epi*-dinor-PGD<sub>1</sub> (*epi*-1) were found to be dual agonists for PPAR $\alpha/\gamma$ , whereas PGD<sub>2</sub> derived from arachidonic acid is selective for PPAR $\gamma$ . Thus, GLA-derived dinor-PGs may have unique biological roles.

© 2013 Elsevier Ltd. All rights reserved.

Polyunsaturated fatty acids (PUFAs) and their derivatives are involved in a variety of biological signaling pathways. Among these compounds, C20 PUFAs, arachidonic acid (AA), dihomo- $\gamma$ -linolenic acid (DGLA), and eicosapentaenoic acid (EPA), are converted to a variety of lipid mediators, such as prostaglandins and leukotrienes,<sup>1</sup> that play critical roles in smooth muscle contraction,<sup>2</sup> sleep,<sup>3</sup> platelet aggregation,<sup>4</sup> tumor suppression<sup>5</sup> and inflammation<sup>6</sup> in mammals.

 $\gamma$ -Linolenic acid (GLA), a representative C18 PUFA, has antitumor activity<sup>7</sup> and transcriptional activation activity.<sup>8–11</sup> GLA is generally recognized as a precursor in the biosynthesis of AA, but its biological activities are not always the same as those of AA, suggesting that GLA itself and/or its metabolites also act as physiological modulators. Interestingly, C18 PG-like molecules have been identified in plasma and urine.<sup>12–14</sup> Although they had been assumed to be formed by auto-oxidation of C18 PUFAs or  $\beta$ -oxidation of C20 PGs, they may themselves have significant biological activity. Indeed, biological activities of some dinor–isoprostanes have been reported.<sup>15</sup> Furthermore, GLA serves as a substrate of human prostaglandinendoperoxide H synthases, raising the possibility that dinor-PGs having 18 carbon atoms may be enzymatically produced from GLA, at least in part.<sup>16</sup> It is noteworthy that C18 prostanoids derived from  $\alpha$ -linolenic acid act as injury signaling factors in the plant kingdom.<sup>17,18</sup> But, as far as we know, no detailed biological studies on GLA-derived dinor-PGs have yet been reported. Therefore, we started a project to examine the chemistry and biology of dinor-PGs. Here, we report the syntheses of 2,3-dinorprostaglandin D<sub>1</sub> (dinor-PGD<sub>1</sub>) and 13-deoxy- $\Delta^{10,12}$ -2,3-dinorprostaglandin J<sub>1</sub> (13-deoxy-dinor-PGJ<sub>1</sub>), and the evaluation studies of their activities towards peroxisome proliferator-activated receptor (PPAR) subtypes.

Peroxisome proliferator-activated receptors (PPAR $\alpha$ ,  $\delta$ , and  $\gamma$ ), which belong to the nuclear receptor superfamily of transcription factors, are key regulators of lipid metabolism and are considered to be important drug targets.<sup>19</sup> Various fatty acids, such as AA and GLA, are known to bind and activate PPARs.<sup>20,21</sup> Furthermore, the relationship between the chain length of fatty acids and their PPAR agonistic activities has been discussed.<sup>8,21</sup> In 1995, prostaglandin D<sub>2</sub> (PGD<sub>2</sub>) and its dehydrated metabolites (Fig. 1) were found to be potent activators of PPARs, and among them, 15-deoxy- $\Delta^{12,14}$ -PGJ<sub>2</sub> was significantly more potent than its precursors, AA and PGD<sub>2</sub>.<sup>22,23</sup> Therefore, we speculated that GLA may also be directly transformed into dinor-PGs and that some of these molecules may activate PPARs. To test this hypothesis, we planned to synthesize and evaluate dinor-PGs corresponding to PGD<sub>2</sub>, PGJ<sub>2</sub> and 15-deoxy- $\Delta^{12,14}$ -PGJ<sub>2</sub>, namely 2,3dinorprostaglandin  $D_1$  (dinor-PGD<sub>1</sub>) (**1**), 2,3-dinor-prostaglandin

<sup>\*</sup> Corresponding author. Tel.: +81 48 467 9373; fax: +81 48 462 4666. E-mail address: sodeoka@riken.jp (M. Sodeoka).

<sup>0960-894</sup>X/\$ - see front matter @ 2013 Elsevier Ltd. All rights reserved. http://dx.doi.org/10.1016/j.bmcl.2013.03.024



Figure 1. Hypothetical metabolic pathway of GLA in parallel to the AA cascade.

J<sub>1</sub> (dinor-PGJ<sub>1</sub>) (**2**), and 13-deoxy- $\Delta^{10,12}$ -2,3-dinor-prostaglandin J<sub>1</sub> (13-deoxy-dinor-PGJ<sub>1</sub>) (**3**) (Fig. 1).

For the synthesis of these molecules, we used Corey's lactone as the starting material.<sup>24</sup> As shown in Scheme 1, after sequential protection of its alcohols with *t*-butyldiphenylsilyl (TBDPS) and ethoxyethyl (EE) groups, **4** was converted to the lactol by reduction with diisobutyl-aluminum hydride (DIBAL). To construct the  $\alpha$ chain, we first attempted to introduce the C-3 unit directly into the lactol by means of a Wittig reaction.<sup>25,26</sup> but this was unsuccessful. Therefore, we next tried to construct the α-chain in a stepwise manner. A C1 unit was successfully introduced by using Ohira-Bestmann homologation<sup>27</sup> in 69% yield over two steps, then acetylation of the resulting secondary alcohol gave alkyne 5 in 86% yield. Next, introduction of a C2 unit was achieved by a Cu-mediated coupling reaction with ethyl iodoacetate to give ester 6 in 89% yield.<sup>28</sup> Hydrogenation of **6** followed by removal of the TBDPS group afforded primary alcohol 7 in 94% yield over two steps. Swern oxidation of 7 and Horner-Wadsworth-Emmons reaction of the resulting aldehyde gave enone **8** in 69% yield over two steps. The diastereoselective reduction of 8 with (S)-BINAL-H<sup>29</sup> was examined, but the stereoselectivity was low (37% de). Other stereoselective reductions of 8 employed previously in prostaglandin synthesis<sup>30,31</sup> also gave unsatisfactory results. Therefore, we decided to employ Luche conditions for large-scale preparation. The reaction proceeded smoothly to give **9a** and **9b** in 49% and 47% yields, respectively.<sup>32</sup> The diastereomers were separated for use in subsequent reactions.

With compounds **9a** and **9b** in hand, we moved to the synthesis of **1–3** (Fig. 1). As shown in Scheme 2, protection of **9a** with TBDPS gave **10a** in 93% yield, then hydrolysis of the acetyl group and ethyl ester gave hydroxyacid **11a** in 96% yield. Protection of the secondary alcohol and carboxylic acid with *t*-butyldimethylsilyl triflate (TBSOTf) afforded stable bis-TBS ether ester, and consecutive treatment with acetic acid afforded carboxylic acid **12a** in quantitative yield over two steps. It was unexpected that the ethoxyethyl group was tolerated under these conditions, and selective removal of this group from **12a** in the presence of the TBS group was troublesome.



**Scheme 1.** Preparations of **9a** and **9b**. Reagents and conditions: (a) TBDPSCI, imidazole, DMF,  $-40 \circ$ C, 2 h, 84%; (b) ethyl vinyl ether, PPTS, CH<sub>2</sub>Cl<sub>2</sub>, 0  $\circ$ C, 1.5 h then rt, 1 h, 84%; (c) DIBAL, toluene,  $-78 \circ$ C, 30 min; (d) MeCOCN<sub>2</sub>PO(OMe)<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub>, MeOH, rt, 14 h then 45  $\circ$ C, 5 h, 69% over two steps; (e) Ac<sub>2</sub>O, Et<sub>3</sub>N, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, rt, 1.5 h, 86%; (f) ICH<sub>2</sub>CO<sub>2</sub>Et, Cul, MeCN, rt, 24 h, 89%; (g) PtO<sub>2</sub>, H<sub>2</sub>. AcoEt, rt, 3 h; (h) TBAF, THF, 0  $\circ$ C to rt, 22.5 h, 94% over two steps; (i) (COCl)<sub>2</sub>, DMSO,  $-78 \circ$ C, 15 min; Et<sub>3</sub>N, 0  $\circ$ C, 45 min; (j) dimethyl-2-oxoheptyl phosphonate, NaH, THF, 0  $\circ$ C to rt, 69% over two steps; (k) NaBH<sub>4</sub>, CeCl<sub>3</sub>-7H<sub>2</sub>O, MeOH,  $-20 \circ$ C, 10 min, 49% for **9a**, 47% for **9b**.

After investigation of several reaction conditions, selective removal of the ethoxyethyl group was accomplished by treatment of **12a** with CeCl<sub>3</sub>·7H<sub>2</sub>O in MeOH at 10 °C for 2 d, affording the desired secondary alcohol **13a** in 84% yield. Oxidation of **13a** using Dess-Martin periodinane (DMP) gave protected dinor-PGD<sub>1</sub> **14a** (67%), and subsequent removal of TBS and TBDPS groups employing aqueous HF solution afforded **1** in 37% yield. By using the same conditions, 13-*epi*-dinor-PGD<sub>1</sub> (*epi*-**1**) was similarly synthesized from **9b**.<sup>33</sup>



**Scheme 2.** Syntheses of dinor-PGD<sub>1</sub> (1) and 13-*epi*-dinor-PGD<sub>1</sub> (*epi*-1). Reagents and conditions: (a) TBDPSCI, imidazole, DMAP, DMF/Et<sub>3</sub>N (5:1), 93% for **10a** and 86% for **10b**; (b) NaOH, MeOH/H<sub>2</sub>O (10:1), 96% for **11a** and 78% for **11b**; (c) TBSOTf, 2,6-lutidine, CH<sub>2</sub>Cl<sub>2</sub>,  $-60 \,^{\circ}$ C, 15 min; (d) THF/AcOH/H<sub>2</sub>O (4:1:1), quant. over two steps for both **12a** and **12b**; (e) CeCl<sub>3</sub>.7H<sub>2</sub>O, MeOH/H<sub>2</sub>O (20:1), 10  $^{\circ}$ C, 2 d, 84% for **13a** and 86% for **13b**; (f) DMP, CH<sub>2</sub>Cl<sub>2</sub>, 0  $^{\circ}$ C, 6 h, 67% for **14a** and 98% for **14b**; (g) HF, MeCN/H<sub>2</sub>O (20:1),  $-10 \,^{\circ}$ C, 6 d, 37% for **1** and 57% for *epi-1*.

Next we examined the synthesis of dinor-PGI<sub>1</sub> ( $\mathbf{2}$ ) and 13deoxy- $\Delta^{10,12}$ -dinor-PGI<sub>1</sub> (**3**). Unfortunately, attempts to synthesize 2 and 3 directly from 14a or 1 by elimination of either silanol or water were unsuccessful. Therefore we decided to try the macrolactone method reported by Bundy et al. (Scheme 3).<sup>34</sup> Lactonization of 11a gave eight-membered lactone 15 in 72% yield. Removal of the ethoxyethyl group gave the secondary alcohol (80%), and subsequent Dess-Martin oxidation afforded dinor-PGD<sub>1</sub> lactone **16** in quantitative yield. Treatment of **16** with silica gel resulted in  $\beta$ -elimination to give the desired carboxylic acid 17 in 67% yield. However, deprotection of the TBDPS group using HF did not give the desired dinor-PGJ $_1$  **2**, and further dehydration proceeded to afford 13-deoxy- $\Delta^{10,12}$ -dinor-PGJ<sub>1</sub> (**3**) in 68% yield. All attempts to synthesize 2 were unsuccessful due to rapid elimination of the hydroxyl group at C13. Interestingly, when the same reactions were performed starting from the epimer 11b, 13-epidinor-PGJ1 was obtained as a minor product (34%), though again **3** was formed as the major product (62%).<sup>35</sup>



**Scheme 3.** Synthesis of 13-deoxy-Δ<sup>10,12</sup>-dinor-PGJ<sub>1</sub> (**3**). Reagents and conditions: (a) PySSPy, PPh<sub>3</sub>, benzene; benzene (4 mM), reflux, 72%; (b) THF/H<sub>2</sub>O/AcOH (2:1:6), 80%; (c) DMP, CH<sub>2</sub>Cl<sub>2</sub>, quant.; (d) silica gel (5000 wt %), *n*-hexane/AcOEt (7:3), 67%; (e) HF, MeCN/H<sub>2</sub>O (20:1), 0 °C, 1 week, 68%.



**Figure 2.** PPAR-agonistic activities of fenofibric acid (10  $\mu$ M, PPAR $\alpha$  agonist), GW50516 (100 nM, PPAR $\delta$  agonist) and ciglitazone (10  $\mu$ M, PPAR $\gamma$  agonist).

To evaluate the activities of these synthetic dinor-PGs towards PPARs, PPAR reporter assay using HEK293 cells was conducted according to a previous report.<sup>36</sup> As controls, we first evaluated the known subtype-selective PPAR ligands, fenofibric acid, GW501516, and ciglitazone. These synthetic PPAR ligands showed the expected subtype-selective activations (Fig. 2). Then, we examined the effects of the synthesized C18 dinor-PGs, compared with those of the parent C20 prostaglandins (Fig. 3). Dinor-PGD<sub>1</sub> (1) and its epimer 13-*epi*-dinor-PGD<sub>1</sub> (*epi*-1) both showed PPAR $\alpha$  agonistic activity, whereas PGD<sub>2</sub> activated PPAR $\gamma$  more strongly than PPAR $\alpha$  or  $\delta$ . According to previous reports,<sup>22,23</sup> PGD<sub>2</sub> is transformed to PGJ<sub>2</sub> and 15-deoxy-PGJ<sub>2</sub> in cells, and 15-deoxy-PGJ<sub>2</sub> shows the most potent PPAR $\gamma$  activation among them. In contrast, 13-deoxy-dinor-PGJ<sub>1</sub> (3), which is the dinor-PG corresponding to 15-deoxy-PGJ<sub>2</sub>, showed only weak activity towards PPAR $\gamma$ , as well as towards PPAR $\alpha/\delta$ .

We next examined the dose-dependent activations of PGD<sub>2</sub>, **1**, and *epi*-**1** (Fig. 4). Both **1** and *epi*-**1** showed stronger PPAR $\alpha$  and weaker PPAR $\gamma$  agonistic activities relative to PGD<sub>2</sub>. These results indicate the change of subtype selectivity from  $\gamma$  to  $\alpha$  in **1** and *epi*-**1**.

Based on the reported crystal structures of PPAR $\alpha$ -LBD complex with synthetic ligand AZ 242<sup>37</sup> and PPAR $\gamma$ -LBD complex with 15deoxy-PGJ<sub>2</sub><sup>38</sup> (Fig. 5a), we speculated that the reason for the PPAR $\alpha$ -selectivity of dinor-PGD<sub>1</sub> is as follows. During PPAR activation, the hydrophobic  $\omega$ -chain of PGs and dinor-PGs is thought to enter the cavity of PPARs, where it is surrounded by hydrophobic amino acid residues. The carboxylic acid moiety of the  $\alpha$ -chain interacts with hydrophilic amino acids (His and Tyr in Fig. 5a) and changes the position of helix-12. The movement of helix-12 induces conformational change and recruitment of coactivators, leading to gene expression. Therefore, the position of the carboxylate should be critical for the activation. Distances between the



Figure 3. PPAR-agonistic activities of PGD<sub>2</sub> (10  $\mu$ M), PGJ<sub>2</sub> (3  $\mu$ M), 15-deoxy-PGJ<sub>2</sub> (3  $\mu$ M), and dinor-PG compounds 1, *epi*-1, and 3 (10  $\mu$ M).

hydrophobic pocket and the hydrophilic amino acid residues are different between the two crystal structures, presumably reflecting the conformation required for activation. Dinor-PGD<sub>1</sub> (**1**) having a shorter  $\alpha$ -chain, fits comfortably into the pocket of PPAR $\alpha$  in comparison to AZ 242. On the other hand, the shorter  $\alpha$ -chain appears to be too short to interact with the hydrophilic amino acid residues of helix-12 in PPAR $\gamma$ . As a result, dinor-PGD<sub>1</sub> clearly showed stronger agonistic activity for PPAR $\alpha$  while at the same time weaker activity for PPAR $\gamma$ , compared to PGD<sub>2</sub>.

Finally, to confirm the binding model, we conducted a docking study of PPAR $\alpha$  (PDB ID 117G) and **1** with AutoDock 4.2 software<sup>39</sup> (Fig. 6). As predicted above, the results indicate that the carboxylic acid moiety of **1** can access the hydrophilic pocket of PPAR $\alpha$ 



Figure 4. Dose-dependent PPAR-agonistic activities of PGD<sub>2</sub>, 1, and epi-1.

through hydrogen bonding (Tyr464, Tyr314 and Ser280), supporting our binding model of **1**.

In this study, we have succeeded in synthesizing dinor-PGD<sub>1</sub> (1), 13-*epi*-dinor-PGD<sub>1</sub> (*epi*-1) and 13-deoxy-dinor-PGJ<sub>1</sub> (3), and we examined their effects on the activity of PPARs. Dinor-PGD<sub>1</sub> (1) and its epimer, 13-*epi*-dinor-PGD<sub>1</sub> (*epi*-1), were found to be PPAR $\alpha/\gamma$  dual agonists. Synthetic methods previously used to obtain PGs were not applicable to dinor-PGs, even though there is only a two-carbon difference in the  $\alpha$ -chain. Our new synthetic routes should be applicable for the synthesis of other dinor-PGs. Further synthetic and biological studies are under way.



**Figure 6.** Docking study of dinor-PGD<sub>1</sub> (1) with PPAR $\alpha$ . Calculated model of PPAR $\alpha$  with 1 constructed by AutoDock 4.2. The image was drawn with PyMOL.



Figure 5. (a) Reported X-ray crystal structures of PPARα (PDB ID 117G) and PPARγ (PDB ID 2ZK1) complexed with agonists. The images were drawn by PyMOL; (b) binding model of dinor-PGD<sub>1</sub> (1) with PPARα and PPARγ.

### Acknowledgment

We thank Dr. M. Ishikawa (The University of Tokyo) for helpful discussions about the docking study.

#### **References and notes**

- 1. Griffiths, G.; Morse, N. J. Am. Oil Chem. Soc. 2006, 83, 171.
- 2. Horton, E. W. Br. Med. Bull. 1979, 35, 295.
- Ueno, R.; Honda, K.; Inoue, S.; Hayashi, O. Proc. Natl. Acad. Sci. U.S.A. 1983, 80, 1735.
- 4. MacIntyre, D. E.; Gordon, J. L. Nature 1975, 258, 337.
- 5. Fukushima, M. Eicosanoids 1990, 3, 189.
- 6. Whelan, C. J. J. Pharm. Pharmacol. 1974, 26, 355.
- 7. Das, U. N. Prostaglandins Leukot. Essent. Fatty Acids 2004, 70, 539.
- 8. Forman, B. M.; Chen, J.; Evans, R. M. Proc. Natl. Acad. Sci. U.S.A. 1997, 94, 4312.
- Mochizuki, K.; Suzuki, T.; Goda, T. J. Nutr. Sci. Vitaminol. 2008, 54, 298.
   Chêne, G.; Dubourdeau, M.; Balard, P.; Escoubet-Lozach, L.; Orfila, C.; Berry, A.; Bernad, J.; Aries, M.-F.; Charveron, M.; Pipy, B. Biochim. Biophys. Acta 2007,
- 1771, 576.
  11. Tsai, N.-P.; Huq, M.; Gupta, P.; Yamamoto, K.; Kagechika, H.; Wei, L.-N. *Biochim. Biophys. Acta* 2009, 1789, 734.
- Mueller, M. J. Chem. Biol. **1998**, 5, R323.
   Burke, A.; Lawson, J. A.; Meagher, E. A.; Rokach, J.; FitzGerald, G. A. J. Biol. Chem. **2000**, 275, 2499.
- 14. Granström, E.; Inger, U.; Samuelsson, B. J. Biol. Chem. 1965, 240, 457.
- Hou, X.; Roberts, L. J.; Taber, D. F.; Morrow, J. D.; Kanai, K.; Gobeil, F., Jr.; Beauchamp, M. H.; Bernier, S. G.; Lepage, G.; Varma, D. R.; Chemtob, S. Am. J. Physiol. Regul. Integr. Comp. Physiol. 2001, 281, R391.
- Laneuville, O.; Breuer, D. K.; Xu, N.; Huang, Z. H.; Gage, D. A.; Watson, J. T.; Lagarde, M.; DeWitt, D. L.; Smith, W. L. J. Biol. Chem. **1995**, 270, 19330.
- 17. Vick, B. A.; Zimmerman, D. C. Plant Physiol. 1984, 75, 458.
- 18. Groenewald, E. G.; van der Westhuizen, A. J. Bot. Rev. 1997, 63, 200.
- Pirat, C.; Farce, A.; Lebègue, N.; Renault, N.; Furman, C.; Millet, R.; Yous, S.; Speca, S.; Berthelot, P.; Desreumaux, P.; Chavatte, P. J. Med. Chem. 2012, 55, 4027.
- Xu, H. E.; Lambert, M. H.; Montana, V. G.; Parks, D. J.; Blanchard, S. G.; Brown, P. J.; Sternbach, D. D.; Lehmann, J. M.; Wisely, G. B.; Willson, T. M.; Kliewer, S. A.; Milburn, M. V. *Mol. Cell* **1999**, 3, 397.
- Krey, G.; Braissant, O.; L'Horset, F.; Kalkhoven, E.; Perroud, M.; Parker, M. G.; Wahli, W. Mol. Endocrinol. 1997, 11, 779.
- 22. Forman, B. M.; Tontonoz, P.; Chen, J.; Brun, R. P.; Spiegelman, B. M.; Evans, R. M. Cell **1995**, 83, 803.
- Kliewer, S. A.; Lenhard, J. M.; Willson, T. M.; Patel, I.; Morris, D. C.; Lehmann, J. M. Cell 1995, 83, 813.

- Corey, E. J.; Weinshenker, N. M.; Schaaf, T. K.; Huber, W. J. Am. Chem. Soc. 1969, 91, 5675.
- 25. Sih, J. C.; Nash, S. A. Prostaglandins 1977, 14, 407.
- 26. Prakash, C.; Saleh, S.; Blair, I. A. Synth. Commun. 1988, 18, 2233.
- 27. Ohira, S. Synth. Commun. 1989, 19, 561.
- 28. Suárez, A.; Fu, G. C. Angew. Chem. Int. Ed. 2004, 43, 3580.
- Noyori, R.; Tomino, I.; Yamada, M.; Nishizawa, M. J. Am. Chem. Soc. 1984, 106, 6717.
- Corey, E. J.; Bakshi, R. K.; Shibata, S.; Chen, C.-P.; Singh, V. K. J. Am. Chem. Soc. 1987, 109, 7925.
- Iguchi, S.; Miyata, Y.; Okuyama, S.; Miyake, H.; Okegawa, T. Chem. Pharm. Bull. 1988, 36, 1128.
- 32. Stereochemistry of the newly formed secondary alcohol at C13 was determined by examination of the MTPA ester. See: Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. J. Am. Chem. Soc. 1991, 113, 4092.
- 33. Dinor-PGD<sub>1</sub> (1): <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  5.54 (dd, J = 15.4, 7.1 Hz, 1H), 5.41 (dd, J = 15.4, 8.1 Hz, 1H), 4.41 (m, 1H), 4.01 (ddd, J = 13.0, 6.4, 6.4 Hz, 1H), 2.68 (dd, J = 11.7, 8.1 Hz, 1H), 2.49 (dd, J = 18.6, 6.4 Hz, 1H), 2.27–2.30 (m, 3H), 1.98 (m, 1H), 1.32–1.68 (m, 16H), 0.90 (t, J = 6.9 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  216.4, 179.0, 135.9, 125.9, 74.4, 68.5, 54.3, 48.5, 37.8, 33.9, 31.7, 27.5, 27.0, 26.7, 24.9, 24.3, 22.5, 14.0; HRMS (ESI) (m/z) calcd for C<sub>18</sub>H<sub>29</sub>O<sub>5</sub> [M–H]<sup>-</sup>: 325.2020, found: 325.2010; 13-epi-dinor-PCD<sub>1</sub> (epi-1): <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  5.51 (dd, J = 15.4, 6.6 Hz, 1H), 5.20 (dd, J = 15.4, 8.2 Hz, 1H), 4.35 (m, 1H), 4.10 (ddd, J = 13.9, 6.4, 6.4 Hz, 1H), 2.60 (dd, J = 11.8, 8.2 Hz, 1H), 2.33 (dd, J = 18.6, 6.4 Hz, 1H), 2.28–2.32 (m, 3H), 1.98–1.94 (m, 2H), 1.34–1.72 (m, 15H), 0.80 (t, J = 9.5 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  216.2, 172.0, 136.1, 125.8, 74.0, 68.6, 54.1, 48.6, 37.7, 31.7, 27.7, 27.0, 26.7, 25.7, 24.9, 24.3, 19.3, 14.0; HRMS (ESI) (m/z) calcd for C<sub>18</sub>H<sub>29</sub>O<sub>5</sub> [M–H]<sup>-</sup>: 325.2020, found: 325.2024.
- Bundy, G. L.; Morton, D. R.; Peterson, D. C.; Nishizawa, E. E.; Miller, W. L. J. Med. Chem. **1983**, 26, 790.
   13-Deoxy-Δ<sup>10,12</sup>-dinor-PGJ<sub>1</sub> (**3**): <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.50 (dd, J = 6.0,
- 35. 13-Deoxy-∆<sup>10,12</sup>-dinor-PGJ<sub>1</sub> (**3**): <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.50 (dd, *J* = 6.0, 0.5 Hz, 1H), 6.92 (d, *J* = 10.5 Hz, 1H), 6.34 (dd, *J* = 6.0, 2.0 Hz, 1H), 6.22 (m, 2H), 3.55 (dddd, *J* = 5.9, 3.9, 2.0, 0.5 Hz, 1H), 2.32 (dt, *J* = 7.1, 3.9 Hz, 2H), 2.20 (dt, *J* = 7.4, 6.9 Hz, 2H), 1.88 (m, 1H), 1.27-1.69 (m, 11H), 0.87 (t, *J* = 6.9 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 197.6, 178.5, 160.8, 147.1, 135.4, 135.2, 131.7, 125.5, 43.3, 33.6, 33.4, 32.5, 31.8, 28.4, 25.3, 24.8, 22.4, 14.0; HRMS (ESI) (*m*/*z*) calcd for C<sub>18</sub>H<sub>25</sub>O<sub>3</sub> [M−H]<sup>-</sup>: 289.1809, found: 289.1811.
- Kasuga, J.; Makishima, M.; Hashimoto, Y.; Miyachi, H. Bioorg. Med. Chem. Lett. 2006, 16, 554.
- Cronet, P.; Petersen, J. F.; Folmer, R.; Blomberg, N.; Sjöblom, K.; Karlsson, U.; Lindstedt, E. L.; Bamberg, K. Structure 2001, 9, 699.
- Waku, T.; Shiraki, T.; Oyama, T.; Fujimoto, Y.; Maebara, K.; Kamiya, N.; Jingami, H.; Morikawa, K. J. Mol. Biol. 2009, 385, 188.
- Morris, G. M.; Goodsell, D. S.; Halliday, R. S.; Huey, R.; Hart, W. E.; Belew, R. K.; Olson, A. J. J. Comput. Chem. 1998, 19, 1639.