This article was downloaded by: [Universitaets und Landesbibliothek] On: 01 December 2013, At: 20:49 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

Facile Syntheses of the Trisaccharide Acceptors—The Key Intermediates for Assembling the Elicitor Hexasaccharide

Wei Wang ^a & Fanzuo Kong ^a

^a Research Center for Eco-Environmental Sciences, Academia Sinica, P.O.Box 287 J, Beijing, 100085, P. R. China Published online: 17 Sep 2007.

To cite this article: Wei Wang & Fanzuo Kong (1999) Facile Syntheses of the Trisaccharide Acceptors—The Key Intermediates for Assembling the Elicitor Hexasaccharide, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 29:18, 3179-3190, DOI: <u>10.1080/00397919908085942</u>

To link to this article: <u>http://dx.doi.org/10.1080/00397919908085942</u>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages,

and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

FACILE SYNTHESES OF THE TRISACCHARIDE ACCEPTORS —THE KEY INTERMEDIATES FOR ASSEMBLING THE ELICITOR HEXASACCHARIDE

Wei Wang, Fanzuo Kong*

Research Center for Eco-Environmental Sciences, Academia Sinica, P.O.Box 2871, Beijing 100085, P. R. China

Abstract: A facile synthesis of trisaccharide 7, the key intermediate for assembling phytoalexin glucohexatose elicitor, was achieved via orthoester formation-rearrangement from stepwise coupling of 1,2-O-ethylidene- α -D-glucopyranose (2) with 2,3,4-tri-O-acetyl-6-O-chloroacetyl- α -D-glucopyranosyl bromide at 6-position and with acetobromoglucose at 3-position respectively. Similar strategy was used for the synthesis of 14.

The hexasaccharide $Glu\beta1\rightarrow 6(Glu\beta1\rightarrow 3)Glu\beta1\rightarrow 6Glu\beta1\rightarrow 6(Glu\beta1\rightarrow 3)Glu$ has been well known as the basic structure for phytoalexin elicitor activity.¹ It has similar action as the corresponding heptasaccharide which is effective in very low doses—approximately 0.1 pmol per cotyledon^{2a} and gives a half-maximal activity at a concentration of 10 nM.^{2b} So far most of the methods for the synthesis of the elicitor oligosaccharides have involved the use of rare and expensive reagents, multifunctional groups protection, and lengthy route of chemical modifications.³

^{*}To whom correspondence should be addressed.





Acceptor

In the synthesis of elicitor hexasaccharide, the trisaccharide acceptor needs to have structurally different B and C unit enabling to make a free hydroxyl group at the C-6 of B unit. In our previous research,⁴ the phytoalexin elicitor hexasaccharide was synthesized convergently through building two trisaccharide blocks as glycosylation donor and acceptor respectively (Scheme 1). Recently we developed a new and effective strategy for highly regio- and stereoselective synthesis of oligosaccharides using un- or less-protected sugar acceptors and *O*-acetylglycosyl bromides as raw materials through orthoester formation-rearrangement.⁵ However, it was found that the reaction with unprotected methyl glucoside as the acceptor was rather slow and difficult to be monitored due to the poor solubility of the acceptor in the reaction media. Glucose 1,2-*O*-ethylidenate (2) is a readily available material with better solubility and it was used as the glycosyl acceptor in the present study for the synthesis of the trisaccharide acceptor **13** bearing 4,6-*O*-benzylidene at the B unit through orthoester formation-rearrangement.

Scheme 1

As shown in scheme 2, coupling⁵ of 2,3,4-tri-O-acetyl-6-O-chloroacetyl- α -Dglucopyranosyl bromide 1 (1.1 equiv) with 1,2-O-ethylidene- α -D-glucopyranose 2 (1 equiv) promoted by AgOTf (1.1 equiv)/lutidine (1.2 equiv) in anhydrous CH₂Cl₂ furnished 1,6-linked disaccharide orthoester (*exo*) 3 (R,S, 82%) after column



Conditions and reagents: a. AgOTf (1 equiv), 2,4-lutidine (1.2 equiv), anhyd CH_2Cl_2 , M.S.(4 Å), RT, N₂, 3 h. b. TMSOTf (0.1 equiv), anhyd CH_2Cl_2 , M.S.(4 Å), -30 °C, N₂, 40 min. c. Ac₂O/pyridine (dry), CH_2Cl_2 . d. Br₂ (3 equiv), Ph₃P (1.5-2 equiv.), and pyridine (dry, 0.25 equiv) in anhyd. CH_2Cl_2 , RT, 2 h. e. 80% aq.CHCl₂COOH/AcOH, 4/50, RT, 3 h.

chromatography with petroleum ether/ethyl acetate (1/1.5) as the eluent. Similar preparation of the disaccharide orthoester from 2 and acetobromoglucose did not give very good regioselectivity perhaps due to the higher reactivity of the latter. Acetylation of 3 (giving 4) confirmed the 6-O-glycosylation as indicated from its 2-D ¹H NMR showing H-6 at δ 3.62 (dd), 3.57 (dd) and H-3 at δ 5.22 (t), H-4 at δ 4.95 (dd) respectively. Rearrangement of 3 with a catalytic amount of TMSOTf in anhydrous CH₂Cl₂ yielded disaccharide 5 (R,S) in a good yield (81%). Glycosylation of 5 with acetobromoglucose under the same condition as described for the preparation of 3 followed by acetylation in anhydrous CH₂Cl₂ gave the 3-linked trisaccharide orthoester (*exo*) 6 in a yield of 78% (two steps) after purification by column chromatography. Trisaccharide 7 was obtained also in a satisfactory yield (76%) by isomerization with a catalytic amount of TMSOTf (0.1 equiv) and 7 gave physical data and ¹H NMR spectrum the same as reported⁴ for the same compound prepared in another way.

An alternative trisaccharide acceptor 13 was also effectively synthesized similarly as shown in Scheme 2. Thus bromination of 2,3-di-O-acetyl-4,6-Obenzylidene- α , β -D-glucopyranose (9)^{6.7} with bromine-triphenyl phosphine gave 2,3di-O-acetyl-4,6-O-benzylidene- α -D-glucopyranosyl bromide (10) in a satisfactory yield (87%). We found that the use of bromine-triphenyl phosphine instead of bromine-triphenyl phosphite⁸ made more easily the bromination of sugar derivatives with C-1 free hydroxyl group. The disaccharide acceptor 11 was prepared readily from condensation of acetobromoglucose with 4,6-O-benzylidene-1,2-O-ethylidene- α -D-glucopyranose followed by debenzylidenation.⁴ Coupling⁵ of 10 with 11 followed by acetylation yielded the desired 6-linked trisaccharide orthoester (*exo*) 12 as the sole product in a very high yield (90%). Rearrangement⁵ of 12 with a catalytic amount of TMSOTf (0.1 equiv) readily yielded trisaccharide 13 also in a high yield (84%). The benzylidene group was stable in both the orthoester formation and rearrangement. The trisaccharide 13 was a key intermediate for the elicitor synthesis as its debenzylidenation with 80% dichloroacetic acid in acetic acid furnished trisaccharide acceptor 14 (R,S, 87%) having 4_{B} , 6_{B} -free hydroxyl groups, while selective coupling of 14 with the trisaccharide donor can afford the desired hexasaccharide.

In summary, here we present an effective method for the synthesis of 3,6branched trisaccharides, the key intermediates for assembling the phytoalexin hexasaccharide, with naked glucose 1,2-O-ethylidenate and O-acetylglycosyl bromide or 4,6-O-benzylidenated glucosyl bromide as the starting materials by selective glycosylation via orthoester formation-rearrangement. In addition, the bromination of sugar derivatives bearing acidic labile groups with bromine-triphenyl phosphine as the reagents was achieved. As many naturally bioactive structures contain $1\rightarrow 6$ and $1\rightarrow 3$ glycosidic linkages, the presented strategy will be useful for the synthesis of complex oligosaccharides.

EXPERIMENTAL

Melting points were determined with a 'Mel-Temp' apparatus. Optical rotations were determined at 20 °C with a Perkin-Elmer Model 241-MC automatic polarimeter. ¹H NMR spectra were recorded with Varian XL-200 or XL-400 spectrometer for solutions in CDCl₃. Chemical shifts are given in ppm downfield from internal Me₄Si. TLC was performed on silica gels G and HF, with detection either by charring with 30% (v/v) H₂SO₄ and MeOH or by UV light. Column chromatography was conducted on columns (16 \times 240 mm, 18 \times 300 mm) of silica gel (100-200 mesh). Solutions were concentrated at <60 °C under diminished pressure. All of the described orthoesters and the oligosaccharides containing 1,2-*O*-ethylidenated sugar residues were obtained as R,S mixtures which were used directly in the next reaction. Intensive purification of the mixtures usually gave one isomer (R or S), which was subjected to NMR spectrometry analysis, but isomerization occurred upon a long term standing or during NMR determination.

3,4-Di-O-acetyl-6-O-chloroacetyl-a-D-glucopyranose 1,2-(1,2-O-Ethylidene- α -D-glucopyranos)-6-yl Orthoacetate (3). To a stirred solution of 2,3,4-tri-Oacetyl-6-O-chloroacetyl-a-D-glucopyranosyl bromide⁹ (488 mg, 1.1 mmol), and 2,4lutidine (140 μL, 1.2 mmol), 1,2-O-ethylidene(R,S)-α-D-glucopyranose 2 (206 mg, 1 mmol) and 4 Å molecular sieves (0.5 g) in anhydrous dichloromethane (20 mL) under nitrogen atmosphere was added silver triflate (283 mg, 1.1 mmol) in a dark room, and the reaction was carried out at room temperature and monitored by TLC (1:1.5 petroleum ether/ethyl acetate). After completion of the reaction, the mixture was partitioned between dichloromethane and water, the organic phase was washed with sat. aq. NaHCO₃, 10% Na₂S₂O₃, and concentrated, dried, then subjected to column chromatography with 1:1.5 petroleum ether/ethyl acetate as the eluent, giving the disaccharide orthoester 3 in a good yield (R,S, 467 mg, 82%, based on acceptor 2); $[\alpha]_{D}^{20}$ +21.4° (R, c 1.0, CHCl₃); ¹H NMR: δ (R) 5.74 (d, 1 H, J_{1.2} 5.2 Hz, H-1'), 5.55 (d, 1 H, J₁₂ 5.1 Hz, H-1), 5.19 (t, 1 H, J 2.7 Hz, H-3'), 5.13 (q, 1 H, J 4.8 Hz, CH₃CH), 4.90 (dd, 1 H, J_{3'4'} 2.7 Hz, J_{4'5'} 8.9 Hz, H-4'), 4.42-4.40 (m, 1 H, H-2'), 4.32-4.30 (m, 2 H, H-6'), 4.13 (s, 2 H, ClCH₂CO), 4.04 (t, 1 H, J 5.1 Hz, H-2), 4.00-3.90 (m, 2 H, H-3, 4), 3.88-3.85 (m, 1 H, H-5'), 3.76 (d, 2 H, J_{5.6} 3.2 Hz, H-6), 3.66-3.62 (m, 1 H, H-5), 2.12, 2.10 (2 s, 6 H, 2 CH₃CO), 1.74 (s, 3 H, CH₃CO₃), 1.48 (d, 3 H, J 4.8 Hz, CH₃CH)

Anal. Calcd for C₂₂H₃₁O₁₅Cl: C, 46.28; H, 5.47. Found: C, 46.41; H, 5.77.

3,4-Di-O-acetyl-6-O-chloroacetyl- α -D-glucopyranose 1,2-(3,4-Di-O-acetyl-1,2-O-ethylidene- α -D-glucopyranos)-6-yl Orthoacetate (4). To a solution of 3 (100 mg, 0.18 mmol) in anhydrous dichloromethane (5 mL) and pyridine (5 mL), was added Ac₂O dropwise and the mixture was stirred at room temperature for 3 h, at the end of which time the reaction was complete as indicated by TLC (2/1 petroleum ether/ethyl acetate). Ice water was added, and the mixture was diluted with CH₂Cl₂. The organic layers were combined, dried, and concentrated to give **4** (R,S, 110 mg, 93%) as a syrup. $[\alpha]_{D}^{20}$ +39.7° (S, *c* 1.0, CHCl₃); ¹H NMR: δ (S) 5.73 (d, 1 H, J_{1'.2'} 5.2 Hz, H-1'), 5.61 (q, 1 H, J 4.9 Hz, CH₃CH), 5.60 (d, 1 H, J_{1.2} 4.7 Hz, H-1), 5.22 (t, 1 H, J 5.7 Hz, H-3), 5.17 (t, 1 H, J 2.7 Hz, H-3'), 4.95 (dd, 1 H, J_{3.4} 5.7 Hz, J_{4.5} 9.4 Hz, H-4), 4.89 (dd, 1 H, J_{3'.4'} 2.7 Hz, J_{4'.5'} 8.9 Hz, H-4'), 4.33-4.30 (m, 3 H, H-2', 6'), 4.22 (dd, 1 H, J_{1.2} 4.7 Hz, J_{2.3} 5.7 Hz, H-2), 4.11 (s, 2 H, ClCH₂CO), 3.95-3.88 (m, 2 H, H-5, 5'), 3.62 (dd, 1 H, J_{5.6a} 2.8 Hz, J_{6a,6b} 10.5 Hz, H-6_a), 3.57 (dd, 1 H, J_{5.6b} 4.6 Hz, J_{6a,6b} 10.5 Hz, H-6_b), 2.12, 2.10, 2.09, 2.06 (4s, 12 H, 4 CH₃CO), 1.70 (s, 3 H, CH₃CO₃), 1.33 (d, 3 H, J 4.9 Hz, CH₃CH).

6-O-(2,3,4-Tri-O-acetyl-6-O-chloroacetyl-β-D-glucopyranosyl)-1,2-O-

ethylidene- α -D-glucopyranose (5). To a stirred solution of sugar-sugar orthoester 3 (365 mg, 0.64 mmol) and 4 Å molecular sieves (0.5 g) in anhydrous dichloromethane (10 mL) was added TMSOTf (12 µL, 0.1 equiv) under nitrogen atmosphere at -30°C, and the reaction was monitored by TLC (1:1.5 petroleum ether/ethyl acetate). After completion of the reaction, triethyl amine (20 µL) was added to the mixture. The mixture was filtered, and the filtrate was washed with CH₂Cl₂. The combined solution was washed with N HCl (10 mL), sat. aq. NaHCO₃ (10 mL), and brine (2×10 mL), then dried over anhydrous Na₂SO₄, concentrated. The residue was subjected to column chromatography with 1:1 petroleum ether/ethyl acetate as the eluent, giving the product 5 (R,S, 296 mg, 81%); [α]_D²⁰ +19.5° (R, *c* 0.6, CHCl₃); ¹H NMR: δ (R) 5.59 (d, 1 H, J_{1.2} 5.1 Hz, H-1), 5.23 (t, 1 H, J 9.4 Hz, H-3'), 5.16-5.07 (m, 2 H, H-4', CHCH₃), 5.02 (dd, 1 H, J_{1.2} 7.8 Hz, J_{2.3}, 9.4 Hz, H-2'), 4.63 (d, 1 H, J_{1.2} 7.8 Hz, H-1'), 4.33-4.30 (m, 1 H, H-6'), 4.15 (s, 2 H, ClCH₂CO), 4.16-3.68 (m, 8 H, H-2, 3, 4, 5, 5', 6, 6'), 2.08, 2.05, 2.02 (3 s, 9 H, CH₃CO), 1.49 (d, 3 H, J 4.8 Hz, CHCH₃).

Anal. Calcd for C₂₂H₃₁O₁₅Cl: C, 46.28; H, 5.47. Found: C, 46.12; H, 5.21.

3,4,6-Tri-*O*-acetyl-α-D-glucopyranose 1,2-{6-*O*-(2,3,4-Tri-*O*-acetyl-6-*O*chloroacetyl-β-D-glucopyranosyl)-4-*O*-acetyl-1,2-*O*-ethylidene-α-D-

glucopyranos}-3-yl Orthoacetate (6). As described for the preparation of 3 and 4, coupling of acetobromoglucose (158 mg, 0.39 mmol) with 3 (200 mg 0.35 mmol) followed by acetylation afforded 6 in a high yield (R,S, 256 mg, 78%, two steps). $[\alpha]_D^{20}$ +28.5° (R, c 1.6, CHCl₃); ¹H NMR: δ (R) 5.70 (d, 1 H, J_{1C,2C} 5.2 Hz, H-1_c), 5.50 (d, 1 H, J_{1.2} 4.9 Hz, H-1_A), 5.30-4.93 (m, 6 H, H-2_B, 3_C, 3_B,4_A, 4_B, CH₃CH), 4.92 (dd, 1 H, J_{3C,4C} 2.7 Hz, J_{4C,5C} 9.6 Hz, H-4_C), 4.82 (d, 1 H, J_{1B,2B} 8.0 Hz, H-1_B), 4.40-3.60 (m, 12 H, H-2_A, 2_C, 3_A, 5_A, 5_C, 5_B, 6_A, 6_C, 6_B), 4.14 (s, 2 H, ClCH₂CO), 2.17-1.98 (m, 21 H, 7 CH₃CO), 1.88 (s, 3 H, CH₃CO₃), 1.48 (d, 3 H, J 4.9 Hz, CHCH₃).

Anal. Calcd for C38H51O25Cl: C, 48.39; H, 5.45. Found: C, 48.17; H, 5.26.

4-O-Acetyl-3-O-(2,3,4,6-tetra-O-acetyl-β-D-glucopyranosyl)-6-O-(2,3,4-tri-O-acetyl-6-O-chloroacetyl-β-D-glucopyranosyl)-1,2-O-(R,S)ethylidene-α-D-

glucopyranose (7). As described for the preparation of 5, isomerization of compound 6 (200 mg, 0.21 mmol) with a catalytic amount of TMSOTF (0.1 equiv) furnished the trisaccharide 7 (R,S, 152 mg) also in a satisfactory yield (76%) and gave physical data the same as reported⁴ for the same compound prepared in another way.

2,3-Di-O-acetyl-4,6-O-benzylidene- α -D-glucopyranosyl bromide (10). To a solution of 9^{6,7} (304 mg, 0.86 mmol), pyridine (90 µL) in 6 mL anhydrous dichloromethane, Br₂ (120 µL, 2.58 mmol) in anhydrous dichloromethane (2mL) and Ph₃P (338 mg, 1.5 mmol) was added at 0°C, then the reaction solution was allowed to regain room temperature. The solution was stirred for 5 h, at the end of which time TLC (petroleum ether/ethyl acetate, 2:1) showed the completion of the reaction. The reaction was quenched with 9 mL aq. 10% Na₂CO₃ and poured into ice-cold water, extracted with dichloromethane (3 × 25mL). The organic layer was washed with 5% CuSO₄ (10 mL), and brine (3×10 mL), dried over Na₂SO₄, concentrated to a

syrup, which was subjected to column chromatography with 1:1 petroleum ether/ethyl acetate as the eluent. Compound 10 was obtained as an amorphous solid (310 mg, 87%); The physical data and ¹H NMR spectrum of compound 10 was identical to that reported in lit.⁸

3-O-Acetyl-4,6-O-benzylidene-α-D-glucopyranose 1,2-{3-O-(2,3,4,6-tetra-O-acetyl-β-D-glucopyranosyl)-4-O-acetyl-1,2-O-(R,S)ethylidene-α-D-

glucopyranos}-6-yl orthoacetate (12). To a stirred mixture of 11⁴ (322 mg, 0.6 mmol), 10 (240 mg, 0.62 mmol), 2,4-lutidine (78 µL, 0.68 mmol), and 4 Å molecular sieves (0.5 g) in anhydrous CH_2Cl_2 (10 mL) was added silver triflate (160 mg, 0.62 mmol) under nitrogen atmosphere in a dark room, and the reaction was carried out at room temperature, and monitored by TLC (1/1.5 petroleum ether/ethyl acetate). After completion of the reaction, the mixture was partitioned between CH₂Cl₂ (15 mL) and brine (15 mL), the organic phase was washed with aq. 10% Na₂S₂O₃ (10 mL) and brine (2×10 mL), dried over Na₂SO₄, then concentrated under reduced pressure. The residual oil was purified by column chromatography with 1/1.5 petroleum ether/ethyl acetate as the eluent giving the orthoester (R,S) which was acetylated with Ac₂O/pyridine quantitatively yielded compound 12 (R.S. 490 mg, 90% in two steps) as a syrup after purification by column chromatography. $[\alpha]_{D}^{20}$ +30.1° (S, c 1.0, CHCl₃); ¹H NMR: δ (S) 7.50-7.31 (m, 5 H, Ph-H), 5.79 (d, 1 H, $J_{1B,2B}$ 5.3 Hz, H-1_B), 5.60 (d, 1 H, $J_{1,2}$ 4.5 Hz, H-1_A), 5.53 (q, 1 H, J 4.9 Hz, CH₃CH), 5.52 (s, 1 H, PhCH=), 5.23-5.18 (m, 2 H, H-3_A, 3_B), 5.08, 5.06, 5.04, 4.95 (4 t, J 9.2 Hz, H-2_c, 3_c, 4_A, 4_c), 4.79 (d, 1 H, J 7.9 Hz, H-1_c), 4.39 (dd, 1 H, J_{5.6a} 5.0 Hz, J_{6a,6b} 10.5 Hz, H-6_{Aa}), 4.30-3.59 (m, 12 H, H-2_A, 2_B, 3_A, 4_B, 5_A, 5_C, 5_B, 6_{Ab}, 6_C, 6_B), 2.13, 2.08, 2.06, 2.06, 2.01, 2.00 (6 s, 18 H, 6 CH₃CO), 1.71 (s, 3 H, CH₃CO₃), 1.36 (d, 3 H, J 4.9 Hz, CH₃CH).

Anal. Calcd for $C_{41}H_{52}O_{23}$: C, 53.95; H, 5.74. Found: C, 54.11; H, 5.55.

3-O-(2,3,4,6-Tetra-O-acetyl-β-D-glucopyranosyl)-6-O-(2,3-di-O-acetyl-4,6-O-benzylidene-β-D-glucopyranosyl)-4-O-acetyl-1,2-O-(R,S)ethylidene-α-D-

glucopyranose (13). To a stirred mixture of 12 (474 mg, 0.52 mmol) and 4 Å molecular sieves (0.5 g) in anhydrous CH₂Cl₂ (20 mL) was added TMSOTf (10 µL, 0.1 equiv) at -30 °C under nitrogen atmosphere, and the reaction was monitored by TLC (1/1.5 petroleum ether/ethyl acetate). After completion of the reaction, to the mixture was added triethylamine (10 μ L), and the reaction was allowed to regain room temperature. The mixture was filtered and the filtrate was treated with aq. 5% AcOH (20 mL), satd. aq. Na₂CO₃ (20 mL), and brine (20 mL), dried, and concentrated. The residue was subjected to column chromatography with 1/1petroleum ether/ethyl acetate as the eluent, giving product 13 (R,S, 398 mg) as a syrup in a yield of 84%. $[\alpha]_{D}^{20}$ +25.2° (R, c 1.5, CHCl₃); ¹H NMR: δ (R) 7.48-7.30 (m, 5 H, Ph-H), 5.54 (d, 1 H, J_{1,2} 4.6 Hz, H-1_A), 5.49 (s, 1 H, PhCH=), 5.33, 5.22 (t, 2 H, J 9.5 Hz, H-3, 3_B), 5.07 (t, 1 H, J 9.5 Hz, H-4_A), 5.06 (q, 1 H, J 4.8 Hz, CH₃CH), 5.00-4.94 (m, 3 H, H-2_c, 2_B, 4_c), 4.79, 4.70 (2 d, J 8.0 Hz, 7.9 Hz, H-1_c, 1_B), 4.36 (dd, 1 H, $J_{5B,6Ba}$ 5.0 Hz, $J_{6Ba,6Bb}$ 10.5 Hz, H-6_{Ba}), 4.21 (d, 2 H, $J_{5C,6C}$ 3.2 Hz, H-6_C), 4.04-3.50 (m, 9 H, H-2_A, 3_A, 4_B, 5_A, 5_C, 5_B, 6_A, 6_{Bb}), 2.09, 2.07, 2.07, 2.04, 2.04, 2.02, 2.00 (7 s, 21 H, 7 CH₃CO), 1.47 (d, 3 H, J 4.8 Hz, CH₃CH).

Anal. Calcd for C₄₁H₅₂O₂₃: C, 53.95; H, 5.74. Found: C, 54.19; H, 5.91.

3-O-(2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl)-6-O-(2,3-di-O-acetyl- β -Dglucopyranosyl)-4-O-acetyl-1,2-O-(R,S)ethylidene- α -D-glucopyranose (14). Compound 13 (360 mg, 0.39 mmol) was dissolved in AcOH (5 mL) and treated with 4/1 Cl₂CHCOOH/H₂O (v/v) (0.4 mL) at room temperature for 6 h, at the end of which time TLC (1/2 petroleum ether/ethyl acetate) indicated that the reaction was complete. The reaction mixture was diluted with H₂O, extracted with CH₂Cl₂ (3 × 5 mL), and the organic phase was washed with satd. aq. Na₂CO₃ (3 × 5 mL) and brine (2×5 mL), dried, concentrated and subjected to column chromatography with 1/2 petroleum ether/ethyl acetate as the eluent, giving 14 (R,S) as a syrup (280 mg, 87%). ¹H NMR: δ (R/S, 1/1) 5.63 (d, 1/2 H, J 4.6 Hz, H-1_s), 5.55-5.50 (m, 1 H, 1/2 CH₃CH, 1/2 H-1_R), 4.81, 4.62 (2 d, 1 H, J 7.0 Hz, 1/2 H-1_{CR}, 1/2 H-1_{BR}), 4.79, 4.54 (2 d, 1 H, J 7.8 Hz, 1/2 H-1_{CS}, 1/2 H-1_{BS}), 2.10-2.00 (m, 21 H, 7 CH₃CO), 1.49 (d, 3/2 H, J 4.8 Hz, CH₃CH_R), 1.36 (d, 3/2 H, J 4.9 Hz, CH₃CH_s).

Anal. Calcd for C₃₄H₄₈O₂₃: C, 49.52; H, 5.87. Found: C, 49.35; H, 5.57.

Acknowledgment

This work was supported by The Chinese Academy of Sciences (Project KJ952J₁510) and by The National Natural Science Foundation of China (Project 29802009).

References

(1) Aldington, S. and Fry, S. C. Adv. Bot. Res. 1992, 19, 1-101.

(2) (a)Sharp, J. K.; Valent, B. and Albersheim, P. J. Biol. Chem. 1984, 259, 11312. (b) Cheong, J. -J. and Hahn, M. G. Plant Cell 1991, 3, 137.

(3) (a) Ossowski, P.; Pilotti, A.; Garegg, P. J. and Lindberg, B. Angew. Chem., Int. Ed. Engl. 1983, 22, 793. (b) Ossowski, P.; Pilotti, A.; Garegg, P. J. and Lindberg, B. J. Biol. Chem. 1984, 259, 11337. (c) Fugedi, P.; Birberg, W.; Garegg, P. J. and Pilotti, A. Carbohydr. Res. 1987, 164, 297. (d) Fugedi, P.; Garegg, P. J.; Kvarnstrom, I. and Pilotti, A. J. Carbohydr. Chem. 1989, 8, 47. (e) Fugedi, P.; Garegg, P. J.; Kvarnstrom, I. and Pilotti, A. J. Carbohydr. Chem. 1989, 8, 47. (e) Fugedi, P.; Garegg, P. J.; Kvarnstrom, I. and Svansson, L. J. Carbohydr. Chem. 1988, 7, 389. (f) Hong, N. and Ogawa, T. Tetrahedron Lett. 1990, 31, 3179. (g) Lorentzen, P. J.; Helpap, B. and Lockhoff, O. Angew. Chem., Int. Ed. Engl. 1991, 30, 1681. (h) Hong, N.; Nakahara, Y. and Ogawa, T. Proc. Jpn. Acad. 1993, 698, 55. (i) Yamada, H.; Harad, T. and Takahashi, T. J. Am. Chem. Soc. 1994, 116, 7919. (j) Nicolaou, K. C.; Winssinger, N.; Pastor, J. and Derosse, F. J. Am. Chem. Soc. 1997, 119, 449.

- (4) Wang, W. and Kong, F. Tetrahedron Lett. 1998, 1937.
- (5) Wang, W. and Kong, F. J. Org. Chem. 1998, 63, 5744.
- (6) Fletcher, Jr. H. G. Methods Carbohydr. Chem. 1963, II, 231.

(7) Chelain, E. and Czernecki, S. J. Carbohydr. Chem. 1996, 15 (5), 571.

(8) Mani, N. S. and Kanakamma, P. P. Synth. Commun. 1992, 22 (15), 2175.

(9) Fugedi, P., Birberg, W., Garegg, P. J. and Pilotti, A. Carbohydr. Res. 1987, 164, 297.

Received in the UK 08 December 1998