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Exclusively endo-Selective Lewis Acid-Catalyzed Hetero Diels-Alder Reactions of (E)-1-Phenylsulfonyl-3-alken-2-ones with Vinyl Ethers

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Abstract: (E)-1-Phenylsulfonyl-3-alken-2-ones as new hetero 1,3-dienes undergo smooth hetero Diels-Alder reactions with vinyl ethers in the presence of a catalytic amount of Lewis acid such as ZnI₂, Eu(fod)₃, and TiCl₂(*i*-PrO)₂. The reactions are absolutely *endo*-selective producing 2,4-*cis*-3,4-dihydro-2*H*-pyrans in excellent yields, the configuration at 3-position depending upon the stereochemistry of the starting vinyl ethers. Reductive ring opening reactions of the 3,4-dihydro-2*H*-pyran cycloadducts with Et₃SiH/TiCl₄ lead to 6-alkoxy-1-phenylsulfonyl-2-hexanones, and the sulfonyl-stabilized carbanions derived from the 3,4-dihydro-2*H*-pyran cycloadducts are alkylated followed by reductive desulfonylation to give 2,4,6-trisubstituted 3,4-dihydro-2*H*-pyran derivatives.

We have recently reported that 2-oxo-3-alkenylphosphonates, phosphoryl-substituted α , β -unsaturated ketones, serve as useful hetero 1,3-dienes in the Lewis acid-catalyzed hetero Diels-Alder reactions using vinyl ether dienophiles.^{1a} They can be activated so effectively by chelate formation with a Lewis acid such as ZnCl₂ or ZnBr₂ that 2-alkoxy-3,4-dihydro-2*H*-pyrans are produced stereoselectively without serious polymerization of vinyl ethers. The resulting heterocycles are useful synthetic equivalents of 5-oxoalkanals through a simple hydrolytic procedure.² Such Lewis acid catalysis must be especially noteworthy since the undesired polymerization of vinyl ethers has been a serious problem in the Lewis acid-catalyzed hetero Diels-Alder reactions of simple α , β -unsaturated aldehydes and ketones.³ Uncatalyzed reactions often require higher reaction temperatures, e.g. above 100 °C in a sealed tube, leading to poor stereoselectivity.⁴

A stoichiometric amount of Lewis acid was needed in the above catalyzed reactions, indicating a lack of efficiency of the catalytic cycle. If effective catalytic efficiency is established, a new entry to catalyzed asymmetric hetero Diels-Alder reaction methodology leading to enantiomers of 2-alkoxy-3,4-dihydro-2*H*-pyran derivatives would be open.⁵ For this purpose, it has been strongly desired to develop new α , β -unsaturated carbonyl compounds which can be highly activated by chelate formation with a Lewis acid and then release the catalyst after the completion of the cycloaddition.⁶

In the present paper, we report the synthesis of (E)-1-phenylsulfonyl-3-alken-2-ones as new effective hetero 1,3-dienes and their use in Lewis acid-catalyzed hetero Diels-Alder reactions with vinyl ethers.⁶ The sulfonyl group accelerates the reactions in the presence of a Lewis acid catalyst, the reactions are exclusively *endo*-selective, and the sulfonyl group can be finally utilized for the further transformation via carbon-carbon bond formation.

Results and Discussion

Synthesis of (E)-1-Phenylsulfonyl-3-alken-2-ones. (E)-1-Phenylsulfonyl-3-alken-2-ones 3a-c, as hetero 1,3-dienes employed in the present work, could be easily prepared by the application of our synthetic method of 2-oxo-3-alkenylphosphonates via dianion intermediates.^{1b} According to the Belletire's method, 1-phenylsulfonylpropanone (1) was treated with two equimolar amounts of lithium diisopropylamide (LDA) in tetrahydrofuran (THF) at 0 °C for 4 h to give a solution of the dianion.⁷ This solution was allowed to react with aldehydes such as acetaldehyde, 2-methylpropionaldehyde, and benzaldehyde at -78 °C to give the corresponding aldol adducts **2a-c** in satisfactory yields based on the starting ketone 1. They were then subjected to dehydration by treatment with a catalytic amount of *p*-toluenesulfonic acid under reflux in benzene to produce the desired (E)-1-phenylsulfonyl-3-alken-2-ones **3a-c** in excellent yields. Since the stereoselectivities at the dehydration steps were almost exclusive in all cases, single purification procedure through a silica gel column chromatography afforded the pure samples of *E*-isomers of **3a-c**.



On the other hand, 1-phenylsulfonyl-3-buten-2-one (3d) as β -unsubstituted enone was prepared through two steps starting from propenal. The 1,2-addition of phenylsulfoylmethyllithium, which was generated from methyl phenyl sulfone and *n*-BuLi in THF at -78 °C, to propenal gave 1-phenylsulfonyl-3-buten-2-ol (2d) in 92% yield. Subsequent Jones oxidation of the resulting alcohol 2d at 0 °C produced 3d in 62% yield. *Hetero Diels-Alder Reactions of (E)-1-Phenylsulfonyl-3-alken-2-ones with Vinyl Ethers*. In order to evaluate the reactivity of 1-sulfonyl-3-alken-2-ones 3a-d and the stereoselectivity in their hetero Diels-Alder reactions, they were allowed to react with vinyl ethers 4a-e in dichloromethane in the absence or presence of Lewis acid. Lewis acids such as ZnI₂, Eu(fod)₃, and TiCl₂(*i*-PrO)₂ were employed (Scheme 2 and Table 1), and stereoselectivity was determined in each case on the basis of ¹H and/or ¹³C NMR spectra of the crude reaction mixtures.

The reaction of enone **3a** as a typical hetero 1,3-diene in the absence of Lewis acid required the use of a large excess of vinyl ether **4b** (20 equiv), and a high reaction temperature and some prolonged reaction time

were needed for the completion of the reaction (48 h at 130-135 °C in a sealed tube). A 34:66 mixture of *cis*and *trans*-isomers **6b**, **5b** of the cycloadduct, which correspond to the *endo*- and *exo*-cycloadducts respectively, was produced in 85% of combined yield (entry 3). The low reactivity and poor selectivity observed resemble those of the related reactions of simple α , β -unsaturated aldehydes and ketones.⁴

To our delight, however, the same reaction using vinyl ether **4b** (5 equiv) was highly accelerated (15 h at ~30 °C) in the presence of a catalytic amount of $TiCl_2(i-PrO)_2$ (10 mol%) and became exclusively stereoselective to give the 2.4-*cis*-isomer of **6b** as a single isomer in 91% yield (entry 4).⁸ No serious polymerization of vinyl ether **4b** was observed, indicating that the Lewis acid effectively accelerated the hetero Diels-Alder reaction rather than the polymerization. The reaction of enone **3a** with vinyl ether **4c** also proceeded stereoselectively in the presence of $TiCl_2(i-PrO)_2$ to give the *cis*-isomer **6c** as a single isomer in 97% yield (entry 5).

Entry	Enone	Vinyl ether	Lewis acid	Reaction conditions		Product (yield/%)b	
		(E/Z, equiv)	(mol%)	Temp/°C Time/h		Isomer ratio ^c	
1	3a	4a (5)	$ZnI_2(3)$	rt	86	6a+5a (92)	92/8
2	3a	4a (5)	TiCl ₂ (<i>i</i> -PrO) ₂ (10)	-50	15	6a+5a (90)	98/2
3d	3a	4b (20)	_	130-135	48	6b+5b (85)	34/66
4	3a	4b (5)	TiCl ₂ (<i>i</i> -PrO) ₂ (10)	-30	15	6b (91)	
5	3a	4c (5)	TiCl ₂ (<i>i</i> -PrO) ₂ (10)	-50	90	6c (97)	
6	3a	4a (5)	$Eu(fod)_3(1)$	5	40	6a (95)	
7	3a	4a (5)	$Eu(fod)_3(0.5)$	-10	72	6a (91)	
8	3b	4a (5)	$Eu(fod)_3(3)$	rt	90	6d (80)	
9	3b	4b (5)	$Eu(fod)_3(3)$	rt	99	6e (71)	
10	3c	4a (5)	$Eu(fod)_3(3)$	n	43	6f (85)	
11	3c	4b (5)	$Eu(fod)_{3}(3)$	rt	92	6g (89)	
12	3a	4d (1/1, 5)	Eu(fod)3 (1)	rt	80	7a+8a (65)	1/1
13	3a	4d (1/19, 5)	$Eu(fod)_3(5)$	5	96	7a+8a (95)	1/19
14	3a	<i>E</i> -4e (3)	$Eu(fod)_3(3)$	rt	65	7b (53)	
15	3a	<i>E</i> -4e (3)	$Eu(fod)_3(3)$	-10	240	7b (80)	
16	3a	<i>E</i> -4e (3)	TiCl ₂ (<i>i</i> -PrO) ₂ (10)	-30	38	7b (97)	
17	3a	Z-4e (3)	TiCl ₂ (<i>i</i> -PrO) ₂ (10)	-30	97	8b (97)	
18	3d	4d (1/1, 5)	$Eu(fod)_3(1)$	20	40	7c+8c (95)	1/1
19	3d	4e (5.5/1, 3)	Eu(fod)3 (10)	rt	38	7d+8d (41)	9/1
20	3d	Z-4e (3)	TiCl ₂ (<i>i</i> -PrO) ₂ (10)	-78	41	7d+8d (57)	4/96
21	3d	4f (5)	$Eu(fod)_3(1)$	-10/rt	73/16	9 (24)	
22	<u>3d</u>	4f (5)	$\operatorname{ZnI}_2(3)$	<u>rı</u>	25	9 (90)	

Table 1. Lewis Acid-Catalyzed Hetero Diels-Alder Reactions of Enones 3a-d with Vinyl Ethers 4a-fa

^aAll reactions were carried out in CH₂Cl₂. ^bYields of isolated cycloadducts. ^cDetermined by ¹H NMR and/or ¹³C NMR. ^dIn a sealed tube in benzene.

Other Lewis acid catalysts such as ZnI_2 and $Eu(fod)_3$ can be effectively used in a catalytic amount (0.5-3 mol%). The titanium catalyst $TiCl_2(i$ -PrO)₂ was a little better than the zinc catalyst ZnI_2 in selectivity (entry 1 vs entry 2). The europium catalyst $Eu(fod)_3$ showed effective catalysis as a catalytic amount exclusively produced *cis*-isomers of the dihydropyran derivatives **6a,d-g** in excellent yields for all combinations of enones **3a-c** and vinyl ethers **4a,b** (entries 6-11). This high efficiency of catalytic cycle makes a striking contrast with the reactions of 2-oxo-3-alkenylphosphonates.^{1a} Although we employed excess (5 equiv) of

vinyl ethers **4a-c** in fear of their partial loss by a possible Lewis acid-catalyzed polymerization, less amounts are probably enough, especially in the reactions performed at low temperatures (entries 2 and 4-7).



Scheme 2.

Such remarkable rate acceleration, high stereoselectivity, and high catalytic efficiency, all observed in the Lewis acid-catalyzed hetero Diels-Alder reactions of 2-oxo-1-sulfonyl-3-alkenes **3a-c**, are no doubt based on their capability of effective chelate formation with a Lewis acid by the aid of the sulfonyl moiety. The coordination makes **3a-c** themselves activated as hetero 1,3-dienes. We further investigated the stereospecificity of the Lewis acid-catalyzed hetero Diels-Alder reactions by using β -substituted vinyl ethers **4d,e**.

The reaction using vinyl ether 4d (*E/Z*-mixture, 5 equiv) was activated in the presence of a catalytic amount (1 mol%) of Eu(fod)₃ at or below room temperature in dichloromethane. A 1:1 stereoisomeric mixture of *r*-2-ethoxy-*t*-3,*c*-4-dimethyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2*H*-pyran (**7a**) and *r*-2-ethoxy*c*-3,*c*-4-dimethyl-6-(phenylsulfonymethyl)-3,4-dihydro-2*H*-pyran (**8a**) was produced from the reaction using a 1:1 mixture of *E/Z* stereoisomers of vinyl ether **4d** (65%, entry 12), and a 1:19 mixture of **7a** and **8a** from a 1:19 mixture of *E/Z* stereoisomers (95%, entry 13). These results indicate that (1) the reactions are exclusively 2,4-*cis*-selective or *endo*-selective, (2) the Eu(fod)₃-catalyzed hetero Diels-Alder reactions of **3a** with 4d are exclusively stereospecific, and (3) the E_{-} and Z-isomers of vinyl ether 4d have comparable reactivities toward enone 3a.

(*E*)-1-Ethoxy-2-phenylethene (*E*-4e) showed comparable reactivity toward enone 3a in the reactions catalyzed by Eu(fod)₃ to give the *endo*-cycloadduct, *r*-2-ethoxy-*c*-4-methyl-*t*-3-phenyl-6-(phenylsulfonyl-methyl)-3,4-dihydro-2*H*-pyran (7b), as a single stereoisomer (entries 14, 15). Use of TiCl₂(*i*-PrO)₂ was even more effective and satisfactory rate enhancement was observed. For example, the reaction of 3a with *E*-4e was completed at -30 °C to produce 7b in the presence of 10 molar percent of TiCl₂(*i*-PrO)₂ without polymerization of vinyl ether 4e (entry 16). On the other hand, the reaction with (*Z*)-1-ethoxy-2-phenylethene (*Z*-4e) provided also the *endo*-cycloadduct, *r*-2-ethoxy-*c*-4-methyl-*c*-3-phenyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2*H*-pyran (8b), as a single stereoisomer (entry 17). The β -unsubstituted enone 3d showed the same reactivity to vinyl ethers 4d,e as that of 3a, while stereospecificity was a little lower than 3a (entries 18-20). With cyclic vinyl ether 4f, the cycloadditions took place (entries 21, 22).



Figure 1. Stereochemical assignment of 2,3-*trans*-3,4-*trans*-cycloadduct 7b and 2,3-*cis*-3,4-*cis*-cycloadducts 8a,b on the basis of NOE spectra and coupling constants.

Stereochemistry of the above 3.4-dihydro-2*H*-pyran derivatives was determined on the basis of spectral data, especially ¹H NMR spectra, typical examples of which are shown in Figure 1. The *endo*-cycloadducts **8a** and **8b** from the Z-isomers of vinyl ether dienophiles **4d** and **4e** could be easily assigned by their NOE spectra. Notable signal enhancement was observed between H-2/H-3 and H-3/H-4, and the small coupling constant between H-2 and H-3 (**8a**: $J_{2.3} = 1.8$ Hz, **8b**: $J_{2.3} = 2.2$ Hz) showed the *cis*-relationship of these hydrogens. On the other hand, the *endo*-cycloadduct **7b** from the *E*-isomer of **4e** bears all the substituents at equatorial positions in its stable conformer. Thus, the relatively large diaxial coupling constants $J_{2.3}$ (8.1 Hz) and $J_{3.4}$ (9.9 Hz) observed between H-2/H-3 and H-3/H-4, respectively, are consistent with the proposed stereochemistry.

Transformation of 2-Alkoxy-3,4-dihydro-2H-dihydropyran Cycloadducts. The aforementioned stereospecific and *endo*-selective hetero Diels-Alder reactions of sulfonyl-substituted enones **3** with *E*- and *Z*-isomers of vinyl ethers **4d,e** offer a convenient route to sterically defined 2-alkoxy-3,4-dihydro-2*H*-pyrans. We already reported that hydrolytic cleavage of the acetal moiety led to 5-oxo-6-sulfonylalkanal intermediates which then underwent ready dehydrative cyclizations to give 2-cyclohexen-1-one derivatives.^{6b} In the present work, we examined other transformations of the 2-alkoxy-3,4-dihydro-2*H*-pyran cycloadducts.





When dihydropyran **6a** was treated with tricthylsilane and titanium tetrachloride in dichloromethane at $-78 \, ^{\circ}C$, ⁹ the reduction occurred at the acetal moiety to give 6-ethoxy-4-methyl-1-phenylsulfonyl-2-hexanone (**10a**). This indicates that the Lewis acid TiCl₄ coordinated at the ring oxygen atom, rather than the ethoxyl oxygen atom, to induce the formation of the ring-opened oxonium intermediate **A** (Scheme 3). Its reduction by the silane reagent led to the ether derivative **10a**. Similarly, dihydropyran **7d** was converted into **10b**.

This facile reductive ring opening reaction could be successfully applied to the 3,4-*trans*-isomer 7b and the 3,4-*cis*-isomer 8b which had been obtained in the stereospecific hetero Diels-Alder reactions of 3d with E-4e and Z-4e, respectively. Thus, 4,5-*syn*-isomer 10c and 4,5-*anti*-isomer 10d were selectively prepared.



Scheme 4.

The sulfonyl moiety of cycloadducts 5-8 has a synthetic advantage since it should mediate the generation of an anionic center at the α -position which would be then utilized for the introduction of a substituent by alkylation. Thus, the sulfonyl-stabilized carbanion of dihydropyran **6a** was generated by treatment with butyllithium at -78 °C and allowed to react with benzyl bromide at the same temperature to provide benzylated product **11a** as a single isomer in 85% yield (Scheme 4). Similar benzylation of **6d** gave **11b** (90% de) in 50% yield. The phenylsulfonyl group in the alkylated **11a,b** and unalkylated **6f** were successfully removed by reduction. Treatment with sodium naphthalenide (4 equiv.), generated from naphthalene and metal sodium under ultrasonic conditions in THF,¹⁰ at -78 °C in THF/n-PrNH₂ (3:1 v/v) gave the desulfonylated products **12a-c** in 55, 52, and 66% yields, respectively.

In conclusion, the exclusively stereoselective formation of 2,4-cis-3,4-dihydro-2H-pyrans **6-9** and **12** has been achieved by a sequence based on the hetero Diels-Alder reactions of 1-phenylsulfonyl-3-alken-2-ones **3** with vinyl ethers **4** in the presence of a catalytic amount of Eu(fod)₃ or TiCl₂(*i*-PrO)₂. The sulfonyl moiety of enones **3** is an excellent reactivity-enhancing auxiliary. A new method of stereocontrolled construction of acyclic framework has been also demonstrated by regiocontrolled reductive ring cleavage reaction of stereochemically defined dihydropyran cycloadducts **7b** and **8b**.

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Experimental

General. Melting points were determined on a Yanagimoto melting point apparatus and are uncorrected. IR spectra were taken with a JASCO A-720 spectrometer. ¹H NMR spectra were recorded on a JEOL JNM-EX90 (90 MHz) or a JEOL GSX-270 instrument (270 MHz) and ¹³C NMR spectra on a JEOL GSX-270 spectrometer (67.94 MHz). Chemical shifts are expressed in parts per million downfield from tetramethylsilane as an internal standard. Mass spectra were measured with a JEOL-01SG-2 or a JEOL JMS-AM 20 spectrometer. Elemental analyses were performed on a Hitachi 026 CHN or a Yanaco CHN Corder MT-5 analyzer. For preparative column chromatography, silica gel 60 (Merk, size: 0.04–0.063 mm) was employed. Flash chromatography was carried out on a Yanazen YFLC 540 apparatus using a column (15 x 350 mm) or (20 x 350 mm) packed with silica gel 60 (Merk). Solvents were evaporated with a Tokyo Rikakikai rotary evaporator type-V at room temperature unless otherwise stated.

Materials. The following reagents were prepared by literature methods: methyl phenyl sulfone,¹¹ 1phenylsulfonylpropanone (1),¹² dichlorodiisopropoxytitanium,¹³ (E)-1-ethoxy-2-phenylethene (E-4e),¹⁴ and (Z)-1-ethoxy-2phenylethene (Z-4e).¹⁵ Complete separation of the (E)- and (Z)-isomers 4e was accomplished by flash chromatography (silica gel, hexane). Dichloromethane was purified by distillation on calcium hydride. THF was distilled on lithium aluminum hydride under nitrogen just before its use.

General Procedure for the Preparation of 4-Hydroxy-1-phenylsulfonyl-2-alkanones 2a-c. As a typical example the preparation of 2a is described as follows: To a solution of LDA, prepared from diisopropylamine (3.36 g, 36 mmol) and butyllithium (in hexane, 23.2 ml, 36 mmol) at -78 °C in THF (20 ml), was added 1-phenylsulfonylpropanone (1) (2.97 g, 15 mmol) in THF (20 ml) under nitrogen. The reaction mixture was warmed to 0 °C and stirred for 4 h. To the resulting THF suspension of the dianion of 1 was added acetaldehyde (792 mg, 1.08 ml, 18 mmol) dropwise at 0 °C. After stirred at 0 °C for 1 h, the mixture was acidified with 1 mol/l hydrochloric acid. The THF was removed in vacuo and the aqueous residue was extracted with dichloromethane (40 ml x 3). The combined extracts were dried (MgSO4) and evaporated in vacuo. The crude product (4 g) was chromatographed on silica gel with hexane–EtOAc (1:1 v/v) to give 2a (3.06 g, 85%). Other β -keto alcohols 2b,c were also prepared in 83 and 75 % yields, respectively, under similar reaction conditions.

4-Hydroxy-1-phenylsulfonyl-2-pentanone (2a): Colorless prisms (Et₂O-hexane); mp 75-77 °C; IR (KBr) 3520, 2950, 2900, 1710, 1450, 1390, 1360, 1305, 1280, 1200, 1150, 1050, 950, 860, 730, and 690 cm⁻¹; ¹H NMR (CDCl₃) δ = 1.22 (3H, d, J_{Me-4} = 6.2 Hz, Me), 2.51 (1H, br. s, OH), 2.80 (1H, dd, J_{gem} = 17.6 and J_{3.4} = 7.7 Hz, one of H-3), 2.87 (1H, dd, J_{gem} = 17.6 and J_{3.4} = 4.4 Hz, the other of H-3), 4.16-4.32 (1H, m, H-4), 4.24 (2H, s, H-1), 7.5-7.7 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (CDCl₃) δ = 22.65 (C-5), 52.57 (C-3), 63.80 (C-4), 67.36 (C-1), 128.26, 129.43, 134.41, 138.71 (each Ph), and 198.70 (C=O); MS (20 eV, rel. intensity, %) *m/z* 227 (M⁺-Me, 13), 224 (M⁺-H₂O, 5), 199 (22), 198 (base peak), 156 (46), 141 (47), 134 (62), 126 (24), 125 (58), 101 (78), 77 (25), 69 (34), and 43 (63). Anal. Calcd for C₁₁H₁₄O₄S: C, 54.53; H, 5.82%. Found: C, 54.70; H, 5.99%.

4-Hydroxy-5-methyl-1-phenylsulfonyl-2-hexanone (2*h*): Separated and purified by silica gel column chromatography using hexane–EtOAc (1:1 v/v). Colorless viscous oil; IR (neat) 3500, 3050, 2970, 1710, 1580, 1445, 1375, 1310, 1240, 1150, 1075, 1040, 1000, 800, 730, and 680 cm⁻¹; ¹H NMR (CDCl₃) $\delta = 0.92$, 0.93 (each 3H, each d, J = 6.9 Hz, Me), 1.69 (1H, oct, $J_{5.4} = J_{5-Me2} = 6.9$ Hz, H-5), 2.14 (1H, br. s, OH), 2.79 (1H, dd, $J_{gem} = 17.2$ and $J_{3.4} = 7.7$ Hz, one of H-3), 2.86 (1H, dd, $J_{gem} = 17.2$ and $J_{3.4} = 4.8$ Hz, the other of H-3), 3.83 (1H, ddd, $J_{4.3} = 7.7$, 4.8, and $J_{4.5} = 6.9$ Hz, H-4), 4.20, 4.26 (each 1H, each d, $J_{gem} = 13.5$ Hz, H-1), 7.5-7.7 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (CDCl₃) $\delta = 17.55$, 18.30 (each 5-Me or C-6), 33.26 (C-5), 48.28 (C-3), 67.40 (C-1), 72.24 (C-4), 128.28, 129.40, 134.48, 138.66 (each Ph), and 199.22 (C=O); MS (20 eV, rel. intensity, %) *m*/z 252 (M⁺+H₂O, 3), 227 (base peak), 199 (31), and 111 (26). Anal. Calcd for C₁₃H₁₈O₄S: C, 57.76; H, 6.71%. Found; C, 58.08; H, 6.95%.

4-Hydroxy-4-phenyl-1-phenylsulfonyl-2-butanone (2c): Separated and purified by silica gel column chromatography using hexane–EtOAc (1:1 v/v). Colorless solids (Et₂O); mp 69-70 °C; IR (KBr) 3350, 3050, 2900, 1710, 1580, 1450, 1300, 1150, 1070, 1050, 900, 750, 730, 700 and 680 cm⁻¹; ¹H NMR (CDCl₃) $\delta = 2.23$ (1H, br. s, OH), 3.06 (1H, dd, $J_{gem} = 17.6$ and $J_{3.4} = 3.7$ Hz, one of H-3), 3.19 (1H, dd, $J_{gem} = 17.6$ and $J_{3.4} = 8.8$ Hz, the other of H-3), 4.19, 4.23 (each 1H, each d, $J_{gem} = 13.2$ Hz, H-1), 5.15 (1H, dd, $J_{4.3} = 8.8$ and 3.7 Hz, H-4), 7.2-7.4 (5H, m, Ph), 7.5-7.7 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (CDCl₃) $\delta = 52.76$ (C-3), 67.40 (C-1), 69.80 (C-3), 125.70, 127.97, 128.28, 128.65, 129.38, 134.38, 138.48, 142.28 (each Ph), and 197.40 (C=O); MS (70 eV, rel. intensity, %) *m/z* 288 (M⁺-16, 14), 287 (26), 164 (21), 163 (76), 162 (24), 107 (52), 104 (62), 103 (33), 97 (22), 92 (29), 90 (24), 80 (25), 79 (34), 78 (46), 76 (96), 75 (26), 74 (24), 65 (50), 52 (32), and 50 (base peak). Anal. Calcd for C₁₆H₁₆O₄S: C, 63.14; H, 5.30%. Found: C, 63.10; H, 5.38%.

Preparation of 1-Phenylsulfonyl-3-buten-2-ol (2d). To a solution of methyl phenyl sulfone (5 g, 32 mmol) in THF (100 ml) was added butyllithium (in hexane, 22.7 ml, 35.2 mmol) by the aid of a syringe at -78 °C under nitrogen. The mixture was stirred at -78 °C for 1 h and the addition of acrolein (1.96 g, 2.3 ml, 35 mmol) was followed. The reaction mixture was stirred at room temperature for 30 min, poured to saturated aqueous NH₄Cl. The THF was removed in vacuo and the aqueous residue was extracted with CH₂Cl₂ (50 ml x 3). The combined extracts were dried (MgSO₄) and evaporated in vacuo. The crude product (6.7 g) was chromatographed on silica gel with hexane-EtOAc (1:1 v/v) to give alcohol 2d (6.22 g, 92%): Colorless oil; IR (neat) 3470, 3050, 2970, 2900, 1630, 1580, 1440, 1300, 1140, 1080, 990, 930, 840, 790, 750, and 680 cm⁻¹; ¹H NMR (CDCl₃) $\delta = 3.25$ (1H, dd, $J_{gem} = 14.3$ and $J_{1-2} = 3.3$ Hz, one of H-1), 3.33 (1H, dd, $J_{gem} = 14.3$ and $J_{1-2} = 8.4$ Hz, the other of H-1), 3.38 (1H, br. s, OH), 4.65-4.74 (1H, m, H-2), 5.18 (1H, ddd, J_{4-3} (cis) = 10.6, $J_{gem} = 1.5$, and $J_{4-2} = 1.1$ Hz, one of H-4), 5.78 (1H, ddd, J_{3-4} (trans) = 17.6, J_{3-4} (cis) = 10.6, and $J_{3-2} = 5.5$ Hz, H-3), 7.5-7.7 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (CDCl₃) $\delta = 61.83$ (C-1), 66.98 (C-2), 116.68 (C-4), 128.00, 129.46, 134.12 (each Ph), 136.78 (C-3), and 139.25 (Ph); MS (70 eV, rel intensity, %) *m/z* 212 (M⁺, 20), 129 (33), 109 (base peak), 108 (80), 107 (35), 81 (29), 57 (33), and 55 (34). Anal. Calcd for C₁₀H₁₂O₃S: C, 56.59; H, 5.70%. Found: C, 56.37; H, 5.72%.

General Procedure for the Dehydration of 2a-c Leading to (E)-1-Phenylsulfonyl-3-alken-2-ones 3a-c. As a typical example the conversion of 2a into 3a is described as follows: To a solution of β -keto alcohol 2a (1.69 g, 6.98 mmol) in dry benzene (60 ml) was added p-toluenesulfonic acid monohydrate (190 mg, 1 mmol). The reaction mixture was heated under reflux for 2 h with continual azeotropic removal of water by aid of a Dean-Stark trap. The mixture was evaporated in vacuo and the residue was chromatographed on silica gel with hexanc-EtOAc (2:3 v/v) to give enone 3a (1.43 g, 92%). Other enones 3b,c were also prepared in 85 and 87 % yields, respectively, under similar reaction conditions.

(*E*)-1-Phenylsulfonyl-3-penten-2-one (3a). This compound was previously prepared¹⁶ and full data are as follows. Colorless needles (Et₂O-hexane); mp 73-75 °C; IR (KBr) 3000, 2910, 1670, 1620, 1450, 1380, 1300, 1150, 1075, 975, 770, 710, and 690 cm⁻¹; ¹H NMR (CDCl₃) δ = 1.96 (3H, dd, J₅₋₄ = 7.0 and J₅₋₃ = 1.5 Hz, Me), 4.28 (2H, s, H-1), 6.31 (1H, dq, J₃₋₄ = 15.8 and J_{3-Me} = 1.5 Hz, H-3), 6.99 (1H, dq, J₄₋₃ = 15.8 and J_{4-Me} = 7.0 Hz, H-4), 7.5-7.7 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (CDCl₃) δ = 18.65 (C-5), 65.25 (C-1), 128.42. 129.27 (each Ph), 131.00 (C-3), 134.24, 138.69 (each Ph),

147.91 (C-4), and 186.86 (C=O); MS (75 eV, rel intensity, %) m/z 224 (M⁺, 8), 160 (28), 77 (39), 69 (base peak), and 41 (21). Anal. Calcd for C₁₁H₁₂O₃S; C, 58.91; H, 5.39%. Found: C, 59.27; H, 5.41%.

(E)-5-Methyl-1-phenylsulfonyl-3-hexen-2-one (3b). Separated and purified by silica gel column chromatography using hexane-EtOAc (2:1 v/v). Colorless prisms (Et₂O-hexane); mp 30-31 °C, IR (neat) 3050, 2950, 1670, 1625, 1590, 1450, 1400, 1325, 1300, 1150, 1070, 980, 900, 755, and 690 cm⁻¹; ¹H NMR (CDCl₃) $\delta = 1.07$ (6H, d, $J_{Me-5} = 7.0$ Hz, 2xMe), 2.28 (1H, doct, $J_{5.4} = J_{5.Me2} = 7.0$ and $J_{5.3} = 1.5$ Hz, H-5), 4.31 (2H, s, H-1), 6.19 (1H, dd, $J_{3.4} = 15.8$ and $J_{3.5} = 1.5$ Hz, H-3), 6.91 (1H, dd, $J_{4.3} = 15.8$ and $J_{4.5} = 7.0$ Hz, H-4), 7.5-7.7 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (CDCl₃) $\delta = 21.02$ (2xMe), 31.43 (C-5), 65.27 (C-1), 126.69 (C-3), 128.45, 129.25, 134.41, 138.86 (each of Ph), 158.41 (C-4), and 187.45 (C=O); MS (20 eV, rel. intensity, %) m/z 252 (M⁺, 10), 188 (33), 111 (base peak), and 110 (21). Anal. Calcd for C₁₃H₁₆O₃S: C, 61.88; H, 6.39%.

(*E*)-4-Phenyl-1-phenylsulfonyl-3-buten-2-one (3c). Separated and purified by silica gel column chromatography using hexane-EtOAc (2:1 v/v). Colorless needles (Et₂O); mp 96-98 °C; IR (KBr) 2970, 2900, 1650, 1575, 1440, 1400, 1280, 1145, 1060, 980, 875, 790, 760, 710, and 680 cm⁻¹; ¹H NMR (CDCl₃) δ = 4.40 (2H, s, H-1), 6.94 (1H, d, J_{3.4} = 15.8 Hz, H-3), 7.4-7.7 (9H, m, H-4 and Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (CDCl₃) δ = 66.22 (C-1), 124.64 (C-3), 128.39, 128.88, 129.04, 129.27, 131.43, 133.70, 134.26, 138.55 (each Ph), 146.46 (C-4), and 186.86 (C=O); MS (75 eV, rel. intensity, %) *m/z* 286 (M⁺, 7), 145 (77), 144 (46), 131 (base peak), 103 (42), 77 (44), and 50 (20). Anal. Calcd for C₁₆H₁₄O₃S: C, 67.11; H, 4.93%. Found: C, 67.24; H, 4.91%.

Preparation of 1-Phenylsulfonyl-3-buten-2-one (3d) by Jones Oxidation of Alcohol 2d. The chromic acid oxidizing reagent is prepared by dissolving chromium trioxide (1.5 g, 15 mmol) in water (3 ml) and adding concentrated sulfuric acid (1.3 ml). To a solution of alcohol 2 (2.12 g, 10 mmol) in acetone (35 ml) was added dropwise the chromic acid reagent at -10-0 °C. The resulting mixture was stirred at 0 °C for 30 min and the remaining oxidizing agent was consumed by addition of isopropyl alcohol (2 ml). The resulting green precipitates were filtered off through Celite 545, the filtrate was concentrated, and extracted with CH₂Cl₂ (50 ml x 3). The combined extracts were dried (MgSO₄) and evaporated in vacuo. The crude product was chromatographed on silica gel with CH₂Cl₂-Et₂O (50:1 v/v) to give enone 3d (1.3 g, 62%). Colorless needles (Et₂O-hexane); mp 36-37 °C; IR (KBr) 3070, 2970, 2920, 1675, 1610, 1450, 1400, 1310, 1150, 1090, 1060, 980, 900, 785, 755, 725, and 690 cm⁻¹; ¹H NMR (CDCl₃) δ = 4.35 (2H, s, H-1), 6.01 (1H, dd, J₄₋₃ (cis) = 10.3 and J_{gem} = 0.7 Hz, one of H-4), 6.33 (1H, dd, J₄₋₃ (trans) = 17.6 and J_{gem} = 0.7 Hz, the other of H-4), 6.52 (1H, dd, J₃₋₄ (trans) = 17.6 and J₃₋₄ (cis) = 10.3 Hz, H-3), 7.5-7.7 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (CDCl₃) δ = 64.75 (C-1), 128.38, 129.33 (each Ph), 132.32 (C-3), 134.35 (Ph), 135.33 (C-4), 138.61 (Ph), and 187.74 (C=O); MS (20 eV, rel intensity, %) *m/z* 210 (M⁺, 4), 146 (base peak), and 55 (47). Anal. Calcd for C₁₀H₁₀O₃S: C, 57.13; H. 4.79%. Found: C. 56.98; H, 4.89%.

Thermal Hetero Diels-Alder Reaction of Enone 3a with Vinyl Ether 4b Leading to 2-Isobutoxy-4-methyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2H-pyran (6b+5b). A mixture of 3a (89 mg, 0.4 mmol) and isobutyl vinyl ether 4b (800 mg, 8 mmol) in benzene (1 ml) was heated in a sealed tube at 130-135 °C for 48 h. The mixture was condensed in a vacuo and the residue was chromatographed on silica gel with hexane-EtOAc (6:1 v/v) to give 6b, 5b (110 mg, 85%) as inseparable mixture of *cis*- and *trans*-isomers. According to the ¹H NMR analysis of crude reaction mixture, the ratio of 6b and 5b was 1:2; *trans*-isomer 5b: ¹H NMR (C₆D₆) δ = 0.69 (3H, d, J_{Me-4} = 7.3 Hz, 4-Me), 0.87, 0.89 (each 3H, each d, J = 6.6 Hz, Me of 2-*i*-BuO), 1.08 (1H, ddd, J_{gem} = 12.8, J₃₋₄ = 10.3, and J₃₋₂ = 2.9 Hz, one of H-3), 1.5-1.9 (2H, m, CH of 2-*i*-BuO and the other of H-3), 2.2-2.4 (1H, m, H-4), 3.04 (1H, dd, J_{gem} = 9.2 and J = 6.6 Hz, the other of CH₂ of 2-*i*-BuO), 4.73 (1H, t, J₂₋₃ = 2.9 Hz, H=2), 6.9-7.0 (3H, m, Ph), 7.8-7.9 (2H, m, Ch): ¹³C NMR (C₆D₆) δ = 19.44 (Me of 2-*i*-BuO), 20.85 (4-Me), 22.74 (C-3), 28.80 (CH of 2-*i*-BuO), 34.56 (C-4), 61.55 (6-CH₂SO₂Ph), 75.17 (CH₂ of 2-*i*-BuO), 97.64 (C-2), 111.52 (C-5), 128.71, 128.98, 133.14 (each Ph), 139.88, and 140.00 (Ph and C-6). The data of *cis*-isomer 6b are presented below.

General Procedure for the Lewis Acid-Catalyzed Hetero Diels-Alder Reactions of Enones 3a-c with Vinyl Ethers 4 Leading to Dihydropyran Cycloadducts 5-9. As a typical example the reaction of 3a with ethyl vinyl ether 4a in the presence of Eu(fod)₃ is described as follows: To a solution of enone 3a (224 mg, 1 mmol) and Eu(fod)₃ (31 mg, 0.03 mmol) in CH₂Cl₂ (3 ml) was added ethyl vinyl ether 4a (360 mg, 5 mmol) at -78 °C under argon. The mixture was warmed to -5 °C and allowed to stand for 40 h. The resulting mixture was quenched with saturated aqueous NaHCO₃ and extracted with CH₂Cl₂ (30 ml x 3). The combined extracts were dried (MgSO₄) and evaporated in vacuo. The crude product (350 mg) was chromatographed on silica gel with hexane-EtOAc (7:1 v/v) to give 6a (276 mg, 95%) as a single isomer. Other hetero Diels-Alder reactions were performed under the reaction conditions shown in Table 1. The yields as well as the diastereoselectivities are also summarized in Table 1. trans-Isomer 5a (entries 1 and 2 in Table 1) was characterized on the basis of ¹H and ¹³C NMR spectra data as a mixture with cis-isomer 6a.

cis-2-Ethoxy-4-methyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2H-pyran (6a). Separated and purified by silica gel column chromatography using hexane–EtOAc (7:1 v/v). Colorless prisms (cyclohexane); mp 88-89 °C; IR (KBr) 2950, 1665, 1580, 1450, 1375, 1290, 1255, 1180, 1155, 1125, 1050, 1025, 980, 900, 855, 810, 750, and 690 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.73 (3H, d, J_{Me-4} = 7.3 Hz, 4-Me), 1.03 (3H, t, J = 7.0 Hz, 2-EtO), 1.29 (1H. ddd, J_{gem} = 13.2, J_{3.4} = 9.5, and J₃₋₂ = 8.1 Hz, one of H-3), 1.64 (1H, dddd, J_{gem} = 13.2, J_{3.4} = 6.0, J_{3.2} = 2.2, and J_{3.5} = 1.1 Hz, the other of H-3), 1.92-2.18 (1H, m, H-4), 3.12 (1H, dq, J_{gem} = 9.5 and J = 7.0 Hz, one of 2-EtO), 3.48 (1H, dd, J_{gem} = 13.2 and J_{CH2-5} = 1.1 Hz, one of 6-CH₂SO₂Ph), 3.55 (1H, dd, J_{gem} = 13.2 and J_{CH2-5} = 0.7 Hz, the other of 6-CH₂SO₂Ph), 3.60 (1H, dq, J_{gem} = 9.5 and J = 7.0 Hz, the other of 2-EtO), 4.35 (1H, br. d, J_{5.4} = 2.6 Hz, H-5), 4.49 (1H, dd, J_{2.3} = 8.1 and 2.2 Hz, H-2), 6.9-7.0 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.29 (2-EtO), 20.88 (4-Me), 26.62 (C-4), 35.85 (C-3), 61.40 (6-CH₂SO₂Ph), 64.26 (2-EtO), 100.15 (C-2), 110.65 (C-5), 128.74, 128.95, 133.01 (each Ph), 140.15, and 140.96 (Ph and C-6); MS (70 eV, rel intensity, %) m/z 296 (M⁺, 25), 155 (40), 111 (23), 109 (59), 85 (24), 81 (48), 77 (90), 72 (100), 69 (31), 57 (27), 55 (27), 51 (29), 45 (64), 44 (84), and 40 (30). Anal. Calcd for C₁₅H₂₀O₄S: C, 60.79; H, 6.80%. Found: C, 60.72; H, 6.97%.

trans-2-Ethoxy-4-methyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2H-pyran (5a). ¹H NMR (C₆D₆) δ = 0.66 (3H, d, J_{Me-4} = 7.3 Hz, 4-Me), 1.09 (3H, t, J = 7.3 Hz, EtO), 2.20-2.40 (1H, m, H-4), 3.28, 3.89 (each 1H, dq, J_{gem} = 9.5 and J = 7.3 Hz, one of 2-EtO), 4.25-4.28 (1H, m, H-5), and 4.73 (1H, t, J₂₋₃ = 2.8 Hz, H-2); ¹³C NMR (C₆D₆) δ = 20.83 (4-Me), 22.71 (C-4), 34.61 (C-3), 61.55 (6-CH₂SO₂Ph), 64.00 (2-EtO), 97.32 (C-2), 111.53 (C-5), 128.68 (Ph), and 140.18 (Ph and C-6). Other signals are overlapped with those of the *cis*-isomer 6a.

cis-2-*isobutoxy*-4-*methyl*-6-(*phenylsulfonylmethyl*)-3,4-*dihydro*-2*H*-*pyran* (**6b**). Separated and purified by silica gel column chromatography using hexane–EtOAc (7:1 v/v). Colorless needles (Et₂O-hexane); mp 57-58 °C; IR (KBr) 2900, 1660, 1445, 1360, 1300, 1260, 1130, 1125, 1060, 975, 825, 745, and 690 cm⁻¹; ¹H NMR (C₆D₆) $\delta = 0.75$ (3H, d, *J*_{Me-4} = 7.0 Hz, 4-Me), 0.84, 0.85 (each 3H, each d, *J*_{Me-CH} = 6.6 Hz, Me of 2-*i*-BuO), 1.28 (1H, ddd, *J*_{gem} = 13.2, *J*₃-2 = 8.1, and *J*₃.4 = 9.2 Hz, one of H-3), 1.64 (1H, dddd, *J*_{gem} = 13.2, *J*₃.2 = 2.2, *J*₃.4 = 6.6, and *J*₃.5 = 1.1 Hz, the other of H-3), 1.75 (1H, ninefold, *J* = 6.6 Hz. CH of 2-*i*-BuO), 1.92-2.18 (1H, m, H-4), 2.82. 3.35 (each 1H, each dd, *J*_{gem} = 9.3 and *J*_{CH2-CH} = 6.6 Hz, CH₂ of 2-*i*-BuO), 3.49, 3.56 (each 1H, each br. d, *J*_{gem} = 14.3 Hz, 6-CH₂SO₂Ph), 4.35 (1H, br. d, *J*₅.4 = 2.6 Hz, H-5), 4.47 (1H, dd, *J*₂₋₃ = 8.1 and 2.2 Hz, H-2), 6.9-7.0 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 19.40, 19.44 (each Me of 2-*i*-BuO), 20.89 (4-Me), 26.58 (C-4), 28.74 (CH of 2-*i*-BuO), 35.71 (C-3), 61.40 (6-CH₂SO₂Ph), 75.56 (CH₂ of 2-*i*-BuO), 100.50 (C-2), 110.65 (C-5), 128.76, 128.95, 133.07 (each Ph), 140.25, and 140.93 (Ph and C-6); MS (70 eV, rel intensity, %) *m/z* 324 (M⁺, 21), 235 (30), 226 (23), 225 (base peak), 184 (76), 182 (36), 140 (30), 128 (22), 126 (46), 112 (34), 108 (30), 101 (32), and 91 (23). Anal. Caled for C₁₇H₂₄O₄S: C, 62.94; H, 7.46%. Found: C, 63.16; H, 7.43%.

cis-4-Methyl-2-phenoxy-6-(phenylsulfonylmethyl)-3,4-dihydro-2H-pyran (6c). Separated and purified by silica gel column chromatography using hexane–EtOAc (7:1 v/v). Colorless needles (Et₂O); mp 133-135 °C; IR (KBr) 3050, 2900, 1660, 1580, 1475, 1440, 1375, 1290, 1225, 1140, 1065, 980, 890, 805, 740, and 675 cm⁻¹; ¹H NMR (CDCl₃) δ = 1.08 (3H, d, J_{Me-4} = 7.3 Hz, 4-Me), 1.65 (1H, ddd, J_{gem} =13.4, J₃₋₂ = 7.9, and J₃₋₄ = 9.0 Hz, one of H-3), 2.11 (1H, dddd, J_{gem} = 13.4, J₃₋₂ = 2.6, J₃₋₄ = 6.4, and J₃₋₅ = 0.7 Hz, the other of H-3), 2.38-2.55 (1H, m, H-4), 3.81 (2H, br. s, 6-CH₂SO₂Ph), 4.76 (1H, br. d, J₅₋₄ = 2.2 Hz, 5-H), 5.43 (1H, dd, J₂₋₃ = 7.9 and 2.2 Hz, H-2), 6.8-6.9 (2H, m, Ph), 7.0-7.1 (1H, m, Ph), 7.2-7.3 (2H, m, Ph), 7.4-7.5 (2H, m, Ph), 7.5-7.6 (1H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 20.81 (4-Me), 25.93 (C-4), 34.97 (C-3), 61.05 (6-CH₂SO₂Ph), 97.55 (C-2), 111.50 (C-5), 116.47, 122.31, 128.53, 128.87, 129.41, 133.66 (each Ph), 138.71, 138.74 (Ph and C-6), and 156.74 (Ph); MS (20 eV, rel intensity, %) *m/z* 344 (M⁺, 13), 251 (23), 249 (30), 235 (base peak), 109 (52), and 108 (62). Anal. Calcd for C₁₉H₂₀O₄S; C, 66.26; H, 5.85%. Found: C, 66.13; H, 5.93%.

cis-2-*Ethoxy*-4-*isopropyl*-6-(*phenylsulfonylmethyl*)-3,4-*dihydro*-2*H*-*pyran* (6d). Separated and purified by silica get column chromatography using hexane–EtOAc (7:1 v/v). Colorless solids (Et₂O-hexane); mp 91-93 °C; IR (KBr) 2850, 1650, 1440, 1350, 1255, 1120, 1025, 900, 850, 825, 800, 770, 745, and 675 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.67, 0.68 (each 3H, each d, J = 6.6 Hz. Me of 4-*i*-Pr), 1.04 (3H, t, J = 7.0 Hz, 2-EtO), 1.31 (1H, oct, J = 6.6 Hz, CH of 4-*i*-Pr), 1.43 (1H, ddd, J_{gem} = 13.0, J₃. 2 = 9.0, and J₃.4 = 10.8 Hz, one of H-3), 1.63 (1H, dddd, J_{gem} = 13.0, J₃.2 = 2.2, and J₃.4 = 6.0 and J₃.5 = 1.0 Hz, the other of H-3), 1.76-1.88 (1H, m, H-4), 3.12 (1H, dq, J_{gem} = 9.5 and J = 7.0 Hz, one of 2-EtO), 3.51 (1H, d, J_{gem} = 13.9 Hz, one of 6-CH₂SO₂Ph), 3.57 (1H, dq, J_{gem} = 9.5 and J = 7.0 Hz, the other of 2-EtO), 3.58 (1H, d, J_{gem} = 13.9 Hz, the other of 6-CH₂SO₂Ph), 4.50 (1H, dd, J_{2.3} = 9.0 and 2.2 Hz, H-2), 6.8-7.0 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.35 (2-EtO), 17.22, 19.64 (each Me of 4-*i*-Pr), 31.00 (CH of 4-*i*-Pr), 31.68 (C-4), 38.43 (C-3), 61.45 (6-CH₂SO₂Ph), 64.25 (2-EtO), 100.82 (C-2), 107.92 (C-5), 128.81, 128.92, 133.06 (each Ph), 140.28, and 142.00 (Ph and C-6); MS (75 eV, rel intensity, %) *m/z* 324 (M⁺, 74), 281 (36), 235 (base peak), and 72 (48). Anal. Calcd for C₁₇H₂₄O₄S: C, 62.94; H, 7.46%. Found: C, 63.00; H, 7.20%.

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cis-2-Isobutoxy-4-isopropyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2H-pyran (6e). Separated and purified by silica gel column chromatography using hexane–EtOAc (7:1 v/v). Colorless needles (Et₂O-hexane); mp 83-84 °C; IR (KBr) 2900, 2850, 1650, 1440, 1350, 1290, 1260, 1125, 1055, 1010, 915, 890, 855, 800, 770, 740, and 675 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.69, 0.70 (each 3H, each d, J = 7.0 Hz. Me of 4-*i*-Pr), 0.85, 0.87 (each 3H, each d, J = 6.6 Hz, Me of 2-*i*-BuO), 1.35 (1H, oct, J = 7.0 Hz, CH of 4-*i*-Pr), 1.42 (1H, ddd, J_{gem} = 13.0, $J_{3,2}$ = 8.8, and $J_{3,4}$ = 10.8 Hz, one of H-3), 1.62 (1H, dddd, J_{gem} = 13.0, $J_{3,2}$ = 8.2, $J_{3,4}$ = 6.0, and $J_{3,5}$ = 1.0 Hz, the other of H-3), 1.76 (1H, ninefold, J = 6.6 Hz, CH of 2-*i*-BuO), 1.78-1.91 (1H, m, H-4), 2.83, 3.33 (each 1H, each dd, J_{gem} = 9.2 and J = 6.6 Hz, CH₂ of 2-*i*-BuO), 3.51, 3.60 (each 1H, each d, J_{gem} = 13.9 Hz, 6-CH₂SO₂Ph), 4.48 (1H, dd, $J_{2,3}$ = 8.8 and 2.2 Hz, H-2), 4.52 (1H, br. s, H-5), 6.8-7.0 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 19.19, 19.42, 19.47, 19.67 (each Me of 2-*i*-BuO) and 4-*i*-Pr), 28.78 (CH of 2-*i*-BuO), 3.0.83 (CH of 4-*i*-Pr), 31.71 (C-4), 38.42 (C-3), 61.45 (6-CH₂SO₂Ph), 75.55 (CH₂ of 2-*i*-BuO), 101.15 (C-2), 107.93 (C-5), 128.81, 128.91, 133.07 (each Ph), 140.43, and 142.00 (Ph and C-6); MS (75 eV, rel intensity, %) *m/z* 352 (M⁺, 8), 309 (44), 235 (base peak), 57 (36), 56 (21), and 41 (24). Anal. Calcd for C19H₂BO₄S: C, 64.74; H, 8.01%. Found: C, 64.71; H, 7.95%.

cis-2-*Ethoxy*-4-*phenyl*-6-(*phenylsulfonylmethyl*)-3,4-*dihydro*-2*H*-*pyran* (**6f**). Separated and purified by silica gel column chromatography using hexane–EtOAc (7:1 v/v). Colorless solids (Et₂O-hexane); mp 88-90 °C; IR (KBr) 2980, 2900, 2870, 1675, 1600, 1580, 1445, 1380, 1320, 1305, 1225, 1175, 1160, 1130, 1080, 1040, 1025, 950, 930, 890, 800, 750, and 690 cm⁻¹; ¹H NMR (C₆D₆) δ = 1.01 (1H, t, *J* = 7.0 Hz, 2-EtO), 1.73 (1H, ddd, J_{gem} = 13.2, J₃₋₂ = 9.2, and J₃₋₄ = 11.0 Hz, one of H-3), 1.94 (1H, dddd, J_{gem} = 13.2, J₃₋₂ = 2.2, J₃₋₄ = 6.6, and J₃₋₅ = 1.1 Hz, the other of H-3), 3.16 (1H, dq, J_{gem} = 9.5 and *J* = 7.0 Hz, one of 2-EtO), 3.09- 3.35 (1H, m, H-4), 3.49 (1H, dd, J_{gem} = 13.9 and *J* = 1.1 Hz, one of 6-CH₂SO₂Ph), 3.59 (1H, br. d, J_{gem} = 13.9 Hz, the other of 6-CH₂SO₂Ph), 3.60 (1H, dq, J_{gem} = 9.5 and *J* = 7.0 Hz, the other of 2-EtO), 4.56 (1H, br. s, H-5), 4.58 (1H, dd, J₂₋₃ = 9.2 and 2.2 Hz, H-2), 6.8-7.2 (8H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.28 (2-EtO), 3.716 (C-4), 38.40 (C-3), 61.35 (6-CH₂SO₂Ph), 64.27 (2-EtO), 100.41 (C-2), 108.43 (C-5), 126.85, 127.50, 127.67, 128.02, 128.38, 133.19 (each Ph), 140.20, 142.79 (Ph and C-6), and 144.19 (Ph); MS (70 eV, rel intensity, %) *m/z* 359 (M⁺+1, 4), 313 (M⁺-EtO, 66), 218 (72), 216 (base peak), 173 (49), 170 (49), 147 (39), 141 (23), 117 (25), 115 (54), 103 (61), 91 (38), 77 (86), and 74 (42). Anal. Calcd for C₂₀H₂₂O₄S: C, 67.02; H, 6.19%. Found: C, 66.82; H, 6.34%.

cis-2-*Isobutoxy*-4-*phenyl*-6-(*phenylsulfonylmethyl*)-3,4-*dihydro*-2*H*-*pyran* (**6g**). Separated and purified by silica get column chromatography using hexane–EtOAc (7:1 v/v). Colorless viscous oil; IR (neat) 3050, 2950, 2850, 1660, 1600, 1445, 1310, 1150, 1075, 1060, 1025, 950, 890, 800, 750, and 680 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.82, 0.83 (each 3H, each d, *J* = 7.0 Hz, Me of 2-*i*-BuO), 1.60-1.85 (2H, m, one of H-3 and CH of 2-*i*-BuO), 1.92 (1H, dddd, *J*_{gem} = 12.8, *J*₃₋₂ = 2.2, *J*₃₋₄ = 6.6, and *J*₃₋₅ = 1.1 Hz, the other of H-3), 2.86 (1H, dd, *J*_{gem} = 9.2 and *J* = 7.0 Hz, one of CH₂ of 2-*i*-BuO), 3.19-3.29 (1H, m, H-4), 3.36 (1H, dd, *J*_{gem} = 9.2 and *J* = 7.0 Hz, the other of CH₂ of 2-*i*-BuO), 3.51 (1H, dd, *J*_{gem} = 13.4 and *J* = 1.1 Hz, one of 6-CH₂SO₂Ph), 3.61 (1H, br, d, *J*_{gem} = 13.4 Hz, the other of 6-CH₂SO₂Ph), 4.56 (1H, dd, *J*₂₋₃ = 9.2 and 2.2 Hz, H-2), 4.61 (1H, dd, *J*₅₋₄ = 2.2 and *J*₅₋₃ = 1.1 Hz, H-5), 6.8-7.2 (8H, m, Ph), and 7.8-7.9 (2H, m, Ph); ⁻¹³C NMR (C₆D₆) δ = 19.35, 19.41 (each Me of 2-*i*-BuO), 28.71 (CH of 2-*i*-BuO), 37.05 (C-4), 38.34 (C-3), 61.38 (6-CH₂SO₂Ph), 75.53 (CH₂ of 2-*i*-BuO), 100.05 (C-2), 108.38 (C-5), 126.84, 127.48, 128.75, 133.13 (each Ph), 140.43, 142.79 (Ph and C-6), and 144.21 (Ph); MS (75 eV, rel intensity, %) *n*/z 386 (M⁺, 5), 245 (25), 244 (62), 171 (71), 170 (30), 145 (base peak), 144 (53), 143 (34), 141 (43), 131 (55), 129 (29), 128 (31), 100 (27), 77 (46), 57 (86), 56 (55), and 41 (29). Anal. Calcd for C₂₂H₂₆O₄S: C, 68.37; H, 6.78%. Found: C, 68.12; H, 6.62%.

r-2-*Ethoxy-t*-3,*c*-4-*dimethyl*-6-(*phenylsulfonylmethyl*)-3,4-*dihydro*-2*H*-*pyran* (**7a**). Purified as an inseparable 1:1 mixture (entry 12 in Table 1) of stereoisomers **7a** and **8a** by sitica gel column chromatography using hexane–EtOAc (5:1 v/v). The compound **7a** was characterized on the basis of ¹H and ¹³C NMR spectral data. ¹H NMR (C₆D₆) δ = 0.77, 0.84 (each 3H, each d, *J* = 7.3 Hz, each 3-Me or 4-Me). 1.02 (3H, t, *J* = 7.0 Hz, 2-EtO), 1.33 (1H, dquint, *J*_{3.2} = *J*_{3.Me} = 7.3 and *J*_{3.4} = 8.1 Hz, H-3), 1.54-1.68 (1H, m, H-4), 3.14 (1H, dq, *J*_{gem} = 9.5 and *J* = 7.0 Hz, one of 2-EtO), 3.51 (2H, s, 6-CH₂SO₂Ph), 3.63 (1H, dq, *J*_{gem} = 9.5 and *J* = 7.0 Hz, one of 2-EtO), 3.51 (2H, s, 6-CH₂SO₂Ph), 3.63 (1H, dq, *J*_{gem} = 9.5 and *J* = 7.0 Hz, the other of 2-EtO), 4.22 (1H, *J*_{2.3} = 7.3 Hz, H-2), 4.33 (1H, d, *J*_{5.4} = 2.6 Hz, H-5), 6.8-7.0 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 14.95, 15.21 (each 2-EtO) or 3-Me), 19.57 (4-Me), 30.04 (C-4), 38.83 (C-3), 61.27 (6-CH₂SO₂Ph), 64.45 (2-EtO), 104.06 (C-2), 110.68 (C-5), 128.91, 128.95, 133.06 (each Ph), and 140.40 (Ph and C-6). Other signals are overlapped with those of the compound **8a**.

r-2-Ethoxy-c-3,c-4-dimethyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2H-pyran (8a). Purified as an inseparable 1:19 mixture (entry 13 in Table 1) of *trans* 7a and *cis* 8a by silica gel column chromatography using hexane–EtOAc (5:1 v/v). Recrystallization of a mixture of stereoisomers 7a and 8a from Et₂O-hexane gave 8a as a single isomer. Colorless prisms (Et₂O-hexane); mp 82-84 °C; IR (KBr) 2850, 1660, 1440, 1300, 1260, 1130, 1020, 940, 840, 810, 740, 675 and 645 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.77, 0.86 (each 3H, each d, *J* = 7.3 Hz, each 3-Me or 4-Me), 1.02 (3H, t, *J* = 7.0 Hz, 2-EtO), 1.68 (1H, dquint, *J*₃₋₂ = 1.8 and *J*_{3-Me} = *J*₃₋₄ = 7.0 Hz, H-3), 1.99-2.15 (1H, m, H-4), 3.12 (1H, dq, *J*_{gem} = 9.5 and *J* = 7.0 Hz, one of 2-EtO), 3.51 (2H, s, 6-CH₂SO₂Ph), 3.65 (1H, dq, *J*_{gem} = 9.5 and *J* = 7.0 Hz, the other of 2-EtO), 4.23 (1H, *J*₅₋₄ = 3.3 Hz, H-5),

4.50 (1H, d, $J_{2,3} = 1.8$ Hz, H-2), 6.8-7.0 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) $\delta = 8.57$ (3-Me), 15.31 (2-EtO), 16.21 (4-Me), 30.70 (C-4), 34.80 (C-3), 61.16 (6-CH₂SO₂Ph), 64.56 (2-EtO), 102.63 (C-2), 109.83 (C-5), 128.75, 128.97, 133.03 (each Ph), and 140.25 (Ph and C-6); MS (20 eV, rel intensity, %) *m/z* 310 (M⁺, 18) and 86 (base peak). Anal. Calcd for C₁₆H₂₂O₄S: C, 61.91; H, 7.14%. Found: C, 61.95; H, 7.14%.

r-2-*Ethoxy-c*-4-methyl-*i*-3-phenyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2*H*-pyran (**7b**). Separated and purified by silica gel column chromatography using hexane–EtOAc (7:1 v/v). Colorless needles (Et₂O-hexane); mp 111-112 °C; IR (KBr) 2900, 1650, 1440, 1300, 1255, 1130, 1075, 995, 820, 745, 680, and 625 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.69 (3H, d, J_{Me-4} = 6.6 Hz, 4-Me), 0.78 (3H, t, J = 7.0 Hz, 2-EtO), 2.22-2.35 (1H, m, H-4), 2.41 (1H, dd, J₃₋₂ = 8.1 and J₃₋₄ = 9.9 Hz, H-3), 3.03, 3.48 (each 1H, each dq, J_{gem} = 9.5 and J = 7.0 Hz, 2-EtO), 3.53, 3.64 (each 1H, each d, J_{gem} = 13.9 Hz, 6-CH₂SO₂Ph), 4.54 (1H, d, J₅₋₄ = 2.2 Hz, H-5), 4.67 (1H, d, J₂₋₃ = 8.1 Hz, H-2), 6.8-7.2 (8H, m, Ph), 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 14.96 (2-EtO), 19.24 (4-Me), 34.59 (C-4), 51.95 (C-3), 61.21 (6'-CH₂SO₂Ph), 64.66 (2-EtO), 103.04 (C-2), 111.26 (C-5), 126.94, 128.55, 128.62, 128.89, 133.16 (each Ph), 140.44, and 140.82 (Ph and C-6); MS (20 eV, rel intensity, %) *m/z* 372 (M⁺, 6), 149 (11), and 148 (base peak). Anal. Calcd for C₂₁H₂₄O₄S: C, 67.72; H, 6.49%. Found: C, 68.08; H, 6.53%.

r-2-*E*thoxy-*c*-4-methyl-*c*-3-phenyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2H-pyran (**8b**). Separated and purified by silica gel column chromatography using hexane–EtOAc (7:1 v/v). Colorless prisms (Et₂O-hexane); mp 91-92 °C; IR (KBr) 2900, 1650, 1440, 1300, 1150, 1125, 1105, 1085, 1050, 1015, 980, 880, 845, 740, and 680 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.81 (3H, d, *J* = 7.3 Hz, 4-Me), 1.03 (3H, t, *J* = 7.0 Hz, 2-EtO), 2.18-2.34 (1H, m, 4-H), 2.93 (1H, dd, J₃₋₂ = 2.2 and J₃₋₄ = 6.8 Hz, 3-H), 3.21 (1H, dq, J_{gem} = 9.5 and *J* = 7.3 Hz, one of 2-EtO), 3.52, 3.62 (each 1H, each d, J_{gem} = 14.3 Hz, 6-CH₂SO₂Ph), 3.86 (1H, dq, J_{gem} = 9.5 and *J* = 7.3 Hz, the other of 2-EtO), 4.39 (1H, br. d, J₅₋₄ = 4.0 Hz, H-5), 4.92 (1H, d, J₂₋₃ = 2.2 Hz, H-2), 6.9-7.4 (8H, m, Ph), and 7.8-8.0 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.18 (2-EtO), 16.60 (4-Me), 31.39 (C-4), 46.48 (C-3), 61.24 (6'-CH₂SO₂Ph), 64.76 (2-EtO) 100.80 (C-2), 111.72 (C-5), 126.88, 127.66, 128.02, 128.38, 130.00, 133.26 (each Ph), 138.53, 139.78, and 139.86 (Ph and C-6); Mass (75 eV, rel. intensity. *%*) *m/z* 372 (M⁺, 2), 148 (base peak), 120 (29), and 91 (28). Anal. Calcd for C₂₁H₂₄O₄S; C, 67.72; H, 6.49%. Found: C, 67.31; H, 6.36%.

trans- and cis-2-Ethoxy-3-Methyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2H-pyran (**7c**) and (**8c**). Purified as an inseparable 1:1 mixture (entry 18 in Table 1) of **7c** and **8c** by silica gel column chromatography using hexane–EtOAc (5:1 v/v). Colorless oil; IR (neat) 2950, 2900, 1675, 1580, 1445, 1370, 1310, 1250, 1210, 1150, 1125, 1080, 1060, 1025, 980, 960, 920, 850, 790, 745, and 680 cm⁻¹; ¹H NMR (C₆D₆) **7c**: $\delta = 0.78$ (3H, d. J = 6.9 Hz, 3-Me), 1.06 (3H, t, J = 7.0 Hz, 2-EtO), 1.33 (1H, dt, $J_{gem} = 12.5$ and $J_{4.3} = J_{4.5} = 4.8$ Hz, one of 4-H), 2.01-2.15 (1H, m, the other of 4-H), 3.24 (1H, dq, $J_{gem} = 9.5$ and J = 7.0 Hz, one of 2-EtO), 3.51 (2H, br. s, 6-CH₂SO₂Ph), 3.78 (1H, dq, $J_{gem} = 9.5$ and J = 7.0 Hz, one of 2-EtO), 3.51 (2H, br. s, 6-CH₂SO₂Ph), 3.78 (1H, dq, $J_{gem} = 9.5$ and J = 7.0 Hz, one of 2-EtO), 3.47, 3.58 (each 1H, each br. d, $J_{gem} = 13.9$ Hz, 6-CH₂SO₂Ph), 3.90 (1H, dq, $J_{gem} = 9.5$ and J = 7.0 Hz, the other of 2-EtO), 4.40 (1H, dd, $J_{5.4} = 4.7$ and 2.5 Hz, 5-H), and 4.56 (1H, d, $J_{2.3} = 2.5$ Hz, H-2). Other signals (H-3 and Ph) of **7c** are overlapped with those (H-3, 4 and Ph) of the compound **8c**: 1.5-1.9 (4H, m, H-3 of **7c** and H-3, 4 of **8c**) and 6.7-7.0 (3H, m, Ph), and 7.7-7.9 (2H, m, Ph): ¹³C NMR (C₆D₆) **7c**: $\delta = 15.26$ (3-Me), 1.6.18 (2-EtO), 25.77 (C-4), 30.05 (C-3), 61.34 (6-CH₂SO₂Ph), 64.09 (2-EtO), 102.40 (C-2), 103.77 (C-5), 128.75, 128.81, 133.06 (each Ph), 140.28 (Ph or C-6), and 140.33 (Ph or C-6); **8c**: $\delta = 15.12$ (3-Me), 15.62 (2-EtO), 25.15 (C-4), 30.58 (C-3), 61.34 (6-CH₂SO₂Ph), 64.109 (2-EtO), 15.62 (2-EtO), 25.15 (C-4), 30.58 (C-3), 61.34 (6-CH₂SO₂Ph), 64.16 (2-EtO), 15.62 (2-EtO), 25.15 (C-4), 30.58 (C-3), 61.34 (6-CH₂SO₂Ph), 64.16 (2-EtO), 15.62 (2-EtO), 25.15 (C-4), 30.58 (C-3), 61.34 (6-CH₂SO₂Ph), 64.16 (2-EtO), 15.62 (2-EtO), 25.15 (C-4), 30.58 (C-3), 61.34 (6-CH₂SO₂Ph), 64.16 (2-EtO), 15.62 (2-EtO), 25.15 (C-4), 30.58 (C-3), 61.34 (6-CH₂SO₂Ph), 64.16 (2-EtO), 102.

trans-2-Ethoxy-3-phenyl-6-(phenylsulfonylmethyl)-3.4-dihydro-2H-pyran (7d). Separated and purified by silica gel column chromatography using hexane–EtOAc (7:1 v/v). Colorless prisms (*i*-PrOH-hexane); mp 102-103 °C; IR (KBr) 2900, 1660, 1430, 1280, 1240, 1120, 1075, 980, 870, 790, 740, and 675 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.93 (3H, t, *J* = 7.0 Hz, 2-EtO), 1.94 (1H, ddd, *J*_{gem} = 18.0, *J*₄₋₃ = 6.6, and *J*₄₋₅ = 4.0 Hz, one of H-4), 2.27 (1H, ddd, *J*_{gem} = 18.0, *J*₄₋₃ = 6.6, *J*₄₋₅ = 4.0 Hz, the other of H-4), 2.82 (1H, dt, *J*₃₋₂ = 5.1 and *J*₃₋₄ = 6.6 Hz, H-3), 3.16 (1H, dq, *J*_{gem} = 9.5 and *J* = 7.0 Hz one of 2-EtO), 3.50 (2H, s, CH₂SO₂Ph), 3.70 (1H, dq, *J*_{gem} = 9.5 and *J* = 7.0 Hz, the other of 2-EtO), 4.53 (1H, t, *J*₅₋₄ = 4.0 Hz, H-5), 4.72 (1H, d, *J*₂₋₃ = 5.1 Hz, H-2), 6.8-7.0 (3H, m, Ph), 7.0-7.3 (5H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.09 (2-EtO), 26.38 (C-4), 41.97 (C-3), 61.28 (6-CH₂SO₂Ph), 64.32 (2-EtO), 101.52 (C-2), 104.61 (C-5), 126.95, 128.15, 128.68, 128.71, 128.81, 132.97 (each Ph), 140.51, 141.28, and 141.59 (Ph and C-6); Mass (70 eV, rel. intensity, %) *m/z* 358 (M⁺, 9), 171 (52), 149 (87), 147 (34), 119 (23), 92 (22), 91 (70), 78 (23), 77 (base peak), 65 (34), 55 (27), and 51 (33). Anal. Calcd for C₂₀H₂₂O₄S: C, 67.02; H, 6.19%. Found: C, 67.19; H, 6.31%.

cis-2-Ethoxy-3-phenyl-6-(phenylsulfonylmethyl)-3,4-dihydro-2H-pyran (8d). Separated and purified by silica gel column chromatography using hexane–EtOAc (7:1 v/v). Colorless prisms (*i*-PrOH-hexane); mp 93-95 °C; IR (KBr) 2900, 1670, 1440, 1305, 1160, 1130, 1100, 1080, 1040, 980, 930, 870, 850, 800, 740, and 690 cm⁻¹; ¹H NMR (C₆D₆) $\delta = 0.97$ (3H, t, J = 7.0 Hz, 2-EtO), 1.77 (1H, dt, $J_{gem} = 16.9$ and $J_{4.3} = J_{4.5} = 5.5$ Hz, one of H-4), 2.52 (1H, br. dd, $J_{gem} = 16.9$ and $J_{4.3} = 12.8$ Hz the

other of H-4), 2.78 (1H, ddd, $J_{3.4} = 12.8$, $J_{3.4} = 5.5$, and $J_{3.2} = 1.8$ Hz, H-3), 3.16 (1H, dq, $J_{gem} = 9.5$ and J = 7.0 Hz, one of 2-EtO), 3.54, 3.66 (each 1H, each d, $J_{gem} = 14.3$ Hz, 6-CH₂SO₂Ph), 3.86 (1H, dq, $J_{gem} = 9.5$ and J = 7.0 Hz, the other of 2-EtO), 4.50 (1H, dd, $J_{5.4} = 5.5$ and 2.2 Hz, H-5), 4.88 (1H, d, $J_{2.3} = 1.8$ Hz, H-2), 6.8-7.3 (8H, Ph), and 7.7-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) $\delta = 14.97$ (2-EtO), 23.70 (C-4), 42.63 (C-3), 61.32 (C-6)', 64.43 (2-EtO), 99.77 (C-2), 105.60 (C-5), 127.17, 128.69, 128.84, 128.94, 129.34, 133.07 (each Ph), 140.13, 140.23, and 140.67 (Ph and C-6); Mass (70 eV, rel. intensity, %) *m/z* 358 (M⁺, 49), 312 (21), 171 (54), 149 (70), 147 (base peak), 143 (35), 142 (22), 141 (69), 129 (24), 128 (68), 127 (21), 121 (32), 119 (45), 115 (69), 105 (37), 104 (31), 103 (20), 92 (26), and 51 (50). Anal. Calcd for C₂₀H₂₂O₄S: C, 67.02; H, 6.19%. Found: C, 67.25; H, 6.35%.

cis-6-(Phenylsulfonylmethyl)-2,3,3a,7a-tetrahydro-4H-furo/2,3-b/pyran (9). Separated and purified by silica gel column chromatography using hexane–EtOAc (4:1 v/v). Colorless prisms (Et₂O-hexane); mp 79-81 °C; IR (KBr) 2900, 1670, 1440, 1270, 1040, 990, 920, 800, 750, 690, and 610 cm⁻¹; ¹H NMR (C₆D₆) δ = 1.20-1.90 (4H, m, H-3, H-3a, and H-4), 3.5-3.6 (1H, m, one of H-2), 3.53 (2H, s, 6-CH₂SO₂Ph), 3.82 (1H, ddd, J_{gem} = 9.2, J₂₋₃ = 8.4, and 2.9 Hz, the other of H-2), 4.45 (1H, dd, J₅₋₄ = 5.1 and 2.6 Hz, H-5), 4.85 (1H, d, J_{7a-3a} = 3.7 Hz, H-7a), 6.9-7.0 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 21.66 (C-4), 27.75 (C-3), 36.59 (C-3a), 61.58 (6-CH₂SO₂Ph), 67.90 (C-2), 99.82 (C-7a), 100.51 (C-5), 128.69, 128.84, 133.17 (each Ph), 140.60, and 141.21 (Ph and C-6); MS (75 eV, rel intensity, %) *m*/z 280 (M⁺, 11), 139 (28), and 70 (base peak). Anal. Calcd for C₁₄H₁₆O₄S: C, 59.98; H, 5.75%. Found: C, 60.18; H, 5.80%.

General Procedure for the Transformation of 2-Ethoxy-3,4-dihydro-2H-pyran Cycloadducts 6a, 7b,d, and 8b to 6-Ethoxy-1-phenylsulfonyl-2-hexanone Derivatives 10a-d. As a typical example the transformation of 6a to 10a is described as follows: To a solution of 6a (148 mg, 0.5 mmol) in CH_2Cl_2 (3 ml) was added TiCl₄ (2 M in CH_2Cl_2 , 0.06 ml, 0.55 mmol) under nitrogen. The mixture was stirred at -78 °C for 10 min and the addition of triethyl silane (64 mg, 0.088 ml, 0.55 mmol) was followed. The reaction mixture was stirred at -78 °C for 1 h. The resulting mixture was quenched with saturated aqueous NaHCO₃ and extracted with CH_2Cl_2 (30 ml x 3). The combined extracts were dried (MgSO₄) and evaporated in vacuo. The crude product (139 mg) was chromatographed on silica gel with hexane-EtOAc (4:1 v/v) to give keto ether 10a (111 mg, 74%). Other keto ethers 10b-d were also prepared in 42, 80, and 80 % yields, respectively, by the similar reaction method.

6-Ethoxy-4-methyl-1-phenylsulfonyl-2-hexanone (10a). Colorless oil; IR (neat) 2900, 1720, 1445, 1375, 1320, 1150, 1080, 1040, 980, 740, and 680 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.81 (3H, d, J_{Me-4} = 6.6 Hz, 4-Me), 1.09 (3H, t, J = 7.0 Hz, 2-EtO), 1.21-1.51 (2H, m, H-5), 2.05-2.22 (1H, m, H-4), 2.26 (1H, dd, J_{gem} = 17.2 and J₃₋₄ = 7.3 Hz, one of H-3), 2.46 (1H, dd, J_{gem} = 17.2 and J₃₋₄ = 5.5 Hz, the other of H-3), 3.15-3.30 (4H, m, H-6 and 6-EtO), 3.64 (2H, s, 1-CH₂SO₂Ph), 6.8-7.0 (3H, m, Ph), and 7.8-7.9 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.48 (6-EtO), 19.97 (4-Me), 26.41 (C-4), 36.50 (C-5), 51.33 (C-3), 66.22 (C-1), 66.99 (6-EtO), 68.54 (C-6), 128.55, 129.08, 133.60, 139.97 (each Ph), and 197.27 (C=O); MS (70 eV, rel intensity, %) *m/z* 254 (M⁺-EtOH, 5), 158 (28), 145 (30), 142 (57), 141 (21), 126 (46), 117 (28), 116 (91), 115 (34), 113 (33), 112 (58), 111 (74), 110 (48), 100 (37), 96 (33), 94 (31), 81 (48), 78 (87), 77 (47), 72 (24), 70 (63), 60 (86), 56 (base peak), 55 (22), and 52 (53). Anal. Calcd for C₁₅H₂₂O₄S: C, 60.38; H, 7.43%. Found: C, 60.48; H, 7.30%.

6-Ethoxy-5-phenyl-1-phenylsulfonyl-2-hexanone (10b). Separated and purified by silica gel column chromatography using hexane-EtOAc (4:1 v/v). Colorless oil; IR (neat) 2850, 1700, 1575, 1440, 1370, 1275, 1220, 1140, 1080, 1020, 875, 825, 725, and 680 cm⁻¹; ¹H NMR (C₆D₆) δ =1.04 (3H, t, *J* = 7.0 Hz, 2-EtO), 1.74-1.78 (1H, m, one of H-4), 2.10-2.43 (3H, m, H-3 and the other of H-4), 2.66-2.80 (1H, m, H-5), 3.21 (2H, q, *J* = 7.0 Hz, 2-EtO), 3.30 (1H, dd, *J*_{gem} = 9.5 and *J*₆₋₅ = 7.0 Hz, one of H-6), 3.36 (1H, dd, *J*_{gem} = 9.5 and *J*₆₋₅ = 5.9 Hz, the other of H-6), 3.43 (2H, s, H-1), 6.8-7.2 (8H, m, Ph), and 7.7-7.8 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.32 (6-EtO), 26.64 (C-4), 42.29 (C-5), 45.60 (C-3), 66.46 (C-1), 66.61 (6-EtO), 75.46 (C-6), 126.92, 128.30, 128.53, 129.08, 129.04, 133.56, 139.23, 142.79 (each Ph), and 197.27 (C=O); MS (70 eV, rel intensity, %) *m/z* 315 (M⁺-EtO, 25), 314 (M⁺-EtOH, 25), 174 (39), 173 (84), 172 (52), 160 (38), 159 (base peak), 158 (59), 141 (50), 140 (25), 117 (38), 116 (23), 104 (58), 103 (41), 91 (45), 77 (51), 59 (63), and 58 (23). Anal. Calcd for C₂₀H₂₄O₄S: C, 66.64; H, 6.71%. Found: C, 66.25; H, 6.71%.

4,5-syn-6-Ethoxy-4-methyl-5-phenyl-1-phenylsulfonyl-2-hexanone (10c). Separated and purified by silica gel column chromatography using hexane–EtOAc (4:1 v/v). Colorless oil; IR (neat) 2950, 2850, 1710, 1575, 1440, 1370, 1320, 1150, 1100, 1025, 900, 730, 700, and 675 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.73 (3H, d, J = 6.2 Hz, 4-Me), 1.02 (3H, t, J = 7.0 Hz, 2-EtO), 2.36 (1H, dd, J_{gem} = 17.6 and J_{3.4} = 7.0 Hz, one of H-3), 2.49-2.70 (2H, m, H-4 and H-5), 2.76 (1H, dd, J_{gem} = 17.6 and J_{3.4} = 7.0 Hz, one of H-3), 2.49-2.70 (2H, m, H-4 and H-5), 2.76 (1H, dd, J_{gem} = 17.6 and J_{3.4} = 7.0 Hz, 2-EtO), 3.41 (1H, dd, J_{gem} = 9.5 and J_{6.5} = 4.8 Hz, one of H-6), 3.49 (1H, dd, J_{gem} = 9.5 and J_{6.5} = 6.6 Hz, the other of H-6), 3.68, 3.74 (each 1H, each d, J_{gem} = 13.6 Hz, H-1), 6.8-7.2 (8H, m, Ph), and 7.7-7.8 (2H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.32 (6-EtO), 17.98 (4-Me), 31.35 (C-4), 49.83 (C-5), 50.81 (C-3), 66.42 (C-1), 67.00 (6-EtO), 73.07 (C-6), 126.79, 128.55, 128.58, 128.88, 129.08, 133.59, 140.04, 142.27 (each Ph), and 197.29 (C=O); MS (70 eV, rel intensity, %) m/z 329 (M⁺-EtO, 74), 328 (M⁺-EtOH, 77), 188 (83), 184 (22), 174 (43), 173 (base peak), 172 (46), 171

(35), 170 (58), 169 (99), 168 (39), 153 (39), 140 (45), 131 (29), 119 (20), 116 (47), 115 (31), 104 (33), 103 (32), 90 (32), 76 (23), and 58 (44). Anal. Calcd for $C_{21}H_{26}O_4S$: C, 67.35 H, 7.00%. Found: C, 67.43; H, 6.98%.

4.5-anti-6-Ethoxy-4-methyl-5-phenyl-1-phenylsulfonyl-2-hexanone (10d). Separated and purified by silica gel column chromatography using hexane-EtOAc (4:1 v/v). Colorless oil; IR (neat) 2950, 2850, 1710, 1575, 1440, 1370, 1320, 1140, 1100, 1025, 880, 730, 700, and 675 cm⁻¹; ¹H NMR (C₆D₆) $\delta = 0.98$ (3H, d, J = 5.5 Hz, 4-Me), 1.02 (3H, t, J = 7.0 Hz, 2-EtO), 2.27 (1H, dd, $J_{gem} = 18.3$ and $J_{3.4} = 9.2$ Hz, one of H-3), 2.49-2.65 (3H, m, the other of H-3, H-4, and H-5), 3.15, 3.20 (each 1H, each dq, $J_{gem} = 9.2$ and J = 7.0 Hz, 6-EtO), 3.42, 3.48 (each 1H, each d, $J_{gem} = 13.2$ Hz, 1-CH₂SO₂Ph), 3.48 (2H, d, $J_{6.5} = 4.8$ Hz, H-6), 6.8-7.2 (8H, m, Ph), and 7.7-7.8 (2H, m, Ph): ¹³C NMR (C₆D₆) $\delta = 15.31$ (6-EtO), 17.98 (4-Me), 31.52 (C-4), 49.48 (C-5), 50.87 (C-3), 66.52 (C-1), 66.90 (6-EtO), 72.19 (C-6), 126.79, 128.31, 128.59, 128.97, 129.05, 133.57, 139.88, 143.06 (each Ph), and 197.46 (C=O); MS (70 eV, rel intensity, %) m/z 329 (M⁺-EtO, 33), 328 (M⁺-EtOH, 53), 188 (25), 187 (84), 186 (base peak), 185 (24), 173 (50), 171 (20), 170 (36), 169 (79), 156 (87), 155 (76), 154 (58), 145 (59), 143 (20), 142 (23), 141 (97), 140 (23), 132 (20), 131 (29), 130 (25), 118 (41), 117 (46), 77 (24), and 59 (41). Anal. Calcd for C₂₁H₂₆O₄S: C, 67.35 H, 7.00%. Found: C, 67.72; H, 7.09%.

General Procedure for the Transformation of 2-Ethoxy-3,4-dihydro-2H-pyran Cycloadducts 6a,d,f to Desulfonylated 6-Ethoxy-1-phenylsulfonyl-2-hexanone Derivatives 12a-c.

Benzylation of Dihydropyran 6a,d. As a example benzylation of **6a** to **11a** is described as follows: To a solution of dihydropyran **6a** (593 mg, 2.0 mmol) in THF (6 ml) was added butyllithium (in hexane, 1.4 ml, 2.2 mmol) by the aid of a syringe under nitrogen at -78 °C. After stirred at room temperature for 20 min, the mixture was again cooled to -78 °C and the addition of benzyl bromide (376 mg, 0.26 ml, 2.2 mmol) was followed. The reaction mixture was stirred at -78 °C for 30 min and quenched with saturated aqueous NH₄Cl. The THF was removed in vacuo and the aqueous residue was extracted with CH₂Cl₂ (30 ml x 3). The combined extracts were dried (MgSO4) and evaporated in vacuo. The crude product (715 mg) was chromatographed on silica gel with hexane-EtOAc (7:1 to 5:1 v/v) to give **11a** (662 mg, 85%) as a single stereoisomer.

Another benzylated dihydropyran 11b (90% de) was also prepared in 50% yield by the similar reaction method.

cis-2-Ethoxy-4-methyl-6-(2-phenyl-1-phenylsulfonylethyl)-3,4-dihydro-2H-pyran (11a). Colorless prisms (Et₂O-hexane); mp 79-81 °C; IR (KBr) 2950, 2850, 1660, 1440, 1280, 1220, 1175, 1125, 1050, 1010, 980, 850, 800, 745, 690, and 630 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.55 (3H, d, J_{Me-4} = 7.0 Hz, 4-Me), 1.07 (3H, t, J = 7.0 Hz, 2-EtO), 1.19 (1H, ddd, J_{gem} = 13.2, J₃₋₄ = 10.3 and J₃₋₂ = 8.8 Hz, one of H-3), 1.63 (1H, br. dd, J_{gem} = 13.2 and J₃₋₄ = 6.6 Hz, the other of H-3), 1.80-1.96 (1H, m, H-4), 3.25 (1H, dg, J_{gem} = 9.5 and J = 7.0 Hz, one of 2-EtO), 3.40 (1H, dd, J_{gem} = 13.2 and J_{CH2-1} = 12.1 Hz, one of 1'-CH₂Ph), 3.58 (1H, dd, J_{gem} = 13.2 and J_{CH2-1} = 3.3 Hz, the other of 1'-CH₂Ph), 3.71 (1H, dq, J_{gem} = 9.5 and J = 7.0 Hz, one of 2-EtO), 3.73 (1H, dd, J_{1'-CH2} = 12.1 and 3.3 Hz, H-1'), 4.16 (1H, dd, J₅₋₄ = 2.2 and J₅₋₃ = 1.1 Hz, H-5), 4.52 (1H, dd, J₂₋₃ = 8.8 and 2.2 Hz, H-2), 6.9-7.2 (8H, m, Ph), and 7.8-7.9 (3H, Ph); ¹³C NMR (C₆D₆) δ = 15.44 (2-EtO), 20.72 (4-Me), 26.80 (C-4), 31.91 (1'-CH2Ph), 36.31 (C-3), 64.61 (2-EtO), 72.68 (C-1'), 100.64 (C-2), 111.53 (C-5), 126.84, 128.49, 128.69, 129.54, 129.61, 133.08, 137.63 (each Ph), 139.41, and 142.60 (Ph and C-6); MS (20 eV, rel intensity, %) *m/z* 386 (M⁺, 43), 245 (base peak), and 199 (61). Anal. Calcd for C₂₂H₂₆O₄S: C, 68.37; H, 6.78%. Found: C, 68.43; H, 6.67%.

cis-2-Ethoxy-4-isopropyl-6-(2-phenyl-1-phenylsulfonylethyl)-3,4-dihydro-2H-pyran (11b). major isomer: Colorless prisms (Et₂O-hexane); mp 87-89 °C; IR (KBr) 2900, 2850, 1660, 1440, 1370, 1280, 1210, 1180, 1120, 1045, 1010, 970, 900, 850, 740, 680, and 630 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.43, 0.49 (each 3H, each d, J = 7.0 Hz, Me of 4-*i*-Pr), 1.11 (3H, t, J = 7.0 Hz, 2-EtO), 1.13 (1H, oct, J = 7.0 Hz, CH of 4-*i*-Pr), 1.35 (1H, ddd, $J_{gem} = 12.8$, $J_{3.4} = 11.2$ and $J_{3.2} = 9.2$ Hz, one of H-3), 1.58 (1H, dddd, $J_{gem} = 12.8$, $J_{3.4} = 6.6$, $J_{3.2} = 2.2$, and $J_{3.5} = 1.1$ Hz, the other of H-3), 1.73-1.85 (1H, m, H-4), 3.29 (1H, dq, $J_{gem} = 9.5$ and J = 7.0 Hz, one of 2-EtO), 3.43 (1H, dd, $J_{gem} = 12.8$ and $J_{CH2-1'} = 11.7$ Hz, one of 1'-CH₂Ph), 3.54 (1H, dd, $J_{gem} = 12.8$ and $J_{CH2-1'}$ = 3.3 Hz, the other of 1'-CH2Pb), 3.72 (1H, dd, $J_{1'-CH2}$ = 11.7 and 3.3 Hz, H-1'), 3.76 (1H, dq, J_{gem} = 9.5 and J = 7.0 Hz, one of 2-EtO), 4.20 (1H, dd, J₅₋₄ = 2.2 and J₅₋₃ = 1.1 Hz, H-5), 4.57 (1H, dd, J₂₋₃ = 9.2 and 2.2 Hz, H-2), 6.9-7.2 (8H, m, Ph), and 7.8-8.0 (3H, Ph); 13 C NMR (C₆D₆) δ = 15.52 (2-EtO), 18.46, 19.18 (each Me of 4-*i*-Pr), 30.63 (C-4), 31.30 (1-CH2Ph), 31.72 (CH of 4-i-Pr), 38.31 (C-3), 64.69 (2-EtO), 73.01 (C-1), 101.32 (C-2), 109.32 (C-5), 126.81, 128.53, 128.75, 129.47, 129.63, 133.08, 137.55 (each Ph), 139.96, and 143.39 (Ph and C-6); MS (75 eV, rel intensity, %) m/z 414 (M⁺, 43), 325 (42), 227 (68), 97 (26), 91 (base peak), 77 (66), and 41 (35). Anal. Calcd for C24H30O4S: C, 69.54; H, 7.29%. Found: C, 69.61 H, 7.31%. minor isomer: This product could not be separated from the mixture with major isomer because of the low yield. Its patial ¹³C NMR spectrum was abstracted (C₆D₆) δ = 15.39 (2-EtO), 19.05, 19.73 (each Me of 4-*i*-Pr), 30.88 (C-4), 31.62 (1'-CH2Ph), 31.88 (CH of 4-i-Pr), 38.08 (C-3), 64.29 (2-EtO), 72.15 (C-1'), 100.67 (C-2), 108.77 (C-5), 128.65, 129.23, and 129.33 (each Ph).

Reductive Desulfonylation of 11a,b and 6f Londing to 12a-c. As a typical example reductive desulfonylation of 11a to 12a is described as follows: A mixture of metal sodium and naphthalene (1:1 mol equiv.) in THF at 5 °C under nitrogen was

ultrasonified for 40 min to afford lithium naphthalenide reducing reagent as a dark green solution. To a solution of 11a (270 mg, 0.7 mmol) in isopropylamine (2ml) was added the freshly prepared sodium naphthalenide (0.5 M in THF, 5.6 ml, 2.8 mmol) at -78 °C under nitrogen. The reaction mixture was stirred at -78 °C for 10 min, quenched with water and extracted with CH₂Cl₂ (30 ml x 3). The combined extracts were dried (MgSO₄) and evaporated in vacuo. The crude product (400 mg) was chromatographed on silica gel with becane-Et₂O (49:1 v/v) to give 12a (95 mg, 55%). Other desulfonylated dihydropyran 12b,c were also prepared in 52 and 66 % yields, respectively, by the similar reaction method.

cis-2-Ethoxy-4-methyl-6-phenethyl-3,4-dihydro-2H-pyran (12a). Separated and purified by silica gel column chromatography using hexane-Et₂O (99:1 v/v). Colorless oil; IR (neat) 3050, 2950, 2900, 2850, 1675, 1610, 1500, 1455, 1380, 1330, 1290, 1260, 1220, 1190, 1130, 1060, 980, 910, 875, 780, 750, and 700 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.90 (3H, d, J_{Me-4} = 7.0 Hz, 4-Me), 1.15 (3H, t, J = 7.0 Hz, 2-EtO), 1.52 (1H, ddd, J_{gem} = 12.8, J₃₋₄ = 9.2 and J₃₋₂ = 8.1 Hz, one of H-3), 1.86 (1H, dddd, J_{gem} = 12.8, J₃₋₄ = 6.6, J₃₋₂ = 2.2, and J₃₋₅ = 1.1 Hz, the other of H-3), 1.99-2.25 (1H, m, H-4), 2.35 (2H, t, J = 7.3 Hz, 6-CH₂CH₂Ph), 2.83 (2H, t, J = 7.3 Hz, 6-CH₂CH₂Ph), 3.40, 3.93 (each 1H, each dq, J_{gem} = 9.5 and J = 7.0 Hz, 2-EtO), 4.28 (1H, br. s, H-5), 4.82 (1H, dd, J₂₋₃ = 8.8 and 2.2 Hz, H-2), and 7.0-7.2 (5H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.54 (2-EtO), 21.80 (4-Me), 26.48 (C-4), 33.92 (6-CH₂CH₂Ph), 36.33, 36.82 (C-3 and 6-CH₂CH₂Ph), 64.10 (2-EtO), 100.05 (C-2), 102.91 (C-5), 126.10, 128.52, 128.84, 142.08 (each Ph), and 150.45 (C-6); MS (75 eV, rel intensity, %) m/z 246 (M⁺, 82), 231 (31), 200 (24), 174 (78), 159 (95), 105 (31), 104 (20), 91 (95), 72 (54), 69 (base peak), 44 (35), and 41 (28). Anal. Calcd for C₁₆H₂₂O₂: C, 78.01; H, 9.00%. Found: C, 78.07; H, 8.86%.

cis-2-Ethoxy-4-isopropyl-6-phenethyl-3,4-dihydro-2H-pyran (12b). Separated and purified by silica gel column chromatography using hexane–Et₂O (99:1 v/v). Colorless oil: IR (neat) 2950, 2850, 1675, 1600, 1490, 1450, 1375, 1300, 1260, 1125, 1055, 975, 910, 890, 860, 775, 740, and 700 cm⁻¹; ¹H NMR (C₆D₆) δ = 0.75, 0.77 (each 3H, each d, *J* = 7.0 Hz, Me of 4-*i*-Pr), 1.18 (3H, t, *J* = 7.0 Hz, 2-EtO), 1.42 (1H, oct, *J* = 7.0 Hz, CH of 4-*i*-Pr), 1.64 (1H, ddd, *J*_{gem} = 12.8, *J*₃₋₄ = 11.0 and *J*₃₋₂ = 8.8 Hz, one of H-3), 1.76-1.86 (1H, m, the other of H-3), 1.91-2.02 (1H, m, H-4), 2.37 (2H, t, *J* = 7.3 Hz, 6-CH₂CH₂Ph), 2.84 (2H, t, *J* = 7.3 Hz, 6-CH₂CH₂Ph), 3.43, 3.98 (each1H, each dq, *J*_{gem} = 9.5 and *J* = 7.0 Hz, one of 2-EtO), 4.34 (1H, br. d, H-5), 4.82 (1H, dd, *J*₂₋₃ = 8.8 and 2.2 Hz, H-2), and 7.0-7.2 (5H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.59 (2-EtO), 19.34, 19.77 (each Me of 4-*i*-Pr), 31.94, 32.14 (each C-4 or CH of 4-*i*-Pr), 33.89 (6-CH₂CH₂Ph), 36.34 (6-CH₂CH₂Ph), 38.40 (C-3), 64.19 (2-EtO), 99.88 (C-2), 100.81 (C-5), 126.10, 128.53, 128.87, 142.03 (each Ph), and 151.54 (C-6); MS (70 eV, rel intensity, %) *m/z* 274 (M⁺, 33), 230 (42), 229 (37), 159 (base peak), 134 (56), 133 (35), 120 (20), 119 (22), and 77 (27). Anal. Calcd for C₁₈H₂₆O₂: C, 78.79; H, 9.55%. Found: C, 78.59; H, 9.35%.

cis-2-Ethoxy-6-methyl-4-phenyl-3,4-dihydro-2H-pyran (12c). Separated and purified by silica gel column chromatography using hexane–Et₂O (99:1 v/v). Colorless oil; IR (neat) 2950, 2900, 2850, 1675, 1600, 1490, 1440, 1375, 1300, 1270, 1175, 1125, 1065, 1030, 940, 885, 865, 760, and 700 cm⁻¹; ¹H NMR (C₆D₆) δ = 1.15 (3H, t, *J* = 7.0 Hz, 2-EtO), 1.77 (1H, dd, *J*_{Me-5} = 2.2 and *J*_{Me-4} = 1.1 Hz, 6-Me), 1.94 (1H, ddd, *J*_{gem} = 12.8, *J*_{3.4} = 10.6 and *J*_{3.2} = 8.8 Hz, one of H-3), 2.11 (1H, dddd, *J*_{gem} = 12.8, *J*_{3.4} = 6.6, *J*_{3.2} = 2.2, and *J*_{3.5} = 1.5 Hz, the other of H-3), 3.34-3.47 (1H, m, H-4), 3.40, 3.93 (each 1H, each dg, *J*_{gem} = 9.5 and *J* = 7.0 Hz, 2-EtO), 4.50 (1H,m, H-5), 4.82 (1H, dd, *J*₂₋₃ = 8.8 and 2.2 Hz, H-2), and 7.0-7.2 (5H, m, Ph); ¹³C NMR (C₆D₆) δ = 15.45 (2-EtO), 19.97 (6-Me), 37.67 (C-4), 38.62 (C-3), 64.07 (2-EtO), 100.14, 100.36 (C-2 and C-5), 126.59, 127.60, 128.69, 145.70 (each Ph), and 149.82 (C-6); MS (70 eV, rel intensity, %) *m*/z 218 (M⁺, 17), 173 (41), 172 (base peak), 171 (31), 161 (39),146 (22), 145 (42), 103 (40), 102 (23), and 85 (24). Anal. Calcd for C₁₄H₁₈O₂: C, 77.03; H, 8.31%. Found: C, 77.08; H, 8.28%.

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