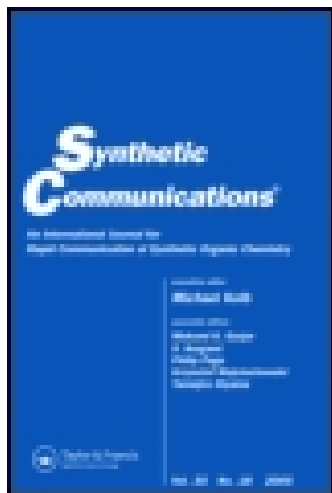


Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages,

and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

SYNTHESES OF 1-ARYLOXYACETYL-4-(3-TOLYLOXYACETYL) THIOSEMICARBAZIDES VIA SOLID-LIQUID PHASE TRANSFER CATALYSIS

Xicun Wang*, Zheng Li, Yuxia Da and Jichou Chen

Department of Chemistry, Northwest Normal University,
Lanzhou, Gansu, 730070, P.R.China

ABSTRACT: A series of new 1-aryloxyacetyl-4-(3-tolyloxyacetyl) thiosemicarbazides is synthesized under the condition of solid-liquid phase transfer catalysis.

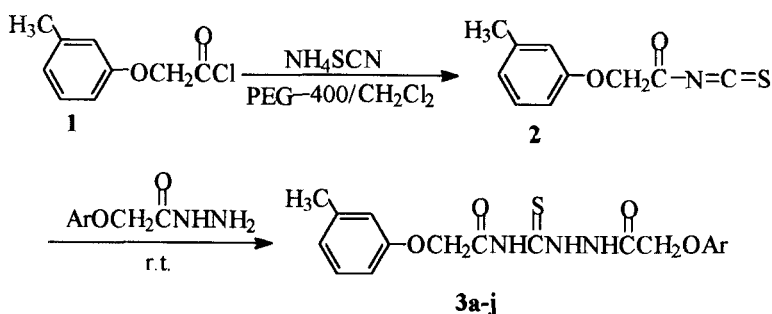
1,4-disubstituted thiosemicarbazides have been found to exhibit wide spectrum of biological activities. Some of them can be used as insecticides, herbicides and plant-growth regulators¹. Meanwhile, aryloxyacetic acid derivatives have also been used as herbicides and plant-growth regulators²⁻⁵. These applications prompt us to synthesize a new series of compounds bearing both thiosemicarbazide and aryloxyacetyl moiety, with the objective of obtaining new biologically active compounds.

* To whom correspondence should be addressed

In this paper, we report a convenient and efficient method for the preparation of new 1,4-disubstituted thiosemicarbazide derivatives under the condition of solid-liquid phase transfer catalysis using polyethylene glycol 400 (PEG-400) as the catalyst.

3- tolyloxyacetyl isothiocyanate (**2**) has been synthesized by the reaction of 3-tolyloxyacetyl chloride (**1**) with ammonium thiocyanate catalyzed by PEG-400 at room temperature. Further, compound **2** on treatment with aryloxyacetic acid hydrazides at room temperature gives 1-aryloxyacetyl-4-(3-tolyloxyacetyl) thiosemicarbazides (**3a-j**) in excellent yields (Scheme).

Scheme



3	Ar	3	Ar
a	C ₆ H ₅	f	1-Naphthyl
b	2-CH ₃ C ₆ H ₅	g	2-Naphthyl
c	4-CH ₃ C ₆ H ₅	h	2-O ₂ NC ₆ H ₅
d	4-CH ₃ OC ₆ H ₅	i	3-O ₂ NC ₆ H ₅
e	4-ClC ₆ H ₅	j	4-O ₂ NC ₆ H ₅

Acyl isothiocyanates are usually prepared by refluxing acyl chloride with potassium thiocyanate in acetone, but the yields are always very low. Harrison⁶ reported that polymer-supported thiocyanate on treatment with acyl chloride in benzene can afford acylisothiocyanate, but the preparation of the polymer-supported reagent required long reaction time and vacuum condition. Reeves and coworkers⁷ carried out the same reactions under the liquid-liquid phase transfer catalysis using tetrabutyl ammonium bromide as phase transfer catalyst, however, hydrolysis reactions of the acyl chloride were taken place frequently because of the presence of water, and the yields of the acylisocyanate were not high yet. In addition, a great excess of ammonium thiocyanate had to be used.

However, acylisocyanate can be easily obtained under the solid-liquid phase transfer catalysis using PEG-400 as catalyst. It is found that acyl chlorides are quantitatively converted to corresponding acyl isothiocyanates, which on reaction with the acyloxyacetic acid hydrazides to afford compound **3a-j** in high yields. From these reactions, it is seen that solid-liquid phase transfer catalytic method as the condition of main step to prepare **3a-j** has great advantages of mild reaction condition, high yield, short reaction time and low cost of catalyst.

EXPERIMENTAL SECTION

IR spectra were recorded using KBr pellets on an Alpha Centauri FTIR spectrophotometer and ¹H NMR spectra on a FT-80A instrument using (CD₃)₂SO

Table 1 Physical data and elemental analyses of **3a-j**

Product	m.p. (°C)	Yield(%)	%C	%H	%N
Found (Calculated)					
3a	118-119	94	57.93(57.89)	5.24(5.13)	11.05(11.25)
3b	162-163	87	59.17(58.90)	5.57(5.46)	10.73(10.85)
3c	167-168	95	58.73(58.90)	5.32(5.46)	10.94(10.85)
3d	117-118	89	56.40(56.56)	5.13(5.25)	10.57(10.41)
3e	137-138	90	52.81(53.00)	4.57(4.45)	10.49(10.30)
3f	208-209	91	62.63(62.40)	4.81(5.00)	10.14(9.92)
3g	182-183	87	62.56(62.40)	4.79(5.00)	10.20(9.92)
3h	180-181	92	51.79(51.67)	4.45(4.34)	13.51(13.39)
3i	197-198	89	51.40(51.67)	4.23(4.34)	13.54(13.39)
3j	201-202	93	51.57(51.67)	4.25(4.34)	13.20(13.39)

as solvent and Me₄Si as internal standard. Elemental analyses were performed on a Carlo-Erba 1106 Elemental Analysis instrument. Melting points were observed in an open capillary tube and uncorrected. 3-tolyloxyacetyl chloride⁸ (**1**) and aryloxyacetic acid hydrazides⁹ were prepared according to literature procedures.

Ammonium thiocyanate and PEG-400 were commercially available and used as received.

General procedure for preparation of **3a-j**

A suspension of 3-tolyloxyacetyl chloride (1.5mmol), ammonium

Table 2 IR data of **3a-j**

Product	IR ν (cm ⁻¹)			
	N-H	C=O	C=S	ArO
3a	3390			
	3275	1712	1184	1243
	3164			
3b	3386			
	3290	1707	1190	1247
	3221			
3c	3382			
	3279	1702	1192	1245
	3172			
3d	3393			
	3271	1713	1189	1250
	3192			
3e	3371			
	3268	1716	1186	1248
	3180			
3f	3386			
	3272	1701	1191	1252
3g	3392			
	3280	1704	1195	1250
3h	3380			
	3287	1708	1187	1247
	3201			
3i	3384			
	3271	1713	1189	1251
	3185			
3j	3392			
	3278	1705	1191	1244
	3189			

Table 3 ^1H NMR data of **3a-j**

Product	N-H ^a	N-H ^b	N-H ^c	Ar-H	OCH ₂	CH ₃	OCH ₃
3a	12.83 (1H,s)	10.06 (1H,s)	9.37 (1H,s)	6.76-7.50 (9H,m)	4.69 (4H,s)	2.33 (3H,s)	
3b	12.85 (1H,s)	10.11 (1H,s)	9.41 (1H,s)	6.80-7.52 (8H,m)	4.65 (4H,s)	2.29 (6H,s)	
3c	12.82 (1H,s)	10.08 (1H,s)	9.38 (1H,s)	6.75-7.54 (8H,m)	4.70 (4H,s)	2.27 (6H,s)	
3d	12.79 (1H,s)	10.03 (1H,s)	9.44 (1H,s)	6.70-7.33 (8H,m)	4.65 (4H,s)	2.31 (3H,s)	3.81 (3H,s)
3e	12.76 (1H,s)	10.10 (1H,s)	9.52 (1H,s)	6.79-7.54 (8H,m)	4.68 (4H,s)	2.33 (3H,s)	
3f	12.75 (1H,s)	10.59 (1H,s)	9.50 (1H,s)	6.75-7.52 (11H,m)	4.71 (4H,s)	2.38 (3H,s)	
3g	12.77 (1H,s)	10.61 (1H,s)	9.48 (1H,s)	6.78-7.53 (11H,m)	4.73 (4H,s)	2.35 (3H,s)	
3h	12.83 (1H,s)	10.09 (1H,s)	9.47 (1H,s)	6.80-7.47 (8H,m)	4.66 (4H,s)	2.20 (3H,s)	
3i	12.79 (1H,s)	10.12 (1H,s)	9.53 (1H,s)	6.85-7.46 (8H,m)	4.71 (4H,s)	2.31 (3H,s)	
3j	12.80 (1H,s)	10.19 (1H,s)	9.58 (1H,s)	6.91-7.58 (8H,m)	4.72 (4H,s)	2.33 (3H,s)	

thiocyanate (2.63mmol) and PEG-400 (3% based on ammonium thiocyanate) in methylene chloride was stirred for 1h at room temperature, then a kind of aryloxyacetic acid hydrazides (1.45mmol) was added. The mixture was stirred for another 0.5h, and a precipitate was observed immediately. The resulting mixture was filtered and washed with water to remove inorganic salts. The residue was recrystallized from DMF-EtOH-H₂O, and crystals of compound **3** were given. The physical and analytical data and spectral results were shown in Table 1-3.

ACKNOWLEDGEMENT

The authors are grateful for financial support from Natural Science Foundation of Gansu Province for this work.

REFERENCES

1. Zhang,Z. Y., Chen,L.M. and Zhang,L.X., *Chem. Res. And Appl. (Chinese)*, **1991**, 3, 3.
2. Baker,B.R.and Hurlbut,J.A., *J. Med. Chem.*, **1969**, 12, 677.
3. Jain,P.K. and Srirastara, S.K., *J. Indian Chem. Soc.*, **1992**, 69, 402.
4. Li,Y.J., Dai Y.J. and Chen,J.C., *Chem. J. Chin. Univ. (Chinese)*, **1988**, 9, 584 (Chem. Abstr., 110: 74986h).
5. Chen,J.C., Zhao,W.Z., Yang,S.Y., Wang,X.C., *Chem. J. Chin. Univ. (Chinese)*, **1991**, 12, 1195 (Chem. Abstr., 116:151263c).

6. Harrison, C. R. and Hodge, P., *Synthesis*, **1980**, 229.
7. Reeves, W. P., Simmons, J. A., Rudis, J. A. and Bothwell, T. C., *Synth. Commun.*, **1981**, *11*, 781.
8. Berliner, J. P. and Richter, S. B., *US Pat.* 3,306,726, **1967** (Chem. Abstr., 67: 81941r).
9. Husain, M. I. and Amir, M., *J. Indian Chem. Soc.*, **1986**, *63*, 317.

(Received in the USA 19 April 1999)