OMBUIN 3-SULPHATE FROM FLAVERIA CHLORAEFOLIA

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(Received 30 October 1987)

Key Word Index-Flaveria chloraefolia; Compositae; ombuin 3-sulphate; flavonol sulphate.

Abstract—One novel flavonoid sulphate ester and two known non-sulphated flavonoids were isolated from the leaves of *Flaveria chloraefolia*. The novel sulphated conjugate is 7,4'-dimethylquercetin (ombuin) 3-sulphate, and the known compounds are 6-methoxyquercetin and its 3-glucoside. The structure of ombuin 3-sulphate was established by UV spectroscopy, acid hydrolysis followed by EIMS of the aglycone, as well as enzymatic synthesis.

INTRODUCTION

The genus Flaveria is known to be a rich source of sulphated flavonols [1-9]. In fact, F. bidentis accumulates a number of flavonoid mono- to tetrasulphate esters, among which are isorhamnetin and quercetin 3-sulphates [1]; isorhamnetin 3,7- [2], quercetin 3,4'- [1] and 3,7-[3] disulphates; quercetin 3,7,4'- [1] and 3,7,3'- [4] trisulphates, as well as quercetin 3-acetyl-7,3',4'-trisulphate [5] and 3,7,3',4'-tetrasulphate [6]. Leaves of F. chloraefolia, on the other hand, contain quercetin 3-sulphate [7], 3,4'and 3,3'-disulphates [8]; patuletin 3-sulphate [7] and 3,3'-disulphate [8], together with 6-methoxy kaempferol, spinacetin, eupalitin, eupatolitin and eupatin 3-monosulphates [9]. In continuation of our phytochemical studies of Flaveria flavonoids, we wish to report the identification of three additional flavonoid compounds from the leaves of F. chloraefolia.

RESULTS AND DISCUSSION

Fresh leaves of *F. chloraefolia* were extracted with 50% aqueous methanol. After concentration under reduced pressure, the aqueous extract was partitioned successively against hexane, chloroform, ethyl acetate and butanol. The ethyl acetate extract contained the non-sulphated flavonoids, which after column chromatography yielded two major compounds, patuletin 1 and its 3-glucoside 2. Compound 1 was identified by co-chromatography with a reference compound, while 2 was characterized on the basis of its UV, ¹H NMR, ¹³C NMR and FABMS spectroscopic data. The butanolic extract after gel filtration on Sephadex LH-20, followed by preparative TLC on polyamide plates, afforded the new sulphated flavonoid 3.

Identification of ombuin 3-sulphate (3)

UV spectral analysis of 3 exhibited no shifts in the presence of aryl-sulphatase reagent [10] and gave a bathochromic shift of 22 nm only after the addition of HCl, both of which are indicative of a flavonol 3-monosulphate ester [10]. On the other hand, the absence of a shift for band II after the addition of sodium acetate as well as the bathochromic shift of 22 nm only in presence of sodium methoxide, demonstrated the substitution of the 7- and 4'-hydroxyl groups, respectively. Acid hydrolysis of 3 gave an aglycone 3a whose chromatographic and UV spectral properties, as well as its EIMS data were identical to a reference sample of ombuin (quercetin 7,4'dimethylether). Furthermore, the enzymatic synthesis of ombuin 3-sulphate was demonstrated using cell-free extracts of F. chloraefolia and [35S] 3'-phosphoadenosine-5'-phosphosulphate (PAPS) as the sulphate group donor. This enzyme preparation is known to catalyse the sulphation of flavonols at position 3 [11]. When ombuin was incubated with the enzyme preparation and the sulphate donor, the labelled reaction product was found to cochromatograph with 3, on polyamide TLC plates, using various solvent systems.

Patuletin 3-glucoside appears to be a characteristic constituent of the genus *Flaveria*, since it has previously been identified in *F. bidentis* [3], *F. linearis* and *F. trinervia* [12]. However, although ombuin 3,3'-disulphate has been identified in *Acrotema uniflorum* (Dilleniaceae) [13] this is the first report of ombuin 3-sulphate as a natural product.

EXPERIMENTAL

Plant material. Seeds of *F. chloraefolia* A. Gray were a generous gift from Professor A. M. Powell (Sul Ross State University, Alpine, TX) and were raised to fully grown plants under greenhouse conditions.

Source of reference compounds. Patuletin was a gift from Dr M. H. Moubasher (University of Texas, Austin, TX). Ombuin was from our laboratory collection. [³⁵S] PAPS was purchased from New England Nuclear, Boston. MA.

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General methods. Acid hydrolysis, ¹HNMR, ¹³CNMR, FABMS and EIMS analyses were carried out according to refs [7, 9]. UV spectra were obtained following standard procedures [14, 15], except for the spectra in presence of HCl and arylsulphatase which were recorded as in [9] and [10]. Prep. reverse-phase HPLC of the non-sulphated flavonoids was performed on a μ -Bondapack C₁₈ semi-prep. column (300 × 7.8 mm), using a UV detector (340 nm), a flow rate of 3 ml/min, and the following solvents: A, 90% aq. MeOH+0.1% HOAc; B, 20% aq. MeOH + 0.1% HOAc. Preparation of cell-free extracts of F. chloraefolia was carried out as in [16], and the enzymatic sulphation of ombuin was demonstrated using the standard flavonoid sulphotransferase assay [16]. Co-chromatography of the labelled ombuin 3-sulphate product with 3 was performed on polyamide-DC 6.6 TLC plates (Macherey-Nagel) using the following solvents, (a) 0.1% aq. tetrabutylammonium hydrogen sulphate-pyridine (4:1), 2 migrations; followed by (b) 0.1% aq. tetrabutylammonium hydrogen sulphate-pyridine (7:3), 2 migrations. The radioactive spots were located on the plate after autoradiography on X-ray film.

Extraction and isolation of the flavonoids. Extraction and liquid-liquid partition of the extracts were carried out as in ref. [7]. Chromatography of the EtOAc extract was performed on a polyamide SC-6 column, using a gradient of C_6H_6 -MeCOEt-MeOH (9:5:5) to (3:1:1), and yielded patuletin (1) and its 3-glucoside (2). Both compounds were purified by HPLC using 60% A + 40% B (1; $R_t = 7.5$ mn) or 40% A + 60% B (2; $R_t = 10.5$ mn). Ombuin 3-sulphate (3) was isolated from the BuOH extract after chromatography on a Sephadex LH-20 column as reported in ref. [7]. It was converted to its tetrabutylammonium salt, and purified by prep. TLC on polyamide plates, according to ref. [9].

Patuletin (1) was identified by co-chromatography with a reference compound on polyamide TLC using C_6H_6 -MeCOEt-MeOH (3:1:1), and cellulose TLC using *n*-BuOH-HOAc-H₂O (3:1:1); ¹H NMR (400.13 MHz, DMSO- d_6): 6.50 (1H, H-8), 6.87 (1H, d, J = 8.3 Hz, H-5'), 7.53 (1H, d, J = 8.3 Hz, H-6'), 7.66 (1H, H-2'). The UV spectral data of 1 was similar to reference patuletin.

Patuletin 3-O-β-glucoside (2). UV λ_{max}^{MeOH} nm: 255, 270 sh, 350; + NaOMe: 249, 305 sh, 382; + AlCl₃: 290, 325 sh, 383; + AlCl₃ + HCl: 258, 280 sh, 300 sh, 363; + NaOAc: 270, 410; + NaOAc $+H_{3}BO_{3}$: 268, 385; ¹H NMR (80 MHz, DMSO- d_{6}): ca 3.00-3.65 (5H, m, H-2", H-3", H-4", H-5" and H-6"), 3.75 (3H, s, Ar-OMe), ca 4.50-5.30 (OH gluc.), 5.45 (1H, d, J = 5.5 Hz, H-1"), 6.50 (1H, s, H-8), 6.83 (1H, d, J = 6.8 Hz, H-5'), ca 7.57 (2H, m, H-2' and H-6'); ¹³C NMR (100.13 MHz, DMSO-d₆): 60.07 (Ar-OMe), 61.10 (C-6"), 70.06 (C-4"), 74.22 (C-2"), 76.61 (C-3"), 77.61 (C-5"), 93.87 (C-8), 101.04 (C-1"), 104.25 (C-10), 115.32 (C-5'), 116.27 (C-2'), 121.32 (C-6'), 121.69 (C-1'), 131.47 (C-6), 133.07 (C-3), 144.90 (C-3'), 148.54 (C-4'), 151.70 (C-9), 152.40 (C-5), 156.30 (C-2), 158.01 (C-7), 177.65 (C-4); Negative FABMS (glycerol): 494 [M-H], 331 [M-glucose-H]. Acid hydrolysis of 2 gave 1 and glucose. The sugar was identified by co-chromatography with reference glucose on silica TLC plate according to [17].

Acknowledgements—This work was supported in part by operating grants from NSERC and the Fonds FCAR for which we are grateful. We wish to thank Dr M. T. Phan-Viet and S. Bilodeau (Montréal Regional High Field NMR Laboratory), for the recording of ¹H and ¹³C NMR spectra. We are indebted to Dr M. Evans (Université de Montréal), R.T.B. Rye (Concordia University) and C. W. Kazakoff (University of Ottawa) for FAB and EI mass spectroscopic analyses.

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