

## A Convenient Annellation of Imines to $\alpha$ -Substituted $\beta$ -Lactams<sup>1</sup>

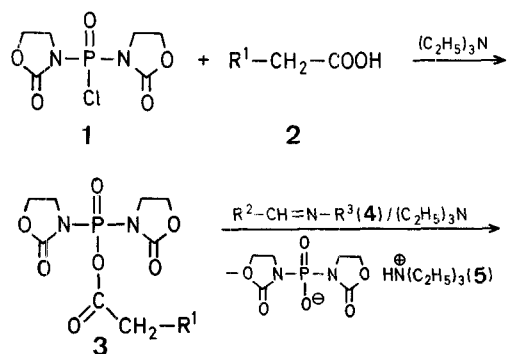
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It is well known that multi-substituted  $\beta$ -lactams exhibit high chemical reactivity and are prone to molecular rearrangements. Therefore, attempts have been made to explore new methods for  $\beta$ -lactam synthesis that operate under milder conditions. Several methods<sup>2-5</sup> are available for the synthesis of  $\beta$ -lactams. In recent years, various phosphorus agents have been used for activating the carboxy group to achieve the amide bond formation. Manhas et al.<sup>6</sup> have utilised such compounds for the formation of  $\beta$ -lactam ring systems. These authors found that phosphorylating agents like diethyl phosphorochloridate, diphenylphosphorochloridate, or *o*-phenylene phosphorochloridate led to the formation of  $\beta$ -lactams whereas diphenylphosphoryl azide, diethyl cyanophosphate, and diphenyl phosphite-pyridine failed to do so.

Palomo et al.<sup>7</sup> have recently described a new reagent *N,N*-bis[2-oxo-3-oxazolidinyl]phosphorodiamidic chloride (**1**) to activate the carboxy group for the synthesis of amides and esters. During our studies directed towards the semi-synthetic  $\beta$ -lactam antibiotics, we became interested in this reagent. In the present communication, we report on the use of this reagent in a new facile method for the conversion of imines to  $\alpha$ -substituted  $\beta$ -lactams.

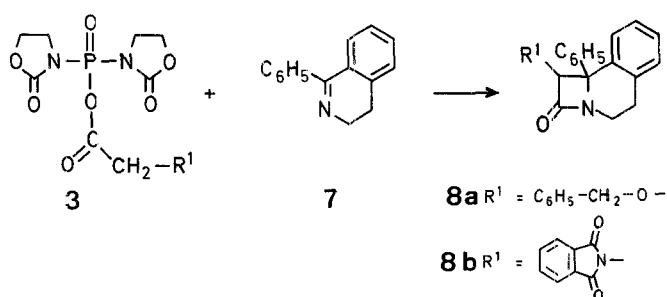
A mixture of imine **4** and an appropriately substituted acetic acid **2** in the presence of triethylamine was reacted with *N,N*-bis[2-oxo-3-oxazolidinyl]phosphorodiamidic chloride (**1**) at 10–20°C for 7–8 h to give the  $\beta$ -lactam **6** in 40–80% yield. The structure of the  $\beta$ -lactam **6** was confirmed by microanalyses, I.R., and <sup>1</sup>H-N.M.R. spectrometry. A possible reaction pathway is shown in Scheme A.



Scheme A

A variety of monocyclic  $\beta$ -lactams (Table) were prepared by this method. The stereochemistry was derived from their  $^1\text{H}$ -N.M.R. spectra. The configuration of C-3 and C-4 protons in all these monocyclic  $\beta$ -lactams was observed to be *cis* ( $J \approx 5$  Hz) except in case of the  $\alpha$ -phthalimido- $\beta$ -lactam which had *trans* disposition ( $J \approx 2$  Hz). The reaction works equally well for the synthesis of polycyclic  $\beta$ -lactams **8a** and **8b** (Scheme B).

The new method works under mild conditions and gives yields comparable to the earlier known methods for the synthesis of  $\beta$ -lactams.



Scheme B

Melting points were determined in open glass capillaries on a Gallenkamp melting point apparatus and are uncorrected. I.R. spectra were taken on Perkin Elmer 237 or 577 spectrophotometer.  $^1\text{H}$ -N.M.R. spectra were recorded on Varian A-60 and Jeol FX-100 instruments using TMS as internal standard. Microanalyses were performed using a Höslti microcombustion apparatus MK 101.

Table.  $\beta$ -Lactams **6** and **8** prepared

Product No.	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	Yield <sup>a</sup> [%]	m.p. [°C]	Molecular formula <sup>b</sup>	I.R. (nujol) $\nu$ [cm <sup>-1</sup> ]	$^1\text{H}$ -N.M.R. (CDCl <sub>3</sub> or CF <sub>3</sub> COOH) $\delta$ [ppm]
<b>6 a</b>				50	156–158°	C <sub>22</sub> H <sub>18</sub> ClNO <sub>2</sub> (363.8)	1750	4.25 (br s, 2H); 4.95 (d, 1H, $J=5$ Hz); 5.15 (d, 1H, $J=5$ Hz); 6.8–7.5 (m, 14H)
<b>6 b</b>				55	269°	C <sub>21</sub> H <sub>16</sub> ClNO <sub>2</sub> (349.8)	1730	insufficiently soluble
<b>6 c</b>				50	137°	C <sub>29</sub> H <sub>24</sub> ClNO <sub>3</sub> (470.0)	1735	4.32 (q, 2H); 5.0–5.2 (m, 4H); 6.9–7.5 (m, 18H)
<b>6 d</b>				45	182°	C <sub>23</sub> H <sub>20</sub> ClNO <sub>3</sub> (393.9)	1730	3.75 (s, 3H); 4.19 (d, 1H, $J=10$ Hz); 4.36 (d, 1H, $J=10$ Hz); 5.10 (d, 1H, $J=5$ Hz); 5.20 (d, 1H, $J=5$ Hz); 6.8–7.4 (m, 13H)
<b>6 e</b>				40	246°	C <sub>21</sub> H <sub>15</sub> Cl <sub>2</sub> NO <sub>2</sub> (384.3)	1730	5.56 (d, 1H, $J=5$ Hz); 5.80 (d, 1H, $J=5$ Hz); 6.7–7.3 (m, 13H)
<b>6 f</b>				45	160°	Ref. <sup>8</sup> , m.p. 159–160°C	1740	2.24 (s, 3H); 3.80 (s, 3H); 4.40 (s, 2H); 5.00 (d, 1H, $J=5$ Hz); 5.20 (d, 1H, $J=5$ Hz); 7.0–7.5 (m, 13H)
<b>6 g</b>				65	219°	C <sub>23</sub> H <sub>15</sub> ClN <sub>2</sub> O <sub>3</sub> (402.8)	1770, 1750, 1710	5.20 (d, 1H, $J=2.5$ Hz); 5.28 (d, 1H, $J=2.5$ Hz); 6.9–7.8 (m, 13H)
<b>8a</b>	—	—	—	60	128°	C <sub>24</sub> H <sub>21</sub> NO <sub>2</sub> (355.4)	1750	2.7 (m, 2H); 3.6 (m, 2H); 4.50 (s, 2H); 5.02 (s, 1H); 7.0–7.6 (m, 14H)
<b>8b</b>	—	—	—	80	273°	C <sub>25</sub> H <sub>18</sub> N <sub>2</sub> O <sub>3</sub> (394.4)	1770, 1750, 1710	2.7 (m, 2H); 3.7 (m, 2H); 5.50 (s, 1H); 6.9–7.9 (m, 13H)

<sup>a</sup> Yield of pure, crystallised product based on **4**.

<sup>b</sup> Satisfactory microanalyses obtained: C  $\pm 0.39$ , H  $\pm 0.34$ , N  $\pm 0.30$ ; exceptions: **6c**, **e**, **8a**, C  $\pm 0.50$ ; **6c**, **d**, H  $\pm 0.50$ .

**$\beta$ -Lactams 6 and 8; General Procedure:**

A mixture of Schiff's base **4** or 1-phenyl-3,4-dihydroisoquinoline **7** (0.01 mol), triethylamine (0.04 mol), and substituted acetic acid **2** (0.01 mol) in dichloromethane (200 ml) is stirred under a nitrogen atmosphere at 10°C and *N,N*-bis[2-oxo-3-oxazolidinyl]phosphorodiamidic chloride (**1**; 0.01 mol) is added. The resulting mixture is stirred at 20°C for 7–8 h, washed with water (50 ml), followed by 5% sodium hydrogen carbonate solution (50 ml), and finally again with water (50 ml). The dichloromethane solution, on evaporation, gives crude **6** or **8**, which is purified by column chromatography on florisil using hexane/dichloromethane (1:1) as eluent. The solvent is evaporated and the residue crystallised from chloroform/ethanol mixture to give pure **6** or **8** (Table).

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<sup>2</sup> R. Graf, *Justus Liebigs Ann. Chem.* **661**, 111 (1963).

<sup>3</sup> A. K. Bose, M. S. Manhas, R. M. Ramer, *Tetrahedron* **21**, 449 (1965).

<sup>4</sup> B. G. Chatterjee, P. N. Moza, S. K. Roy, *J. Org. Chem.* **28**, 1418 (1963).

<sup>5</sup> J. C. Sheehan, J. J. Ryan, *J. Am. Chem. Soc.* **73**, 1204, 4367 (1951).

<sup>6</sup> M. S. Manhas, Bansi Lal, S. G. Amin, A. K. Bose, *Synth. Commun.* **6**, 435 (1976).

<sup>7</sup> J. Diago-Meseguer, A. L. Palomo-Coll, *Synthesis* **1980**, 547.

<sup>8</sup> M. S. Manhas, S. G. Amin, B. Ram, A. K. Bose, *Synthesis* **1976**, 689.