= 128 Hz, CH₃), 38.1 (q, J = 132 Hz, C(O)CH₃), 62.7 (t, J = 128 Hz, CH₂), 82.3 (s, S-C-S), 201.8 (s, ketone C=O), 208.1, and 208.2 (s, $Fe(CO)_3$). These data can be rationalized only in terms of the methylene-bridged structure shown in Table I.

It is clear from these results that the $Fe_2(CO)_6$ system is a useful template for the designed construction of cyclic organosulfur ligands. In the Michael-type addition which is the subject of the present communication, activating functional groups (CO₂R, CN, C(O)R, etc.) must be present. Whether a one-carbon, two-carbon, or a three-carbon bridge is formed between the two sulfur atoms depends upon the type and the substitution pattern of the α,β unsaturated substrate that is used. Clearly, further variations are possible.

Further work based on this novel concept for organosulfur compound synthesis is in progress.

Acknowledgment. We are grateful to the National Science Foundation for support of this work and to R. M. Weinstein and M. K. Gallagher for assistance in obtaining NMR and mass spectra.

Copper(I) Catalysis of Olefin Photoreactions. 11. Synthesis of Multicyclic Furans and Butyrolactones via Photobicyclization of Homoallyl Vinyl and Diallyl Ethers¹

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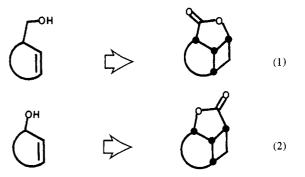
Since functional substituents enhance the potential synthetic utility of olefin photoreactions,² we are exploring photobicyclization of unsaturated ethers.⁵ We now report that copper(I) trifluoromethanesulfonate (CuOTf) catalyzes intermolecular photocycloadditions of homoallyl vinyl ethers to produce 2-oxabicyclo[3.2.0]heptanes. In conjunction with remarkably selective

(1) For previous papers in this series see: Salomon, R. G.; Ghosh, S.; Zagorski, M. G.; Reitz, M. J. Org. Chem. 1982, 47, 829-36 and references cited therein.

(2) Previously, this concept led to our discovery of intermolecular³ and intramolecular^{1,4} photocycloadditions of allylic alcohols. Copper(I)-catalyzed photocycloadditions of simple alkenes, i.e., those lacking functional substitu-ents, are known only for cyclic olefins: (a) Srinivasan, R. J. Am. Chem. Soc. J. E.; Greeley, R. H. *Ibid.* **1965**, 87, 4514–16. (d) Trecker, D. J.; Henry, J. P.; McKeon, J. E. Ibid. 1965, 87, 3261-63. (e) Haller, I.; Srinivasan, R. Ibid. 1966, 88, 5084-86. (f) Trecker, D. J.; Foote, R. S.; Henry, J. P.; McKeon, J. E. Ibid. 1966, 88, 3021-26. (g) Meinwald, J.; Kaplan, B. E. Ibid. 1967, 89, 2611-18. (h) Whitesides, G. M.; Goe, G. L.; Cope, A. C. Ibid. 1969, 91, 2608-15. (i) Angel, A. C. Ph.D. Thesis, University of Arizona, 1972. (j) Salomon, R. G.; Kochi, J. K. Tetrahedron Lett. 1973, 2529-32. (k) Salomon, R. G.; Kochi, J. K. J. Am. Chem. Soc. 1974, 96, 1137-44. (1) Salomon, R G.; Folting, K.; Streib, W. E.; Kochi, J. K. Ibid. 1974, 96, 1145-51. (m) Schwendiam, D. P.; Kutal, C. Inorg. Chem. 1977, 16, 719-21. (n) Kutal, C. Schwendiam, D. P.; Kutal, C. *Inorg. Chem.* 1977, 16, 719–21. (n) Rutal, C.;
Schwendiman, D. P.; Grutsch, P. A. Sol. Energy 1977, 19, 651–55. (o)
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Grutsch, P. A.; Kutal, C. Ibid. 1977, 99, 6460–63. (q) Evers, J. Th. M.;
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P. A.; Kutal, C. J. Am. Chem. Soc. 1979, 101, 4228–33. (s) Evers, J. Th.
M.; Mackor, A. Tetrahedron Lett. 1980, 21, 415–18. (t) Sterling, R. F.;
Kutal, C. Inorg. Chem. 1980, 19, 1402–05. (u) Maruyama, K.; Terada, K.;
Naruka V. Varmento, V. Chem. Lett. 1980, 1250–62. Naruta, Y.; Yamamoto, Y. Chem. Lett. 1980, 1259-62.

 (3) Salomon, R. G.; Sinha, A. *Tetrahedron Lett.* 1978, 1367-70.
 (4) (a) Salomon, R. G.; Coughlin, D. J.; Easler, E. M. J. Am. Chem. Soc.
 1979, 101, 3961-2. (b) Salomon, R. G.; Coughlin, D. J.; Ghosh, S.; Zagorski, M. G. Ibid. 1982, 104, 998-1007. (c) Shih, C.; Fritzen, E. L.; Swenteon, J. S. J. Org. Chem. 1980, 45, 4462-71. (d) McMurry, J. E.; Choy, W. Tetrahedron Lett. 1980, 21, 2477-80.

(5) The seminal discovery that copper(I) trifluoromethanesulfonate catalyzes photobicyclization of diallyl ether to produce cis-3-oxabicyclo[3.2.0]heptane was reported by Evers and Mackor (Evers, J. Th. M.; Mackor, A. *Tetrahedron Lett.* **1978**, 821-4). An analogous reaction, photobicyclization of 1.6-heptadiene to produce *cis*-bicyclo[3.2.0]heptane, is sensitized by mercury: (a) Srinivasan, R.; Hill, K. A. J. Am. Chem. Soc. 1965, 87, 4988-92. (b) Srinivasan, R.; Carlough, K. H. Ibid. 1967, 89, 4932-6.



ruthenium-catalyzed oxidation of these tetrahydrofuran derivatives, the new photoreaction provides a novel annulation of multicyclic butyrolactones that incorporate a cis-fused cyclobutane ring, e.g., eq 1. Similar annulations, e.g., eq 2, are also reported that involve 3-oxabicyclo[3.2.0]heptane intermediates generated by copper-(I)-catalyzed photobicyclization of unsymmetrical diallyl ethers.

Ultraviolet irradiation of the homoallyl vinyl ethers⁶ 1a-f in the presence of CuOTf⁷ produces cycloadducts 2a-f (Table I).^{8,9} The structure of **2a** was confirmed by ¹H NMR spectral comparison with a sample prepared by catalytic hydrogenation of 2-oxabicyclo[3.2.0]hept-6-ene.¹⁰ The structures of **2b-f** are assumed to be derivatives of 2-oxabicyclo[3.2.0]heptane in analogy with the $1a \rightarrow 2a$ conversion and the generation of multicyclic derivatives of bicyclo[3.2.0]heptane from various derivatives of 1,6-heptadiene by copper(I)-catalyzed photobicyclization.¹ The new photoreaction provides a general and topologically unique route to a variety of polycyclic ethers.¹¹

Synthetically useful transformations of the photoproducts 2 are facilitated by the ether functional group. These transformations are most valuable if they can be achieved selectively at one of the two carbon centers which may be activated by the ether oxygen. Moderate selectivity for oxidation of the methylene group in 1-methyltetrahydrofuran was noted previously with ruthenium tetraoxide.¹⁴ We now find that the fused multicyclic tetrahydrofuran photoproducts 2c-k undergo highly selective oxidation

(7) Salomon, R. G.; Kochi, J. K. J. Am. Chem. Soc. 1973, 95, 1889-1900. (8) Representative procedure: To 2-cyclopentenylmethyl vinyl ether (1d) (1.16 g, 9.3 mmol) in ether (200 mL, freshly distilled from LiAlH₄) was added (CuOTf)₂C₆H₆ (0.21 g, 0.4 mmol). The homogeneous solution is irradiated for 24 h in a cylindrical Pyrex vessel with an internal 450-W Hanovia mercury vapor lamp in a water-cooled quartz well. After completion of the irradiation, the mixture is poured into a slurry of crushed ice (25 g) in concentrated aqueous NH_4OH (25 mL) and shaken well. The deep blue aqueous phase is separated and extracted with ether (2 × 50 mL). The combined ether solutions are washed with saturated aqueous NaCl (50 mL) and dried (Mg-SO₄). Distillation provides 2d (92%, bp 70-72 °C (12 mmHg)).

(9) Satisfactory elemental analyses were obtained for all new compounds, which also exhibited ¹H NMR spectra consistent with the structures assigned (10) Paquette, L. A.; Barrett, J. H.; Spitz, R. P.; Pitcher, R. J. Am. Chem. Soc. 1965, 87, 3417-22.

(11) Previous routes to 2-oxabicyclo[3.2.0]heptanes include photoelectrocyclization of 2,3-dihydrooxepins followed by catalytic hydrogenation,¹¹ termolecular photocycloaddition of alkenes with 4,5-dihydrofurans,¹² or intermolecular photocycloaddition of alkynes with 4,5-dihydrofurans followed by catalytic hydrogenation.^{12,13}

(12) Rosenberg, H. M.; Servé, M. P. J. Org. Chem. 1971, 36, 3015-18.

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(14) Smith, A. B., III; Scarborough, R. M., Jr. Synth. Commun. 1980, 10, 205 - 11

(15) General procedure: To a vigorously stirred two-phase mixture of CCl₄ (10 mL) and water (10 mL) containing NaIO₄ (0.50 g) is added RuO₂ (20 mg). The black dioxide dissolves after stirring about 10 min at 22 °C to produce a yellow solution of RuO_4 in the CCl_4 layer. To this is added the tetrahydrofuran derivative (0.65 mmol). The mixture is stirred at 22 °C until complete consumption of the starting material is observed by thin-layer chromatography on silica gel. The mixture is then filtered through Celite, which is then washed thoroughly with CH₂Cl₂. The aqueous layer is saturated with NaCl and extracted with CH₂Cl₂ (3×5 mL). Isopropanol (0.20 mL) is added to the combined organic extracts, which are shaken vigorously to reduce RuO_4 to RuO_2 , dried (MgSO₄), filtered, and stripped of solvent to afford the lactone, which is usually quite pure by ¹H NMR. Analytical samples were prepared by gas-liquid-phase chromatography on a 4 ft \times $^1/_4$ in. column filled with 15% FFAP on 60/80 Chromosorb P.

⁽⁶⁾ These were prepared by mercury(II)-catalyzed transvinylation of homoallyl alcohols with ethyl vinyl ether. Details will be reported in a full account of this work

Table I. Copper(I)-Catalyzed Photobic yclization of Homoallyl Vinyl Ethers 8

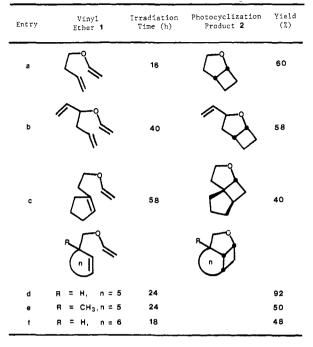
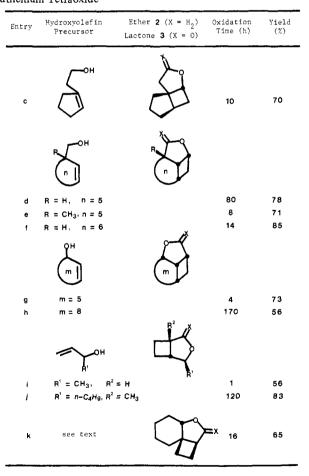
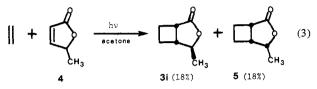


Table II.	Synthesis of Multicyclic Butyrolactones by Selective
Oxidation	of Tetrahydrofuran Derivatives with
Ruthenium	n Tetraoxide



of the methylene group to afford butyrolactones 3c-k with sodium periodate and a catalytic amount of ruthenium tetraoxide (Table II).^{9,15} Table II entries c-f demonstrate novel annulations of multicyclic butyrolactones from homoallyl alcohols. The cyclobutyl ring in the products **3d-f** is necessairly incorporated cis to

the original alcohol substituent. Similar annulations from allylic alcohols via unsymmetrical diallyl ethers are demonstrated by Table II entries g-j. The requisite diallyl ethers for photoannulation of **2g-j** are readily available by Williamson synthesis from the hydroxy olefin precursors indicated. It is noteworthy that the stereochemistry of the substituent \mathbb{R}^1 in **3i** and **3j** is exclusively exo.¹⁶ This synthetically valuable stereoselectivity is in sharp contrast with a topologically different previous synthesis of **3i** by intermolecular photocycloaddition of ethylene with **4**, which affords an equal yield of the endo-epimer **5** (eq 3).¹⁷ The requisite diallyl



ether 6 for photoannulation of 2k was prepared from cyclohexene oxide and allyl alcohol according to eq 4. Other synthetically

useful transformations of the photoproducts 2 as well as copper(I)-catalyzed photocyclizations of unsaturated ethers incorporating additional functionality are under investigation and will be described in a full account.

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(16) A discussion of the mechanistic basis of this stereoselectivity is deferred to a future full paper.

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Conrotatory Ring Opening of Cis-Fused 3-Aminocyclobutenes. X-ray Analysis of a *cis,trans-2H-Thiocin*

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The thermal valence isomerization of (hetero) cyclobutenes to (hetero) 1,3-butadienes is a classic example of a stereospecific reaction, the conrotatory mode of which is in agreement with the principle of conservation of orbital symmetry in electrocyclic reactions.^{1,2} A large amount of work has been carried out on the thermal isomerization of compounds in which the cyclobutene molety is cis annulated to another ring system. It is generally accepted that when the annulated ring possesses less than eight atoms, ring opening *must* occur by way of the symmetry-forbidden disrotatory mode³ or by homolytic⁴ or heterolytic⁵ pathways, all

⁽¹⁾ Woodward, R. B.; Hoffmann, R. "The Conservation of Orbital Symmetry"; Verlag Chemie: Weinheim, 1971.

⁽²⁾ Pennings, M. L. M.; Reinhoudt, D. N.; Harkema, S.; van Hummel, G. J. J. Am. Chem. Soc. 1980, 102, 7570-7571.