

Enantioselective Total Synthesis of Lankacidin C

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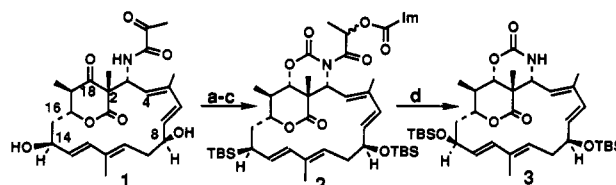
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The lankacidins, represented by lankacidin C (**1**), comprise a group of structurally unique, orally active antibiotics with substantial *in vivo* antitumor activity.¹ Because of the lankacidins' instability to both acids and bases,^{1c,2} chemical transformations of the intact antibiotics have been limited, and only a few approaches to their total synthesis have been reported.³ We now describe the first total synthesis of natural (-)-lankacidin C (**1**) by a convergent, enantioselective sequence starting from D-arabinose and L-aspartic acid, proceeding through the tricyclic carbamate **3** as an advanced relay intermediate. Structure **3** was chosen because it precluded the known degradative chemistry of this system.^{1c,2} To this end, natural **1** was silylated and reduced (Scheme I) to give a 1:1 mixture of C(2')-diol epimers, of which the less-polar isomer⁴ was reacted with Im₂CO to yield **2**. Selective deacylation of **2** with LiOOH⁵ gave a 98% yield of the stable relay **3**, mp 186–187 °C, $[\alpha]^{22}_D = -68.3^\circ$.

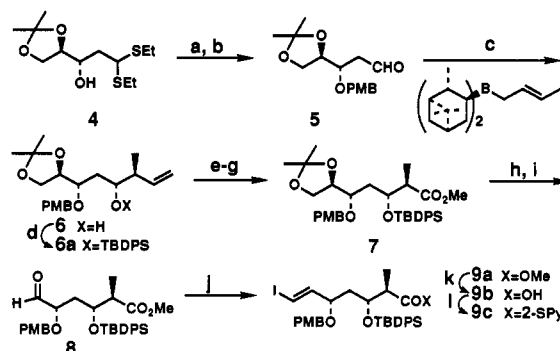
The enantiopure C(12)–C(18) segment was prepared (Scheme II) from the known dithioacetal **4**, derived in 43% yield from D-arabinose.⁶ The aldehyde **5** reacted with the crotylborane shown to give 58% of the adduct **6**,⁷ which was smoothly transformed to the ester **7**. Oxidative cleavage produced the unstable noraldehyde **8**, which was directly converted by the Takai method⁸ to the iodoalkene **9a** and then to the acid **9b**.

Stereoselective acylation by **9c** of the Li enolate **10**⁹ gave a β -ketolactam, reduced by KET₃BH to the single carbinol **11** (Scheme III).¹⁰ As explored earlier by Koch, **11** was desilylated and subjected to MeSO₃H-catalyzed N \rightarrow O acyl migration and then Im₂CO trapping to yield **12**.^{3c,9} Hydrolysis, Dess–Martin oxidation,¹¹ and PMB scission gave the stable iodoaldehyde **13**.

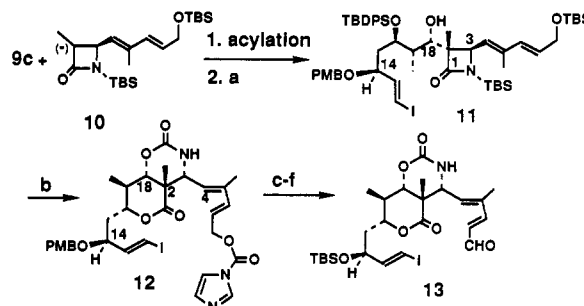
Lynchpin closure of **13** to relay **3** was achieved (Scheme IV) by Stille coupling of **13** with the stannane **14**¹² to give the tetraene **15a**. The chloride **15b** was reacted with TMSCN and then cyclized with LiHMDS at –78 °C to yield on hydrolysis the tetraenone **16**.¹³ The stereospecific reduction at C(8) was achieved by the (*R*)-CBS method¹⁴ to give 89% of the 8 β -ol, which on silylation gave crystalline **3**, mp 187–188 °C, $[\alpha]^{22}_D = -69.9^\circ$,

Scheme I^a

^a (a) Imidazole, TBSCl, DMF, rt, 100%. (b) NaBH₄, MeOH, rt, 99%. (c) 1,1'-Carbonyldiimidazole, LiHMDS, THF, –78 °C, 92% (from the less polar isomer). (d) LiOOH, THF–H₂O (3:1), 98%.

Scheme II^a

^a (a) NaH, PMBCl, DMF, rt, 91%. (b) HgCl₂, CaCO₃, MeCN–H₂O, 77%. (c) Chiral borane reagent, NaOH, H₂O₂, THF, 55%. (d) TBDPSCl, imidazole, DMF, rt, 48 h, 84%. (e) O₃, Sudan III, Me₂S, CH₂Cl₂–CH₃OH (1:1), –78 °C. (f) NaClO₂, rt, 78% (two steps). (g) CH₂N₂, 87%. (h) CuCl₂, MeOH, reflux for 1 h, 97%. (i) Pb(OAc)₄, THF, 0–5 °C. (j) CrCl₂, CHCl₃, THF, 62% (two steps). (k) LiOH, THF–H₂O–MeOH (6:3:2), rt, 12 h. (l) PySSPy, Ph₃P, THF, rt, 15 h, 79% (two steps).

Scheme III^a

^a (a) KET₃BH, Et₂O, –78 °C. (b) Bu₄NF, THF, rt, 2 h; MsOH, rt, 2 h; 1,1'-carbonyldiimidazole, NEt₃, rt, 12 h, 75% (two steps). (c) HCl (0.14 M), H₂O–dioxane (1:1), rt, 8 h, 70%. (d) Dess–Martin periodinone, CH₂Cl₂, 85%. (e) CAN, MeCN–H₂O, 97%. (f) TBSCl, imidazole, 79%.

indistinguishable by mmp, TLC, ¹H NMR, ¹³C NMR, IR, and FAB HRMS from **3** made from natural **1**.

(9) Koch, K. Ph.D. Thesis, Department of Chemistry, University of Rochester, New York, 1988; *Diss. Abstr. Int. B* 1989, 50(4), 1416. An account of our β -lactam rearrangement strategy toward lankacidin C, describing a successful prototype rearrangement to form a hydroxypyranone, was reported in August 1987 in Budapest, as cited in ref 3c. For an analogous and independently conceived approach, see refs 3b and 3e. The β -ketolactam corresponding to **10** was synthesized by an efficient sequence from the known 1-(TBS)-4-formylazetidin-2-one, itself derived from L-aspartic acid (Labia, R.; Morin, C. *Chem. Lett.* 1984, 1007. Salzmann, T. H.; Ratcliffe, R. W.; Christensen, B. G.; Bouffard, F. A. *J. Am. Chem. Soc.* 1980, 102, 6161). Reaction of the above aldehyde with the Li salt of *t*-BuN=CHCH(SiEt₃)–CH₃ (Schlessinger, R. H.; Poss, M. A.; Richardson, S.; Lin, P. *Tetrahedron Lett.* 1985, 26, 2391), followed by (PhS)₂/AIBN equilibration of the resulting crude enals, produced the pure (*E*)- α,β -unsaturated aldehyde. A second Schlessinger–Peterson condensation with the Li salt of *t*-BuN=CHCH₂SiEt₃ gave on workup the (*E,E*)-dienal, which was reduced with LiBH₄ (THF, –30 °C), O-silylated (TBSCl, Im, DMF), and C-methylated (LDA, MeI, –78 °C) to give the neutral form of **10** in 30% yield over six steps.

(1) (a) Uramoto, M.; Ōtake, H.; Ogawa, Y.; Yonehara, H.; Marumo, F.; Saito, Y. *Tetrahedron Lett.* 1969, 27, 2249. (b) Harada, S.; Kishi, T. *Chem. Pharm. Bull.* 1974, 22, 99. (c) Harada, S. *Chem. Pharm. Bull.* 1975, 23, 2201 and earlier citations in the above references. (d) Ootsu, K.; Matsumoto, T.; Harada, S.; Kishi, T. *Cancer Chemother. Rep., Part 1* 1975, 59, 919.

(2) For an illustration of this instability, see: McFarland, J. W.; Pirie, D. K.; Retsema, J. A.; English, A. R. *Antimicrob. Agents Chemother.* 1984, 25, 226.

(3) (a) Fray, M. J.; Thomas, E. J. *Tetrahedron* 1984, 40, 673. (b) Thomas, E. J.; Williams, A. C. *J. Chem. Soc., Chem. Commun.* 1987, 992. (c) Kende, A. S.; Luzzio, M. J.; Koch, K. In *Chemistry and Biotechnology of Biologically Active Natural Products, Proceedings of the Fourth International Conference*; Szántay, C., Ed.; Budapest, Hungary, Aug 10–15, 1987; Akad Kiado: Budapest, 1988; p 93, *Chem. Abstr.* 1989, 111, 214771m. (d) Rieger, D. L. Ph.D. Thesis, Department of Chemistry, Indiana University, 1989, *Chem. Abstr.* 1990, 113, 58758w. (e) Roe, J. M.; Thomas, E. J. *Syn. Lett.* 1990, 727.

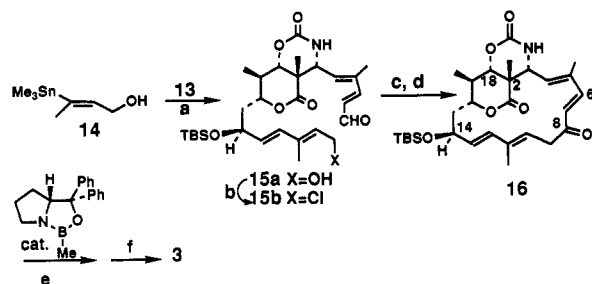
(4) Formation of both C(2')-carbinols on controlled reduction of **1** has been reported in ref 1b. The C(2') stereochemistry of the less polar carbinol was subsequently found to be *S* by showing its identity with the synthetic diol **18** made from relay **3** (Scheme V).

(5) Evans, D. A.; Britton, T. C.; Ellman, J. A. *Tetrahedron Lett.* 1987, 28, 6141.

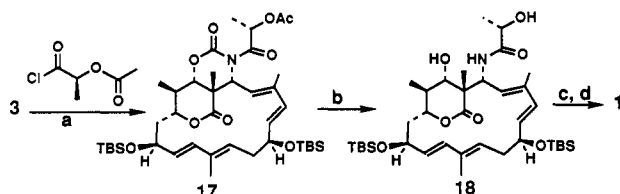
(6) Wong, M. Y. H.; Gray, R. G. *J. Am. Chem. Soc.* 1978, 100, 3548. (b) Maehr, H.; Perrotta, A.; Smallheer, J. *J. Org. Chem.* 1988, 53, 832.

(7) Cf.: Brown, H. C.; Bhat, K. S. *J. Am. Chem. Soc.* 1986, 108, 293.

(8) Takai, K.; Natta, K.; Utimoto, K. *J. Am. Chem. Soc.* 1986, 108, 7408. This reaction was performed under sonication.

Scheme IV^a

^a (a) Catalytic PdCl₂(CH₃CN)₂, DMF, rt, 90%. (b) 2,6-Lutidine, LiCl, MsCl, DMF, 0 °C. (c) Catalytic KCN/18-crown-6, TMSCN. (d) LiHMDS, THF, -78 °C; AcOH, THF-H₂O, rt, 20 h; 1% aqueous NaOH, 61% from 15a. (e) Oxazaborole catalyst, BH₃-THF, THF, -10 °C, 89%. (f) TBSCl, imidazole, DMF, rt, 95%.

Scheme V^a

^a (a) LiHMDS, THF, -78 °C, 85%. (b) LiOH, THF-H₂O (3:1), 0 °C, 82%. (c) Dess-Martin periodinane, CH₂Cl₂, rt, 96%. (d) HCOOH-THF-H₂O (3:6:1), rt, 3 h, 82%.

The final relay conversion of 3 to 1 by direct alkaline hydrolysis failed. However, when relay 3 (from natural 1) was acylated as in Scheme V, the *N*-acylcarbamate 17 was formed. Aqueous LiOH at 0 °C gave 82% of the bicyclic amide 18, which on Dess-Martin oxidation and careful desilylation (aqueous HCO₂H-THF)

gave 80% of the target molecule 1, identical in all respects with the natural antibiotic. This first total synthesis of (-)-1 proceeds in 30 steps from D-arabinose to relay 3 and proceeds from 3 to 1 over four steps in 55% yield.

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Supplementary Material Available: Physical and analytical data for 1, 3, 6, 9a, 10, 12, 13, 15a, 16, and 17 (14 pages). Ordering information is given on any current masthead page.

(10) The C(18)-S stereochemistry of 11 is assigned from NOE studies on the derived carbamate 12. Irradiation of the C(2)-Me in 12 gave an NOE of 7% on the *cis*-C(18)-H and one of 9% on the *cis*-C(3)-H. Together with the observed vicinal $J_{17,18}$ = 9.4 Hz, these data suggest a half-chair lactone conformation in 12, with a H(17)-H(18) dihedral angle of ca. 160° (cf. Fray, M. J.; Thomas, E. J.; Williams, D. J. *J. Chem. Soc., Perkin Trans. 1* 1985, 2763). This C(18)-S assignment would imply that KEt₃BH-Et₂O reduction of the β -ketolactam derived from acylation of 10 gives a configuration opposite that observed for a structurally related thienamycin precursor lacking the angular methyl substituent (Bouffard, F. A.; Christensen, B. G. *J. Org. Chem.* 1981, 46, 2208).

(11) For the preparation, use, and possible hazards of the Dess-Martin periodinane, see: Dess, D. B.; Martin, J. C. *J. Am. Chem. Soc.* 1991, 113, 7277.

(12) Piers, E.; Morton, H. E. *J. Org. Chem.* 1980, 45, 4263.

(13) Takahashi, T.; Nagashima, T.; Tsuji, J. *Tetrahedron Lett.* 1981, 22, 1359.

(14) Corey, E. J.; Bakshi, R. K.; Shibata, S.; Chen, C.-P.; Singh, V. *J. Am. Chem. Soc.* 1987, 109, 7925. A 10:1 ratio of chromatographically separable 8 β /8 α epimers was obtained using the (*R*)-CBS reagent.