

Novel Synthesis of 4,5-Bis(arylthio)-2,3,4,5-tetrahydro-1-benzothiepins: Noteworthy Cyclization by the Reaction of 2-Butynediol with Arenethiols in the Presence of Zinc Iodide

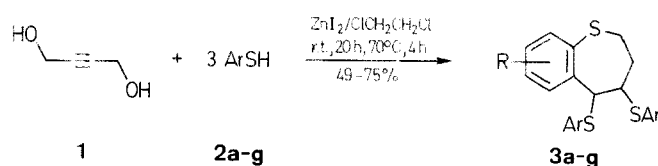
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New cyclization products, 4,5-bis(arylthio)-2,3,4,5-tetrahydro-1-benzothiepins, were obtained in good yield by the reaction of 2-butyndiol with arenethiols in the presence of zinc iodide. The reaction is assumed to proceed through intramolecular cyclization of a cationic intermediate generated *in situ*.

Base-catalyzed additions of arylthio groups to acetylenic bonds have been reported as typical examples of hitherto known reactions of alkynols with thiols,¹ while it has recently been found that the hydroxy group of benzyl and allylic alcohols is efficiently replaced by an arylthio group under catalysis by zinc iodide.² To our knowledge, a Lewis acid-catalyzed reaction of alkynols with arenethiols has not yet been reported.³

In the course of our synthetic and mechanistic studies on the reactivity and utilization of acetylenic alcohols, especially 2-butyndiols,⁴ we investigated the synthesis of heterocyclic compounds as potential biologically active substances. We now report a novel and convenient synthesis of 4,5-bis(arylthio)-2,3,4,5-tetrahydro-1-benzothiepins (**3**) from the reaction of 2-butyndiol (**1**) with arenethiols (**2**) in the presence of zinc iodide.



This selective formation of the cyclization product **3** may be assumed to proceed through the stepwise involvement of at least three different types of processes, that is, addition of thiol **2** to the $\text{C}\equiv\text{C}$ bond of **1**, substitution of one hydroxy group of the intermediate 2-arylthio-2-butene-1,4-diol by **2**, and intramolecular cyclization of a cationic intermediate generated *in situ*.

The structural assignment of products **3** is based on microanalyses, mass spectra, IR- and ^1H -NMR-spectral data and by comparison of the spectral data with those of dihalobenzothiepins.⁵ Although it should be expected that the two *vicinal* arylthio groups of the tetrahydro-1-benzothiepins **3** may be *cis* or *trans* to each other, the products **3** were found to consist of almost only one stereoisomer (GLC, TLC, ^1H -NMR, ^{13}C -NMR). The analogous isomers of dihalobenzothiepins deriva-

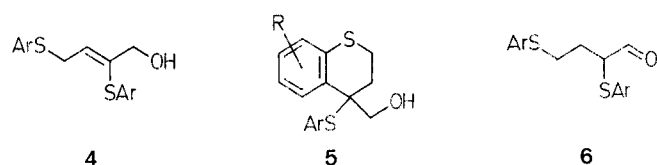
Table. 4,5-Bis(arylthio)-2,3,4,5-tetrahydro-1-benzothiepins **3** from 2-Butynediol (**1**) and Arenethiols **2**

2, 3	Ar	Yield ^{a,b,c} (%)	n_D^{25g}	Molecular Formula ^d	MS (M^+) ^e m/e	¹ H-NMR (CDCl ₃ /TMS) ^f δ , J (Hz)
a	C ₆ H ₅	72	oil 1.6600	C ₂₂ H ₂₀ S ₃ (380.4)	380	2.02 (m, 1H); 2.53 (m, 1H); 3.19 (m, 2H); 3.67 (m, 1H); 4.49 (d, 1H, J = 2.88); 7.19 (m, 14H)
b	p-ClC ₆ H ₄	72	oil 1.6597	C ₂₂ H ₁₇ Cl ₃ S ₃ (483.7)	482	2.04 (m, 1H); 2.43 (m, 1H); 3.18 (m, 2H); 3.53 (d t, 1H, J = 10.80, 3.60); 4.32 (d, 1H, J = 3.30); 7.17 (m, 11H)
c	p-CH ₃ C ₆ H ₄	75	oil 1.6332	C ₂₃ H ₂₆ S ₃ (422.5)	422	1.96 (m, 1H); 2.34 (s, 9H); 2.52 (m, 1H); 3.12 (m, 2H); 3.54 (d t, 1H, J = 11.16/3.24); 4.38 (d, 1H, J = 3.24); 7.04 (m, 11H)
d	o-CH ₃ C ₆ H ₄	56	oil 1.6440	C ₂₃ H ₂₆ S ₃ (422.5)	422	2.21 (s, 3H); 2.25 (m, 2H); 2.32 (s, 3H); 2.44 (s, 3H); 3.16 (m, 2H); 3.58 (m, 1H); 4.38 (d, 1H, J = 3.24); 7.08 (m, 11H)
e	m-CH ₃ C ₆ H ₄	69	oil 1.6384	C ₂₃ H ₂₆ S ₃ (422.5)	422	2.32 (m, 12H); 3.14 (m, 2H); 3.68 (m, 1H); 4.50 (d, 1H, J = 3.24); 7.05 (m, 11H)
f	4- <i>t</i> -C ₄ H ₉ C ₆ H ₄	65	oil 1.5846	C ₃₄ H ₄₄ S ₃ (548.7)	548	1.28 (s, 9H); 1.29 (s, 18H); 1.96 (m, 1H); 2.50 (m, 1H); 3.14 (m, 2H); 3.64 (m, 1H); 4.46 (d, 1H, J = 3.24); 7.26 (m, 11H)
g	2-naphthyl	49	oil 1.6046	C ₃₄ H ₂₆ S ₃ (530.4)	530	2.11 (m, 1H); 2.71 (m, 1H); 3.28 (m, 2H); 3.91 (m, 1H); 4.76 (d, 1H, J = 3.24); 7.39 (m, 20H)

^a Yield of isolated product.^b Under similar conditions, the reaction of **1** with a variety of aliphatic mercaptans resulted in the formation of complex mixture.^c The results in the absence of zinc iodide were: **3**, 2%; **4**, trace amount.^d Satisfactory microanalyses obtained: C \pm 0.32, H \pm 0.25.^e Recorded on a Jeol JMS-07.^f Recorded on a Jeol FX-90Q.^g n_D^{25} recorded on a Shimadzu Bausch & Lomb-3L.

tives have been clearly distinguished by means of ¹H-NMR techniques.⁵ However, the stereochemistry of **3** could not be definitely established to be either *cis* or *trans* from the spectral data of **3**.

Addition of 10 vol% of tetrahydrofuran to the reaction solvent lead to the formation of a mixture of 1-benzothiepin derivatives **3**, 2,4-diarylthio-2-butenols (**4**), 4-arylthio-4-hydroxymethyl-3,4-dihydro-2H-1-benzothiopyrans (**5**), and 2,4-bis(arylthio)-butanals (**6**). For example, in the case of reaction with 4-*t*-butylbenzenethiol (**2f**), products **3f**, **4f**, **5f**, and **6f** were obtained in yields of 8%, 25%, 8%, and 17%, respectively.



Each of the products **4**, **5**, and **6** was separated and efficiently transformed to **3** by further reaction with arenethiols and zinc iodide. These results suggest that **3** may be formed by intramolecular cyclization of **4** and **6**, or ring-enlargement of **5**. Although a detailed mechanism of the present reaction has not yet been, it may be assumed that **4** is initially formed and that **5** and **6** are intermediates formed by the cationic cyclization of **4** or by double-bond shift to the terminal position, respectively. The compound **3** is then formed by intramolecular cyclization of cationic intermediates generated by electrophilic attack of zinc iodide to carbonyl or hydroxy groups.

The present one-pot reaction thus provides a new and convenient method for the synthesis of benzothiepin derivatives which have hitherto been difficult to synthesize.⁶

7-Chloro-4,5-bis(4-chlorophenylthio)-2,3,4,5-tetrahydro-1-benzothiepin (**3b**); Typical Procedure:

Dried zinc iodide (8.00 g, 0.025 mol) is added to a stirred solution of 2-butyne-1,3-diol (**1**; 0.86 g, 0.01 mol) in dry 1,2-dichloroethane (40 mL) at

room temperature under N₂ atmosphere. Stirring is continued for 30 min and to the resultant suspension is added dropwise a solution of 4-chlorobenzenethiol (**2b**; 4.50 g, 0.04 mol) in 1,2-dichloroethane (10 mL). The mixture is stirred for ~ 20 h at room temperature and for 4 h at ~ 70°C. The reaction is then quenched with water (20 mL), and extracted with Et₂O (3 × 50 mL). The combined organic extracts were washed with ~ 20% NaOH solution (2 × 50 mL), and dried (MgSO₄). The solvent is removed in vacuo and the residue is purified by column chromatography on silica gel (hexane/acetone, 7:3); yield of **3b**: 3.50 g (72%) oil.

C₂₂H₁₇Cl₃S₃ calc. C 54.77 H 3.52
(483.7) found 54.49 3.71

MS: m/e = 482 (M^+).

IR (neat, major peaks): ν = 1580, 1480, 1390, 1100, 1015, 825 cm⁻¹.

¹H-NMR (CDCl₃): δ = 2.04 (m, 1H, 3,3-H₂); 2.43 (m, 1H, 3,3-H₂); 3.18 (m, 2H, 2,2-H₂); 3.53 (d t, J = 10.80, 3.60 Hz, 1H, 4-H); 4.32 (d, J = 3.30 Hz, 1H, 5-H); 7.17 (m, 11H_{arom}).

¹³C-NMR (CDCl₃): δ = 30.8 (t, C-3); 33.8 (t, C-2); 54.5 (d, C-4); 66.9 (d, C-5); ~ 130 (C_{arom}).

Spectral Data of Compound **4f**:

C₂₄H₃₂OS₂ (400.5)

MS: m/e = 400 (M^+).

IR (neat): ν = 3375 cm⁻¹.

¹H-NMR (CDCl₃): δ = 1.29 (s, 9H, 3 CH₃); 1.32 (s, 9H, 3 CH₃); 1.78 (br. s, 1H, OH); 3.59 (d, J = 8.28 Hz, 0.31H, S-CH₂, *E* isomer); 3.89 (d, J = 7.20 Hz, 1.7H, S-CH₂, *Z* isomer); 4.00 (br. s, 2H, OCH₂); 5.90 (t t, J = 8.28, 1.20, 0.14 Hz, CH=C, *E* isomer); 6.28 (t t, J = 7.20, 1.20, 0.86 Hz, CH=H, *Z* isomer); 7.28 (m, 8H_{arom}).

Spectral Data of Compound **5f**:

C₂₄H₃₂OS₂ (400.5)

MS: m/e = 400 (M^+).

IR (neat): ν = 3460 cm⁻¹.

¹H-NMR (CDCl₃): δ = 1.28 (s, 9H, 3 CH₃); 1.30 (s, 9H, 3 CH₃); 2.00 (m, 2H, 3,3-H₂); 2.38 (t, J = 6.48 Hz, 1H, OH); 3.30 (m, 2H, 2,2-H₂); 3.39 (d, J = 6.48 Hz, OCH₂); 7.32 (m, 8H_{arom}).

Spectral Data of Compound **6f**:

C₂₄H₃₂OS₂ (400.5)

MS: m/e = 400 (M^+).

IR (neat): ν = 1722 cm⁻¹.

¹H-NMR (CDCl₃): δ = 1.28 (s, 9 H, 3 CH₃); 1.31 (s, 9 H, 3 CH₃); 2.00 (m, 2 H, 3,3-H₂); 3.10 (t, J = 7.20 Hz, 2 H, SCH₂); 3.74 (t d, J = 7.20, 3.24 Hz, 1 H, SCH); 7.28 (m, 8 H_{arom}); 9.48 (d, J = 3.24 Hz, 1 H, CHO).

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- (4) For example: Ishino, Y., Wakamoto, K., Hirashima, T. *Chem. Lett.* **1984**, 765.
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