

# A Convenient Synthesis of Benzyloxycarbonyl-L-amino Acid 4-Methylcoumaryl-7-amides

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4-Methylcoumaryl-7-amides of amino acids or peptides have been employed as fluorogenic substrates for a variety of proteolytic enzymes<sup>1-6</sup>. In several cases these substrates gave more sensitive assays than were possible with analogous chromogenic substrates of the *p*-nitroanilide type due to the high fluorescence of the released 7-amino-4-methylcoumarin<sup>1,3</sup>.

We were particularly interested in the *N*- $\alpha$ -benzyloxycarbonyl derivative of L-arginine 4-methylcoumaryl-7-amide (**5a**). This compound is not only an excellent substrate for trypsin and papain, but may be de-benzyloxycarbonylated and elaborated to give a variety of specific peptide substrates for diagnostically important enzymes. Examples of enzymes which cleave peptidyl-arginine 4-methylcoumaryl-7-amide substrates are  $\alpha$ -thrombin, factor X<sub>a</sub>, plasma kallikrein, urokinase, and plasminogen activator<sup>4,6</sup>.

The synthesis of **5a** was reported by two groups using dicyclohexylcarbodiimide, with yields of 29%<sup>3</sup> and 13%<sup>2</sup>, respectively. However, repeated attempts in our laboratory to obtain the material by the latter method or by mixed anhy-

dride coupling failed to yield any product. In the search for alternative methods, we have found that a modification of the phosphorus pentoxide method of Schramm and Wissmann<sup>7</sup> gives the product in moderate yield with a minimum of purification necessary. This seldom-used coupling method involves an initial activation of the amine as a phosphoramidate monoethyl ester **3**. The activated amine is then coupled with a carboxylic acid **4** to yield a carboxamide **5** as shown in the Scheme.

This method was also successfully employed for other benzyloxycarbonyl-amino acids **4b-d** to give the amides **5b-d** as optically active crystalline solids in yields ranging from 35–58%. The optimum conditions for reaction were found to be addition of 1 equivalent of 7-amino-4-methylcoumarin (**1**) and 1–2 equivalents of tertiary amine to a hot solution of phosphorus pentoxide in diethyl phosphonate (**2**; diethyl phosphite). After allowing for formation of the activated intermediate **3**, the protected amino acid **4** was added. The coupling was carried out for 1–2 h at 110 °C.

## *N*- $\alpha$ -Benzyloxycarbonyl-L-arginine 4-Methylcoumaryl-7-amide Hydrochloride (**5a**):

A suspension of 7-amino-4-methylcoumarin (**1**; 17 g, 0.09 mol) in diethyl phosphonate (**2**; 92 ml) containing triethylamine (12.5 ml, 0.09 mol) is added to a 110 °C solution of phosphorus pentoxide (25.5 g, 0.2 mol) in diethyl phosphonate (92 ml). After 20 min a 90 °C solution of benzyloxycarbonyl-L-arginine (**4a**; 27.7 g, 0.09 mol) in diethyl phosphonate (92 ml) containing 85% phosphoric acid (10.4 g, 0.09 mol) is added. After 2 h at 110 °C, the reaction is evaporated (60 °C/1 torr) to give an oil. The oily residue is stirred with 1 normal hydrochloric acid (910 ml) while heating at 90 °C. Upon cooling, the product crystallizes. After 18 h at 4 °C, the product is collected and dried. Further purification is accomplished by dissolving the product in boiling methanol (2500 ml), decolorizing with charcoal, filtering, and concentrating until incipient crystallization; yield: 23.4 g (52%); m.p. 213 °C (dec.); [ $\alpha$ ]<sub>D</sub><sup>20</sup>: –17.0° (DMF); Ref.<sup>3</sup> m.p. 210–211 °C; [ $\alpha$ ]<sub>D</sub><sup>20</sup>: –17°.

C <sub>24</sub> H <sub>26</sub> CIN <sub>5</sub> O <sub>5</sub>	calc.	C 57.43	H 5.62	N 13.95
(502.2)	found	57.39	5.64	13.87

## Benzyloxycarbonyl-L-alanine 4-Methylcoumaryl-7-amide (**5b**):

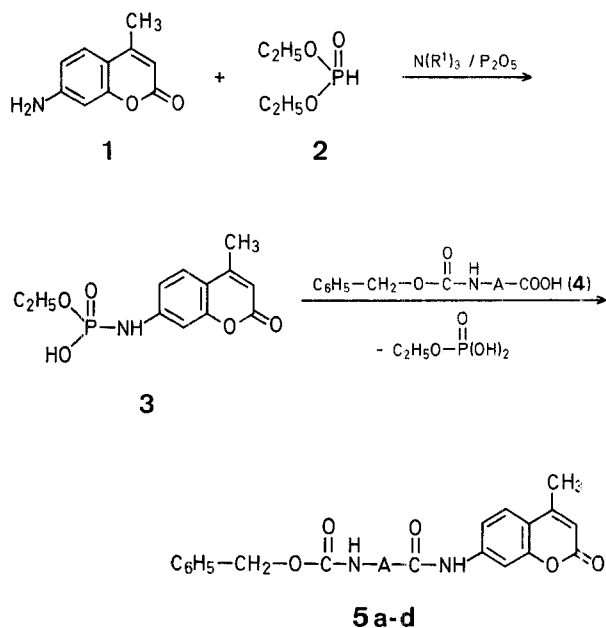
In a good hood, a suspension of 7-amino-4-methylcoumarin (**1**; 21.2 g, 0.11 mol) in diethyl phosphonate (**2**; 100 ml) containing triethylamine (31.2 ml, 0.22 mol) is added to a 110 °C solution of phosphorus pentoxide (31.8 g, 0.22 mol) in diethyl phosphonate (100 ml). After 20 min at 110 °C, benzyloxycarbonyl-L-alanine (**4b**; 25 g, 0.1 mol) is added to the reaction suspension. Stirring is continued for 2 h at 110 °C, then the reaction mixture is evaporated (60 °C/1 torr) to give a semisolid residue. The residue is triturated with water (1000 ml), collected, and dried in vacuum. The crude solid is dissolved in hot, 1:1 chloroform/methanol (2500 ml) to give a turbid solution which is decolorized with carbon, filtered hot, and concentrated until incipient crystallization. After standing at 4 °C, the colorless product is collected, washed with hexane, and dried to give pure **5b**; yield: 25.2 g (57%); m.p. 219–220 °C; [ $\alpha$ ]<sub>D</sub><sup>20</sup>: –33.5° (DMF).

C <sub>21</sub> H <sub>20</sub> N <sub>2</sub> O <sub>5</sub>	calc.	C 66.31	H 5.31	N 7.36
(380.4)	found	66.53	5.20	7.21

## Benzyloxycarbonyl L-phenylalanine 4-Methylcoumaryl-7-amide (**5c**):

Prepared as above for **5a** from benzyloxycarbonyl-L-phenylalanine (**4c**; 17.6 g, 0.06 mol). Product crystallizes upon concentration of a 1:1 chloroform/methanol solution; yield: 10 g (35%); m.p. 198–203 °C; [ $\alpha$ ]<sub>D</sub><sup>20</sup>: +56° (DMF).

C <sub>27</sub> H <sub>24</sub> N <sub>2</sub> O <sub>5</sub>	calc.	C 71.04	H 5.30	N 6.14
(456.5)	found	70.84	5.50	6.41



R' = C<sub>2</sub>H<sub>5</sub>, n-C<sub>4</sub>H<sub>9</sub>;

5	A
a	$\text{CH}-(\text{CH}_2)_3-\text{NH}-\text{C}(\text{NH}_2)=\text{NH} \cdot \text{HCl}$
b	$\text{CH}-\text{CH}_3$
c	$\text{CH}-\text{CH}_2-\text{C}_6\text{H}_5$
d	$\text{CH}-\text{C}(=\text{O})-\text{O}-\text{CH}_2-\text{C}_6\text{H}_5$

**Benzyloxycarbonyl- $\gamma$ -L-glutamyl 4-Methylcoumaryl-7-amide  $\alpha$ -Benzyl Ester (5d):**

Prepared as above for 5a from benzyloxycarbonyl-L-glutamic acid  $\alpha$ -benzyl ester (4d; 21.5 g, 0.13 mol) except with tributylamine as base. The product is crystallized by concentration of a 1:1 chloroform/methanol solution; yield: 17.9 g (58%); m.p. 193–195 °C;  $[\alpha]_D^{20}$ : –22.5° (DMF).

$C_{30}H_{28}N_2O_7$	calc.	C 68.17	H 5.34	N 5.30
(528.6)	found	66.84 <sup>8</sup>	5.65	5.37

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<sup>8</sup> Result of only 1 analysis.