

PALLADIUM-CATALYZED COUPLING REACTIONS OF TRIFLUOROACETIMIDOYL IODIDES WITH OLEFINS AND 1-ALKYNES

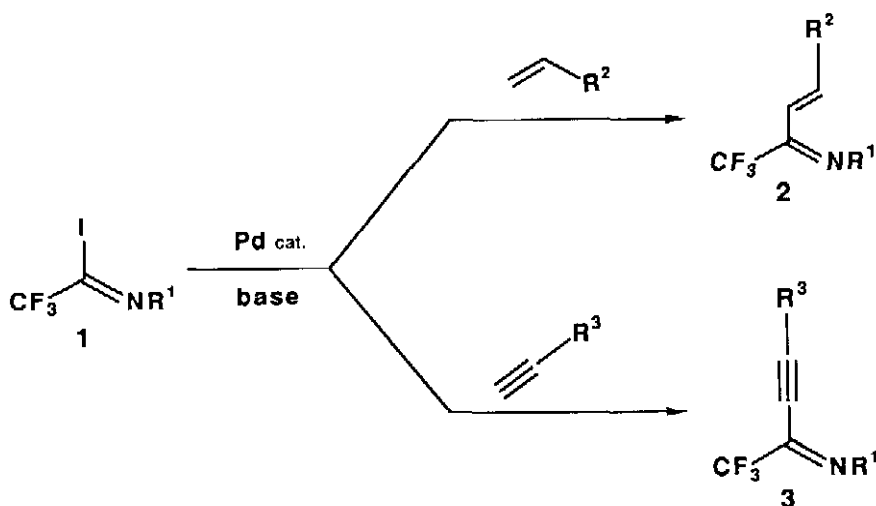
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Summary: Palladium-catalyzed coupling of trifluoroacetimidoyl iodides with olefins and 1-alkynes affords trifluoromethylated α , β -unsaturated imines, which are transformed into nitrogen heterocycles bearing CF_3 -group.

Methods for the synthesis of trifluoromethylated compounds have received a growing interest in the recent years, because many of these compounds exhibit interesting biological activity and high performance in material science.¹⁾ Therefore, development and transformation of trifluoromethylated building blocks are the important synthetic needs in this area.²⁾ Trifluoroacetimidoyl chlorides³⁾ have been found to be useful synthetic blocks for trifluoromethylated heterocycles. They behave as imidoyl carbocation and displace chlorine with both carbon⁴⁾ and nitrogen nucleophiles.⁵⁾ No reaction with electrophiles has been known so far. Here we describe palladium-catalyzed coupling reactions of trifluoroacetimidoyl iodides (**1**) with active olefins (H Heck type reaction) and 1-alkynes, providing the trifluoromethylated α , β -unsaturated imines which should be potential precursors of nitrogen heterocycles bearing CF_3 -group.

Very few is known about Heck type reaction with imidoyl halides although numerous Pd-catalyzed coupling reactions of aryl and vinyl halides with olefins and 1-alkynes have been reported.⁶⁾ Recently,

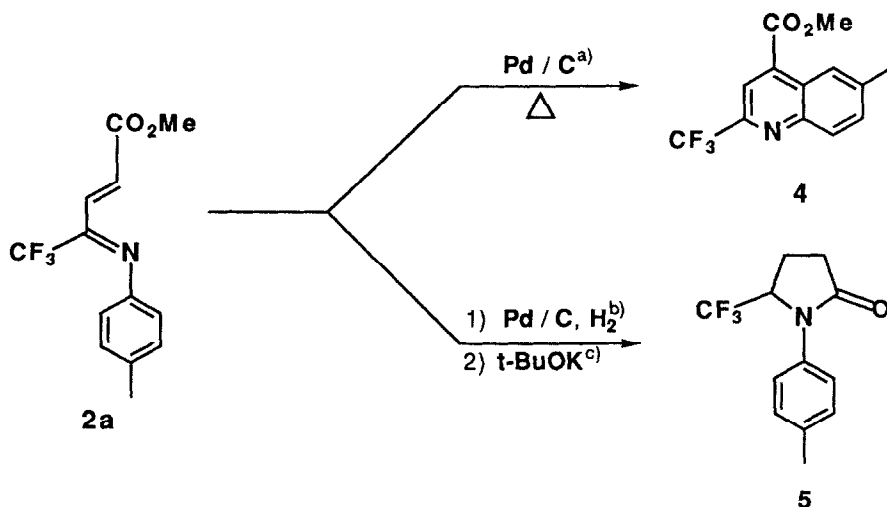


Tanaka, and Kosugi and Migita reported the Pd-catalyzed coupling reaction of benzimidoyl chlorides with alkenyl- and alkynylstannanes.⁷⁾ Although this procedure provided the coupling products in reasonable yields, it necessitated activation of olefins and 1-alkynes by stannilation.

We have found that N-aryl trifluoroacetimidoyl iodides (**1**)⁸⁾ undergo smooth palladation with $\text{Pd}_2(\text{dba})_3 \cdot \text{CHCl}_3$ although the corresponding chlorides are recovered intact in the same conditions. During the palladation process elimination of β -fluorine did not occur⁹⁾ and coupling of **1** with olefins and 1-alkynes afforded **2** and **3** in good yields in the presence of 5 mole % equivalent of palladium catalyst. The results are summarized in Table-1.¹⁰⁾

1-Alkynes react faster than the corresponding 1-alkenes (entry 1 and 10, 6 and 12). Activated carbon-carbon double bond reacts smoothly (entry 1~4), but non-activated olefin such as 1-octene provides the desired product in a poor yield (48%), while 1-hexyne reacts satisfactorily although the reaction rate is slower than those of the activated alkynes (entry 9 and 10). The $^1\text{H-NMR}$ revealed that the geometries of carbon-carbon double bond of **2** were trans.

The α, β -unsaturated imines **2** and **3** are promising building blocks for the trifluoromethylated nitrogen heterocycles. Thus, **2a** ($\text{R}^1 = \text{p-tolyl}$, $\text{R}^2 = \text{CO}_2\text{Me}$) was subjected to the palladium-catalyzed cyclization-dehydrogenation reaction, affording quinoline carboxylic acid ester (**4**)¹¹⁾ ¹²⁾ in a quantitative yield. The same compound (**2a**) was hydrogenated with Pd / C and then lactamized with t-BuOK to give the lactam (**5**)¹²⁾ in 91 % yield.



a) at 190-200 $^{\circ}\text{C}$ in nitrobenzene for 5 hr.

b) Pd/C in MeOH at room temperature for 1 day, H_2 (3 atm.).

c) t-BuOK(1 equiv to **2a**) in DMF at room temperature for 10 min.

We are currently trying to apply these coupling reactions to the syntheses of heterocycles and amino acids bearing trifluoromethyl groups.

Table-1. Pd-Catalyzed Coupling of Trifluoroacetimidoyl iodides (**1**) with Olefins^{a)} and 1-Alkynes.^{b)}

entry	R ¹	R ² or R ³ (equiv)		time (hr)	temp (°C)	yield of 2 or 3 ^{c)} (%)
1	p-CH ₃ C ₆ H ₄	CO ₂ Me	(3)	31	60	2a (94)
2	p-CH ₃ C ₆ H ₄	Ph	(3)	26	65	2b (95)
3	p-CH ₃ C ₆ H ₄	CN	(3)	45	65	2c (71) ^{d)}
4	p-CH ₃ C ₆ H ₄	SO ₂ Ph	(2)	17	65	2d (54)
5	p-CH ₃ C ₆ H ₄	(CH ₂) ₅ CH ₃	(3)	42	65	2e (48) ^{e)}
6	o-BrC ₆ H ₄	CO ₂ Me	(2)	48	65	2f (54)
7	p-MeOC ₆ H ₄	CO ₂ Me	(3)	52	65	2g (86)
8	p-CH ₃ C ₆ H ₄	CH ₂ OCOPh	(1.1)	12	r.t.	3a (70)
9	p-CH ₃ C ₆ H ₄	(CH ₂) ₃ CH ₃	(2)	24	r.t.	3b (71)
10	p-CH ₃ C ₆ H ₄	CO ₂ Et	(1.1)	2	r.t.	3c (66)
11	p-CH ₃ C ₆ H ₄	Ph	(1.1)	1	65	3d (92)
12	o-BrC ₆ H ₄	CO ₂ Et	(1.1)	4	r.t.	3e (47)

a) In a typical experimental procedure, a two-necked flask fitted with a septum cap, stir bar and condenser topped with a nitrogen inlet was charged with 0.017g(0.016 mmol) of Pd₂(dba)₃·CHCl₃, 0.088g(0.64 mmol) of K₂CO₃, and 0.2 ml of toluene. Then, 0.086ml(0.96 mmol) of methyl acrylate and 0.1g(0.32 mmol) of **2**, 2-trifluoroacetimidoyl iodide (**1**) in 0.5 ml of toluene were added to the catalyst mixture through a syringe. The mixture was stirred and filtered with florisil column (CH₂Cl₂). After evaporation of the solvent, the residue was purified by silica gel column chromatography to give 0.081g(94 %) of **2a**. b) A typical experimental procedure is similar to that described except that PdCl₂ (0.05 mol%), PPh₃ (0.1 mol%), and CuI (0.02 mol%) as catalysts and a mixture of CH₃CN and toluene(1:1) as solvents were employed. c) Isolated yield based on **1**. d) The ratio (E/Z) is 3/1. e) mixture of isomers.

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- 8) N-Arylacetimidoyl iodides (**1**) are prepared by displacement of chlorine of N-arylacetimidoyl chlorides with iodine in NaI-acetone system (yield 91 %).
- 9) A facile elimination of fluoride from β -metalated trifluoro-compounds is well known. T. Nakai, K. Tanaka, and N. Ishikawa, *Chem. Lett.*, 1263, (1976).
- 10) Structures of **2**, **3**, **4**, and **5** are identified by spectroscopic analyses (^1H , ^{13}C , and ^{19}F -NMR, IR, and MS).
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- 12) **4**: IR (Nujol, cm^{-1}) 1738 (CO_2Me); ^1H -NMR (δ , CDCl_3) 8.62 (s, 1H), 8.21 (s, 1H) 8.18 (d, $J=10.2\text{Hz}$, 1H), 7.72 (dd, $J_1=8.7\text{Hz}$, $J_2=1.9\text{Hz}$, 1H), 4.08 (s, 3H), 2.63 (s, 3H); ^{19}F -NMR (δ , CDCl_3 , C_6F_6) 94.28 (s). **5**: IR (Nujol, cm^{-1}) 1698 (CONR_2); ^1H -NMR (δ , CDCl_3) 7.20 (br.s, 4H), 4.48~4.65 (m, 1H), 2.23~2.88 (m, 4H), 2.35 (s, 3H); ^{19}F -NMR (δ , CDCl_3 , C_6F_6) 86.58 (d, $J=6.9\text{Hz}$).

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