

Synthesis of 3-Indolylarylmethanamides by Samarium Triiodide Catalyzed Friedel–Crafts Amidoalkylation

Hui Mao, Xiaoxia Wang,* Wencun Wang, Liang He, Lichun Kong, Junhua Liu

Zhejiang Key Laboratory for Reactive Chemistry on Solid Surfaces, College of Chemistry and Life Sciences, Zhejiang Normal University, Jinhua 321004, P. R. of China

Fax +86(579)82282595; E-mail: wangxiaoxia@zjnu.cn

Received 31 January 2008; revised 9 May 2008

Abstract: The amidoalkylation of indoles with *N*-(α -benzotriazol-1-ylalkyl)amides was smoothly realized with samarium triiodide as a catalyst to give a series of novel 3-indolyl arylmethanamides with good yields and selectivities.

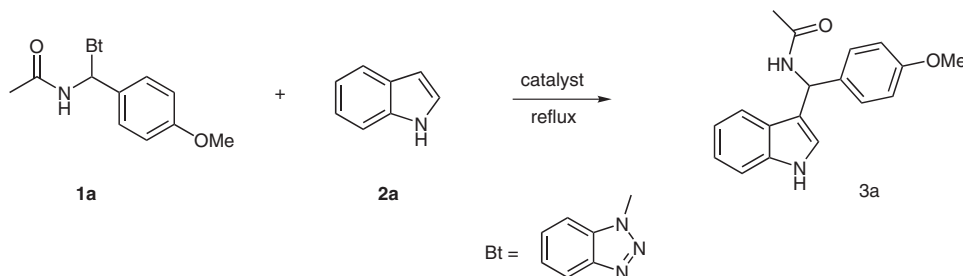
Key words: amidoalkylation, *N*-(α -benzotriazol-1-ylalkyl)amides, indoles, samarium triiodide, catalysis

3-Substituted indoles usually possess numerous biological activities, act as important intermediates for the synthesis of indole-based alkaloids or pharmaceutical compounds, and constitute a key structural unit for many natural products.¹ Among them, the 3-indolyl arylmethanamine derivatives exist in a number of natural and unnatural products with significant biological activities² and have attracted the attention of many synthetic chemists in recent years.

The aza-Friedel–Crafts reaction of indole with imines catalyzed by Lewis acids provide straightforward access to 3-indolylarylmethanamines, however, only moderate yields have been obtained^{3a,b} and the bisindolylmethanes were always formed as a by-product in significant amounts^{3a,b} or even as the sole product.^{3c} Very recently, a three-component aza-Friedel–Crafts reaction^{3d} of aldehyde, amine, and indole in water was developed, catalyzed by carboxylic acid, which provided the 3-indolylarylmethanamine. Despite the mild reaction conditions and avoidance of the bisindolyl compound formation, the amine was limited only to *o*-anisidine. Good yields of the 3-indolylarylmethanamine derivatives as well as high enantioselectivities could be obtained by an alternative method involving the reaction of indoles with

N-protected imines, such as: 1) modification of the imine nitrogen with sulfonyl^{4a–4d} using a chiral copper-complex or organic phosphoric acid as the catalyst; 2) introduction of trifluoromethyl^{4e,f} to the imino nitrogen, catalyzed by boron trifluoride; 3) employment of *N*-alkoxycarbonyl protected α -imino esters^{4g,h} with a copper complex or bi-phenylphosphine–palladium(II) complex as the catalyst. In addition, treatment of 3-lithio-1-TBDMS-indole with aromatic tosylaldimines offered another approach to the functionalized 3-indolylarylmethanamines.⁵ The ald-imines may also be transformed into *N*-acyl iminium salts in the presence of acyl chlorides,^{6a,b} which has realized the amidoalkylation of indoles smoothly without any catalyst. Although the method seems simple and practical, *N*-acyl iminium salts are generally hygroscopic and sensitive to hydrolysis^{6a} and must be prepared under extreme anhydrous conditions, along with the involvement of environmentally toxic media. The aminoalkylbenzotriazoles have been used to alkylate the 3-position of indoles to afford 3-indolylmethylethylamines in the presence of such Lewis acids as AlCl₃ or ZnCl₂.^{6c} Unfortunately, the method was limited to the preparation of compounds without aryl substitution on the methylene. Herein we wish to report the synthesis of 3-indolylarylmethanamides by amidoalkylation of indoles with *N*-(α -benzotriazol-1-ylalkyl)amides.

N-(α -Benzotriazol-1-ylalkyl)amides are readily accessible amidoalkylation reagents, which are prepared from the three-component condensation of a primary (or secondary) amide, benzotriazole and an aldehyde.⁷ The reagents have been used for amidoalkylation of many compounds, such as active methylene compounds,^{7a,8} alcohols and thiols,⁹ active aromatic compounds¹⁰ and amines.¹¹ In the



Scheme 1

SYNTHESIS 2008, No. 16, pp 2582–2588

Advanced online publication: 24.07.2008

DOI: 10.1055/s-2008-1067192; Art ID: F02708SS

© Georg Thieme Verlag Stuttgart · New York

Table 1 Lewis Acid Catalyzed Friedel–Crafts Amidoalkylation of Indoles with **1a**^a

Entry	Lewis acid	Catalyst (mol%)	Solvent	Time (h)	Yield (%) ^b
1	ZnCl ₂	100	THF	20	— ^c
2	AlCl ₃	100	THF	20	— ^c
3	FeCl ₃	100	THF	20	— ^c
4	SmI ₃	50	THF	16	85
5	SmI ₃	20	THF	16	84
6	SmI ₃	20	MeCN	16	56
7	SmI ₃	20	CH ₂ Cl ₂	16	63
8	SmI ₃	20	Toluene	16	34

^a All reactions were carried out under reflux unless otherwise specified.

^b Isolated yield.

^c No reaction.

process of the amidoalkylation reaction, *N*-(α -benzotriazol-1-ylalkyl)amides furnish the azomethine electrophile by acid-promoted elimination of the benzotriazolyl group, which reacts with nucleophiles.¹² In this paper, we present the results of our studies on the reaction between *N*-(α -benzotriazol-1-ylalkyl)amides and indoles catalyzed by samarium triiodide, providing 3-indolylarylmethanamine derivatives in good yields.

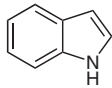
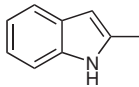
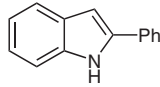
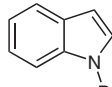
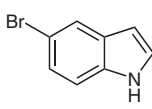
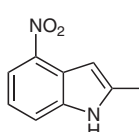
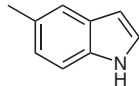
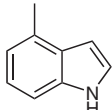
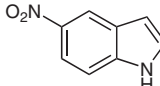
The reaction of *N*-[(α -benzotriazol-1-yl)(4-methoxyphenyl)methyl]acetamide (**1a**) with indole catalyzed by Lewis acids was initially performed in order to optimize the reaction conditions (Scheme 1).

A series of Lewis acids were screened to promote the reaction shown in Scheme 1, and the results are listed in Table 1. Surprisingly, commonly used and less expensive Lewis acids (ZnCl₂, AlCl₃ and FeCl₃) tested in refluxing tetrahydrofuran showed no catalytic activity here; however, the employment of 50 mol% of samarium triiodide (SmI₃) afforded the desired product **3a** in 85% yield (Table 1, entry 4). Reducing the catalyst load to 20 mol% did not decrease the reaction efficiency and afforded a comparable yield in the same period of time (Table 1, entry 5).

The effect of the solvent was also investigated; the reaction of **1a** and **2a** catalyzed by samarium triiodide (20 mol%) was performed in tetrahydrofuran, acetonitrile, dichloromethane and toluene, respectively (Table 1, entries 5–8). Product **3a** could be obtained in 34–84% yields in 16 hours. The reaction carried out in tetrahydrofuran afforded the highest yield (Table 1, entry 5), thus, reflux in THF with 20 mol% of SmI₃ catalyst was established as the optimized reaction conditions.

A variety of indoles and *N*-(α -benzotriazol-1-ylalkyl)amides reacted under the established conditions, demonstrating the generality of the method for the prepa-

Table 2 Samarium Triiodide Catalyzed Friedel–Crafts Amidoalkylation of Indoles with *N*-(α -Benzotriazol-1-ylalkyl)amides

Entry	R ¹	R ²	Indole	Product	Time (h)	Yield (%) ^a
1	Me	4-MeOC ₆ H ₄		3a	16	84
2	Me	Ph	2a	3b	16	86
3	Ph	Ph	2a	3c	20	73
4	Ph	4-O ₂ NC ₆ H ₄	2a	3d	16	87
5	Me	4-MeOC ₆ H ₄		3e	16	75
6	Me	Ph	2b	3f	20	80
7	Me	4-MeOC ₆ H ₄		3g	20	70
8	Me	Ph	2c	3h	24	76
9	Me	4-MeOC ₆ H ₄		3i	20	72
10	Me	Ph	2d	3j	20	75
11	Ph	Ph		3k	20	63
12	Me	Ph	2e	3l	20	65
13	Ph	Ph		— ^b	20	—
14	Ph	4-O ₂ NC ₆ H ₄		3m	20	76
15	Me	4-MeOC ₆ H ₄	2g	3n	20	68
16	Me	4-MeOC ₆ H ₄		3o	16	78
17	Me	4-MeOC ₆ H ₄		— ^b	20	—

^a Yield of pure isolated product.

^b No reaction.

ration of 3-indolylarylmethanamine derivatives (Scheme 2). The results are listed in Table 2.

As can be seen, unsubstituted indole **2a** underwent 3-amidoalkylation smoothly to afford **3a–d** in 73–87% yields (Table 2, entries 1–4). 2-Methylindole also showed good reactivity, leading to **3e** and **3f** in 75% and 80% yield, respectively, in 16–20 hours (Table 2, entries 5 and 6). However, with a bulky group such as phenyl at the 2-position of the indole, the reaction required longer times to afford satisfactory yields (Table 2, entries 7 and 8). Longer reaction times were also necessary when *N*-benzylindole was used (Table 2, entries 9 and 10). The electronic effect of the substituent on the indole also exerted influence on the amidoalkylation reaction; with electron-donating groups such as methyl, the reaction gave 68–78% yields (Table 2, entries 14, 15 and 16), while with electron-withdrawing groups such as bromo, it afforded lower yields (Table 2, entries 11 and 12). Although the reaction seems general, we did encounter two failures: substrates **2f** and **2i** with a strongly electron-withdrawing nitro group located at the 4- or 5-position of the indole failed to produce the expected amidoalkylation products and no reaction occurred (Table 2, entries 13 and 17). In all the cases, the R¹ group in amidoalkylation reagents **1** could be either methyl or phenyl, and R² should be an aryl. When R² is an alkyl such as propyl, the amidoalkylation could not occur and the starting materials were recovered almost quantitatively.

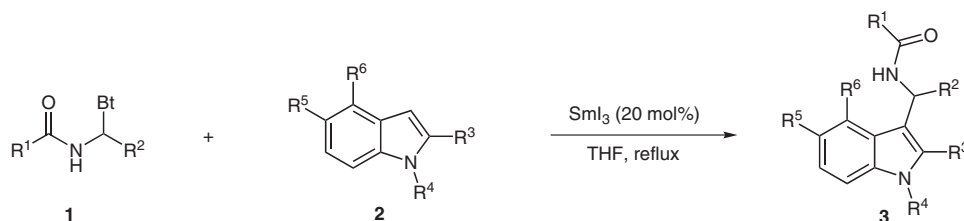
To extend the application scope of the reaction, methyl 1-benzotriazol-1-yl phenylmethanamate (**4**) was reacted with indole under the same conditions (Scheme 3). Unfor-

tunately, instead of the desired amidoalkylation product **5**, the bisindole compound **6** was formed (determined by comparison with an authentic sample).

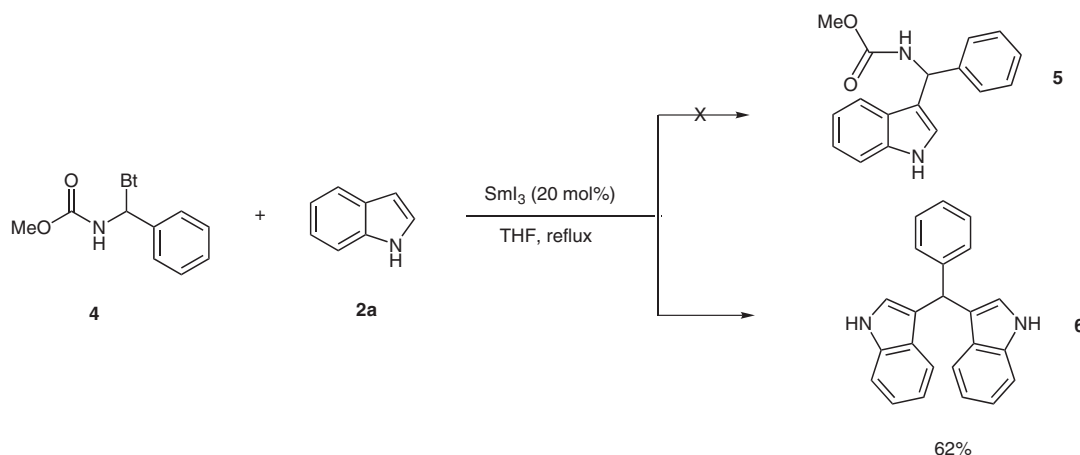
A control experiment in the absence of indole showed that decomposition of **4** took place under the reaction conditions in one hour (benzaldehyde and benzotriazole were detected). Apparently, compound **4** underwent samarium triiodide catalyzed decomposition in preference to amidoalkylation of the indole. It was reported that indoles were prone to dimerization with aldehydes or ketones in the presence of a Lewis acid.¹³ Therefore, the formation of **6** could be ascribed to the reaction of indole with the in situ produced benzaldehyde.

A possible mechanism may also be proposed for the formation of **3**. According to previous reports that, assisted by Lewis acids, benzotriazole adducts could form a benzotriazole anion and the corresponding carbocation that could then react with nucleophiles,^{10,14} the samarium triiodide catalyzed amidoalkylation of indoles with *N*-(α -benzotriazol-1-ylalkyl)amides could be described as shown in Scheme 4. It should be mentioned that when R² is an aryl group, loss of the benzotriazole anion involves the cleavage of a more active benzylic C–N bond and could facilitate the amidoalkylation process. However, in cases where R² is an alkyl group such as propyl, breaking the C–N bond becomes quite difficult even with the assistance of catalyst, therefore no reaction could be observed.

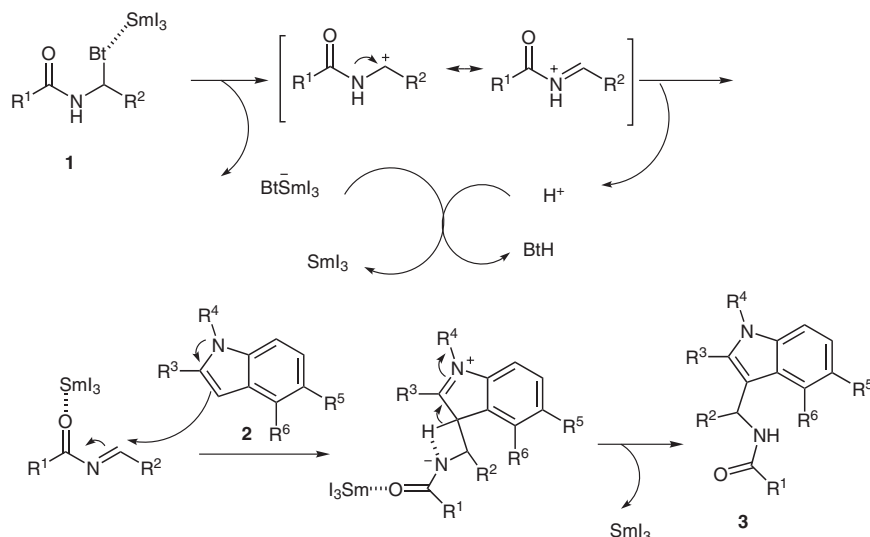
In conclusion, we have developed a samarium triiodide catalyzed amidoalkylation of indoles with *N*-(α -benzotriazol-1-ylalkyl)amides, thus providing a facile and clean synthesis of 3-indolyl arylmethanamides.



Scheme 2



Scheme 3



Scheme 4

THF was distilled from sodium-benzophenone immediately prior to use. ^1H NMR (400 MHz) spectra were recorded on a Bruker AV400 NMR instrument as DMSO or CDCl_3 solutions using TMS as internal standard. Chemical shifts (δ) are reported in ppm and coupling constants (J) are given in Hz. IR spectra were recorded as films or using KBr disks with a NEXUS 670 FTIR spectrometer. Mass spectra were performed on a Thermo Finnigan GC-MS instrument (Trace DSQ). Elemental analyses were performed on a Vario-ELIII instrument. *N*-(α -Benzotriazol-1-ylalkyl)amides **1** were easily prepared from benzotriazole, aldehydes and an amide in toluene by a previously described method.⁷ Methyl 1-benzotriazol-1-yl phenylmethylcarbamate (**4**) was prepared according to the general procedure for the preparation of benzyloxycarbonylamino-1-(1-benzotriazolyl)alkanes.¹⁵ Petroleum ether (PE), where used, had a boiling range of 60–90 °C.

Friedel–Crafts Amidoalkylation of Indoles with *N*-(α -Benzotriazol-1-ylalkyl)amides Catalyzed by SmI_3 ; General Procedure

To samarium powder (0.03 g, 0.2 mmol) in a flask was added anhydrous THF (10 mL) and I_2 (0.076 g, 0.3 mmol) and the mixture was stirred at r.t. for 1 h. To the SmI_3 –THF suspension thus prepared was added *N*-(α -benzotriazol-1-ylalkyl)amide **1** (1.0 mmol) and indole **2** (1.0 mmol). The reaction mixture was stirred until the disappearance of **1** was observed, then the reaction was quenched with aq HCl (0.1 M, 3 mL) and extracted with EtOAc (3 \times 30 mL). The combined organic extracts were washed with sat. $\text{Na}_2\text{S}_2\text{O}_3$ (10 mL), sat. Na_2CO_3 (10 mL), then with brine (2 \times 10 mL) and were dried over anhydrous Na_2SO_4 . Removal of the solvent under reduced pressure afforded the crude product, which was purified by column chromatography (EtOAc–PE, 1:2) to give the desired product **3**.

N-[(1*H*-Indol-3-yl)(4-methoxyphenyl)methyl]acetamide (**3a**)

White powder; mp 162–165 °C.

IR (KBr): 3370, 3300, 1642, 1513, 1400, 743 cm^{-1} .

^1H NMR (400 Hz, CDCl_3): δ = 2.03 (s, 3 H), 3.80 (s, 3 H), 6.12 (d, J = 8.0 Hz, 1 H), 6.46 (d, J = 8.0 Hz, 1 H), 6.77 (s, 1 H), 6.86 (d, J = 8.0 Hz, 2 H), 7.07 (t, J = 8.0 Hz, 1 H), 7.19 (t, J = 8.0 Hz, 1 H), 7.25–7.29 (m, 1 H), 7.34 (d, J = 8.0 Hz, 1 H), 7.44 (d, J = 8.0 Hz, 1 H), 8.24 (br s, 1 H).

^{13}C NMR (100 Hz, CDCl_3): δ = 23.1, 49.0, 55.5, 112.0, 114.0, 117.4, 119.0, 119.4, 121.7, 123.8, 126.4, 128.7, 135.5, 137.1, 158.5, 168.6.

MS (EI, 70 eV): m/z (%) = 295 (5.4) $[\text{M} + \text{H}]^+$, 177 (99.2) $[\text{M} - \text{indole}]^+$, 162 (100) $[177 - \text{CH}_3]^+$, 134 (99.5), 107 (29.7), 92 (46.3), 77 (57.1).

Anal. Calcd for $\text{C}_{18}\text{H}_{18}\text{N}_2\text{O}_2$: C, 73.45; H, 6.16; N, 9.52. Found: C, 73.63; H, 5.90; N, 9.63.

N-[(1*H*-Indol-3-yl)(phenyl)methyl]acetamide (**3b**)

White powder; mp 176–179 °C.

IR (KBr): 3407, 3381, 3305, 1646, 1538, 742 cm^{-1} .

^1H NMR (400 Hz, CDCl_3): δ = 2.01 (s, 3 H), 6.13–6.22 (m, 2 H), 6.52–6.71 (m, 1 H), 6.69 (s, 1 H), 7.07–7.11 (m, 1 H), 7.19–7.34 (m, 5 H), 7.44 (m, 1 H), 8.41 (s, 1 H).

^{13}C NMR (100 Hz, CDCl_3): δ = 23.4, 50.4, 111.5, 117.2, 119.3, 119.9, 122.5, 123.6, 125.9, 127.0, 127.2, 128.5, 136.7, 141.3, 169.2.

MS (EI, 70 eV): m/z (%) = 265 (75.1) $[\text{M} + \text{H}]^+$, 264 (100) $[\text{M}]^+$, 222 (98.9) $[\text{M} + \text{H} - \text{CH}_3\text{CO}]^+$, 205 (99.1), 145 (52.5), 118 (98.3), 104 (90.7), 77 (30.0).

Anal. Calcd for $\text{C}_{17}\text{H}_{16}\text{N}_2\text{O}$: C, 77.25; H, 6.10; N, 10.60. Found: C, 77.47; H, 5.87; N, 10.43.

N-[(1*H*-Indol-3-yl)(phenyl)methyl]benzamide (**3c**)

White powder; mp 178–180 °C.

IR (KBr): 3410, 3299, 3309, 1642, 1524, 701 cm^{-1} .

^1H NMR (400 Hz, $\text{DMSO}-d_6$): δ = 6.63 (d, J = 8.0 Hz, 1 H), 6.88 (s, 1 H), 6.97 (d, J = 8.0 Hz, 1 H), 7.07–7.10 (t, J = 8.0 Hz, 1 H), 7.28–7.53 (m, 10 H), 7.93 (d, J = 8.0 Hz, 2 H), 9.19 (d, J = 8.0 Hz, 1 H), 11.0 (s, 1 H).

^{13}C NMR (100 Hz, $\text{DMSO}-d_6$): δ = 50.3, 112.1, 116.9, 119.2, 121.7, 124.4, 126.6, 127.3, 127.8, 128.0, 128.1, 128.6, 128.8, 131.6, 135.1, 137.0, 143.3, 166.2.

MS (EI, 70 eV): m/z (%) = 207 (16.3) $[\text{M} - \text{indolyl}]^+$, 206 (100) $[\text{M} - \text{PhCONH}]^+$, 105 (32.0), 91 (38.2).

Anal. Calcd for $\text{C}_{22}\text{H}_{18}\text{N}_2\text{O}$: C, 80.96; H, 5.56; N, 8.58. Found: C, 81.25; H, 5.68; N, 8.66.

N-[(1*H*-Indol-3-yl)(4-nitrophenyl)methyl]benzamide (**3d**)

White powder; mp 107–110 °C.

IR (KBr): 3415, 3299, 1642, 1520, 744 cm^{-1} .

^1H NMR (400 Hz, DMSO- d_6): δ = 6.68–6.73 (m, 1 H), 6.91–6.98 (m, 2 H), 7.06–7.10 (m, 1 H), 7.43–7.51 (m, 4 H), 7.76 (d, J = 8.0 Hz, 2 H), 7.90 (d, J = 8.0 Hz, 2 H), 8.22 (d, J = 8.0 Hz, 2 H), 9.33 (d, J = 8.0 Hz, 1 H), 11.00 (s, 1 H).

^{13}C NMR (100 Hz, DMSO- d_6): δ = 50.2, 112.2, 115.5, 119.0, 119.4, 122.0, 124.0, 124.7, 126.4, 128.1, 128.7, 129.0, 131.8, 134.7, 137.0, 147.0, 151.1, 166.5.

MS (EI, 70 eV): m/z (%) = 207 (100) $[\text{M} + \text{H} - \text{NO}_2 - \text{PhCONH}_2]^+$, 178 (16.5), 129 (26.4), 117 (99.2), 91 (96.3), 77 (61.1).

Anal. Calcd for $\text{C}_{22}\text{H}_{17}\text{N}_3\text{O}_3$: C, 71.15; H, 4.61; N, 11.31. Found: C, 70.99; H, 4.46; N, 11.15.

***N*-(4-Methoxyphenyl)(2-methyl-1*H*-indol-3-yl)methyl]acetamide (3e)**

White powder; mp 160–163 °C.

IR (KBr): 3405, 3268, 3058, 1630, 1544, 696 cm^{-1} .

^1H NMR (400 Hz, DMSO- d_6): δ = 1.92 (s, 3 H), 2.34 (s, 3 H), 3.35 (s, 3 H), 6.33 (d, J = 8.0 Hz, 1 H), 6.85–6.93 (m, 2 H), 7.18–7.37 (m, 6 H), 8.63 (s, 1 H), 10.80 (s, 1 H).

^{13}C NMR (100 Hz, DMSO- d_6): δ = 21.8, 23.1, 40.0, 49.6, 111.7, 116.5, 118.8, 123.3, 124.0, 126.6, 127.1, 127.5, 128.5, 135.4, 143.6, 168.7.

MS (EI, 70 eV): m/z (%) = 218 (30.1) $[\text{M} - \text{H} - \text{OCH}_3 - \text{NHCOCH}_3]^+$, 207 (78.9), 117 (96.3), 91 (100) $[\text{PhCH}_2]^+$, 77 (22.6).

Anal. Calcd for $\text{C}_{19}\text{H}_{20}\text{N}_2\text{O}_2$: C, 74.00; H, 6.54; N, 9.08. Found: C, 74.13; H, 6.40; N, 9.17.

***N*-(2-Methyl-1*H*-indol-3-yl)(phenyl)methyl]acetamide (3f)**

White powder; mp 160–163 °C.

IR (KBr): 3382, 3324, 3057, 1639, 1536, 744 cm^{-1} .

^1H NMR (400 Hz, CDCl_3): δ = 2.05 (s, 3 H), 2.37 (s, 3 H), 6.25 (d, J = 8.0 Hz, 1 H), 6.54 (d, J = 8.0 Hz, 1 H), 6.92–6.96 (m, 1 H), 7.09 (d, J = 8.0 Hz, 2 H), 7.21–7.30 (m, 5 H), 8.23 (s, 1 H).

^{13}C NMR (100 Hz, CDCl_3): δ = 12.0, 23.4, 49.2, 110.7, 111.5, 118.5, 119.6, 121.3, 126.6, 126.9, 128.4, 129.1, 133.3, 135.5, 141.2, 169.5.

MS (EI, 70 eV): m/z (%) = 279 (30.6) $[\text{M} + \text{H}]^+$, 278 (100) $[\text{M}]^+$, 235 (40.5), 220 (95.2), 204 (31.7), 148 (99.3), 132 (45.0), 106 (94.3).

Anal. Calcd for $\text{C}_{18}\text{H}_{18}\text{N}_2\text{O}$: C, 77.67; H, 6.52; N, 10.06. Found: C, 77.48; H, 6.64; N, 9.89.

***N*-(4-Methoxyphenyl)(2-phenyl-1*H*-indol-3-yl)methyl]acetamide (3g)**

White powder; mp 200–202 °C.

IR (KBr): 3421, 3270, 1662, 1510, 747 cm^{-1} .

^1H NMR (400 Hz, CDCl_3): δ = 2.03 (s, 3 H), 3.77 (s, 3 H), 6.30 (d, J = 8.0 Hz, 1 H), 6.65 (d, J = 8.0 Hz, 1 H), 6.80 (d, J = 8.0 Hz, 2 H), 7.02 (t, J = 8.0 Hz, 1 H), 7.18–7.26 (m, 3 H), 7.38–7.46 (m, 4 H), 7.57 (d, J = 8.0 Hz, 2 H), 8.30 (s, 1 H).

^{13}C NMR (100 Hz, CDCl_3): δ = 23.4, 49.3, 55.3, 111.4, 112.0, 113.8, 119.7, 120.1, 122.5, 126.9, 128.0, 128.48, 128.53, 129.1, 132.2, 133.6, 136.2, 136.5, 158.5, 169.4.

MS (EI, 70 eV): m/z (%) = 193 (99.5) $[\text{2-phenyl-3-indolyl}]^+$, 163 (68.8), 115 (26.4), 92 (98.9), 77 (29.7), 71 (69.3).

Anal. Calcd for $\text{C}_{24}\text{H}_{22}\text{N}_2\text{O}_2$: C, 77.81; H, 5.99; N, 7.56. Found: C, 77.65; H, 6.12; N, 7.38.

***N*-(Phenyl(2-phenyl-1*H*-indol-3-yl)methyl]acetamide (3h)**

White powder; mp 250 °C (dec.).

IR (KBr): 3418, 3269, 3060, 1657, 1495, 743 cm^{-1} .

^1H NMR (400 Hz, CDCl_3): δ = 2.07 (s, 3 H), 6.30 (s, 1 H), 6.70 (d, J = 8.0 Hz, 1 H), 7.01 (t, J = 8.0 Hz, 1 H), 7.19–7.31 (m, 6 H), 7.40–7.49 (m, 4 H), 7.59 (d, J = 8.0 Hz, 2 H), 8.25 (s, 1 H).

^{13}C NMR (100 Hz, CDCl_3): δ = 22.9, 49.3, 111.8, 112.6, 119.2, 120.9, 121.9, 126.9, 127.1, 127.2, 128.3, 128.6, 129.0, 129.2, 132.9, 136.3, 136.8, 142.9, 169.2.

MS (EI, 70 eV): m/z (%) = 220 (100) $[\text{M} - \text{Ph} - \text{COCH}_3]^+$, 205 (99.8), 177 (95.3), 145 (98.3), 105 (66.8), 57 (99.7).

Anal. Calcd for $\text{C}_{23}\text{H}_{20}\text{N}_2\text{O}$: C, 81.15; H, 5.92; N, 8.23. Found: C, 80.96; H, 6.07; N, 8.32.

***N*-(1-Benzyl-1*H*-indol-3-yl)(4-methoxyphenyl)methyl]acetamide (3i)**

White powder; mp 174–176 °C.

IR (KBr): 3327, 3131, 1693, 1543, 753 cm^{-1} .

^1H NMR (400 Hz, DMSO- d_6): δ = 1.87 (s, 3 H), 3.72 (s, 3 H), 5.34 (s, 2 H), 6.29 (d, J = 8.0 Hz, 1 H), 7.39–6.87 (m, 14 H), 8.62 (d, J = 8.0 Hz, 1 H).

^{13}C NMR (100 Hz, DMSO- d_6): δ = 23.1, 48.9, 49.4, 55.5, 110.7, 114.0, 117.3, 119.4, 119.7, 122.0, 127.0, 127.4, 127.6, 127.8, 128.7, 129.0, 135.2, 136.9, 138.8, 158.5, 168.6.

MS (EI, 70 eV): m/z (%) = 327 (3.1) $[\text{M} + \text{H} - \text{NHCOCH}_3]^+$, 221 (98.3), 205 (100) $[\text{M} + \text{H} - \text{PhCH}_2 - \text{OCH}_3 - \text{CH}_3\text{CONH}]^+$, 177 (66.4), 145 (75.9), 119 (99.1), 91 (99.7), 64 (99.5).

Anal. Calcd for $\text{C}_{25}\text{H}_{24}\text{N}_2\text{O}_2$: C, 78.10; H, 6.29; N, 7.29. Found: C, 78.32; H, 6.16; N, 7.17.

***N*-(1-Benzyl-1*H*-indol-3-yl)(phenyl)methyl]acetamide (3j)**

White powder; mp 181–184 °C.

IR (KBr): 3300, 3059, 1641, 1538, 741 cm^{-1} .

^1H NMR (400 Hz, CDCl_3): δ = 2.05 (s, 3 H), 5.23 (s, 2 H), 6.10 (d, J = 8.0 Hz, 1 H), 6.54 (d, J = 8.0 Hz, 1 H), 6.74 (s, 1 H), 7.08 (t, J = 6.8 Hz, 3 H), 7.19 (s, 1 H), 7.25–7.29 (m, 5 H), 7.32–7.40 (m, 4 H), 7.48 (d, J = 8.0 Hz, 1 H).

^{13}C NMR (100 Hz, CDCl_3): δ = 23.4, 50.1, 50.3, 110.1, 116.5, 119.6, 119.8, 122.4, 126.7, 127.0, 127.3, 127.55, 127.59, 127.7, 128.5, 128.8, 137.1, 137.3, 141.3, 169.2.

MS (EI, 70 eV): m/z (%) = 341 (100) $[\text{M} + 2\text{H} - \text{CH}_3]^+$, 298 (51.3), 204 (73.1), 117 (34.6), 104 (70.9), 91 (18.3).

Anal. Calcd for $\text{C}_{24}\text{H}_{22}\text{N}_2\text{O}$: C, 81.33; H, 6.26; N, 7.90. Found: C, 81.54; H, 6.32; N, 7.79.

***N*-(5-Bromo-1*H*-indol-3-yl)(phenyl)methyl]benzamide (3k)**

White powder; mp 179–181 °C.

IR (KBr): 3310, 3127, 1656, 1510, 826 cm^{-1} .

^1H NMR (400 Hz, CDCl_3): δ = 6.64 (d, J = 8.0 Hz, 1 H), 6.73 (d, J = 7.6 Hz, 1 H), 6.81 (s, 1 H), 7.21–7.52 (m, 8 H), 7.62 (s, 1 H), 7.79–7.84 (m, 3 H), 8.36 (s, 1 H).

^{13}C NMR (100 Hz, CDCl_3): δ = 50.8, 112.9, 113.3, 116.9, 121.9, 124.9, 125.5, 125.9, 127.0, 127.1, 127.2, 127.6, 128.7, 131.7, 134.2, 135.4, 140.8, 166.7.

MS (EI, 70 eV): m/z (%) = 220 (67.4), 205 (100) $[\text{M} - \text{Br} - \text{PhCONH}]^+$, 177 (30.7), 145 (32.0), 115 (15.8), 105 (21.1), 81 (21.1).

Anal. Calcd for $\text{C}_{22}\text{H}_{17}\text{BrN}_2\text{O}$: C, 65.20; H, 4.23; N, 6.91. Found: C, 65.53; H, 4.21; N, 6.81.

***N*-(5-Bromo-1*H*-indol-3-yl)(phenyl)methyl]acetamide (3l)**

White powder; mp 140–143 °C.

IR (KBr): 3298, 3122, 1668, 1517, 1401, 750 cm^{-1} .

^1H NMR (400 Hz, DMSO- d_6): δ = 1.90 (s, 3 H), 6.32 (d, J = 8.0 Hz, 1 H), 7.00 (s, 1 H), 7.17–7.27 (m, 2 H), 7.34–7.36 (m, 5 H), 7.49 (s, 1 H), 8.68 (d, J = 8.0 Hz, 1 H), 11.2 (s, 1 H).

^{13}C NMR (100 Hz, DMSO- d_6): δ = 23.0, 49.3, 111.7, 114.1, 116.8, 121.6, 124.2, 125.7, 127.2, 127.5, 128.1, 128.7, 135.7, 143.0, 168.8.

MS (EI, 70 eV): m/z (%) = 342 (2.3), 220 (64.1), 205 (100) [$\text{M} - \text{Br} - \text{PhCONH}$] $^+$, 177 (20.9), 145 (25.0), 115 (8.1), 105 (15.5), 81 (16.7), 57 (42.8).

Anal. Calcd for $\text{C}_{17}\text{H}_{15}\text{BrN}_2\text{O}$: C, 59.49; H, 4.41; N, 8.16. Found: C, 59.28; H, 4.52; N, 8.02.

***N*-[5-Methyl-1*H*-indol-3-yl](4-nitrophenyl)methyl]benzamide (3m)**

Red powder; mp 205–209 °C.

IR (KBr): 3419, 3098, 1641, 1521, 720 cm^{-1} .

^1H NMR (400 Hz, DMSO- d_6): δ = 2.30 (s, 3 H), 6.69 (d, J = 8.0 Hz, 1 H), 6.90 (m, 2 H), 7.26 (d, J = 8.0 Hz, 2 H), 7.43–7.57 (m, 3 H), 7.75 (d, J = 8.0 Hz, 2 H), 7.90–7.96 (m, 2 H), 8.22 (d, J = 8.0 Hz, 2 H), 9.29 (d, J = 8.0 Hz, 1 H), 10.93 (s, 1 H).

^{13}C NMR (100 Hz, DMSO- d_6): δ = 21.8, 50.1, 111.9, 114.9, 118.5, 124.0, 124.8, 126.3, 126.7, 127.9, 128.1, 128.7, 128.9, 131.8, 134.7, 135.3, 146.9, 151.2, 166.5.

MS (EI, 70 eV): m/z (%) = 340 (3.8) [$\text{M} + \text{H} - \text{NO}_2$] $^+$, 250 (100) [$\text{M} - \text{CH}_3 - \text{PhCONH}$] $^+$, 235 (98.2), 220 (98.5), 143 (70.6), 134 (99.1), 117 (99.4).

Anal. Calcd for $\text{C}_{23}\text{H}_{19}\text{N}_3\text{O}_3$: C, 71.67; H, 4.97; N, 10.90. Found: C, 71.78; H, 5.06; N, 10.78.

***N*-(4-Methoxyphenyl)(5-methyl-1*H*-indol-3-yl)methyl]acetamide (3n)**

White powder; mp 194–197 °C.

IR (KBr): 3399, 3285, 2650, 1648, 1511, 745 cm^{-1} .

^1H NMR (400 Hz, DMSO- d_6): δ = 1.89 (s, 3 H), 2.67 (s, 3 H), 3.69 (s, 3 H), 6.29 (d, J = 8.0 Hz, 1 H), 6.77–6.84 (m, 3 H), 6.93 (d, J = 8.0 Hz, 1 H), 7.14–7.22 (m, 4 H), 8.48 (d, J = 8.0 Hz, 1 H), 10.83 (s, 1 H).

^{13}C NMR (100 Hz, DMSO- d_6): δ = 12.2, 23.0, 48.1, 55.5, 110.9, 111.9, 113.8, 118.6, 119.1, 120.4, 127.3, 128.1, 133.2, 135.0, 135.6, 158.3, 168.8.

MS (EI, 70 eV): m/z (%) = 310 (3.2) [$\text{M} + 2\text{H}$], 220 (99.3), 205 (100) [$\text{M} + \text{H} - \text{CH}_3 - \text{CH}_2\text{O} - \text{CH}_2\text{CONH}$] $^+$, 177 (44.7), 161 (16.1), 145 (52.5), 115 (9.8), 105 (32.6), 57 (98.9).

Anal. Calcd for $\text{C}_{19}\text{H}_{20}\text{N}_2\text{O}_2$: C, 74.00; H, 6.54; N, 9.08. Found: C, 74.26; H, 6.65; N, 8.86.

***N*-(4-Methoxyphenyl)(4-methyl-1*H*-indol-3-yl)methyl]acetamide (3o)**

White powder; mp 199–202 °C.

IR (KBr): 3401, 3380, 3003, 2955, 2838, 1652, 1511, 1253, 748 cm^{-1} .

^1H NMR (400 Hz, CDCl_3): δ = 2.02 (s, 3 H), 2.58 (s, 3 H), 3.79 (s, 3 H), 6.09 (d, J = 8.0 Hz, 1 H), 6.59 (d, J = 6.8 Hz, 2 H), 6.85–6.88 (m, 3 H), 7.07–7.11 (m, 1 H), 7.18 (d, J = 8.0 Hz, 1 H), 7.25 (d, J = 6.8 Hz, 2 H), 8.24 (s, 1 H).

^{13}C NMR (100 Hz, CDCl_3): δ = 20.1, 23.4, 50.8, 55.3, 109.2, 113.8, 118.4, 121.7, 122.6, 124.3, 124.9, 128.1, 130.9, 134.4, 137.2, 158.7, 168.6.

MS (EI, 70 eV): m/z (%) = 220 (91.8), 205 (100) [$\text{M} + \text{H} - \text{CH}_3 - \text{CH}_2\text{O} - \text{CH}_2\text{CONH}$] $^+$, 177 (51.2), 145 (36.6), 117 (20.0), 57 (96.3).

Anal. Calcd for $\text{C}_{19}\text{H}_{20}\text{N}_2\text{O}_2$: C, 74.00; H, 6.54; N, 9.08. Found: C, 74.19; H, 6.37; N, 9.15.

Acknowledgment

We thank the Department of Science and Technology, Zhejiang Province (Project No. 2006C11262) for financial support.

References

- (1) (a) Sundberg, R. J. *The Chemistry of Indoles*; Academic Press: New York, **1970**, 105. (b) Sundberg, R. J. *Indoles*; Academic Press: San Diego, **1996**, 113. (c) Saxton, J. E. *Nat. Prod. Rep.* **1997**, *14*, 559. (d) Lounasmaa, M.; Tolvanen, A. *Nat. Prod. Rep.* **2000**, *17*, 175. (e) Somei, M.; Yamada, F. *Nat. Prod. Rep.* **2005**, *22*, 73. (f) Rajeswaran, W. G.; Labroo, R. B.; Cohen, L. A.; King, M. M. *J. Org. Chem.* **1999**, *64*, 1369. (g) Jiang, B.; Yang, C.-G.; Wang, J. *J. Org. Chem.* **2001**, *66*, 4865. (h) Wang, T.; Cook, J. M. *Org. Lett.* **2000**, *2*, 2057. (i) Chataigner, I.; Hess, E.; Toupet, L.; Piettre, S. R. *Org. Lett.* **2001**, *3*, 515. (j) Molina, P.; Alcantara, J.; Lopez-Leonardo, C. *Tetrahedron* **1996**, *52*, 5833.
- (2) (a) Somei, M.; Yamada, F. *Nat. Prod. Rep.* **2004**, *21*, 278. (b) Faulkner, D. J. *Nat. Prod. Rep.* **2002**, *19*, 1. (c) Bosch, J.; Bennasar, M.-L. *Synlett* **1995**, 587.
- (3) (a) Mi, X. L.; Luo, S. Z.; He, J. Q.; Cheng, J. P. *Tetrahedron Lett.* **2004**, *45*, 4567. (b) Xie, W.; Bloomfield, K. M.; Jin, Y.; Dolney, N. Y.; Wang, P. G. *Synlett* **1999**, 498. (c) Ke, B.; Qin, Y.; He, Q.; Huang, Z.; Wang, F. *Tetrahedron Lett.* **2005**, *46*, 1751. (d) Shirakawa, S.; Kobayashi, S. *Org. Lett.* **2006**, *8*, 4939.
- (4) (a) Wang, Y.-Q.; Song, J.; Hong, R.; Li, H.; Deng, L. *J. Am. Chem. Soc.* **2006**, *128*, 8156. (b) Jia, Y.; Xie, J.; Duan, H.; Wang, L.; Zhou, Q. *Org. Lett.* **2006**, *8*, 1621. (c) Kang, Q.; Zhao, Z. A.; You, S. L. *J. Am. Chem. Soc.* **2007**, *129*, 1484. (d) Esquivias, J.; Arrayás, R. G.; Carretero, J. C. *Angew. Chem. Int. Ed.* **2006**, *45*, 629. (e) Gong, Y.; Kato, K.; Kimoto, H. *Synlett* **2000**, 1058. (f) Gong, Y. *Tetrahedron: Asymmetry* **2001**, *12*, 2121. (g) Johannsen, M. *Chem. Commun.* **1999**, 2233. (h) Hao, J.; Taktak, S.; Aikawa, K.; Yusa, Y.; Hatano, M.; Mikami, K. *Synlett* **2001**, 1443.
- (5) Wynne, J. H.; Stalick, W. M. *J. Org. Chem.* **2002**, *67*, 5850.
- (6) (a) Venkov, A. P.; Statkova-Abeghe, S. M.; Donova, A. K. *Cent. Eur. J. Chem.* **2004**, *2*, 234. (b) Sheinkman, A. K.; Kucherenko, A. P. *Chem. Heterocycl. Compd.* **1974**, *9*, 1296. (c) Katritzky, A. R.; Yang, Z.; Lam, J. N. *Tetrahedron* **1992**, *48*, 4971.
- (7) (a) Katritzky, A. R.; Drewniak, M. *J. Chem. Soc., Perkin Trans. I* **1988**, 2339. (b) Katritzky, A. R.; Pernak, J.; Fan, W. Q.; Saczewski, F. *J. Org. Chem.* **1991**, *56*, 4439. (c) Katritzky, A. R.; Drewniak, M. *Tetrahedron Lett.* **1988**, *29*, 1755.
- (8) Wang, X. X.; Mao, H.; Liu, J. H.; Du, J. X. *Synth. Commun.* **2007**, *37*, 3751.
- (9) (a) Katritzky, A. R.; Ignatchenko, A. V.; Lang, H. *J. Org. Chem.* **1995**, *60*, 4002. (b) Katritzky, A. R.; Fan, W.-Q.; Black, M.; Pernak, J. *J. Org. Chem.* **1992**, *57*, 547. (c) Katritzky, A. R.; Takahashi, I.; Fan, W.-Q.; Pernak, J. *Synthesis* **1991**, 1147.
- (10) Katritzky, A. R.; Pernak, J.; Fan, W. Q. *Synthesis* **1991**, 868.
- (11) Katritzky, A. R.; Urogdi, L.; Mayence, A. *J. Org. Chem.* **1990**, *55*, 2206.
- (12) Katritzky, A. R.; Manju, K.; Singh, S. K.; Meher, N. K. *Tetrahedron* **2005**, *61*, 2555.

- (13) (a) Zhang, Z. H.; Yin, L.; Wang, Y. M. *Synthesis* **2005**, 1949. (b) Du, D. M.; Meng, S. M.; Wang, Y. M.; Meng, J. B.; Zhou, X. Z. *Chin. J. Chem.* **1995**, *13*, 520. (c) Babu, G.; Sridhar, N.; Perumal, P. T. *Synth. Commun.* **2000**, *30*, 1609. (d) Nagarajan, R.; Perumal, P. T. *Tetrahedron* **2002**, *58*, 1229.
- (14) Katritzky, A. R.; Yao, G.; Lan, X.; Zhao, X. *J. Org. Chem.* **1993**, *58*, 2086.
- (15) Katritzky, A. R.; Kirichenko, K.; Elsayed, A. M.; Ji, Y.; Fang, Y.; Steel, P. J. *J. Org. Chem.* **2002**, *67*, 4957.