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Communication

An Organocatalytic Asymmetric Double Michael Cascade Reaction of Unsaturated Ketones and Unsaturated Pyrazolones: Highly Efficient Synthesis of Spiropyrazolones Derivatives[†]

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The first organocatalytic double Michael cascade reaction between unsaturated ketones and unsaturated pyrazolones has been developed which provides spiropyrazolones core 10 structures containing two interval or three consecutive stereogenic centers with excellent diastereo- (>20:1) and enantioselectivities (up to 99% ee). Moreover, a pair of enantiomers 5a and 5a' can be achieved via different catalyst.

The heteroatom-containing spirocyclic subunits are featured in a ¹⁵ number of naturally occurring products as well as biologically, pharmaceutically, and medically active molecules. As a consequence, finding simple and effective synthetic methods to realize functional diversity, architectural complexity, and stereochemical unicity of the core draws increasing attention of ²⁰ organic chemists and a number of excellent achievements have been reported during the past decades especially based on the flourishing development of asymmetric organocatalysis.¹ To date, many efforts are concentrated on the synthesis of chiral spirooxindoles² while other spiroheterocyclic motifs are rarely ²⁵ involved.³

Pyrazol-3-one derivatives, because of their approved antiinflammatory, antiviral, antitumor, and antibacterial properties, exist widely in clinical and listed drugs.⁴ For example, the very common structure of compound 4, which is known as 30 edavarone, is used in the treatment of blood brain barrier injury. Although various synthetic protocols have been achieved on the construction of 4-substitute pyrazolones, there are few reports concerning on the synthesis of chiral pyrazolones with quaternary stereogenic center at the C-4 position and intriguing 35 spiropyrazolones combining of multistereogenic cyclohexanone and pyrozolone motifs. In 2010, Yuan and co-workers reported an enantioselective pyrazolone Michael addition to nitrostyrenes by organocatalysis.⁵ In 2011, Feng's group disclosed a highly enantioselective reaction of pyrazolones and azodicarboxylates ⁴⁰ by metal catalysis.⁶ Shortly after, Rios reported the first example of Michael-Michael-aldol condensation of pyrazolone 4 and unsaturated aldehydes for the preparation of chiral spiropyrazolones.⁷ In 2012, Wang's group reported a double Michael reaction between pyrazolones and divinyl ketones that gave

45 spiropyrazolones with excellent yields and enantioselectivities.⁸

Recently, Rios and co-workers disclosed an organocatalytic enantioselective pyrazol-3-one addition to maleimides catalyzed by bifunctional thiourea catalysts. ⁹ All of the above methodologies employ pyrazolones as nucleophiles. Considering 50 the similar electronic and structural properties between pyrazolone and oxindole and the successful application of unsaturated oxindole, the use of an unsaturated pyrazolone as an electrophile is also promising. However, so far only a single asymmetric catalytic entry currently available to spiropyrazolones 55 was described by Rios and co-workers, consisting of a domino cyclization reaction among aldehyde, unsaturated aldehyde and unsaturated pyrazolone.¹⁰ No results have been reported in the literature for constructing chiral spiropyrazolones using unsaturated ketones and unsaturated pyrazolones. As part of our 60 ongoing research program toward the exploitation of organocatalytic cascade reaction and our interests in seeking enantioselective approach toward the construction of spirocyclic motif,¹¹ we will in this context present the first asymmetric organocatalytic double Michael cascade cycloaddition reaction of 65 unsaturated ketones and unsaturated pyrazolones, providing spiropyrazolones derivatives in moderate yields and excellent diastereo- and enantioselectivities.



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As early as in 2002, Barbas's group firstly introduced dienamine activation mode of unsaturated ketones with reactive hydrogen at C-terminal.¹² They used secondary amines derived from proline to catalyze the DA reaction of unsaturated ketones 5 and nitro olefins with excellent yields and no asymmetric catalysis was referred to. Chen and Melchiorre's group independently¹³developed this concept and proved that quinine or quinidine alkaloid-derived primary amine can act as powerful catalyst to activate these ketones to complete chiral induction. 10 Inspired by these experimental results, we envisaged the following reaction mechanism: Unsaturated ketone 3 should first intercept the primary amine catalyst via HOMO-activation concept to generate the activated dienamine ion intermediate i. The resulting prochiral carbon nucleophile i should be reactive 15 enough to initiate an intermolecular Michael addition to unsaturated pyrazolones 2, and the LUMO-active intermediate ii adduct would further undergo an intramolecular conjugate addition, thus affording the desired spiropyrazolone 5 in a one-pot fashion

20 Table 1. Screening of the Catalysts, Solvents, and Reaction Conditions for the Reactions^[a]



[a] Unless noted, the reaction was performed on 0.1 mmol scale with 2a (1.0 equiv), 3a (2.0 equiv) and catalyst 1a (20 mol%) at rt (17 ° C) for ²⁵ displayed time. [b] Yield of isolated product. [c] Determined by HPLC analysis on a chiral stationary phase. [d] BzOH (40 mol%) was used as cocatalyst. [e] The reaction was performed at 40 °C. [f] The enantiomers 5' of compound 5 was obtained using the same steps while catalyst 1b was employed during the reaction process.

- ³⁰ To identify the validation of our strategy, we began the investigation by testing the model reaction between **2a** and **3a** catalyzed by primary amine **1a** in toluene at rt. However, no product was observed after 120h (Table 1, entry 1). Under the same condition, the using of protic acid BzOH as cocatalyst to the
- ³⁵ reactive system triggered the reaction and provided the desired product **5a** in 87% yield and >99% ee value after 72h (entry 2). Encouraged by this outcome, we further explored **1b** and **1c** as catalysts for this reaction, and the results were shown in Table 1.

- Using catalyst **1b**, we were delighted to obtain spirocyclic ⁴⁰ product **5a'** with comparable ee value, yield and reversed configuration (entry 3). Although an acceptable ee value (93%) was obtained with catalyst **1c**, the reaction needed longer time and afforded the product in lower yield (entry 4). Considering the enantioselectivity as well as yield, catalyst **1a** and **1b** both ⁴⁵ delivered ideal results and a pair of enantiomers **5a** and **5a'** can be achieved via different catalysis. However, in these two cases the reaction was found to be somewhat sluggish, which demanded 72h to completion. Therefore, taking primary amine **1a** as catalyst, the reaction conditions were further optimized. ⁵⁰ Toluene was selected as the optimal solvent in terms of both yield and enantioselectivity (entries 2 and 6-9). The temperature had a pronounced effect on the reaction rate and would shorten the
- reaction time from 72h to 3.5h without sacrificing of either yield or enantioselectivity (entries 2 and 10) through the elevation of ⁵⁵ temperature from rt to 40 °C.

Table 2. Substrate Scope of Cat 1a/ BzOH-Catalyzed Double MichaelAddition of Unsaturated Pyrazolones 2a-j and Unsaturated Ketones 3a-jwith Same Substituents^[a]

(Ph N-N R	+	R -	1a (20 mol%) BzOH (40 mol%) Tol, 40 ⁰ C	Ph O= R	N-N R
	2		3		5	Ö
-	entry	5	R	yield[%] ^b	dr [%] ^c	ee^{d}
-	1	5a	Ph	87	>20:1	>99
	2	5b	$4-CH_3C_6H_4$	75	>20:1	>99
	3	5c	$4-FC_6H_4$	81	>20:1	>99
	4	5d	$4-ClC_6H_4$	88	>20:1	>99
	5	5e	$4-BrC_6H_4$	84	>20:1	>99
	6	5f	4-MeOC ₆ H ₄	68	>20:1	>99
	7	5g	3-ClC ₆ H ₄	83	>20:1	>99
	8	5h	3-MeOC ₆ H ₄	71	>20:1	>99
	9	5i	2-naphthyl	64	>20:1	>99
_	10	5j	2-furyl	55	>20:1	nd ^e

⁶⁰ [a] Unless noted, the reaction was performed on 0.1 mmol scale with 2a (1.0 equiv), 3a (2.0 equiv), BzOH (40 mol%) and catalyst 1a (20 mol%) at 40 °C for 3-12 h. The control comparative experiments details, see the Supporting Information. [b] Yield of isolated product. [c] Measured by ¹H NMR spectroscopic analysis. [d] Determined by HPLC analysis on a ⁶⁵ chiral stationary phase. [e] The ee value can not be determined by HPLC analysis.

Having established the optimal reaction conditions (Table 1, entry 10), we next examined the scope and limitations of the above system with variants of reactants 2 and 3. Considering the 70 product 5 with two interval (same substituent on reactants 2 and 3) or three consecutive stereocenters (different substituent on reactants 2 and 3), we will describe them separately. We first tested reactions of compound 2 and 3 with same substituent and the results were summarized in Table 2. Unsaturated pyrazolones 75 2 and unsaturated ketones 3 with same substituents in meta- and para- positions could be tolerated and underwent smooth cyclization reactions, providing spiropyrazolones 5b-5h in good yields and with excellent diastereo- and enantioselectivities (entries 2-8). The electronic properties of the substituents seemed 80 to have no influence on the enantioselectivity, although the reaction yields were a little lower for reactants possessing

(Scheme 1).

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electron-rich substituents (entries 6 and 8). Conjugated aromatic (entry 9) and heteroaromatic (entry 10) substrates could also be applied to the present process, despite the ee value of product **5j** could not be detected using HPLC.

- Next, we were concentrated on constructing spiropyralones 5 with three consecutive stereocenters. Under optimized conditions, we choosed reactants 2 and 3 with different substituent optionally to manipulate the reaction, the results were somehow complicated mainly displaying of poor dr value (1:1) and difficult separation
- ¹⁰ on HPLC (Table 3, entry 14). When an ortho-substituented unsaturated pyrazolone or unsaturated ketone was used, the diastereoselectivity was improved greatly and the results were described in Table 3. When 2-Cl-benzalacetone was utilized, electron-neutral (entry 2), electron-withdrawing (entries 3-4),
- ¹⁵ electron-donating (entries 5-6), and double-substituted (entry 7) unsaturated pyrazolones 2 on the aryl ring were well tolerated and provided reaction products in moderate yields and excellent enantio- and diastereoselectivities; moreover, fused aromatic unsaturated pyrazolone was also suitable substrate for this
 ²⁰ reaction (entry 8). When 2-Cl-substituted unsaturated pyrazolone was employed, a wide range of unsaturated ketones were also tested. In most cases, the reactions were completed with excellent ee's (>99%) and dr's (>20:1) (entries 9-12) except in the occasion
- of alkyl-substituted unsaturated ketone with a 2:1 dr value (entry ²⁵ 13). The absolute configuration of the products was determined by X-ray crystallography analysis of compound **5s** (see Supporting Information).¹⁴

 Table 3.
 Substrate Scope of Cat 1a/ BzOH-Catalyzed Double Michael

 Addition of Unsaturated Pyrazolones 2 and Unsaturated Ketones 3 with

 30
 Different Substituents^[a]

Ph N O R ₁	H-N + R ₂	0 1a (2) BzOH(Tol, 4	0 mol%) 40 mol% 10 ⁰ C)	Ph N- R ₂	N III IIIR1
	2	3			5 C)
en	R_1	R_2	5	yield	ee	dr"
try	DI	2 (10)		[%]	[%]	. 20.1
1	Ph	2-CIPh	5k	67	>99	>20:1
2	4-CH ₃ Ph	2-ClPh	51	63	>99	>20:1
3	4-ClPh	2-ClPh	5m	65	>99	>20:1
4	3-ClPh	2-ClPh	5n	64	>99	>20:1
5	4-MeOPh	2-ClPh	50	48	>99	>20:1
6	3-MeOPh	2-ClPh	5p	52	>99	>20:1
7	3,4-(CH ₃) ₂ Ph	2-ClPh	5q	54	>99	>20:1
8	2-naphthyl	2-ClPh	5r	52	>99	>20:1
9	2-ClPh	4-BrPh	5s	69	>99	>20:1
10	2-ClPh	4-MeOPh	5t	52	>99	>20:1
11	2-ClPh	Ph	5u	68	>99	>20:1
12	2-ClPh	2-furyl	5v	50	>99	>20:1
13	2-ClPh	i-Bu	5w	67	98	2:1
14	Ph	4-CH ₃ Ph	5x	82	>99	1:1

[a] Unless noted, the reaction was performed on 0.1 mmol scale with **2a** (1.0 equiv), **3a** (2.0 equiv), BzOH (40 mol%) and catalyst **1a** (20 mol%) at 40 °C for 3-12 h. The control comparative experiments details, see the ³⁵ Supporting Information. [b] Yield of isolated product. [c] Determined by HPLC analysis on a chiral stationary phase. [d] Measured by ¹H NMR spectroscopic analysis. [d] Determined by HPLC analysis on a chiral stationary phase.

In the initial screening of catalysts, we found that catalyst **1b** ⁴⁰ also resulted in spiropyrazolones **5**' with comparable yield,

enantio- and diastereoselectivity and inversed configuration, so we also tested the generality of the catalytic system and the results were depicted in Table 4. The desired spirocyclic products **5g'-5h'**, **5l'**, **5r'-5x'** were obtained in 42-64% yields with 1:1- $_{45} > 20:1$ dr's and 98- >99% ee's (Table 4, entries 1-10).

 $\begin{array}{l} \textbf{Table 4. Substrate Scope of Cat 1b/ BzOH-Catalyzed Double Michael}\\ \textbf{Addition of Unsaturated Pyrazolones 2} \ \textbf{and Unsaturated Ketones 3}^{[a]} \end{array}$

Ph N O R ₁	N-N +) (20 mol%) H (40 mol ol, 40 ⁰ C	b) %) ➡	N-	Ph N O R ₂
	2	3			5' Ö	
en	\mathbf{R}_1	R_2	5'	yield	ee	dr ^d
try				$[\%]^{b}$	$[\%]^{c}$	
1	3-ClPh	3-ClPh	5g'	57	>99	>20:1
2	3-MeOPh	3-MeOPh	5h'	63	>99	>20:1
3	4-CH ₃ Ph	2-ClPh ₄	51'	61	>99	>20:1
4	2-naphthyl	2-ClPh	5r'	50	>99	>20:1
5	2-ĈlPh	4-BrPh	5s'	55	>99	>20:1
6	2-ClPh	4-MeOPh	5ť	64	>99	>20:1
7	2-ClPh	Ph	5u'	48	>99	>20:1
8	2-ClPh	2-furyl	5v'	52	>99	>20:1
9	2-ClPh	i-Bu	5w'	54	98	2:1
10	Ph	4-CH ₃ Ph	5x'	42	>99	1:1

[a] Unless noted, the reaction was performed on 0.1 mmol scale with 2a
 (1.0 equiv), 3a (2.0 equiv), BzOH (40 mol%) and catalyst 1a (20 mol%) at 40 °C for 3-12 h. The control comparative experiments details, see the Supporting Information. [b] Yield of isolated product. [c] Determined by HPLC analysis on a chiral stationary phase. [d] Measured by ¹H NMR spectroscopic analysis.

55 Conclusions

In summary, we have developed the first organocatalytic double Michael cascade reaction of unsaturated ketones and unsaturated pyrazolones that provides spiropyrazolones core structures with two interval or three consecutive stereogenic centers, including a ⁶⁰ spiro quaternary center with excellent diastereo- (>20:1) and enantioselectivities (up to 99% ee). Moreover, a pair of enantiomers **5a** and **5a**' can be achieved via different catalysit. This simple and effective reaction provides rapid entry to stereochemically complex spiropyrazolones derivatives.

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Notes and references

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- 14 CCDC 907183 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif.

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l'able 1	•				
entry	Cat	solvent	Time[h]	yield[%] ^b	ee ^c
1	1 a	Toluene	120	-	-
2^d	1 a	Toluene	72	87	>99
$3^{d,f}$	1b	Toluene	72	81	>99
4^d	1c	Toluene	120	76	93
6^d	1a	MTBE	72	59	96
7^d	1 a	THF	72	55	86
8^d	1 a	CH_2Cl_2	72	79	93
9^d	1 a	MeCN	72	46	84
$10^{d,e}$	1 a	Toluene	3.5	85	>99
$11^{d,e,f}$	1b	Toluene	3.5	80	>99

Table 2.

entry	5	R	yield[%] ^b	dr [%] ^c	ee ^d
1	5a	Ph	87	>20:1	>99
2	5b	$4\text{-}CH_3C_6H_4$	75	>20:1	>99
3	5c	$4\text{-}FC_6H_4$	81	>20:1	>99
4	5d	$4\text{-}ClC_6H_4$	88	>20:1	>99
5	5e	$4-BrC_6H_4$	84	>20:1	>99
6	5f	$4-MeOC_6H_4$	68	>20:1	>99
7	5g	$3-ClC_6H_4$	83	>20:1	>99
8	5h	3-MeOC ₆ H ₄	71	>20:1	>99
9	5i	2-naphthyl	64	>20:1	>99
10	5j	2-furyl	55	>20:1	nd ^e

Table 3.

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entry	\mathbf{R}_1	R_2	5	yield	ee [%] ^c	dr^{d}
				$[\%]^{b}$		
1	Ph	2-ClPh	5k	67	>99	>20:1
2	4-CH ₃ Ph	2-ClPh	51	63	>99	>20:1
3	4-ClPh	2-ClPh	5m	65	>99	>20:1
4	3-ClPh	2-ClPh	5n	64	>99	>20:1
5	4-MeOPh	2-ClPh	50	48	>99	>20:1
6	3-MeOPh	2-ClPh	5р	52	>99	>20:1
7	3,4-(CH ₃) ₂ Ph	2-ClPh	5q	54	>99	>20:1
8	2-naphthyl	2-ClPh	5r	52	>99	>20:1
9	2-ClPh	4-BrPh	5 s	69	>99	>20:1
10	2-ClPh	4-MeOPh	5t	52	>99	>20:1
11	2-ClPh	Ph	5u	68	>99	>20:1
12	2-ClPh	2-furyl	5v	50	>99	>20:1
13	2-ClPh	i-Bu	5w	67	98	2:1
14	Ph	4-CH ₃ Ph	5x	82	>99	1:1

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entry	R_1	R_2	5'	yield	ee [%] ^c	dr^{d}
				$[\%]^{b}$		
1	3-ClPh	3-ClPh	5g'	57	>99	>20:1
2	3-MeOPh	3-MeOPh	5h'	63	>99	>20:1
3	4-CH ₃ Ph	2-ClPh ₄	51'	61	>99	>20:1
4	2-naphthyl	2-ClPh	5r'	50	>99	>20:1
5	2-ClPh	4-BrPh	5s'	55	>99	>20:1
6	2-ClPh	4-MeOPh	5ť	64	>99	>20:1
7	2-ClPh	Ph	5u'	48	>99	>20:1
8	2-ClPh	2-furyl	5v'	52	>99	>20:1
9	2-ClPh	i-Bu	5w'	54	98	2:1
10	Ph	4-CH ₃ Ph	5x'	42	>99	1:1