ORGANIC LETTERS

2001
Vol. 3, No. 3
457−459

Enantioselective Rhodium(I)-Catalyzed Hydrogenation of Trifluoromethyl Ketones

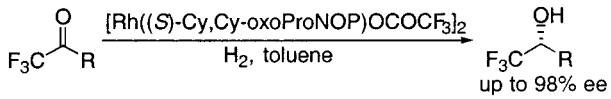
Yoshichika Kuroki,* Yuko Sakamaki, and Katsuhiko Iseki

Chemical Division, Daikin Industries, Ltd., Miyukigaoka, Tsukuba, Ibaraki 305-0841, Japan

yoshichika.kuroki@daikin.co.jp

Received December 5, 2000

ABSTRACT



The asymmetric hydrogenation of trifluoromethyl ketones to yield chiral α-trifluoromethyl alcohols with enantiomeric excesses up to 98% was achieved in the presence of chiral rhodium-(amidephosphine-phosphinite) complexes.

The catalytic asymmetric synthesis of chiral organofluorine compounds has played an important role in the development of medicines and materials based on the influence of fluorine's unique properties.[1] Homochiral α-trifluoromethyl alcohols are versatile intermediates for the synthesis of antiferroelectric liquid crystalline molecules.[2] Although a few asymmetric catalyses for preparing the alcohols have been reported,[3] their synthesis has drawbacks such as insufficient levels of enantioselectivity, low catalytic efficiencies, and limited scope of the substrates. Recently, we reported that the highly enantioselective synthesis of 1,1,1-trifluoroalkan-2-ols can be successfully achieved by hydrogenating 1,1,1-trifluoroalkan-2-one enol acetates in the presence of chiral ruthenium catalysts.[4] This paper discloses the asymmetric hydrogenation of trifluoromethyl ketones catalyzed by chiral rhodium-(amidephosphine-phosphinite) complexes to provide chiral α-trifluoromethyl alcohols with up to 98% ee.

Recently, we found that chiral rhodium-(amidephosphine-phosphinite) complexes, prepared from $[Rh(COD)OCOCF_3]_2$ and oxoProNOP ligands,[5] catalyze the hydrogenation of 2,2-difluoro-3-oxocarboxylates and 4,4,4-trifluoroacetoacetate to give the corresponding β-hydroxy esters with good-to-excellent enantioselectivity.[6] The stereochemical outcome from the latter β-keto ester indicated that the trifluoromethyl group has a significant influence on the enantiotopic face selection, prompting us to examine the hydrogenation of the trifluoromethyl ketones using the chiral rhodium-(amidephosphine-phosphinite) complexes.

(1) (a) Bravo, P.; Resnati, G. *Tetrahedron: Asymmetry* **1990**, *1*, 661−692. (b) Iseki, K. *Tetrahedron* **1998**, *54*, 13887−13914. (c) *Enantiocontrolled Synthesis of Fluoro-Organic Compounds*; Soloshonok, V. A., Ed.; Wiley: New York, 1999. (d) *Asymmetric Fluoroorganic Chemistry: Synthesis, Application and Future Directions*; Ramachandran, P. V., Ed.; American Chemical Society: Washington, DC, 2000.

(2) (a) Suzuki, Y.; Hagiwara, T.; Kawamura, I.; Okamura, N.; Kitazume, T.; Kakimoto, M.; Imai, Y.; Ouchi, Y.; Takezoe, H.; Fukuda, A. *Liq. Cryst.* **1989**, *6*, 167. (b) Mikami, K. In *Asymmetric Fluoroorganic Chemistry: Synthesis, Application and Future Directions*; Ramachandran, P. V., Ed.; American Chemical Society: Washington, DC, 2000; pp 255−269.

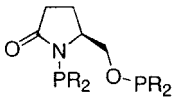
(3) See, for instance: (a) Koenig, K. E.; Bachman, G. L.; Vineyard, B. D. *J. Org. Chem.* **1980**, *45*, 2362−2365. (b) Burk, M. J. *J. Am. Chem. Soc.* **1991**, *113*, 8518−8519. (c) Corey, E. J.; Bakshi, R. K. *Tetrahedron Lett.* **1990**, *31*, 611−614. (d) Ohkuma, T.; Koizumi, M.; Doucet, H.; Pham, T.; Kozawa, M.; Murata, K.; Katayama, E.; Yokozawa, T.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1998**, *120*, 13529−13530. (e) Mikami, K.; Yajima, T.; Terada, M.; Uchimaru, T. *Tetrahedron Lett.* **1993**, *34*, 7591−7594. (f) Poras, H.; Matsutani, H.; Yaruva, J.; Kusumoto, T.; Hiyama, T. *Chem. Lett.* **1998**, 665−666. (g) Bennai, Y. L.; Vanhessche, K. P. M.; Sharpless, K. B. *Tetrahedron: Asymmetry* **1994**, *5*, 1473−1476. (h) Iseki, K.; Nagai, T.; Kobayashi, Y. *Tetrahedron Lett.* **1994**, *35*, 3137−3138. (i) Kuroki, Y.; Iseki, K. *Tetrahedron Lett.* **1999**, *40*, 8231−8234.

(4) Kuroki, Y.; Asada, D.; Sakamaki, Y.; Iseki, K. *Tetrahedron Lett.* **2000**, *41*, 4603−4607.

(5) (a) Roucoux, A.; Agbossou, F.; Mortreux, A.; Petit, F. *Tetrahedron: Asymmetry* **1993**, *4*, 2279−2282. (b) Roucoux, A.; Devocelle, M.; Carpentier, J.-F.; Agbossou, F.; Mortreux, A, *Synlett* **1995**, 358−359. (c) Roucoux, A.; Thieffry, L.; Carpentier, J.-F.; Devocelle, M.; Méliet, C.; Agbossou, F.; Mortreux, A.; Welch, A. J. *Organometallics* **1996**, *15*, 2440−2449. (d) Carpentier, J.-F.; Mortreux, A. *Tetrahedron: Asymmetry* **1997**, *8*, 1083−1099.

(6) Kuroki, Y.; Asada, D.; Iseki, K. *Tetrahedron Lett.* **2000**, 41, 9853−9858.

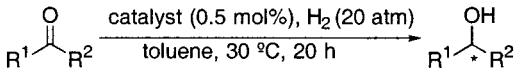
10.1021/ol006962s CCC: $20.00
Published on Web 01/17/2001



(*S*)-Cy,Cy-oxoProNOP: R = *c*-C_6H_{11}
(*S*)-*i*-Pr,*i*-Pr-oxoProNOP: R = *i*-Pr

1,1,1-Trifluoro-2-decanone was hydrogenated using 0.5 mol % [Rh((*S*)-Cy,Cy-oxoProNOP)$OCOCF_3]_2$ (**1**) or 0.1 mol % [Rh((*R*)-*i*-Pr,*i*-Pr-oxoProNOP)$OCOCF_3]_2$ (**2**) under 10 or 20 atm of hydrogen in toluene at 30 °C for 20 h to give 1,1,1-trifluoro-2-decanol[7] with 97% ee in a nearly quantitative yield (Table 1, entries 1 and 2). The reactions of 1,1-difluoro-2-decanone and 1-fluoro-2-decanone under the same conditions dramatically diminished the enantioselectivity despite the quantitative yields (entries 3 and 4). However, the catalyst **1** scarcely promoted the hydrogenation of 2-decanone (entry 5). A nonfluorinated aromatic ketone, acetophenone, was also hydrogenated into 1-phenylethanol with only 8% ee in only 2% yield (entry 6), while the reaction of α,α,α-trifluoroacetophenone led to a 93% yield of the corresponding (*R*)-product[8] with 73% ee (entry 7). 1,1,1,2,2-Pentafluoro-3-dodecanone underwent hydrogenation to yield the corresponding α-pentafluoroethyl alcohol with 97% ee in quantitative yield (entry 8).

Table 1. Asymmetric Hydrogenation of Ketones Using Rhodium-(Amidephosphine-phosphinite) Complexes (**1** and **2**)

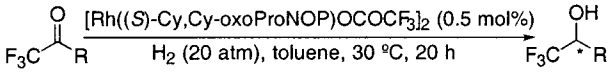


entry	R^1	R^2	catalyst [a]	yield (%)[b]	ee (%)
1	CF_3	C_8H_{17}	**1**	99	97[c] (*R*)[d]
2[e]	CF_3	C_8H_{17}	**2**	100	97[c] (*S*)[d]
3	CHF_2	C_8H_{17}	**1**	100	27[c]
4	CH_2F	C_8H_{17}	**1**	100	15[c]
5	CH_3	C_8H_{17}	**1**	<1	
6	CH_3	Ph	**1**	2	8[f]
7	CF_3	Ph	**1**	93	73[f] (*R*)[d]
8	C_2F_5	C_9H_{19}	**1**	100	97[g] (*R*)[h]

[a] **1**, [Rh((*S*)-Cy,Cy-oxoProNOP)$COCF_3]_2$; **2**, [Rh((*R*)-*i*-Pr,*i*-Pr-oxoProNOP)$COCF_3]_2$. [b] Isolated yield. [c] Determined by GLC analysis of the corresponding acetate with CP-Cyclodex-β-236M. [d] Assigned by comparing the sign of the optical rotations with literature data. See refs 7 and 8. [e] Carried out using 0.1 mol % of **2** under 10 atm H_2. [f] Determined by HPLC analysis with CHIRALCEL OD-H. [g] Determined by HPLC analysis of the corresponding 3,5-dinitrobenzoate with CHIRALCEL OD-H. [h] Assigned by the modified Mosher method.[9]

A variety of trifluoromethyl ketones were hydrogenated in the presence of 0.5 mol % of the catalyst **1** under 20 atm of hydrogen in toluene at 30 °C for 20 h (Table 2). The hydrogenations of 1,1,1-trifluoro-2-octanone and cyclohexyl trifluoromethyl ketone provided the (*R*)-products[7,10] with 97% ee in 98% and 90% yields, respectively (entries 1 and 2).

Table 2. Asymmetric Hydrogenation of Trifluoromethyl Ketones



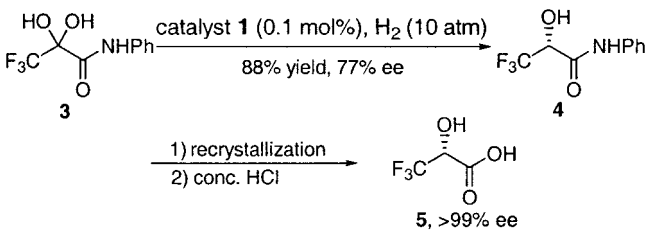
entry	R	yield (%)[a]	ee (%)
1	C_6H_{13}	98	97[b] (*R*)[c]
2	*c*-C_6H_{11}	90	97[d] (*R*)[c]
3	*c*-$C_6H_{11}CH_2$	97	98[b]
4	$PhCH_2$	97	97[e]
5	$PhCH_2CH_2$	99	96[e]
6	$PhCH_2OCH_2$	100	86[b]
7	*p*-ClPh	8	38[e]
8	*p*-CH_3OPh	100	83[e]

[a] Isolated yield. [b] Determined by GLC analysis of the corresponding acetate with CP-Cyclodex-β-236M. [c] Assigned by comparing the sign of the optical rotations with literature data. See refs 7 and 10. [d] Determined by HPLC analysis of the corresponding 3,5-dinitrobenzoate with CHIRALCEL OD-H. [e] Determined by HPLC analysis with CHIRALCEL OD-H.

The cyclohexylmethyl trifluoromethyl ketone was also hydrogenated with high enantioselectivity and good catalyst activity (entry 3). While the reactions of the two ketones having a benzene ring also proceeded with high enantiomeric excesses (entries 4 and 5), the catalyst **1** failed in high asymmetric induction (86% ee) with 1,1,1-trifluoro-5-phenyl-4-oxa-2-pentanone (entry 6). The introduction of a chlorine atom into the *p*-position of α,α,α-trifluoroacetophenone drastically reduced the chemical and optical yields (entry 7 vs Table 1, entry 7), while the hydrogenation of α,α,α-trifluoro-*p*-methoxyacetophenone significantly improved the enantioselectivity (entry 8).

Finally, the asymmetric hydrogenation of 3,3,3-trifluoro-2,2-dihydroxypropionanilide (**3**), prepared from hexafluoropropene oxide,[11] using 0.1 mol % of the catalyst **1** was also examined in toluene under 10 atm H_2 at 70 °C for 20 h, and (*R*)-3,3,3-trifluoro-2-hydroxypropionanilide (**4**) was obtained in 77% ee and 88% yield. A single recrystallization of **4** afforded (*R*)-**4** with >99% ee. Hydrolysis of the amide (*R*)-**4** with concentrated HCl at 80 °C provided (*R*)-trifluorolactic acid (**5**) in 84% yield and >99% ee (Scheme 1).

Scheme 1



In summary, the catalytic asymmetric hydrogenation of trifluoromethyl ketones has been accomplished using the rhodium-oxoProNOP catalysts, providing a variety of opti-

(7) Ramachandran, P. V.; Teodorovic, A. V.; Brown, H. C. *Tetrahedron* **1993**, *49*, 1725−1738.
(8) Peters, H. M.; Feigl, D. M.; Mosher, H. S. *J. Org. Chem.* **1968**, *33*, 4245−4250.

cally active α-trifluoromethyl alcohols with up to 98% ee. Applications of this method to the synthesis of versatile chiral fluorinated molecules are currently under investigation.

Supporting Information Available: Detailed experimetal procedures and characterization data for chiral α-trifluoromethyl alcohols. This material is available free of charge via the Internet at http://pubs.acs.org.

OL006962S

(9) (a) Xiao, L.; Yamazaki, T.; Kitazume, T.; Yonezawa, T.; Sakamoto, Y.; Nogawa, K. *J. Fluorine Chem.* **1997**, *84*, 19−23. (b) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. *J. Am. Chem. Soc.* **1991**, *113*, 4092−4096.

(10) Ramachandran, P. V.; Teodorovic, A. V.; Gong, B.; Brown, H. C. *Tetrahedron: Asymmetry* **1994**, *5*, 1075−1086.

(11) Sianesi, D.; Pasetti, A.; Tarli, F. *J. Org. Chem.* **1966**, *31*, 2312−2316.