

# Highly stereoselective total synthesis of PM-Toxin B, a corn host-specific pathotoxin

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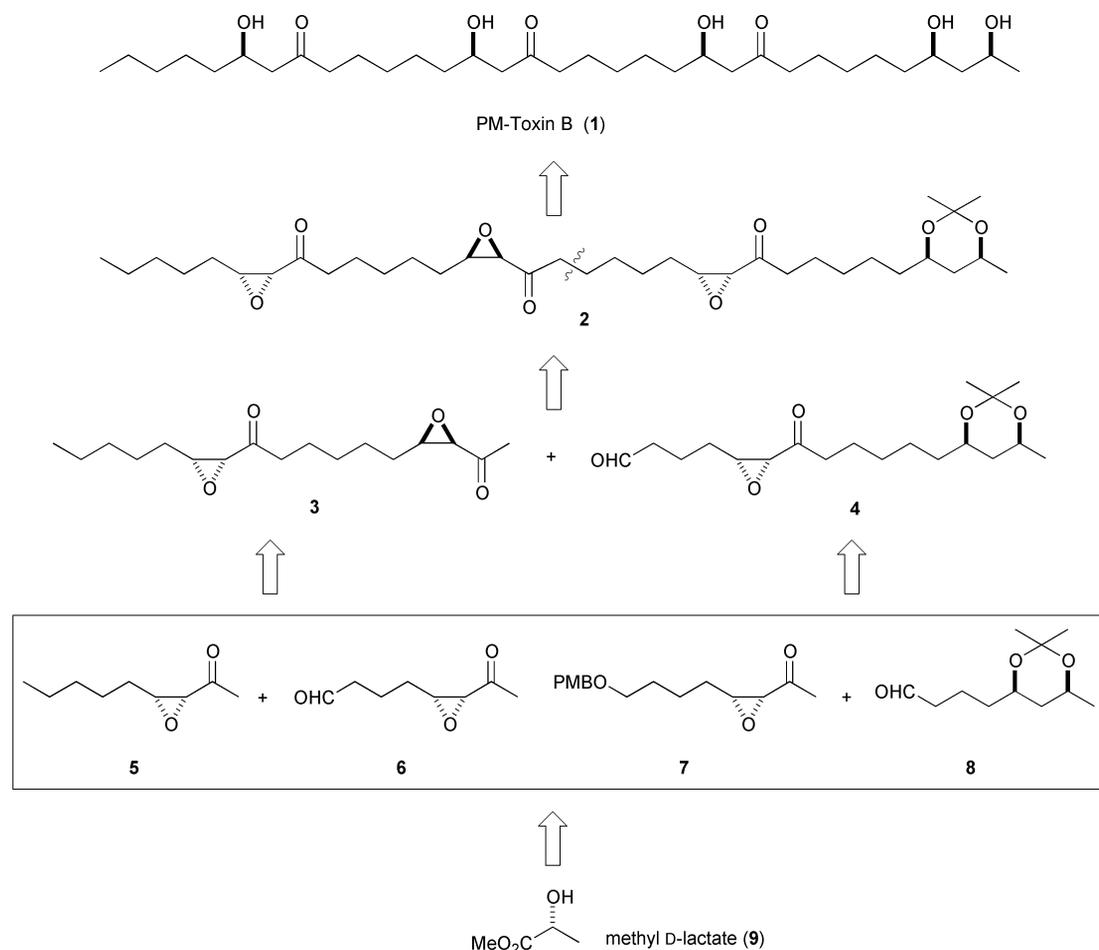
The highly stereoselective total synthesis of PM-toxin B, a corn host-specific toxin produced by the fungal pathogen *Phyllosticta maydis*, has been achieved by a convergent synthetic strategy which involves cross-aldol coupling of four key segments and the regioselective reductive cleavage of three  $\alpha,\beta$ -epoxy ketone functionalities by an organo-selenium reagent as key steps.

The cause of major epidemics of Northern T-corn leaf blight disease in the United States in 1970 has been shown to be PM-toxin, a corn host-specific pathotoxin produced by the fungal pathogen *Phyllosticta maydis*.<sup>1</sup> PM-toxin<sup>2</sup> and HMT-toxin<sup>3</sup> produced by the fungal pathogen *Helminthosporium maydis*, race T, are representative corn host-specific toxins.<sup>4</sup> Among 10–15 PM-toxin components, the four major ones, PM-toxin A, B, C, and D, have been isolated so far and found to be linear C<sub>33</sub> and C<sub>35</sub> compounds containing a number of characteristic  $\beta$ -ketol (aldol) structures.<sup>2</sup> The unique structures of these

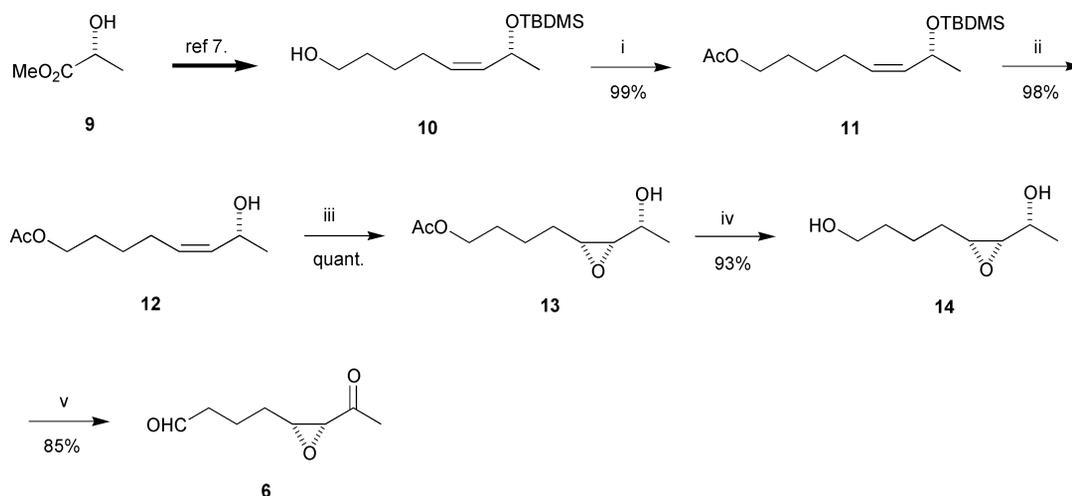
corn pathotoxins as well as their marked host-specific toxicity have elicited much attention from biologists and synthetic chemists.<sup>4,5</sup> So far, only the synthesis of an isomeric mixture of PM-toxin B in a racemic form by Daly and co-workers has been reported.<sup>6</sup> Recently, we reported the first asymmetric total synthesis of PM-toxin A, containing a characteristic sequence of four aldol structures, by a linear synthetic strategy which involves four tandem aldol reactions as key steps;<sup>7</sup> however, the yield of the cross aldol reactions was greatly reduced as the carbon chain length of the substrate became longer.<sup>7</sup>

We describe herein the first and highly stereoselective total synthesis of PM-toxin B (**1**) by a convergent synthetic strategy which involves cross-aldol coupling of four key segments and the organoselenium-mediated regioselective reductive cleavage of three  $\alpha,\beta$ -epoxy ketone units as key steps.

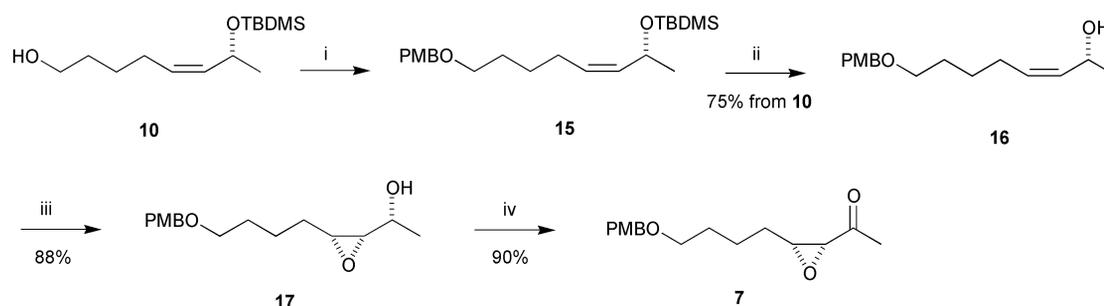
The retrosynthetic analysis for **1** is shown in Scheme 1. Taking into consideration chemically labile aldol structures, we designed a synthetic route which involves construction of



Scheme 1 The retrosynthetic analysis for the synthesis of **1**.



**Scheme 2** Reagents: i, Ac<sub>2</sub>O, DMAP, pyridine; ii, Bu<sub>4</sub>NF, THF; iii, MCPBA, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C; iv, NH<sub>3</sub>, MeOH; v, (COCl)<sub>2</sub>, DMSO, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C, then Et<sub>3</sub>N.



**Scheme 3** Reagents: i, PMBOCH(=NH)CCl<sub>3</sub>, PPTS, CH<sub>2</sub>Cl<sub>2</sub>; ii, Bu<sub>4</sub>NF, THF; iii, MCPBA, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C; iv, (COCl)<sub>2</sub>, DMSO, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C, then Et<sub>3</sub>N.

three aldol structures by regioselective reductive cleavage of tris-epoxy ketone **2** at the final stage of the synthesis. For the synthesis of the key tris-epoxy ketone **2**, **2** was divided into the left-half segment, bis-epoxy ketone **3**, and the right-half segment, epoxy aldehyde **4**, and both segments were designed to be assembled from the key fragments **5** and **6**, **7** and **8**, respectively. Furthermore, all these key fragments were designed to be synthesized from the same starting material, *i.e.*, methyl D-lactate (**9**), as shown in Scheme 1. The first epoxy ketone fragment **5** was efficiently and highly stereoselectively synthesized from **9** according to our previous route (63% overall yield for 6 steps).<sup>7</sup> Similarly, the second epoxy aldehyde **6** was also stereoselectively synthesized from **9** employing a modified procedure<sup>7</sup> as shown in Scheme 2. Thus, the (*Z*)-alkenyl alcohol **10** readily obtainable from **9** was converted to the acetate **11**, which was then treated with Bu<sub>4</sub>NF in THF to give the (*Z*)-allylic alcohol **12** in 97% overall yield from **10**. Upon epoxidation of the allylic alcohol **12** with MCPBA in CH<sub>2</sub>Cl<sub>2</sub> the α-epoxy alcohol **13** was produced exclusively and quantitatively. The epoxy alcohol **13** thus obtained was converted to the epoxy aldehyde **6** by the two-step reaction sequence: (1) hydrolysis of the acetate with NH<sub>3</sub> in MeOH (93%) and (2) Swern oxidation (85%).

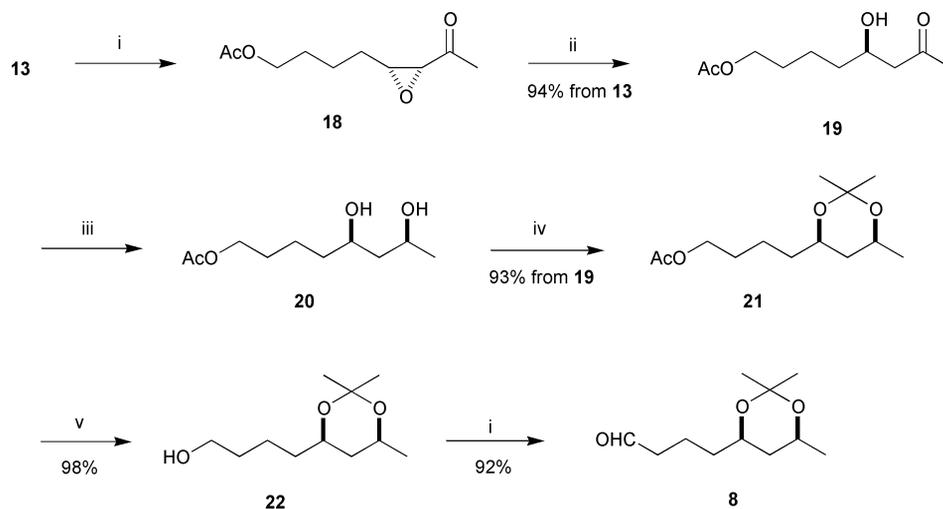
On the other hand, the third key fragment **7** was prepared from the common intermediate **10** as shown in Scheme 3. Protection of the primary hydroxy group of **10** with 4-methoxybenzyl (PMB) trichloroacetimidate followed by treatment of the resulting TBDMS ether **15** with Bu<sub>4</sub>NF in THF furnished the allyl alcohol **16** in 75% overall yield. Subsequent epoxidation of **16** with MCPBA in CH<sub>2</sub>Cl<sub>2</sub> proceeded cleanly giving rise to the single α-epoxy alcohol **17** in 88% yield. The epoxy alcohol **17** thus obtained was converted to the epoxy ketone **7** by Swern oxidation (90%).

The fourth fragment **8** containing a *syn*-1,3-diol moiety was efficiently and highly stereoselectively synthesized according to Scheme 4. First, the epoxy alcohol **13** used in the synthesis of the fragment **6** was transformed into the epoxy ketone **18** by Swern oxidation, which was then subjected to the organo-

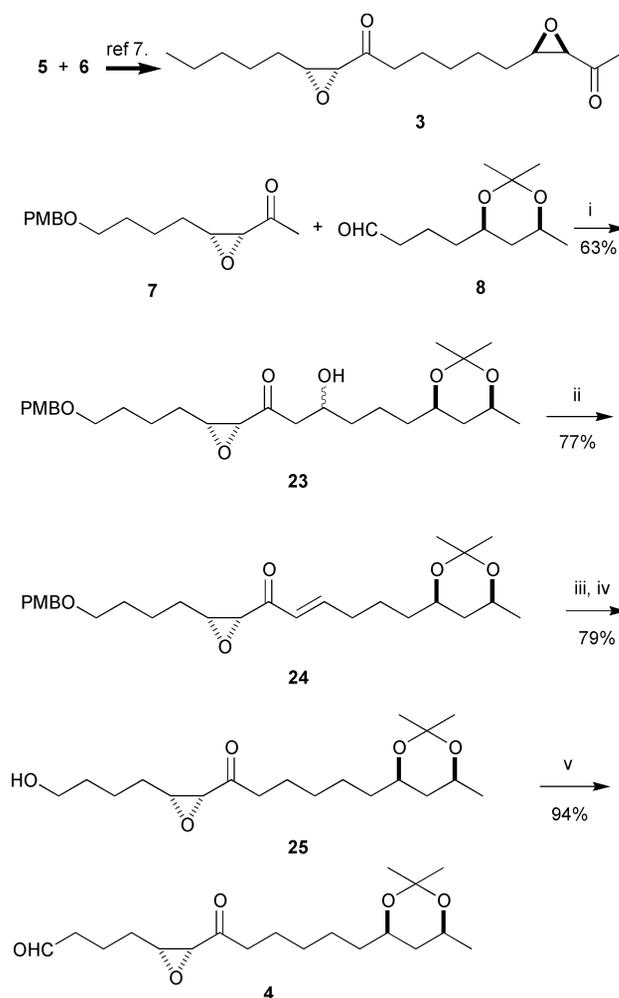
selenium reduction.<sup>8</sup> Thus, on treatment of **18** with sodium (phenylseleno)triethylborate (Na[PhSeB(OEt)<sub>3</sub>])<sup>8</sup> (2.5 equiv.) in ethanol in the presence of AcOH (3 equiv.) at 0 °C, the regioselective reductive cleavage of the epoxide moiety occurred to give the β-hydroxy ketone **19** as a single product in 94% isolated yield. It should be noted that the chemo- and regioselective reduction of an epoxy ketone moiety is possible by organoselenium reduction.<sup>8</sup> Construction of the requisite *syn*-1,3-diol **20** from the resulting β-hydroxy ketone **19** was successfully performed by diethylmethoxyborane–sodium borohydride reduction<sup>9</sup> in THF–MeOH (5 : 1) in excellent yield, and no diastereomer was detected, although other reducing agents inevitably formed a mixture of diastereomers in variable ratios. The 1,3-diol **20** thus obtained was converted to the fourth key segment **8** by the three-step reaction sequence: (1) protection of *syn*-1,3-diol with 2,2-dimethoxypropane (93%); (2) hydrolysis of the acetate with NH<sub>3</sub> in MeOH (98%); (3) Swern oxidation (92%).

With the four key fragments **5–8** in hand, we next focused on the cross-aldol reaction in the synthesis of the left segment **3** and the right segment **4**, respectively. Initially, the left segment **3** was synthesized by the aldol reaction of **5** and **6** as in the synthesis of PM-toxin A<sup>7</sup> (Scheme 5). On the other hand, the right segment **4** was successfully synthesized according to Scheme 5. Thus, the cross-aldol reaction of **7** (1.3 equiv.) and **8** (1 equiv.) by the use of lithium hexamethyldisilazide (LiHMDS, 1.3 equiv.) as base smoothly proceeded in THF at -78 °C giving rise to the hydroxy ketone **23** as a diastereomeric mixture in 63% yield. Subsequent treatment of **23** with methanesulfonyl chloride in CH<sub>2</sub>Cl<sub>2</sub> in the presence of DMAP directly afforded the enone **24** in 77% yield, which was submitted to catalytic hydrogenation over 10% Pd-C in ethyl acetate followed by treatment with DDQ to give the epoxy alcohol **25** in 79% overall yield. Swern oxidation of the resulting alcohol **25** furnished the desired segment **4** in 94% yield.

With the left and right segments **3** and **4** in hand, we set out the final coupling reaction in the total synthesis of PM-toxin B



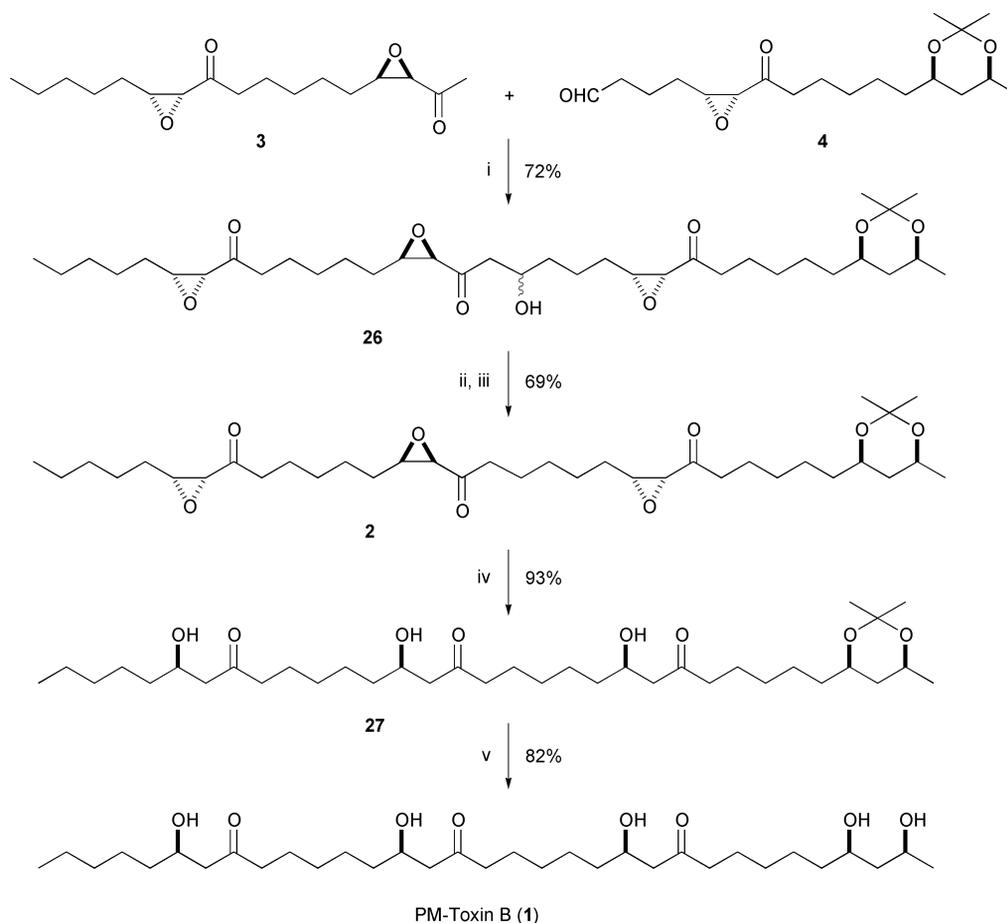
**Scheme 4** Reagents: i,  $(\text{COCl})_2$ , DMSO,  $\text{CH}_2\text{Cl}_2$ ,  $\text{Et}_3\text{N}$ ; ii,  $\text{Na}[\text{PhSeB}(\text{OEt})_3]$ , AcOH, EtOH,  $0^\circ\text{C}$ , 1 h; iii,  $(\text{CH}_3\text{CH}_2)_2\text{BOCH}_3$ ,  $\text{NaBH}_4$ , THF–MeOH; iv,  $(\text{CH}_3)_2\text{C}(\text{OCH}_3)_2$ , PPTS,  $\text{CH}_2\text{Cl}_2$ ; v,  $\text{NH}_3$ , MeOH.



**Scheme 5** Reagents: i, LiHMDS (1.3 equiv.), THF,  $-78^\circ\text{C}$ , then **8**; ii,  $\text{CH}_3\text{SO}_2\text{Cl}$ , DMAP,  $\text{CH}_2\text{Cl}_2$ ; iii,  $\text{H}_2$ , 10% Pd-C, AcOEt; iv, DDO,  $\text{CH}_2\text{Cl}_2$ – $\text{H}_2\text{O}$  (10:1); v,  $(\text{COCl})_2$ , DMSO,  $\text{CH}_2\text{Cl}_2$ ,  $\text{Et}_3\text{N}$ .

(**1**) (Scheme 6). Fortunately, the key aldol reaction of **3** and **4** cleanly occurred with the use of LiHMDS in THF at  $-78^\circ\text{C}$ , giving rise to the  $\beta$ -hydroxy ketones **26** as a diastereomeric mixture in 72% isolated yield, although we were afraid that the yield of this cross-aldol reaction might decrease greatly as in the synthesis of PM-toxin A.<sup>7</sup> We were pleased, therefore, to obtain optimal results. The products **26** thus obtained were converted to the tris-epoxy ketone **2** by the two-step reaction sequence: (1) mesylation (79%) and (2) hydrogenation (87%). Thus, the key tris-epoxy ketone **2** for the synthesis of **1** was secured in an optically pure form based on the convergent

aldol strategy. The critical organoselenium reduction of **2** was successfully performed by treatment with benzeneselenol ( $\text{PhSeH}$ ) generated *in situ* from  $\text{Na}[\text{PhSeB}(\text{OEt})_3]$  (15 equiv.)<sup>8</sup> and acetic acid (18 equiv.) in EtOH at  $0^\circ\text{C}$ , whereupon the reductive cleavage of three epoxy ketone moieties occurred regioselectively at the  $\alpha$ -position, giving rise to the crystalline tris- $\beta$ -ketol **27** in 93% yield. Finally, on treatment of the resulting tris- $\beta$ -hydroxy ketone **27** with 1 M HCl in THF at  $0^\circ\text{C}$ , PM-toxin B (**1**) was obtained as crystals in 82% yield. Physical properties of the synthetic compound: mp  $126$ – $126.5^\circ\text{C}$  (from acetone);  $[\alpha]_D^{25} -29.7$  ( $c$  0.38,  $\text{CHCl}_3$ ),  $-6.8$



**Scheme 6** Reagents: i, LiHMDS (1.3 equiv.), THF,  $-78\text{ }^{\circ}\text{C}$ , then **4**; ii,  $\text{CH}_3\text{SO}_2\text{Cl}$ , DMAP,  $\text{CH}_2\text{Cl}_2$ ; iii,  $\text{H}_2$ , 10% Pd-C, AcOEt; iv,  $\text{Na}[\text{PhSeB}(\text{OEt})_3]$ , AcOH, EtOH; v, 1 M HCl, THF,  $0\text{ }^{\circ}\text{C}$ .

( $c$  0.35, MeOH); UV  $\lambda_{\text{max}}$  280 nm ( $\epsilon$  108, MeOH); CD  $\lambda_{\text{max}}$  281 nm ( $\Delta\epsilon$  0.36, MeOH); FD-MS  $m/z$  587 ( $\text{M} + \text{H}^+$ ),  $m/z$  609 ( $\text{M} + \text{Na}^+$ ), 625 ( $\text{M} + \text{K}^+$ ). These results are in agreement with the reported data except melting point  $97\text{ }^{\circ}\text{C}$ ,  $[\alpha]_{\text{D}}^{25} -10$  (MeOH), UV  $\lambda_{\text{max}}$  275 nm ( $\epsilon$  138, MeOH), CD  $\lambda_{\text{max}}$  281 nm ( $\Delta\epsilon$  0.36, MeOH).<sup>2b</sup>  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra of the synthetic compound were identical with those of natural PM-toxin B.<sup>6,10</sup>

In conclusion, the highly stereoselective total synthesis of PM-toxin B (**1**) has been achieved by a convergent strategy involving cross-aldol reactions and the regioselective reductive cleavage of  $\alpha,\beta$ -epoxy ketone units as key steps.

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- $^1\text{H}$  NMR (500 MHz,  $\text{C}_6\text{D}_6\text{N}$ ): 0.82 (t,  $J=7.0$  Hz, 3H), 1.12–1.78 (m, 38H), 1.37 (d,  $J=6.2$  Hz, 3H), 1.87 (dt,  $J=13.8, 9.5$  Hz, 1H), 2.54 (t,  $J=7.2$  Hz, 2H), 2.55 (t,  $J=7.3$  Hz, 4H), 2.61 (dd,  $J=15.0, 4.1$  Hz, 1H), 2.62 (dd,  $J=15.0, 4.1$  Hz, 1H), 2.63 (dd,  $J=15.0, 4.1$  Hz, 1H), 2.79 (dd,  $J=15.0, 8.6$  Hz, 2H), 2.81 (dd,  $J=15.0, 8.6$  Hz, 1H), 4.01–4.12 (m, 1H), 4.24–4.47 (m, 4H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{C}_6\text{D}_6\text{N}$ ): 210.3, 210.3, 210.3, 71.3, 67.8, 67.7, 67.6, 51.3, 51.3, 51.3, 46.5, 43.8, 43.8, 43.8, 38.6, 38.3, 38.1, 38.1, 32.0, 29.6, 29.5, 29.5, 25.9, 25.9, 25.8, 25.7, 24.6, 23.9, 23.9, 23.9, 22.9, 14.1.