A New Synthetic Approach to Novel Spiro-β-lactams

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An operationally simple and efficient approach for the synthesis of novel spiro- β -lactams is described. The key reaction is a halogen-mediated intrasulfenyl cyclization of a *cis*-3-benzylthio-3-(prop-2-ynyloxy/-enyloxy)- β -lactam procured

through a Lewis acid-mediated C-3-alkylation of the *trans*-3-benzylthio-3-chloro-β-lactam carbocation equivalent. (© Wiley-VCH Verlag GmbH & Co. KGaA, 69451 Weinheim, Germany, 2006)

Introduction

After the discovery of clinically highly useful β-lactam antibiotics such as penicillins, cephalosporins and monobactams,^[1] the past few decades have witnessed a remarkable growth in the field of β -lactam chemistry. Apart from being antibacterial agents, these compounds have been found to be useful synthons for the preparation of a variety of molecules of biological and medicinal interest.^[2] Very recently, β-lactam antibiotics have been shown to offer neuroprotection by increasing expression of glutamate transporters through gene activation,^[3] whilst in addition, the discoveries of new biologically active β-lactams such as cholesterol acyl transferase inhibitors,^[4] thrombin inhibitors,^[5] human cytomegalovirus protease inhibitors,[6] matrixmetalloprotease inhibitors.^[7] human leukocyte elastase.^[8] cvsteine protease^[9] and apoptosis inductors^[10] have provided motivation for the development of new β -lactam systems. In particular, spirocyclic β -lactams have become centres of attraction as they behave as β -turn mimetics^[11] and precursors of α, α -disubstituted β -amino acids.^[12] In addition, the spiranic β -lactam moiety is also present in the chartellins, a family of marine natural products.^[13] Among more specific examples of spiro-β-lactams, I has been found to be an inhibitor of both poliovirus and human rhinovirus 3C-proteinases,^[14] II exhibits cholesterol absorption inhibiting activity,^[15] whereas the proline-derived spiro- β -lactam III serves as an efficient β -turn nucleator (Figure 1).^[16]



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Figure 1. Biologically active spiro-β-lactams.

Several syntheses of spiro- β -lactams have been described in the literature,^[17] and many researchers have in recent years accomplished the synthesis of spiro- β -lactams through cycloaddition reactions employing different ketenes and imines.^[18] Recently, Alcaide et al.^[19] have developed a metal-assisted synthesis of enantiopure spirocyclic



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 β -lactams from azetidin-2-ones. In view of the ever growing applications of spiro-*β*-lactams, ranging from their biological activity to their utility as synthetic intermediates in organic synthesis, we envisaged syntheses of novel spiro-βlactams through halogen-mediated intrasulfenyl additions to alkynes/alkenes through the use of cis-3-(prop-2-ynyloxy/-enyloxy)-β-lactams (3, 4; Scheme 1). Such spiro-β-lactams are not easily accessible by the classical ketene-imine cycloaddition, the Staudinger reaction, whilst it is well known that intramolecular addition of a heteronucleophile to a carbon-carbon double or triple bond in the presence of an electrophilic reagent is one of the most fundamental methods for construction of heterocyclic rings.^[20] Thus, in continuation to our efforts relating to the C-3 functionalization of β -lactams through the use of β -lactam carbocation equivalents of type $2^{[21-26]}$ we report here a novel, operationally simple and efficient approach for the synthesis of spiro-β-lactams.

Results and Discussion

The β -lactams **1**a–e required for this study were prepared from benzylthioethanoic acid and the appropriate Schiff's bases by the reported procedure.^[24,27] Furthermore, the trans-3-benzylthio-3-chloroazetidin-2-ones 2a-e, the appropriate β -lactam carbocation equivalents, were obtained by stereospecific chlorination of 1a-e by treatment with Nchlorosuccinimide (NCS) in the presence of catalytic amounts of AIBN by the reported procedure.^[24-25] On treatment^[25,27] with propargyl alcohol or allyl alcohol in the presence of $ZnCl_2/SiO_2$ in chloroform at reflux, these β lactam carbocation equivalents 2a-e afforded the cis-3-benzylthio-3-(prop-2-ynyloxy/-enyloxy)azetidin-2-ones 3а-е and 4a-b, considered potentially capable of undergoing halogen-mediated intrasulfenyl cyclization reactions (Scheme 1, Table 1).



Scheme 1. Synthesis of *cis*-3-benzylthio-3-(prop-2-ynyloxy/-enyl-oxy)azetidin-2-ones **3** and **4**.

The structures of the *cis*-3-(prop-2-ynyloxy/-enyloxy)- β -lactams **3a**–e and **4a–b** were established spectroscopically, by IR, ¹H NMR, ¹³C NMR, and DEPT 135 ¹³C NMR techniques. However, the stereochemistry of **3** and **4** at C-3 was tentatively assigned as *cis*, in view of the structure assignment of *cis*-3-benzylthio-3-methoxyazetidin-2-ones in single-crystal X-ray crystallographic studies.^[25]

The initial studies of halogen-mediated intrasulfenyl addition reactions were carried out by treatment of *cis*-3-(prop-2-ynyloxy)- β -lactam (**3a**) with one equiv. of iodine in dichloromethane at room temperature. The reaction resulted in the exclusive formation of the five-membered ring, affording a spiro product 7-iodomethylene-2-(4-methoxyphenyl)-3-phenyl-5-oxa-8-thia-2-azaspiro[3.4]octan-1one (**5a**). To investigate the role of halogen and selectivity in product formation, this reaction was performed with replacement of the iodine with bromine, resulting in a similar product profile, although it was observed that with bromine

Table 1. cis-3-Benzylthio-3-(prop-2-ynyloxy/-enyloxy)azetidin-2-ones 3a-e and 4a-b.

Entry	Substrate 2	r ³ oh	R ¹	R ²	Product	% Yield ^[a]
1	2a	€ОН	Ph	4-MeO-C ₆ H ₄	3a	76
2	2 b	€ОН	4-MeO-C ₆ H ₄	4-MeO-C ₆ H ₄	3b	57
3	2c	ОН	Ph	Ph	3e	74
4	2d	€ОН	Ph	$4-Cl-C_6H_4$	3d	71
5	2e	€OH	4-Cl-C ₆ H ₄	4-Me-C ₆ H ₄	3e	54
6	2a	Solution	Ph	4-MeO-C ₆ H ₄	4a	84
7	2b	S∽он	4-MeO-C ₆ H ₄	4-MeO-C ₆ H ₄	4b	62

[a] Yields quoted are for the isolated products characterized by IR, ¹H NMR, and ¹³C NMR spectroscopy.

the reaction proceeded more slowly, giving a lower yield of product than had been obtained in the reaction using iodine. The reaction was found to be general with several substrates (Scheme 2), and the results are summarized in Table 2.



Scheme 2. Synthesis of spiro-β-lactams 5a-e and 6a-e.

Table 2. Spiroazetidin-2-ones 5a-e and 6a-e.

Entry	Subs	trate	Product (% yield)[a]		
	3	\mathbb{R}^1	\mathbb{R}^2	5 (X = I)	$6 (\mathbf{X} = \mathbf{B}\mathbf{r})$
1	3a	Ph	4-MeO-C ₆ H ₄	5a (81)	6a (74)
2	3b	4-MeO-C ₆ H ₄	4-MeO-C ₆ H ₄	5b (65)	6b (54)
3	3c	Ph	Ph	5c (69)	6c (61)
4	3d	Ph	$4-Cl-C_6H_4$	5d (70)	6d (69)
5	3e	$4-Cl-C_6H_4$	4-Me-C ₆ H ₄	5e (61)	6e (46)

[a] Yields quoted are for the isolated products characterized by IR, $^1\mathrm{H}$ NMR, $^{13}\mathrm{C}$ NMR and MS.

The structures of spiro- β -lactams **5a–e** and **6a–e** were established by spectroscopic means such as IR, ¹H NMR, MS, ¹³C NMR, DEPT 135 ¹³C NMR, proton-proton COSY (¹H, ¹H COSY), and proton-carbon COSY (¹H, ¹³C COSY). Finally, the exclusive formation of the five-membered ring cycloadducts and the stereochemistry at the C-3 spiro junctions in **5a–e** and **6a–e** were established through single-crystal X-ray crystallographic analysis of **6d** (Figure 2).

A plausible mechanism for the formation of the fivemembered ring spiro- β -lactams is presented in Scheme 3, with the substrate 3 cyclizing to give exclusively the fivemembered ring cycloadduct by a 5-exo ring-closure process, rather than 6-endo ring-closure. The 5-exo-dig cyclizations appear to be kinetically controlled and are believed to proceed through the addition of a nucleophilic sulfur to electrophilically activated triple/double bonds.^[28] The reaction first involves the coordination by halogen to the triple bond of β -lactam 3 to produce a π complex (A). Subsequently, nucleophilic addition of the sulfide centre to the halogenolefin complex gives the cyclic sulfonium ion (C), which undergoes further dealkylation by halide ion to produce benzyl halide and the five-membered ring spiro[3.4]-β-lactams 5 and 6. The faster rate of ring-closure for the 5-exodig pathway than for the 6-endo-dig mode may, in part, be



Figure 2. An ORTEP view of compound 6d.

due to the internal *p*-lobe of the acetylene- $X_2 \pi$ complex (**B**) having a more favourable alignment with the lone pair of the incoming sulfur nucleophile. No six-membered ring regioadducts are formed in these halogen-mediated intrasulfenyl reactions of alkynyloxy substrates **3a**–e, which is reasonable to accept, since equilibrium between five- and six-membered ring isomers (**5**, **6** and **7**) would necessarily proceed through an unsaturated episulfonium intermediate (**D**), which is quite unlikely to be formed (Scheme 3).

In continuation of this study, we further examined the behaviour of the *cis*-3-benzylthio-3-(prop-2-enyloxy)azetidin-2-ones **4a–b** under the same conditions as described above (Scheme 4). Initially, **4a** was treated with one equiv. of bromine in dichloromethane at room temperature, resulting in the formation of a mixture of two diastereomeric five-membered ring spiro- β -lactams **8a** and **9a** as the major products, along with a single isomer of the six-membered ring spiro- β -lactam **10a** as a minor one.

Furthermore, the major isomeric cyclized five-membered ring adducts **8a** and **9a** were formed in 1:1 ratio, as was evident from ¹H NMR spectroscopy. One of the isomers – **8a** – crystallized in pure form from dichloromethane/hexanes and was identified as 7*a*-bromomethyl-2-(4-methoxyphenyl)-3-phenyl-5-oxa-8-thia-2-azaspiro[3.4]octan-1one, whereas the other isomer **9a** remained as a liquid product and was identified as 7 β -bromomethyl-2-(4-methoxyphenyl)-3-phenyl-5-oxa-8-thia-2-azaspiro[3.4]octan-1one. Similar results were obtained with iodine as the halogenating reagent, although the reaction proceeded very slowly relative to the bromination reaction. The reaction was also found to give the same profile of products with the other substrate **4b** and results are summarized in Table 3.

The isomeric spiro- β -lactams 8 and 11 were separated by crystallization, whereas the other isomeric spiro- β -lactams 9 and 12 remained as liquids. The six-membered spiro- β -lactams 10 and 13 were easily separated by chromatography and the structures of the spiro- β -lactams 8–13(a–b) were established by spectroscopic techniques such as IR, ¹H NMR, MS, ¹³C NMR, DEPT 90 ¹³C NMR, DEPT 135 ¹³C NMR, proton–proton COSY (¹H, ¹H COSY) and hetero-nuclear single quantum correlation (HSQC). In addition,



Scheme 3. Plausible mechanism for the formation of spiro- β -lactams **5a**–e and **6a**–e.



Scheme 4. Synthesis of spiroazetidin-2-ones 8a-b to 13a-b.

double irradiation studies and NOE measurements confirmed the assignments made. The formation of five-membered ring spiro- β -lactams and the stereochemical assignments at the C-7 and C-3 spiro junctions in 8 and 11 were established by single-crystal X-ray crystallographic analysis of 8a (Figure 3). The geometries of the 9 and 12 epimers were tentatively assigned by comparison of their data with those relating to 8 and 11, respectively, whilst the steric arrangements of the six-membered ring spiro- β -lactams 10 and 13 were tentatively assigned on the basis of various NMR spectral studies.



Figure 3. An ORTEP view of compound 8a.

The formation of five- and six-membered ring spiro- β -lactams in this case can be explained by the plausible

Table 3. Synthesis of spiroazetidin-2-ones 8a-b to 13a-b.

Entry	Substrate				Five-membere	ed ring cycloadduct	Six-membered ring cycloadduct	
	4	X_2	\mathbb{R}^1	R ²	α-isomer (% yield) ^[a]	β-isomer (% yield) ^[a]	α-isomer (% yield) ^[a]	
1	4a	Br ₂	Ph	4-MeO-C ₆ H ₄	8a (35)	9a (32)	10a (16)	
2	4b	Br_2	$4-MeO-C_6H_4$	$4-\text{MeO-C}_6\text{H}_4$	8b (27)	9b (25)	10b (12)	
3	4a	I_2	Ph	$4-\text{MeO-C}_6\text{H}_4$	11a (31)	12a (29)	13a (13)	
4	4b	I_2	$4-MeO-C_6H_4$	$4-MeO-C_6H_4$	11b (24)	12b (21)	13b (10)	

[a] Yields quoted are for the isolated products characterized by IR, ¹H NMR, ¹³C NMR and MS.



Scheme 5. Plausible reaction pathway for the formation of spiro- β -lactams 8a-b to 13a-b.

mechanism presented in Scheme 5. The reaction is believed to proceed by a similar pathway as for the spiro- β -lactams 5 and 6.

The formation of the five-membered ring spiro[3.4]- β lactams **8** as major products suggests that the difference in the thermodynamic stabilities of the five- and six-membered ring cycloadducts must be significantly large to force the equilibrium in the direction of the five-membered ring. It would appear that the exocyclic positioning of the halomethyl moiety on the five-membered ring is energetically more favourable than having the halo group on the sixmembered ring cycloadduct. It is further believed that the attack of the halogen on the episulfonium intermediate (**D**) occurs from the less hindered side, avoiding any steric repulsion with the phenyl group on C-4, thus favouring the formation of the α -isomer of the six-membered ring spiro[3.5]- β -lactam.

These studies have revealed some interesting features of the halocyclization reactions of cis-3-alkoxy- β -lactams 3 and 4. The regio- and stereochemistries of the products formed after the intramolecular addition of the heteronucleophile depend on the type of unsaturation within the substrate. Initiated by a halogen, the alkynyloxy and alkenyloxy sulfides cyclize to give the five-membered ring products by a 5-exo-ring closure process. The regiospecificities of these ring closures may in part be due to a kinetic preference for formation of the five-membered rings; these results are consistent with observations reported by Turos et al.^[28] Moreover, the type of unsaturation in the substrate dictates which halogenating reagent should be used as the initiating agent. For alkynyloxy sulfide cyclizations, iodine generally gives better results than bromine because the iodocycloadducts are stable to the reaction conditions and can be obtained in quantitative yields. On the other hand, bromine is preferred over iodine for promoting the cyclization of alkenvloxy sulfides because the bromination reaches completion more rapidly.

Conclusion

In conclusion, a facile route to novel spiro- β -lactams through intramolecular halogen-mediated cyclization reactions of *cis*-3-benzylthio-3-(prop-2-ynyloxy/-enyloxy)azeti-din-2-ones has been developed. Furthermore, the alk-ynyloxy sulfide ring closures result in the exclusive formation of five-membered ring spiro[3.4]- β -lactams, whereas alkenyloxy sulfide ring closures result in the formation of mixtures of five-membered ring spiro[3.4]- β -lactams as the major products along with six-membered ring spiro[3.5]- β -lactams as minor products.

Experimental Section

General: Compounds 1a-e, 2a-e,^[24] 3a-e and 4a-b^[25] were prepared as described previously. Spectroscopic data for compounds 1a-b and 2a-b^[24] were reported previously. NMR spectra were recorded on Jeol 300 and Bruker Avance II 400 spectrometers. Chemical shifts are given in ppm relative to tetramethylsilane as an internal standard ($\delta = 0$ ppm) for ¹H NMR and CDCl₃ ($\delta =$ 77.0 ppm) for ¹³C NMR spectra. IR spectra were measured on a Perkin-Elmer 430 FTIR spectrometer. MS spectra were recorded on a Shimadzu GCMS-QT 5000 instrument and elemental analyses (C,H,N) were performed on a Perkin-Elmer 2400 elemental analyzer. Column chromatography was carried out on silica gel (60-120 mesh, Merck). Thin-layer chromatography (TLC) was performed on silica gel G (Merck). For visualization, TLC plates were stained with iodine vapour. Melting points are uncorrected. All commercially available compounds/reagents were used without further purification. Dichloromethane, carbon tetrachloride and chloroform, distilled over P2O5, were redistilled over CaH2 before use.

trans-3-Benzylthio-1,4-diphenylazetidin-2-one (1c): Yield 1.500 g (40%), white solid flakes, m.p. 136–137 °C (dec.). ¹H NMR (300 MHz, CDCl₃): δ = 7.31–7.00 (m, 15 H, ArH), 4.51 (d, *J* = 2.1 Hz, 1 H, C3-H), 3.95 (d, *J* = 4.8 Hz, 2 H, CH₂S), 3.85 (d, *J* = 2.4 Hz, 1 H, C4-H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 162.1, 137.4, 136.8, 129.6, 129.3, 129.1 (2), 128.9, 128.6, 128.5, 128.3, 128.2, 128.1, 127.6, 127.2, 125.7, 123.9, 117.1, 63.0, 58.9, 35.2 ppm.

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IR (KBr): $\tilde{v}_{max} = 1739$ (C=O) cm⁻¹. C₂₂H₁₉NOS (345.37): calcd. C 76.49, H 5.54, N 4.05; found C 76.31, H 5.43, N 3.99.

trans-3-Benzylthio-3-chloro-1,4-diphenylazetidin-2-one (2c): Yield 0.350 g (53%), white solid flakes, m.p. 147–148 °C (dec.). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.37$ –7.06 (m, 15 H, ArH), 5.42 (s, 1 H, C4-H), 4.29 (d, J = 11.4 Hz, 1 H, CH_aH_bS), 3.99 (d, J = 11.2 Hz, 1 H, CH_aH_bS) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 160.3$, 136.7, 135.5, 131.5 (2), 129.7, 129.5 (2), 129.3 (2), 128.6 (2), 128.5, 127.9 (2), 127.5 (2), 124.8, 117.8 (2), 113.8, 80.7, 71.5, 34.9 ppm. IR (KBr): $\tilde{v}_{max} = 1769$ (C=O) cm⁻¹. C₂₂H₁₈ClNOS (379.89): calcd. C 69.56, H 4.77, N 3.69; found: C 69.45, H 4.68, N 3.62.

cis-3-Benzylthio-1-(4-methoxyphenyl)-4-phenyl-3-(prop-2-ynyloxy)azetidin-2-one (3a): Yield 0.040 g (76%), brownish-yellow oil. ¹H NMR (300 MHz, CDCl₃): δ = 7.35–6.69 (m, 14 H, ArH), 5.34 (s, 1 H, C4-H), 4.50 (dd, *J* = 2.4, 2.4 Hz, 1 H, CH_aH_bO), 4.22 (dd, *J* = 2.7, 2.4 Hz, 1 H, CH_aH_bO), 3.87 (d, *J* = 12.2 Hz, 1 H, CH_aH_bS), 3.71 (s, 3 H, OCH₃), 3.69 (d, *J* = 12.2 Hz, 1 H, CH_aH_bS), 2.47 (t, 1 H, HC=) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 161.2, 156.4, 137.4, 133.1, 130.5, 129.2, 128.9, 128.4, 128.1, 127.0, 119.0, 114.4, 98.2, 79.9, 75.4, 67.7, 55.1, 52.8, 32.4 ppm. ¹³C NMR (DEPT 135) (75.5 MHz, CDCl₃): δ = 129.2 (+), 128.9 (+), 128.4 (+), 128.1 (+), 127.0 (+), 119.0 (+), 114.4 (+), 79.9 (+), 75.4 (+), 67.7 (+), 55.1 (+), 52.8 (-), 32.4 (-) ppm. IR (CHCl₃): \tilde{v}_{max} = 1760 (C=O) cm⁻¹. C₂₆H₂₃NO₃S (429.51): calcd. C 72.71, H 5.39, N 3.26; found: C 72.67, H 5.33, N 3.29.

cis-3-Benzylthio-1-(4'-methoxyphenyl)-4-phenyl-3-(prop-2-enyloxy)azetidin-2-one (4a): Yield 0.044 g (84%), yellowish-white solid, m.p. 70–71 °C (dec.). ¹H NMR (300 MHz, CDCl₃): δ = 7.31–6.70 (m, 14 H, ArH), 5.92–5.81 (m, 1 H, H₂C=CH–), 5.37–5.30 (m, 1 H, CH_aH_bO), 5.21–5.17 (m, 1 H, CH_aH_bO), 5.10 (s, 1 H, C4-H), 4.42– 4.35 (m, 1 H, $H_bH_aC=CH_{-}$), 4.26–4.19 (m, 1 H, $H_bH_aC=CH_{-}$), 3.88 (d, J = 12.3 Hz, 1 H, CH_aH_bS), 3.70 (s, 3 H, OCH₃), 3.66 (d, J = 12.3 Hz, 1 H, CH_aH_bS) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 162.3, 156.3, 137.2, 133.3, 133.2, 132.6, 130.2, 129.1, 128.9,$ 128.4, 128.2, 127.9, 127.0, 119.0, 117.8, 114.3, 77.2, 68.0, 66.3, 55.4, 32.2, 28.0 ppm. ¹³C NMR (DEPT 135) (75.5 MHz, CDCl₃): δ = 130.2 (+), 129.1 (+), 128.9 (+), 128.4 (+), 128.2 (+), 127.9 (+), 127.0 (+), 119.0 (+), 117.8 (-), 114.3 (+), 68.0 (+), 66.3 (-), 55.4 (+), 32.2 (-), 28.0 (+) ppm. IR (CHCl₃): $\tilde{v}_{max} = 1762$ (C=O) cm⁻¹. $C_{26}H_{25}NO_3S$ (431.53): calcd. C 72.36, H 5.83, N 3.24; found: C 72.27, H 5.79, N 3.21.

Preparation of Spiro-β-lactams 5-13. General Procedure: Iodine/bromine (1.2 mmol) was added at room temperature to a stirred solution of the *cis*-3-(prop-2-ynyloxy)-β-lactam **3** or the *cis*-3-(prop-2enyloxy)-β-lactam **4** (1.0 mmol) in dry dichloromethane (10 mL). The reaction mixture was stirred for 4–7 h, the progress of the reaction being monitored by TLC. After TLC indicated complete consumption of the starting substrate, the reaction mixture was poured into aqueous 5% Na₂S₂O₃/5% Na₂S₂O₅ solution (10 mL) and stirred until the purplish coloration of iodine/reddish coloration of bromine dissipated. The aqueous mixture was extracted with dichloromethane (3×10 mL) and the combined organic layer was washed with brine (2×5 mL) and dried with anhydrous Na₂SO₄. After evaporation of the solvent under vacuum, the residue was purified by column chromatography on silica gel in hexanes/ethyl acetate (93:7).

7-Iodomethylene-2-(4-methoxyphenyl)-3-phenyl-5-oxa-8-thia-2-aza-spiro[3.4]octan-1-one (5a): Yield 0.044 g (81%), white solid flakes, m.p. 128–129 °C (dec.). ¹H NMR (300 MHz, CDCl₃): δ = 7.25–6.69 (m, 9 H, ArH), 5.74 (t, 1 H, ICH), 5.23 (s, 1 H, C3-H), 4.79 (dd, J = 2.1, 2.4 Hz, 1 H, CH_aH_bO), 4.75 (dd, J = 2.4, 2.1 Hz, 1 H, CH_aH_bO), 3.73 (s, 3 H, OCH₃) ppm. ¹³C NMR (75.5 MHz,

CDCl₃): δ = 163.0, 156.7, 142.5, 134.0, 130.1, 129.4, 129.1, 126.5, 119.0, 114.4, 106.7, 79.5, 67.5, 58.5, 55.2 ppm. ¹³C NMR (DEPT 135) (75.5 MHz, CDCl₃): δ = 129.4 (+), 129.1 (+), 126.5 (+), 119.0 (+), 114.4 (+), 79.5 (-), 67.5 (+), 58.5 (+), 55.2 (+) ppm. IR (KBr): \tilde{v}_{max} = 1754 (C=O) cm⁻¹. MS (70eV, EI): *m*/*z*: 465 (29) [M]⁺, 316 (100), 211 (65), 196 (55), 189 (24), 167 (26), 161 (22), 149 (9), 128 (31), 118 (68), 90 (46), 77 (24), 45 (48). C₁₉H₁₆INO₃S (465.29): calcd. C 49.05, H 3.46, N 3.01; found: C 48.98, H 3.39, N 2.93.

7-Bromomethylene-2-(4-chlorophenyl)-3-phenyl-5-oxa-8-thia-2-aza-spiro[3.4]octan-1-one (6d): Yield 0.037 g (69%), colourless, crystal-line solid, m.p. 142–143 °C (dec.). ¹H NMR (300 MHz, CDCl₃): δ = 7.44–7.19 (m, 9 H, ArH), 5.81 (t, 1 H, BrCH), 5.27 (s, 1 H, C3-H), 4.93 (dd, J = 2.4, 2.4 Hz, 1 H, CH_aH_bO), 4.84 (dd, J = 2.7, 2.4 Hz, 1 H, CH_aH_bO) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 163.5, 138.7, 135.2, 133.6, 130.1, 129.7, 129.4, 129.3, 126.5, 118.9, 106.2, 90.6, 76.2, 67.2 ppm. ¹³C NMR (DEPT 135) (75.5 MHz, CDCl₃): δ = 129.7 (+), 129.4 (+), 129.3 (+), 126.5 (+), 118.9 (+), 90.6 (+), 76.2 (-), 67.2 (+) ppm. IR (KBr): \tilde{v}_{max} = 1771 (C=O) cm⁻¹. C₁₈H₁₃BrClNO₂S (422.71): calcd. C 51.14, H 3.10, N 3.31; found: C 51.06, H 3.03, N 3.24.

7α-Bromomethyl-2-(4-methoxyphenyl)-3-phenyl-5-oxa-8-thia-2-azaspiro[3.4]octan-1-one (8a): Yield 0.022 g (35%), colourless, crystalline solid, m.p. 148–149 °C (dec.). $R_{\rm f}$ (8a) = 0.40, $R_{\rm f}$ (4a) = 0.45. ¹H NMR (400 MHz, CDCl₃): δ = 7.39–6.78 (m, 9 H, ArH), 5.24 (s, 1 H, C3-H), 4.52 (dd, J = 1.8, 1.5 Hz, 1 H, CH_aH_bO), 4.16– 4.12 (m, 1 H, $CH_{\alpha}H_{\beta}O$), 3.80–3.74 (m, 1 H, $CH_{\beta}S$), 3.74 (s, 3 H, OCH₃), 3.72 (dd, J = 1.8, 1.5 Hz, 1 H, BrCH_aH_b), 3.54–3.52 (m, 1 H, BrCH_aH_b) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 164.3, 156.6, 134.5, 130.1, 129.2, 129.1, 126.8, 119.3, 114.5, 103.5, 75.5, 70.0, 55.5, 50.8, 33.1 ppm. ¹³C NMR (DEPT 135) (100 MHz, $CDCl_3$): $\delta = 129.2 (+), 129.1 (+), 126.8 (+), 119.3 (+), 114.5 (+),$ 75.5 (-), 70.0 (+), 55.5 (+), 50.8 (+), 33.1 (-) ppm. ¹³C NMR (DEPT 90) (100 MHz, CDCl₃): δ = 129.2 (+), 129.1 (+), 126.8 (+), 119.3 (+), 114.5 (+), 70.0 (+), 50.8 (+) ppm. IR (KBr): $\tilde{v}_{max} = 1752$ $(C=O) \text{ cm}^{-1}$. MS (70eV, EI): m/z: 421 (31), 419 (20) $[M]^+$, 272 (100), 271 (97), 211 (75), 196 (57), 167 (17), 121 (28), 118 (78), 90 (25), 88 (19), 77 (21), 73 (44), 44 (33), 41 (97). C₁₉H₁₈NO₃BrS (420.30): C 54.30, H 4.31, N 3.33; found: C 54.23, H 4.25, N 3.29.

7β-Bromomethyl-2-(4-methoxyphenyl)-3-phenyl-5-oxa-8-thia-2-aza-spiro[3.4]octan-1-one (9a): Yield 0.020 g (32%), yellow oil. $R_{\rm f}$ (**9a**) = 0.40, $R_{\rm f}$ (**4a**) = 0.45. ¹H NMR (400 MHz, CDCl₃): δ = 7.41–6.77 (m, 9 H, ArH), 5.25 (s, 1 H, C3-H), 4.53 (dd, J = 1.8, 1.5 Hz, 1 H, CH_aH_βO), 4.40–4.37 (m, 1 H, CH_aH_βO), 3.83–3.81 (m, 1 H, CH_aS), 3.75–3.74 (m, 1 H, BrCH_aH_β), 3.74 (s, 3 H, OCH₃), 3.21 (dd, J = 1.8, 1.5 Hz, 1 H, BrCH_aH_β) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 164.2, 156.6, 135.1, 130.3, 129.1, 126.6, 119.1, 114.4, 104.0, 74.9, 68.2, 55.2, 49.6, 32.9 ppm. ¹³C NMR (DEPT 135) (100 MHz, CDCl₃): δ = 129.1 (+), 126.6 (+), 119.1 (+), 114.4 (+), 74.9 (-), 68.2 (+), 55.2 (+), 49.6 (+), 32.9 (-) ppm. IR (CHCl₃): $\tilde{\nu}_{max}$ = 1765 (C=O) cm⁻¹. C₁₉H₁₈BrNO₃S (420.30): C 54.30, H 4.31, N 3.33; found: C 54.19, H 4.22, N 3.24.

7α-Bromo-2-(4-methoxyphenyl)-3-phenyl-5-oxa-9-thia-2-azaspiro-[3.5]nonan-1-one (10a): Yield 0.010 g (16%), white solid flakes, m.p. 146–147 °C (dec.). $R_{\rm f}$ (**10a**) = 0.50, $R_{\rm f}$ (**4a**) = 0.45. ¹H NMR (400 MHz, CDCl₃): δ = 7.33–6.74 (m, 9 H, ArH), 4.90 (s, 1 H, C3-H), 4.61 (t, 1 H, CH_αH_βO), 4.34–4.24 (m, 1 H, BrCH_β), 4.20–4.14 (m, 1 H, CH_αH_βO), 3.90 (t, 1 H, CH_αH_βS), 3.72 (s, 3 H, OCH₃), 2.98–2.92 (m, 1 H, CH_αH_βS) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 162.4, 156.5, 132.0, 130.3, 129.4, 128.6, 127.6, 118.9, 114.5, 91.4, 71.1, 67.6, 55.2, 43.0, 32.9 ppm. ¹³C NMR (DEPT 135) (100 MHz, CDCl₃): δ = 129.4 (+), 128.6 (+), 127.6 (+), 118.9 (+), 114.5 (+), 71.1 (–), 67.6 (+), 55.2 (+), 43.0 (+), 32.9 (–) ppm. IR (CHCl₃): $\tilde{v}_{max} = 1756$ (C=O) cm⁻¹. C₁₉H₁₈BrNO₃S (420.30): C 54.30, H 4.31, N 3.33; found: C 54.18, H 4.20, N 3.26.

7α-Iodomethyl-2-(4-methoxyphenyl)-3-phenyl-5-oxa-8-thia-2-aza-spiro[3.4]octan-1-one (11a): Yield 0.020 g (31%), white solid flakes, m.p. 159–160 °C (dec.). $R_{\rm f}$ (**11a**) = 0.42, $R_{\rm f}$ (**4a**) = 0.45. ¹H NMR (300 MHz, CDCl₃): δ = 7.41–6.75 (m, 9 H, ArH), 5.19 (s, 1 H, C3-H), 4.83 (dd, J = 1.8, 1.5 Hz, 1 H, CH_aH_βO), 4.23–4.20 (m, 1 H, CH_aH_βO), 3.84–3.78 (m, 1 H, CH_βS), 3.75 (s, 3 H, OCH₃), 3.73 (dd, J = 1.8, 1.8 Hz, 1 H, ICH_aH_β), 3.43–3.39 (m, 1 H, ICH_aH_β) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 164.5, 156.4, 134.7, 134.4, 130.0, 129.1, 126.6, 119.1, 114.3, 103.6, 77.4, 69.8, 55.3, 51.0, 29.1 ppm. ¹³C NMR (DEPT 135) (75.5 MHz, CDCl₃): δ = 129.1 (+), 126.6 (+), 119.1 (+), 114.3 (+), 77.4 (-), 69.8 (+), 55.3 (+), 51.0 (+), 29.1 (-) ppm. IR (CHCl₃): \tilde{v}_{max} = 1763 (C=O) cm⁻¹. C₁₉H₁₈NO₃IS (467.30): C 48.83, H 3.87, N 2.99; found: C 48.77, H 3.82, N 2.92.

7β-Iodomethyl-2-(4-methoxyphenyl)-3-phenyl-5-oxa-8-thia-2-aza-spiro[3.4]octan-1-one (12a): Yield 0.018 g (29%), yellow oil. R_f (**12a**) = 0.42, R_f (**4a**) = 0.45. ¹H NMR (300 MHz, CDCl₃): δ = 7.41–6.75 (m, 9 H, ArH), 5.21 (s, 1 H, C3-H), 4.43 (dd, J = 1.4, 1.4 Hz, 1 H, CH_αH_βO), 4.12–4.09 (m, 1 H, CH_αH_βO), 3.94–3.88 (m, 1 H, CH_αS), 3.75 (s, 3 H, OCH₃), 3.69–3.62 (m, 1 H, ICH_αH_β), 3.01 (dd, J = 1.8, 1.5 Hz, 1 H, ICH_αH_β) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 164.3, 156.4, 134.8, 134.3, 130.1, 129.0, 126.6, 119.0, 114.3, 104.2, 77.1, 68.5, 55.1, 50.0, 28.9 ppm. ¹³C NMR (DEPT 135) (75.5 MHz, CDCl₃): δ = 129.0 (+), 126.6 (+), 119.0 (+), 114.3 (+), 77.1 (-), 68.5 (+), 55.1 (+), 50.0 (+), 28.9 (-) ppm. IR (CHCl₃): $\tilde{\nu}_{max}$ = 1768 (C=O) cm⁻¹. C₁₉H₁₈INO₃S (467.30): C 48.83, H 3.87, N 2.99; found: C 48.71, H 3.75, N 2.86.

 7α -Iodo-2-(4-methoxyphenyl)-3-phenyl-5-oxa-9-thia-2-azaspiro[3.5]nonan-1-one (13a): Yield 0.008 g (13%), colourless oil. $R_{\rm f}$ (13a) =

Table 4. Crystallographic data for 6d and 8a.

Compound	6d	8a
Empirical formula	C ₁₈ H ₁₃ BrClNO ₂ S	C ₁₉ H ₁₈ BrNO ₃ S
Formula mass	422.71	420.31
Temperature [K]	293(2)	293(2)
Crystal size [mm]	$0.25 \times 0.19 \times 0.18$	$0.21 \times 0.18 \times 0.11$
Crystal system	monoclinic	monoclinic
Space group	$P2_{1}/c$	$P2_{1}/c$
a [Å]	11.201(1)	10.8578(12)
<i>b</i> [Å]	16.014(2)	17.0488(19)
<i>c</i> [Å]	9.978(1)	9.8850(11)
a [°]	90	90
β [°]	99.37(1)	98.437(2)
γ [°]	90	90
V [Å ³]	1765.9(15)	1810.0(3)
Ζ	4	4
$d_{\rm c} [{\rm mg}{\rm m}^{-3}]$	1.590	1.542
F(000)	848	856
μ (Mo- K_{α}) [mm ⁻¹]	2.607	2.404
θ range [°]	1.84 to 24.00	2.24 to 25.00
Index range	$-12 \le h \le 12,$	$-5 \le h \le 12,$
	$-18 \le k \le 0,$	$-19 \le k \le 20,$
	$-11 \leq l \leq 0$	$-11 \leq l \leq 11$
No. of reflns. collected	2927	8721
No. of independent reflns.	2752	3174
	$[R_{\rm int} = 0.0238]$	$[R_{\rm int} = 0.0219]$
No. of reflues. with $I > 2\sigma(I)$	1741	2805
$R_1; wR_2 [I > 2\sigma(I)]$	0.0566; 0.1312	0.0474; 0.1152
R_1 ; wR_2 (all data)	0.0990; 0.1535	0.0704; 0.1284
Data/restraints/parameters	2752/0/217	317/4/0/226
GOF on F^2	1.014	1.034
Largest diff. peak/hole [e·A ⁻³]	0.436/-0.665	1.313/-0.832

0.55, $R_{\rm f}$ (**4a**) = 0.45. ¹H NMR (300 MHz, CDCl₃): δ = 7.27–6.68 (m, 9 H, ArH), 4.81 (s, 1 H, C3-H), 4.70 (t, 1 H, $CH_{\alpha}H_{\beta}O$), 4.45– 4.36 (m, 1 H, IC H_{β}), 4.19–4.14 (m, 1 H, $CH_{\alpha}H_{\beta}O$), 4.01 (t, 1 H, $CH_{\alpha}H_{\beta}S$), 3.68 (s, 3 H, OCH₃), 2.99-2.93 (m, 1 H, $CH_{\alpha}H_{\beta}S$) ppm. IR (CHCl₃): $\tilde{v}_{\rm max}$ = 1757 (C=O) cm⁻¹. $C_{19}H_{18}INO_3S$ (467.30): C 48.83, H 3.87, N 2.99; found: C 48.73, H 3.79, N 2.89.

X-ray Structure Determination: Data were collected on diffractometers: Siemens P4 (for **6d**) and Bruker SMART 1 K CCD (for **8a**). For details see Table 4. The data were collected by φ and ω scan mode and corrected by Lorentz and polarization factors, but no absorption correction was applied. The structure was solved by direct methods by use of the SHELX-97 program (G. M. Sheldrick, University of Göttingen, Germany, 1997) and also refined on F^2 by use of the same program suite.

CCDC-606118 and -606119 (for **6d** and **8a**, respectively) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Supporting Information (see also the footnote on the first page of this article): 2D NMR spectra (¹H,¹H COSY, ¹H,¹³C COSY, and HSQC) for compounds **5a**, **8a** and **10a**. Spectroscopic data for compounds **1d–e**, **2d–e**, **3b–e**, **4b**, **5b–e**, **6a–c**, **e**, **8b**, **9b**, **10b**, **11b**, **12b** and **13b**.

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