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TETRAHEDRON: ASYMMETRY

Highly enantioselective addition of diethylzinc to N-diphenylphosphinylimines using dendritic chiral ligands with hydrocarbon backbones

Itaru Sato, Ryo Kodaka, Takanori Shibata, Yutaka Hirokawa, Nobuaki Shirai, Koji Ohtake and Kenso Soai*

Department of Applied Chemistry, Faculty of Science, Science University of Tokyo, Kagurazaka, Shinjuku-ku, Tokyo, 162-8601 Japan

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Abstract

Chiral dendrimers bearing three or six chiral β -amino alcohols on the hyperbranched hydrocarbon chain-ends act as efficient chiral ligands for the enantioselective addition of diethylzinc to *N*-diphenylphosphinylimines to afford enantiomerically enriched *N*-diphenylphosphinylamines with up to 94% e.e. \mathbb{C} 2000 Elsevier Science Ltd. All rights reserved.

Dendrimer, an orderly hyperbranched macromolecule, is defined as one of the polymers with a certain molecular weight and a molecular architecture. Its characteristics are usually determined by the functional groups at the chain-ends.^{1–3} When the well-designed chiral functionalities are loaded on the chain-ends of the dendritic backbone, those functionalities are positioned on the surface and the resulting chiral dendrimer is expected to act as a chiral ligand in asymmetric synthesis. Dendritic chiral ligands are prominent in their features among other polymeric chiral ligands⁴ such as: (1) all of the chiral sites are placed on the surface of a dendrimer; and (2) individual chiral environments of these functionalities are approximately the same, while those of the chiral functionalities on other polymers are not. However, dendrimers modified by chiral functionalities at the terminal positions have rarely been utilized in asymmetric synthesis.^{5–7} Our recent report shows that the dendrimers loaded with chiral amino alcohols on hydrocarbon backbones can be used as effective chiral ligands in the enantioselective alkylation of aldehydes.⁸

Meanwhile, the enantioselective alkylation of imines represents one of the challenges in asymmetric synthesis.⁹ Compared with the asymmetric addition of dialkylzincs to aldehydes,¹⁰ an enantioselective addition of dialkylzincs to imines has rarely been examined. We previously reported the enantioselective alkylation of *N*-diphenylphosphinylimines in the presence of chiral

^{*} Corresponding author. Tel: +81-3-3260-4271; fax: +81-3-3235-2214; e-mail: ksoai@ch.kagu.sut.ac.jp

 β -amino alcohols which provides a direct and convenient method for the formation of enantiomerically enriched secondary *N*-diphenylphosphinylamines.^{11,12} Subsequent removal of the diphenylphosphinyl group by acid hydrolysis affords enantiomerically enriched secondary amines.¹³

We report here a highly enantioselective addition of diethylzinc (Et₂Zn) to *N*-diphenylphosphinylimines in the presence of dendritic chiral ligands possessing hydrocarbon backbones. Each of chiral ligands **1a** and **1b** bears three chiral sites on the *para*- and *meta*-positions to the backbone, respectively, and a second-generation chiral ligand **2** possesses six chiral β -amino alcohols on the phenylacetylene chain-end (Fig. 1).



Figure 1.



N-Diphenylphosphinylimines were treated with Et_2Zn in the presence of dendritic chiral ligands 1a,⁸ 1b,¹⁴ and 2^8 (Eq. (1)). All the reactions were carried out in toluene at room temperature. The results are summarized in Table 1. In the presence of *para*-substituted chiral ligand 1a (0.34 mol equiv.), *N*-diphenylphosphinylimine 3a was ethylated to give *N*-diphenylphosphinylamine (*R*)-4a with 89% e.e. (Entry 1).¹⁵ Enantioselective alkylations of *N*-diphenylphosphinylimines 3b and 3c in the presence of chiral ligand 1a gave the corresponding chiral (*R*)-amines 4b and 4c with 94 and 89% e.e., respectively (Entries 2 and 3). *N*-Diphenylphosphinylimine

3d with heteroaromatic ring was also ethylated to afford enantiomerically enriched (*R*)-4d (Entry 4). On the other hand, *meta*-substituted chiral dendrimer 1b drove asymmetric ethylation of *N*-diphenylphosphinylimines with 73 to 85% e.e. (Entries 5–7). The enantioselectivity of chiral ligand 1a was higher than that of 1b. This is probably due to the fact that the environments of active sites of 1a have enough space to work as a chiral ligand, while active sites of 1b are too close to the backbone chains. Highly enantioselective ethylation also proceeded using dendritic chiral ligand 2 bearing six amino alcohols. In the presence of 0.17 mol equiv. of 2, imine 3a was converted to (*R*)-amine 4a with 87% e.e. in good yield (Entry 8). The reaction was complete within 20 h. The rate of the reaction was almost the same as in the cases using 0.34 mol equiv. of dendritic ligand 1a (Entry 1) or 1.0 mol equiv. of *N*-benzylephedrine.¹⁵ The ligand 2 could be recovered (80%) by purification on silica gel TLC (Entry 8) and could be reused without a considerable change in the enantioselectivity (81% e.e.). Similarly, enantioselective addition of Et₂Zn to imines 3b and 3c gave amines 4b and 4c in 90 and 85% e.e., respectively (Entries 9 and 10).

Entrya R Mol. Equiv. Yield / % 1 phenyl $3a$ $1a$ 0.34 $4a$ 73 2 p -tolyl $3b$ $4b$ 77 3 2 -naphthyl $3c$ $4c$ 80 4 2 -furyl $3d$ $4d$ 77 5 phenyl $3a$ $1b$ 0.34 $4a$ 62 6 p -tolyl $3b$ $4b$ 70 70 70 7 2 -naphthyl $3c$ $4c$ 84 $8d$ 70 7 2 -naphthyl $3c$ $4c$ 84 70 9 p -tolyl $3b$ 2 0.17 $4a$ 77	• 4 ^b	R)-Amine 4	(1	Chiral ligand		Imine 3		
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	76 E.e. / %	Yield / %		Mol. Equiv.			R	Entry ^a
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	89	73	4 a	0.34	1a	3a	phenyl	1
3 2-naphthyl $3c$ $4c$ 80 4 2-furyl $3d$ $4d$ 77 5 phenyl $3a$ $1b$ 0.34 $4a$ 62 6 p -tolyl $3b$ $4b$ 70 7 2 -naphthyl $3c$ $4c$ 84 8 phenyl $3a$ 2 0.17 $4a$ 77 9 p -tolyl $3b$ $4b$ 79	94	77	4b			3b	<i>p</i> -tolyl	2
	89	80	4 c			3c	2-naphthyl	3
	71	77	4d			3d	2-furyl	4
6 p -tolyl 3b 4b 70 7 2-naphthyl 3c 4c 84 8 phenyl 3a 2 0.17 4a 77 9 p -tolyl 3b 4b 79	85	62	4 a	0.34	1b	3a	phenyl	5
7 2-naphthyl 3c 4c 84 8 phenyl 3a 2 0.17 4a 77 9 p-tolyl 3b 4b 79	83	70	4b			3b	<i>p</i> -tolyl	6
8 phenyl 3a 2 0.17 4a 77 9 <i>p</i> -tolyl 3b 4b 79	73	84	4c			3c	2-naphthyl	7
9 <i>p</i> -tolyl 3b 4b 79	87	77	4a	0.17	2	3a	phenyl	8
	90	79	4b			3b	<i>p</i> -tolyl	9
10 2-naphthyl 3c 4c 74	85	74	4 c			3c	2-naphthyl	10

 Table 1

 Enantioselective alkylation of various N-diphenylphosphinylimines 3a–d using dendritic chiral ligands 1a,b and 2

^a Reactions were run in toluene for 20-27 h at room temperature using 3.0 molar equiv. of diethylzinc.
 ^b Isolated yields. E.e. was determined by HPLC analysis using a chiral stationary phase. For the absolute configuration, see Ref. 11a.

Typical experimental procedure is as follows (Table 1, Entry 9): To a solution of *N*-diphenylphosphinylimine **3b** (63.9 mg, 0.20 mmol) and dendritic chiral ligand **2** (68.2 mg, 0.033 mmol) in toluene (4 ml) was added a 1 M toluene solution of Et_2Zn (0.6 ml, 0.6 mmol) at 0°C under an argon atmosphere. After the mixture was stirred at room temperature for 20 h, saturated aq. ammonium chloride (5 ml) was added. The mixture was filtered using Celite, and the filtrate was extracted with dichloromethane. The combined organic layer was dried over anhydrous sodium sulfate and evaporated. Purification of the residue by silica gel TLC (developing solvent, hexane:ethyl acetate:methanol = 5:4:1) gave *N*-diphenylphosphinylamine (*R*)-**4b** (51.1 mg, 79%). Enantiomeric excess was determined as 90% by HPLC analysis using a chiral stationary phase (Chiralcel OD). In conclusion, chiral dendrimers bearing three or six β -amino alcohols on the hyperbranched hydrocarbon chain-ends work as efficient chiral ligands for the enantioselective addition of diethylzinc to *N*-diphenylphosphinylimines. The corresponding *N*-diphenylphosphinylamines are obtained with up to 94% e.e.

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- 14. Dendritic chiral ligand **1b** was synthesized by the Sonogashira coupling [cat. PdCl₂(PPh₃)₂ cat. CuI, *i*-Pr₂NH, 50°C] between (1*R*,2*S*)-*N*-(3-ethynylbenzyl)ephedrine and 1,3,5-tribromobenzene; $[\alpha]_D^{24}$ +30.0 (*c* 1.0, CH₃OH).
- 15. The enantioselective addition of diethylzinc to imine 3a in the presence of 1.0 mol equiv. of (1R,2S)-*N*-benzylephedrine for 24 h gave the corresponding (*R*)-amine 4a with 92% e.e. in the yield of 91%. It should be noted that, in the reaction using chiral ligand 2, a lesser amount of chiral ligand 2 (0.17 mol equiv.) was utilized (Table 1, Entry 8).