

## Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for  
authors and subscription information:

<http://www.tandfonline.com/loi/lcyc20>

### A Convenient Route to Condensed-Ring Aromatic Acetylenes

Geoffrey T. Crisp<sup>a</sup> & Yu-Lin Jiang<sup>a</sup>

<sup>a</sup> Department of Chemistry, The University of  
Adelaide, Adelaide, S. A. Australia, 5005

Version of record first published: 20 Aug 2006.

To cite this article: Geoffrey T. Crisp & Yu-Lin Jiang (1998): A Convenient Route to Condensed-Ring Aromatic Acetylenes, *Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry*, 28:14, 2571-2576

To link to this article: <http://dx.doi.org/10.1080/00397919808004825>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or

up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## A CONVENIENT ROUTE TO CONDENSED-RING AROMATIC ACETYLENES

Geoffrey T. Crisp\* and Yu-Lin Jiang\*

Department of Chemistry, The University of Adelaide, Adelaide, S. A.  
Australia 5005

**Abstract:** Palladium catalysed coupling of aryl halides with 3-methylbut-1-yn-3-ol and a subsequent base-induced retro-Favorsky reaction gave the corresponding ethynyl aromatics in good to excellent yields.

### INTRODUCTION

Condensed-ring aromatic acetylenes have found applications in an array of areas including inhibition and modulation of animal microsomes by interactions with specific peptides<sup>1-4</sup>, as intermediates in the synthesis of lipoxygenase inhibitors<sup>7-8</sup> and as precursors for the synthesis of nonlinear optical polymers<sup>9-10</sup>.

---



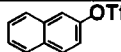
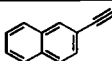
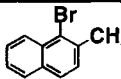
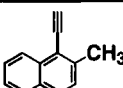
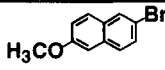
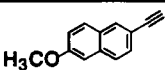
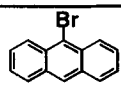
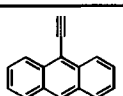
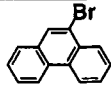
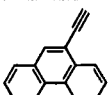
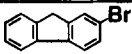
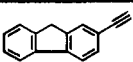
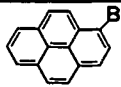
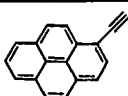
\* To whom correspondence should be addressed.

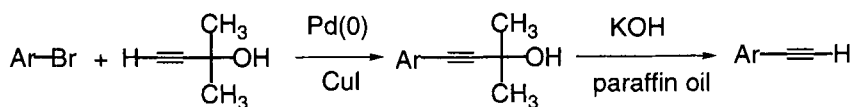
<sup>10</sup>. Arylacetylenes can be prepared by (i) dehydrochlorination of 2-chlorovinyl aromatics with potassium *tert*-butoxide<sup>11</sup>, (ii) dehydrochlorination of 1,2-dichloroethyl aromatics with sodium amide<sup>12</sup>, (iii) palladium-catalysed coupling of arylhalides with trimethylsilylacetylene with subsequent desilylation<sup>10,13,14</sup> and by (iv) base-catalysed retro-Favorsky elimination of acetone from  $\text{ArC}\equiv\text{CC}(\text{CH}_3)_2\text{OH}$ <sup>15</sup>. The use of trimethylsilylacetylene is convenient but expensive whereas the coupling of 3-methylbut-1-yn-3-ol to arylhalides and base-catalysed retro-Favorsky elimination is less expensive but the high temperatures and basic conditions often results in polymerization of the alkyne unless the product can be distilled from the reaction mixture,<sup>15</sup> or a one pot procedure used to generate diarylalkynes<sup>13, 16</sup>.

## RESULTS AND DISCUSSION

Using a modification to a reported procedure<sup>15</sup> we have prepared condensed aryl acetylenes in good yield by initially coupling the arylhalide with 3-methylbut-1-yn-3-ol and performing the base-catalysed retro-Favorsky elimination using powered potassium hydroxide in paraffin oil at 150°C. The formation of the expected arylacetylene was rapid under these conditions and the reaction was stopped as soon as the visual bubbling from the formation of acetone had ceased. The product was separated by flash chromatography with hexanes as an eluant. The condensed arylacetylenes polymerize on storage so the alkynyl precursor  $\text{ArC}\equiv\text{CC}(\text{CH}_3)_2\text{OH}$  was a convenient means of storing the material and converting as needed. The Table summarises the range of arylacetylenes which can be conveniently prepared by this methodology using the procedure outlined below.

Table

Aryl Halide	Product	Mp °C (lit)	M/z	Yield %	$^1\text{H } \delta$ HC≡	$^{13}\text{C } \delta$ C≡C
		oil (oil) <sup>17</sup>	152	63	3.45	81.8, 82.0
		35-36 (36) <sup>18</sup>	152	86	3.13	77.4, 84.0
		oil	166	63	3.69	80.6, 86.3
		110-111	182	83	3.10	76.7, 84.2
		74-75 (76-76.5) <sup>19</sup>	202	75	3.95	80.4, 88.2
		61-62 (62.5-63) <sup>12</sup>	202	88	3.45	81.7, 81.9
		82-83 (83) <sup>20</sup>	190	89	3.10	77.1, 84.4
		116-117.5 (lit) <sup>21</sup>	226	60	3.59	82.5, 82.8



A typical procedure: To a stirred solution of 2-bromo-6-methoxynaphthalene (0.44g, 1.86 mmol) and 3-methylbut-1-yn-3-ol (0.38ml 3.72 mmol) in piperidine (1ml) under nitrogen was added  $\text{Pd}(\text{PPh}_3)_4$  (5mg, 0.0047 mmol),  $\text{PPh}_3$  (2mg, 0.0076 mmol) and the solution of copper bromide (2mg, 0.014 mmol) and lithium bromide (7.8mg, 0.09 mmol) in THF (0.5ml). The clear solution was stirred at  $90^\circ\text{C}$  for 2hr. The reaction mixture was diluted with  $\text{CH}_2\text{Cl}_2$  and washed with 5% HCl to remove piperidine and the residue was chromatographed with (1:1:4,  $\text{CH}_2\text{Cl}_2$ :acetone:hexanes) to give 4-(6-methoxy-2-naphthyl)-2-methyl-3-butyn-2-ol which was suspended in paraffin oil (1ml) and finely powdered KOH (0.2g) was added then the flask was attached to a vacuum pump at  $\sim 30\text{mm Hg}$ . The resulted mixture was heated in an oil bath at  $150^\circ\text{C}$ . When bubbles ceased to be produced the reaction flask was immediately removed from the oil bath (2-3 min). The resulting mixture was diluted with dichloromethane and washed with water and subjected to flash chromatography with hexanes as eluant. Paraffin oil was eluted initially followed by 2-ethynyl-6-methoxynaphthalene 0.28g (83%). Mp  $110\text{--}111^\circ\text{C}$ ,  $M/z(M^+)$ : 182.  $^1\text{H}$  NMR:  $\delta$  3.10(s,  $\equiv\text{C-H}$ ); 3.85(s,  $\text{OCH}_3$ ); 7.04(bs, 1H); 7.12(ddd,  $J = 1.5, 2.7, 8.7\text{Hz}$ , 1H); 7.47(ddd,  $J = 1.5, 2.8, 8.4\text{Hz}$ , 1H); 7.58-7.66(m, 2H); 7.91(s, H1).  $^{13}\text{C}$  NMR:  $\delta$  55.2, 76.7, 84.2, 105.8, 117.0, 119.5, 126.9, 128.4, 129.2, 129.4, 132.1, 134.5, 158.6.

**Acknowledgement:** We would like to thank the Australian Research Council for the award of an Overseas Postgraduate Research Scholarship to Y.-L. J.

## REFERENCES

1. Lee, P. C.; and Dasmahaptra, A. K. *Chem.-Biol. Interact.* **1994**, *93*, 1-10.
2. Roberts, E. S.; Ballou, D. P.; Hopkins, N. E.; Alworth, W. L. and Hollenberg, P. F. *Arch. Biochem. Biophys.* **1995**, *323*, 303-312.

3. Roberts, E. S.; Hopkins, N. E.; Zaluzec, E. J.; Gage, D. A.; Alworth, W. L.; and Hollenberg, P. F. *Arch. Biochem. Biophys.* **1995**, 323, 295-302.
4. Beebe, L. E.; Roberts, E. S.; Fornwald, L. W.; Alworth, W. L. and Hollenberg, P. F. *Biochem. Pharmacol.* **1996**, 52, 1507-1513.
5. Chan, W. K.; Sui, Z. and Ortiz de Montellano, P. R. *Chem. Res. Toxicol.* **1993**, 6, 36-45.
6. Hopkins, N. E.; Foroozesh, M. K. and Alworth, W. L. *Biochem. Pharmacol.* **1992**, 44, 787-796.
7. Ohemeng, K. A.; Appollina, M. A.; Nguyen, V. N.; Schwender, C. F.; Singer, M.; Steber, M.; Ansell, J.; Argentieri, D. and Hageman, W. *J. Med. Chem.* **1994**, 37, 3663-3667.
8. Brooks, D. W.; Stewart, A. O.; Kerkman, D. J.; Bhatia, P. A. and Basha, A. *PCT Int. Appl. WO 92 01,682 (Chem. Abstr.* **1992**, 117, 26346b).
9. Crisp, G. T. and Bubner, T. P. *Tetrahedron* **1997**, 53, 11899-11912.
10. Ng, M.-K.; Chow, H.-F.; Chan, T.-L. and Mak, T. C. W. *Tetrahedron Lett.* **1996**, 37, 2979-2982.
11. Rodriguez, J. G.; Martin-Villamil, R.; Cano, F. H. and Fonseca, I. *J. Chem. Soc., Perkin Trans.* **1997**, 709-714.
12. Akiyama, S. and Nakagawa, M. *Bull. Chem. Soc. Jpn.* **1971**, 44, 2237-2248.
13. Crisp, G. T. and Bubner, T. P. *Tetrahedron* **1997**, 53, 11881-11898.
14. Vogtle, F.; Koch, H. and Rissanen, K. *Chem. Ber.* **1992**, 125, 2129-2135.
15. Mal'kina, A. G.; Brandsma, L.; Vasilevsky, S. F. and Trofimov, B. A. *Synthesis* **1996**, 589-590.
16. Carpita, A.; Lessi, A and Rossi, R. *Synthesis* **1984**, 571-572.
17. Stephens, E. B. and Tour, J. M. *Macromolecules* **1993**, 26, 2420-2427.

18. Leroy, J. A. *Bull. Soc. Chim. Fr.* **1982**, 648.
19. Rappoport, Z.; Shulman, P. and Thuval, M. *J. Am. Chem. Soc.* **1978**, *100*, 7041-7051
20. Golgolab, H. and Lalezari, I. *J. Heterocycl. Chem.* **1975**, *12*, 801-804.
21. Korshun, V. A.; Manasova, E. V.; Balakin, K. V.; Prokhorenko, I. A.; Buchatskii, A. G. and Berlin, Y. A. *Bioorg. Khim.*, **1996**, *22*, 923-925.

(Received in the UK 21 November 1997)