

Synthesis of a Polyhydroxyquinolizidine bearing a Polyhydroxylated Carbon Side-Chain

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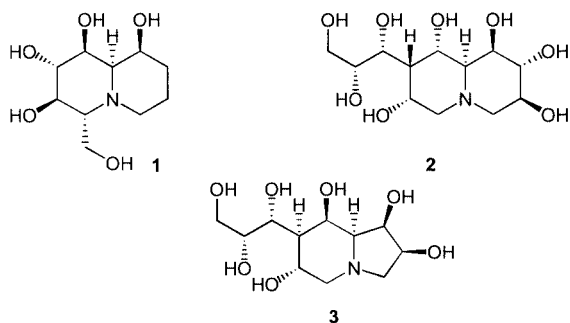
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Abstract: The lithium enolate of (-)-6-endo-chloro-5-exo-(phenylseleno)-7-oxabicyclo[2.2.1]heptan-2-one ((-)-**4**) added to (+)-(3aR,4aR,7aR,7bS)-3a,4a,7a,7b-tetrahydro-6,6-dimethyl[1,3]dioxolo-[4,5]furo[2,3-d]isoxazole-3-carbaldehyde ((+)-**5**) giving a major aldol (+)-**6** in 90% yield that was converted into (+)-(1R,4R,5R,6S,7R)-4-exo-(4-bromobenzenesulfonyloxy)-6-exo-((S)-[(*tert*-butyl)dimethylsilyloxy]-[(3aR,4aR,7aR,7bS)-3a,4a,7a,7b-tetrahydro-6,6-dimethyl[1,3]dioxolo[4,5]furo[2,3-d]isoxazol-3-yl)methyl]-7-endo-(methoxymethoxy)-2,8-dioxabicyclo[3.2.1]octan-3-one ((+)-**15**). Methanolysis of (+)-**15**, followed by reduction and protection steps provided (-)-1,4-anhydro-3-{6S-5-[(benzyloxy)carbonyl]amino-5-deoxy-β-L-idohexofuranos-6-C-yl}-3-deoxy-2,6-bis-O-(methoxymethyl)-α-D-galactopyranose ((-)-**21**). Acidic treatment of (-)-**21** followed by catalytical hydrogenation generated (-)-(1R,2R,3S,7R,8S,9S,9aS)-1,3,4,6,7,8,9, 9a-octahydro-8-[(1'R,2'R)-1',2',3'-trihydroxypropyl]-2H-quinolizine-1,2,3,7,9-pentol ((-)-**2**) characterized as its octaacetate (+)-**26**.

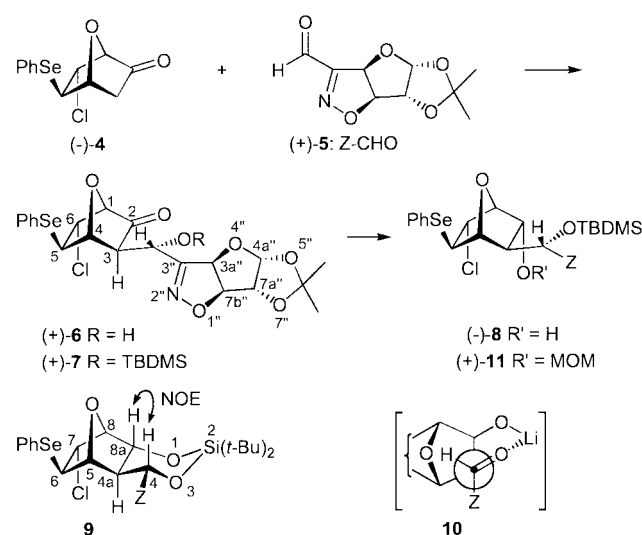
Key words: aldol condensation, "naked sugars", isoxazolines, aminodeoxysugars

Polyhydroxylated indolizidine alkaloids and their analogues, including ring contracted derivatives have shown interesting glycosidase inhibiting activities and as a consequence have a great potential as drugs against cancer and viruses.¹ Ring expanded analogues of polyhydroxyindolizidines, the polyhydroxyquinolizidines, emerge as a new class of potential glycosidase inhibitors, a few derivatives have been reported already.² Among them **1**, an analogue of α-homonojirimycin, is a potent inhibitor of α-glucosidase I from pig kidney (IC₅₀ = 0.15 μM).³



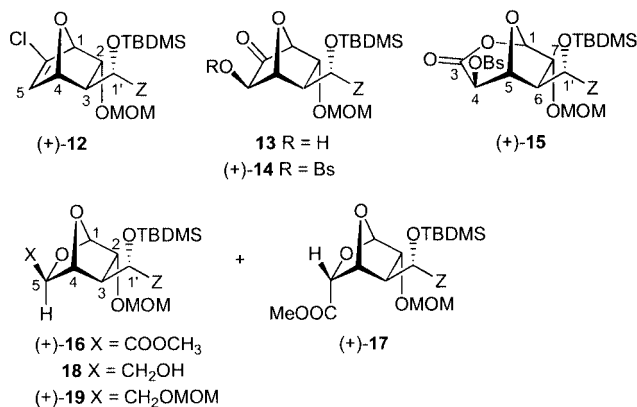
We report here the synthesis of (-)-(1R,2R,3S,7R,8S,9S,9aS)-1,3,4,6,7,8,9,9a-octahydro-8-[(1'R,2'R)-1',2',3'-trihydroxypropyl]-2H-quinolizine-1,2,3,7,9-pentol ((-)-**2**) the first example of a new kind of polyhydroxyquinolizidines bearing a polyhydroxylated side-chain. Our synthetic approach has been inspired

from that we had used to prepare **3**, the first example a 7-(1,2,3-trihydroxypropyl)-octahydroindolizine-1,2,6,8-tetrol.⁴ It features the diastereoselective cross-aldolization⁵ of the enantiomerically pure 7-oxabicyclo[2.2.1]heptan-2-one ((-)-**4**) derived from a "naked sugar of the first generation"⁶ with (+)-(3aR,4aR,7aR,7bS)-3a,4a,7a,7b-tetrahydro-6,6-dimethyl[1,3]dioxolo-[4,5]furo[2,3-d]isoxazole-3-carbaldehyde ((+)-**5**) that has been obtained recently enantiomerically pure.⁸



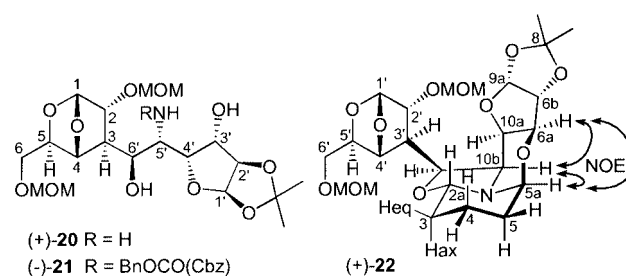
The lithium enolate of (-)-**4**^{6c,9} obtained by reaction of (-)-**4** with (Me₃Si)₂NLi in THF at -5 °C was cooled to -78 °C and a solution of aldehyde (+)-**5**⁸ in THF was added to it. After stirring at -78 °C for 3 h, acidic quenching with AcOH/MeOH and flash chromatography on silica gel provided pure aldol (+)-**6** in 90% yield¹⁰ which was silylated into (+)-**7** (92%)¹¹ with (*t*-Bu)Me₂SiOSO₂CF₃ and 2,6-lutidine (CH₂Cl₂, 0 °C for 1 h, then 25 °C for 5 h). Reduction of ketone (+)-**7** with NaBH₄ in MeOH (5 °C) was highly *exo* face selective^{6c} giving the corresponding *endo* alcohol (-)-**8** (91%).¹² The structures of (+)-**6**, (+)-**7** and (-)-**8** were deduced from their spectral data including 2D (NOESY, COSY) ¹H-NMR spectra and were confirmed by the formation of the 5,8-epoxy-4H-1,3,2-benzodioxasiline **9**^{13,14} for which ³J(H-8,H-8a) = 4.2 Hz,¹⁵ ³J(H-4a,H-5) ≅ 0 Hz,¹⁵ ³J(H-4,H-4a) = 10.8 Hz and ³J(H-4a,H-8a) = 3.5 Hz were measured in its ¹H-NMR spectrum. Furthermore, a strong NOE was observed between the signals at δ_H = 4.78 and 4.48 ppm assigned to protons Hexo-8a and H-4, respec-

tively. Compound **9** was obtained in 36% yield on treatment of (-)-**8** first with (*n*-Bu)₄NF in THF, then with (*t*-Bu)₂Si(OSO₂CF₃)₂ and 2,6-lutidine in CH₂Cl₂ (0–20 °C, 4 h). The data reported above confirm that the cross-aldolization (-)-**4** + (+)-**5** → (+)-**6** adopts a Zimmerman-Traxler transition state¹⁶ of type **10** in which the *exo* face of the lithium enolate is preferred for the aldehyde attack by its *Si* face.

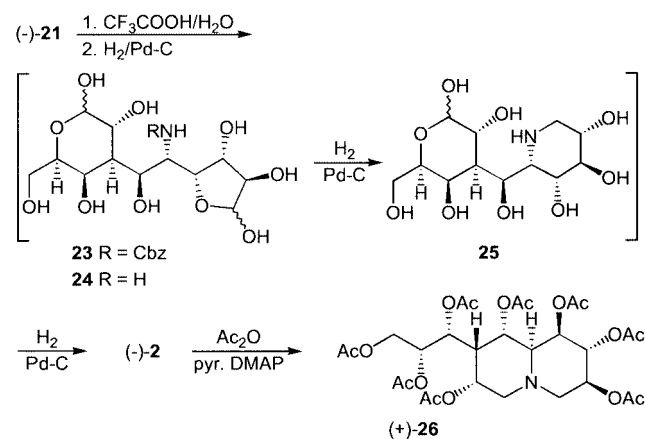


Conversion of the 7-oxabicyclo[2.2.1]heptane moiety of (-)-**8** into a uronic moiety followed a procedure similar to that developed by us earlier for the synthesis of (±)-**3**.⁴ The *endo* alcohol (-)-**8** was protected as a methoxymethyl ether on treatment with CH₃OCH₂Cl and (*i*-Pr)₂NEt in CH₂Cl₂ (0–25 °C, 20 h) giving (+)-**11** (86% yield) which underwent oxidative elimination of the benzeneselenenyl group on treatment with *meta*-chloroperbenzoic acid (*m*CPBA 70%) in CH₂Cl₂/THF (-78 °C for 2 h, then 25 °C for 14 h) to furnish the chloroalkene (+)-**12** in 95% yield.¹⁷ Double hydroxylation of (+)-**12** with Me₃NO·2 H₂O in THF/H₂O catalyzed with OsO₄ (NaHCO₃, 25 °C, 2 h) was highly *exo* face selective giving the corresponding α-hydroxyketone **13** that was not isolated, but esterified directly with *para*-bromobenzenesulfonyl chloride and Et₃N (0 °C for 1 h, then 25 °C for 2 h) into the brosylate (+)-**14** (88%).¹⁸ Baeyer-Villiger oxidation of ketone (+)-**14** with *m*CPBA/NaHCO₃ in CH₂Cl₂ (0 °C for 5 h, then 25 °C for 15 h) was, as expected,¹⁹ highly regioselective providing uronolactone (+)-**15** (92%).²⁰ Methanolysis (MeOH, K₂CO₃, DMF, 5 °C for 30 min, then 25 °C for 90 min) led to a 5:1 mixture (90%) of methyl 1,5-anhydro-α-D-galactofuranuronate (+)-**16** and methyl 1,5-anhydro-β-L-altrofuranuronate (+)-**17** that could be separated by column chromatography on silica gel.²¹ Reduction of (+)-**16** with LiAlH₄ in THF at 0 °C (10 min) provided the 1,5-anhydro-α-D-galactofuranose **18** which was not isolated but directly treated with MeOCH₂Cl and (*i*-Pr)₂NEt to give (+)-**19** (92%).²² Under these conditions the reduction of the isoxazoline moiety does not occur. The latter reaction requires ether as solvent, higher temperature (25 °C) and prolonged exposure (5 days) to a concentrated solution of LiAlH₄. This generated the aminodiols (+)-**20** (desilylation occurs concomitantly with the imine reduction) which

was selectively protected as the *N*-benzylcarbamate (-)-**21** (88%)²³ following a modified Oku's procedure.²⁴



The *L*-ido configuration of the 5-amino-5-deoxy-aldose moiety of (+)-**20** was expected by analogy with the LiAlH₄ reductions of simpler isoxazolines that favor hydride delivery on the less sterically hindered face of the imine functions.^{7,8,25} Definitive proof for that configuration was given by ¹H-NMR spectra of the 1*H*,6*aH*,8*H*-2,6,7,9,10-pentaoxa-10*c*-azapentaleno[2,3-*d*]acenaphthylene derivative (+)-**22** obtained by reaction of (+)-**20** (65%) with glutaraldehyde in MeOH²⁶ (25 °C, 12 h). For steric reasons and because of the conformational anomeric effect expected²⁷ for the aminoacetal moiety in (+)-**22**, this compound is the most stable stereomer resulting from the reversible condensation (+)-**20** + glutaraldehyde. The ¹H-NMR spectra of the crude product showed the presence of 6–10% only of stereomers of (+)-**22**.²⁸ The vicinal coupling constants observed in the ¹H-NMR spectrum of (+)-**22** were consistent with a "cis-decalin" type of conformation for its 1-aza-5-oxabicyclo[4.4.0]decane moiety (³*J*(H-2*a*,H*a*-3) = 9.7 Hz, ³*J*(H-2*a*,Heq-3) = 2.5 Hz, ³*J*(H-5*a*,H-5) = 2.9, 2.4 Hz, ³*J*(H-10*a*,H-10*b*) = 3.3 Hz, ³*J*(H-6*a*,H-10*a*) = 1.8 Hz). The 2D-NOESY ¹H-NMR spectrum of (+)-**22** showed correlation peaks confirming the structure and conformation shown for (+)-**22**.¹⁴



Treatment of (-)-**21** with CF₃COOH/H₂O (25 °C, 10 h) gave a complex mixture of dialdoses **23** that was hydrogenated over 10% Pd on charcoal (25 °C, H₂O, 15 h) to give the polyhydroxylated quinolizidine (-)-**2** as unique prod-

uct. Debenzylation was expected to generate **24**. The primary amine of **24** equilibrates with imines resulting from its reaction with the aldose moieties, imines that are reduced into the corresponding piperidines under our hydrogenation conditions. A possible piperidine intermediate is **25** which undergoes formation of an intermediate iminium salt with the second aldose moiety that is reduced into **2**. This polyol was fully characterized as its octaacetate (+)-**26** obtained in 40% yield (based on (-)-**21**) on treating crude **2** with Ac₂O/pyridine and 4-dimethylaminopyridine as catalyst (25°C, 15 h).²⁹

This work demonstrates that the cross-aldolization of aldehydes such as (+)-**5** with 7-oxabicyclo[2.2.1]heptanones can be highly stereoselective, thus allowing one to construct complicated long-chain amino-deoxysugars via the stereoselective hydride reductions of the isoxazolines so-obtained. Deprotection under hydrogenation conditions converts these systems into quinolizidines. The first example (**2**) of a 8-(1',2',3'-trihydroxypropyl)-2H-quinolizine-1,2,3,7,9-pentol has been reached by this method. Further efforts should make possible the obtention of stereomers of **2** and of unprotected long-chain amino-deoxycarbohydrates and analogues.

Acknowledgement

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References and Notes

- 1) Tyler, P. C.; Winchester, B. G. in "Iminosugars as Glycosidase Inhibitors", Stütz, A. E., Ed., Wiley-VCH, Weinheim, 1999, Chapter 7, p. 125-156; Ekhardt, C. E.; Fechter, M. H.; Hadwiger, P.; Mlaker, E.; Stütz, A. E.; Tauss, A.; Wrodnigg, T. M. *Ibid.* 1999, p. 253-390.
- 2) Hamana, H.; Ikota, N.; Ganem, B. *J. Org. Chem.* **1987**, 52, 5492; Liu, P. S. *Ibid.* **1987**, 52, 4717; Gradnig, G.; Berger, A.; Grassberger, V.; Stütz, A. E.; Legler, G. *Ibid.* **1991**, 32, 4889; Dax, K.; Fechter, M.; Gradnig, G.; Grassberger, V.; Illaszewicz, C.; Ungerank, M.; Stütz, A. E. *Carbohydr. Res.* **1991**, 217, 59; Rassu, G.; Casiraghi, G.; Pinna, L.; Spanu, P.; Ulgheri, F.; Cornia, M.; Zanardi, F. *Tetrahedron* **1993**, 49, 6627; Pearson, W. H.; Hembre, E. J. *Tetrahedron Lett.* **1993**, 34, 8221; Pearson, W. H.; Hembre, E. J. *J. Org. Chem.* **1996**, 61, 5537; Herczegh, P.; Kovacs, I.; Szilagyi, L.; Sztaricskai, F.; Bericibar, A.; Riche, C.; Chiaroni, A.; Olesker, A.; Lukacs, G. *Tetrahedron* **1995**, 51, 2969; Carretero, J. C.; Arrayas, R. G.; De Gracia, I. S. *Tetrahedron Lett.* **1997**, 38, 8537; Overkleeft, H. S.; Bruggeman, P.; Pandit, U. K. *Tetrahedron Lett.* **1998**, 39, 3869.
- 3) Liu, P. S.; Rogers, R. S.; Kang, M. S.; Sunkara, P. S. *Tetrahedron Lett.* **1991**, 32, 5853.
- 4) Kraehenbuehl, K.; Picasso, S.; Vogel, P. *Helv. Chim. Acta* **1998**, 81, 1439.
- 5) Chen, Y.; Vogel, P. *J. Org. Chem.* **1994**, 59, 2487; Emery, F.; Vogel, P. *Ibid.* **1995**, 60, 5843; Baudat, A.; Vogel, P. *Ibid.* **1997**, 62, 6252.
- 6) a) Vieira, E.; Vogel, P. *Helv. Chim. Acta* **1983**, 66, 1865; b) Reymond, J.-L.; Vogel, P. *Tetrahedron: Asymmetry* **1990**, 1, 729; c) Vogel, P.; Fattori, D.; Gasparini, F.; Le Drian, C. *Synlett* **1990**, 173; d) see also: Saf, R.; Faber, K.; Penn, G.; Griengl, H. *Tetrahedron* **1988**, 44, 389; Ronan, B.; Kagan, H. B. *Tetrahedron: Asymmetry* **1991**, 2, 75; Corey, E. J.; Loh, J.-L. *Tetrahedron Lett.* **1993**, 34, 3979; Forster, A.; Mosimann, H.; Renaud, P.; Vogel, P. *Tetrahedron: Asymmetry* **1999**, 10, 567.
- 7) Jäger, V.; Müller, I. *Tetrahedron* **1985**, 41, 3519; see also: Jäger, V.; Müller, R.; Leibold, T.; Hein, M.; Schwarz, M.; Fengler, M.; Jaroskova, L.; Pätz, M.; Le Roy, P.-Y. *Bull. Chem. Soc. Belg.* **1994**, 103, 491.
- 8) Schaller, C.; Vogel, P.; Jäger, V. *Carbohydr. Res.* **1998**, 314, 25.
- 9) Black, K. A.; Vogel, P. *J. Org. Chem.* **1986**, 57, 5341.
- 10) Data for (+)-**6**: m.p. 111-112 °C; [α]_D²⁵ +50 (c=1.0, CHCl₃); ¹H-NMR (400 MHz, CDCl₃) δ _H: 7.61-7.65 (m, 2H), 7.35-7.32 (m, 3H), 5.77 (d, ³J=3.6, H-4a"), 5.68 (d, ³J=6.4, H-3a"), 4.96 (d, ³J=6.4, H-7b"), 4.84 (dd, ³J=8.3, 2.6, H-1'), 4.78-4.76 (m, H-7a", H-4), 4.54 (d, ³J=5.6, H-1), 4.29 (ddd, ³J=5.6, 2.8, ⁴J=1.0, H-6), 3.65 (d, ³J=2.8, H-5), 3.35 (d, ³J=2.6, HO-C(1')), 2.78 (d, ³J=8.3, H-3), 1.50, 1.37 (2s, 2 Me).
- 11) Data for (+)-**7**: [α]_D²⁵ +28 (c=1.1, CHCl₃).
- 12) Data for (-)-**8**: [α]_D²⁵ -10 (c=0.43, CHCl₃).
- 13) Data for **9**: ¹H-NMR (400 MHz, CDCl₃) δ _H: 7.59-7.57 (m, 2H), 7.34-7.31 (m, 3H), 5.70 (d, ³J=3.5, H-4a'), 5.51 (d, ³J=6.4, H-3a'), 4.83 (d, ³J=6.4, H-7b'), 4.78 (dd, ³J=4.2, 3.8, H-8a), 4.76 (d, ³J=3.5, H-7a'), 4.49-4.47 (m, H-4, H-8), 4.30 (s, H-5), 4.15 (dd, ³J=5.3, 3.5, H-7), 3.51 (d, ³J=5.3, H-6), 2.23 (dd, ³J=10.8, 3.3, H-4a), 1.49, 1.37 (2s, 2 Me), 1.05 (s, 2 t-Bu); ¹³C-NMR (100.6 MHz, CDCl₃) δ _C: 158.6 (s), 134.0 (d, 2C), 133.9 (s), 129.2 (2d), 127.8, 113.9 (s), 106.1, 87.1, 86.8, 86.5, 84.6, 79.4, 77.1, 68.6, 60.4, 56.6, 51.5 (11d), 27.7, 27.3, 27.1, 26.8 (4q), 24.3, 20.8 (2s).
- 14) All new compounds presented here gave satisfactory elemental analyses.
- 15) Gagnaire, D.; Payo-Subiza, E. *Bull. Chem. Soc. Fr.* **1963**, 2627; Nelson, W. L.; Allen, D. R. *J. Heterocycl. Chem.* **1972**, 9, 561; Kienzle, F. *Helv. Chim. Acta* **1975**, 58, 1180.
- 16) Zimmerman, H. E.; Traxler, M. D. *J. Am. Chem. Soc.* **1957**, 79, 1920; see also: Kleschick, W. A.; Buse, C. T.; Heathcock, C. H. *Ibid.* **1977**, 99, 247; Fellmann, P.; Dubois, J. E. *Tetrahedron* **1978**, 34, 1349; Evans, D. A.; Vogel, E.; Nelson, J. V. *J. Am. Chem. Soc.* **1979**, 101, 6120; Heathcock, C. H.; Buse, C. T.; Kleschick, W. A.; Pirrung, M. C.; Sohn, J. E.; Rampe, J. *J. Org. Chem.* **1980**, 45, 1066; Seebach, D.; Prelog, V. *Angew. Chem., Int. Ed. Engl.* **1982**, 21, 654.
- 17) Data for (+)-**12**: colorless oil; [α]_D²⁵ = +129 (c=1.1, CHCl₃); UV (MeCN) λ _{max}: 209 nm (ϵ =7000); ¹H-NMR (400 MHz, CDCl₃) δ _H: 6.37 (d, ³J=2.2, H-5), 5.87 (d, ³J=3.6, H-4a"), 5.44 (d, ³J=5.8, H-3a"), 4.97 (d, ³J=2.2, H-4), 4.92 (d, ³J=6.5, H-1'), 4.84 (d, ³J=5.8, H-7b"), 4.82 (d, ³J=3.6, H-7a"), 4.70-4.68 (m, H-1), 4.69, 4.66 (AB, ²J=6.7, CH₂(MOM)), 4.17 (dd, ³J=4.4, 2.4, H-2), 3.39 (s, MeO), 2.02 (dd, ³J=6.5, 2.4, H-3), 1.49, 1.36 (2s, 2 Me), 0.92 (s, t-Bu), 0.11, 0.10 (2s, Me₂Si).
- 18) Data for (+)-**14**: mp 55-56 °C, [α]_D²⁵ +13 (c=1.0, CHCl₃).
- 19) Arvai, G.; Fattori, D.; Vogel, P. *Tetrahedron* **1992**, 48, 10621.
- 20) Data for (+)-**15**: mp 67-68 °C, [α]_D²⁵ +18 (c=1.0, CHCl₃); IR (KBr) ν : 2930, 1770, 1575, 1390, 1375, 1205, 1190, 1105, 1060, 1010, 840 cm⁻¹; ¹H-NMR (400 MHz, CDCl₃) δ _H: 7.83, 7.73 (2d, ³J=8.8, 4H), 5.82 (d, ³J=3.5, H-1), 5.79 (d, ³J=3.5, H-4a"), 5.56 (d, ³J=6.7, H-3a"), 4.95 (d, ³J=6.7, H-7b"), 4.78 (d, ³J=5.7, H-1'), 4.74 (d, ³J=3.5, H-7a"), 4.73 (s, H-4), 4.64, 4.50 (AB, ²J=7.0, CH₂(MOM)), 4.55 (br. s, H-5), 4.19 (dd, ³J=4.5, 3.5, H-7), 3.32 (s, MeO), 2.36 (ddd, ³J=5.7, 4.5, 1.9, H-6), 1.44, 1.32 (2s, 2 Me), 0.84 (s, t-Bu), 0.06, 0.05 (2s, Me₂Si).
- 21) Data for (+)-**16**: oil, [α]_D²⁵ +64 (c=0.36, CHCl₃); ¹H-NMR (400 MHz, CDCl₃): ³J(H-4,H-3) = ³J(H-4,H-5) $\approx$ 0, ³J(H-1,H-2) = 2.3. Data for (+)-**17**: oil, [α]_D²⁵ +75 (c=0.25, CHCl₃); ³J(H-3,H-4) $\approx$ 0, ³J(H-4,H-5) = 4, ³J(H-1,H-2) = 2.0.

- (22) Data for (+)-**19**: oil, $[\alpha]_D^{25} + 54$ ($c=0.65$, CHCl_3); $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ_{H} : 5.86 (d, $^3J=3.6$, H-4a"), 5.55 (d, $^3J=6.0$, H-3a"), 5.47 (d, $^3J=2.4$, H-1), 4.89 (d, $^3J=6.0$, H-7b"), 4.82 (d, $^3J=3.6$, H-7a"), 4.72 (d, $^3J=7.0$, H-1'), 4.71 (AB, $^2J=6.8$, 1H, $\text{CH}_2(\text{MOM})$), 4.64–4.60 (m, 4H, H-4, MOM), 3.93 (dd, $^3J=8.0$, 5.2, H-5), 3.81 (dd, $^3J=2.8$, 2.4, H-2), 3.46 (dd, $^2J=10.3$, $^3J=5.2$, H-6), 3.41–3.39 (m, H'-6), 3.40, 3.35 (2s, 2 MeO), 2.05 (dd, $^3J=7.0$, 2.8, H-3), 1.51, 1.36 (2s, Me_2C), 0.91 (s, $t\text{-Bu}$), 0.11, 0.09 (2s, Me_2Si).
- (23) Data for (+)-**20**: $[\alpha]_D^{25} + 37$ ($c=1.1$, CHCl_3); Data for (-)-**21**: m.p. 49–50°C, $[\alpha]_D^{25} - 2.0$ ($c=0.5$, CHCl_3); $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ_{H} : 7.34–7.30 (m, 5 H), 5.89 (d, $^3J=3.7$, H-1'), 5.53 (d, $^3J=2.5$, H-1), 5.42 (d, $^3J(\text{H-N, H-5}')=9.0$, HN), 5.12, 5.08 (AB, $^2J=12.4$), 4.72, 4.66 (AB, $^2J=6.9$), 4.60, 4.58 (AB, $^2J=6.4$), 4.54 (s, H-4), 4.50 (d, $^3J=3.7$, H-2'), 4.38 (dd, $^3J=5.0$, 2.7, H-4'), 4.24 (d, $^3J=2.7$, H-3'), 4.12 (m, H-5'), 3.98 (dd, $^3J=2.8$, 2.5, H-2), 3.91 (dd, $^3J=7.7$, 5.3, H-5), 3.80 (dd, $^3J=8.1$, 5.8, H-6'), 3.55 (br. s, 2 HO), 3.45 (dd, $^3J=10.3$, 5.3, H-6), 3.41 (s, MeO), 3.38–3.34 (m, H'-6), 3.34 (s, MeO), 2.01 (dd, $^3J=8.1$, 2.8, H-3), 1.47, 1.30 (2s, Me_2C).
- (24) Kayakiri, H.; Oku, T.; Hashimoto, M. *Chem. Pharm. Bull.* **1991**, 39, 1397 and ref. cited therein.
- (25) Müller, I.; Jäger, V. *Tetrahedron Lett.* **1982**, 23, 4777; Jäger, V.; Müller, I.; Paulus, E. F. *Ibid.* **1985**, 26, 2997; Jäger, V.; Müller, I.; Schohe, R.; Frey, M.; Ehrler, R.; Häfele, B.; Schröter, D. *Sect. Heterocycl. Chem.* **1985**, 8, 79.
- (26) Schipchandler, M. J. *Heterocycl. Chem.* **1977**, 14, 305; Fülöp, F.; Lazar, L.; Pelczar, I.; Bernath, G. *Mag. Kem. Foly* **1989**, 95, 212; *Chem. Abstr.* **1989**, 112, 77073.
- (27) "The Anomeric Effect and Associated Stereoelectronic Effects", Thatcher, G. R. J., Ed., ACS Symposium Series 539, 1993; Juaristi, E.; Cuevas, G. "The Anomeric Effect", CRC Press, 1995; Juaristi, E.; Cuevas, G. *Tetrahedron* **1992**, 48, 5019.
- (28) Data for (+)-**22**: mp 54–55 °C, $[\alpha]_D^{25} + 17$ ($c=0.35$, CHCl_3); $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ_{H} : 5.82 (d, $^3J=3.6$, H-9a), 5.53 (d, $^3J=2.4$, H-1'), 4.87, 4.67 (AB, $^2J=6.6$, 2H), 4.84 (dd, $^3J=9.7$, 2.5, H-2a), 4.69 (dd, $^3J=2.9$, 2.4, H-5a), 4.55 (s, 2H), 4.48 (d, $^3J=3.6$, H-6b), 4.28 (s, H-4'), 4.21 (dd, $^3J=11.6$, 6.5, H-1), 4.17 (br. s, H-6a), 4.03 (dd, $^3J=2.6$, 2.4, H-6'), 4.01 (dd, $^3J=8.8$, 4.8, H-3'), 3.98 (dd, $^3J=3.3$, 1.8, H-10a), 3.73 (dd, $^3J=6.5$, 3.3, H-10b), 3.46 (dd, $^3J=9.7$, 4.8, H-8'), 3.41, 3.31 (2s, 2 MeO), 3.32–3.29 (m, H'-8'), 2.78 (dd, $^3J=11.6$, 2.6, H-5'), 1.96–1.92 (m, H-3), 1.75–1.70 (m, 2H, H-4), 1.69–1.55 (m, 2H, H-5), 1.44, 1.31 (2s, Me_2C), 1.29–1.23 (m, H'-3).
- (29) Data for (+)-**26**: oil, $[\alpha]_D^{25} + 40$ ($c=0.3$, CHCl_3); UV (MeCN): 199 (8500); IR (film) ν : 2920, 2850, 1750, 1435, 1375, 1230, 1045, 740 cm^{-1} ; $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ_{H} : 5.51 (dd, $^3J=7.8$, 4.1, H-1'), 5.33 (ddd, $^3J=6.8$, 4.1, 4.0, H-2'), 5.25 (ddd, $^3J=9.7$, 4.9, 4.8, H-7), 5.07 (dd, $^3J=4.8$, 4.7, H-9), 5.03 (br. s, H-2), 4.98 (br. s, H-1), 4.76 (br. s, H-3), 4.28 (dd, $^2J=11.9$, $^3J=4.0$, H-3'), 3.93 (dd, $^2J=11.9$, $^3J=6.8$, H'-3'), 3.37–3.33 (m, Heq-6), 3.01 (dm, $^2J=13.5$, Heq-4), 2.95 (m, H-9a), 2.85 (dm, $^2J=13.5$, Hax-4), 2.67–2.60 (m, 2H, Hax-6, H-8), 2.15, 2.11, 2.10, 2.09, 2.07, 2.03 (6s, 8 AcO). $^{13}\text{C-NMR}$ (100.6 MHz, CDCl_3) δ_{C} : 170.5, 170.3, 170.0, 169.8, 169.5, 169.2, 168.2 (7s, 8 COO), 70.0, 68.2 (2C), 67.8 (2C), 66.8, 66.3 (5d), 62.6 (t), 57.1 (d), 54.0, 52.7 (2t), 37.8 (d), 21.2, 21.0, 20.9, 20.8, 20.5 (5q, 8 CH_3); CI-MS (NH_3) m/z : 647 ($\text{M}+1$, 7), 646 (M^+ , 5). Anal. calcd. for $\text{C}_{28}\text{H}_{39}\text{NO}_{16}$ (645.61): C 52.09, H 6.09, N 2.17; found: C 52.09, H 6.42, N 2.30.

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