

# Ru(II)-Catalyzed Oxidative Olefination of Benzamides: Switchable Aza-Michael and Aza-Wacker Reaction for Synthesis of Isoindolinones

Manoj Kumar,<sup>†</sup> Shalini Verma,<sup>†</sup> and Akhilesh K. Verma\*

Cite This: <https://dx.doi.org/10.1021/acs.orglett.0c01237>

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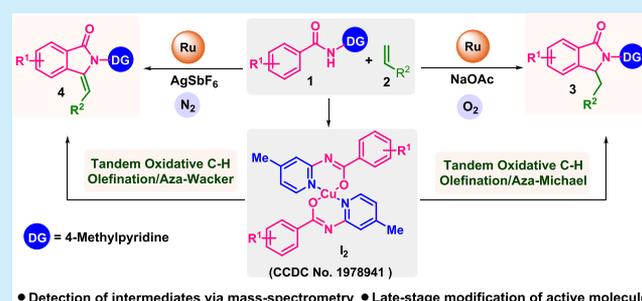
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**ABSTRACT:** Selective tandem oxidative C–H olefination–aza-Michael/aza-Wacker reaction of *N*-arylbenzamides is achieved by fine-tuning between base and additive to access valuable 3-oxoisindolinyls and 3-oxoisindolinylidenes, respectively. Careful optimization and control experiments provides a guiding principle in the design of a proposed catalytic cycle. The copper–iminium complex acting as a precursor for the binding of Ru catalyst was isolated and confirmed by X-ray diffraction. The versatility of this catalytic system has been demonstrated by the synthesis of biologically relevant molecules.



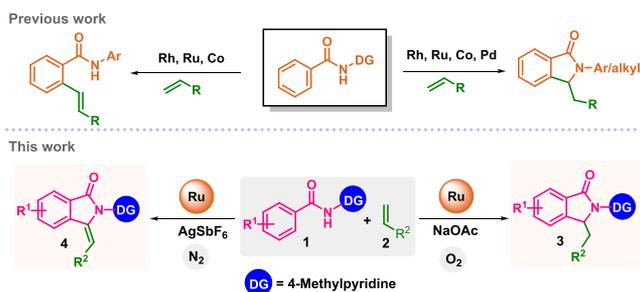
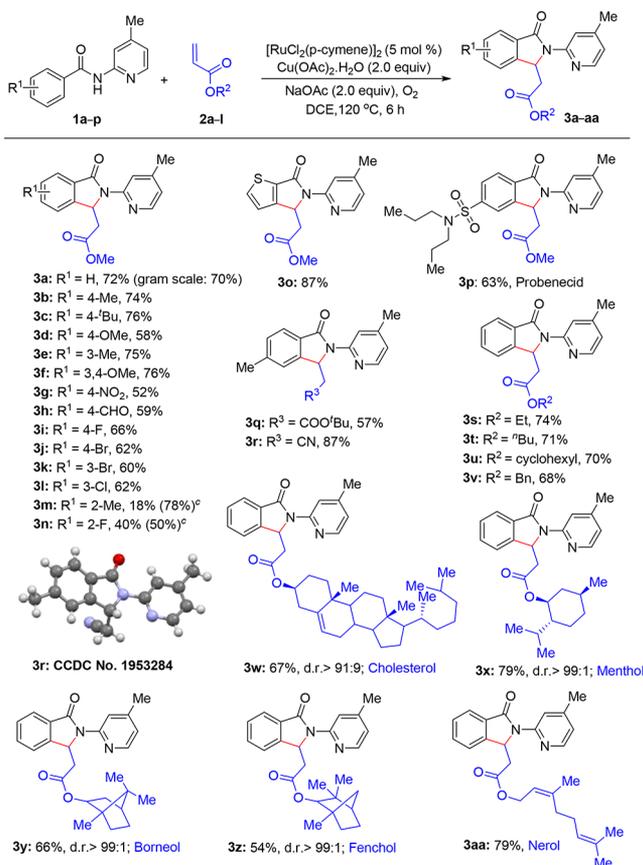
Isoindoline represents an important structural motif owing to its fascinating biological and physiological properties. The isoindoline core has been found to serve as a key precursor for the synthesis of valuable drug molecules and complex natural products.<sup>1</sup> Considering the enormous potential of isoindolines, immense effort toward the development of new synthetic strategies for their synthesis is not surprising.<sup>2</sup> In the ensuing years, auxiliary-assisted transition-metal-catalyzed C–H olefination followed by oxidative annulation has been executed successfully in this area. Although extensive studies have been done on *ortho*-C–H olefination of unactivated arenes to access alkenylated<sup>3</sup> and alkylated<sup>4</sup> arenes, only a modest success has been achieved for the tandem oxidative annulation of alkenylated arenes to synthesize isoindolines. In this regard, most of the established methods rely on use of expensive metal-based catalytic system.<sup>5</sup> In particular, the precious Rh catalyst gains considerable attention due to its high catalytic activity in C–H functionalization.<sup>6</sup> Despite having impressive advances, the use of expensive rhodium metal necessitates the development of an efficient strategy utilizing relatively inexpensive, yet sustainable, catalyst. Recently, the use of ruthenium catalyst in C–H activation has spurred considerable interest in a plethora of organic transformations.<sup>7</sup> In particular, the use of Ru(II) species for C–H olefination of unactivated arenes is currently under development.<sup>8</sup> In 2015, Jeganmohan et al. reported the only Ru catalyzed synthesis of isoindolines by cyclization of *N*-alkylbenzamides with allylic alcohols.<sup>9a</sup> Subsequently, Ackermann et al. introduced bidentate 8-aminoquinoline and tosyl auxiliaries for Co- and Ru-catalyzed oxidative C–H alkenylation.<sup>9b,c</sup> More recently, Zhang et al. developed a multi-component synthesis of isoindolinones by Rh(III) relay

catalysis.<sup>6c</sup> The reaction does not require the preparation of amide substrate and demonstrates the use of *N*-pyridin-2-yl benzamide as an effective directing group.

To date, we are unaware of any precedent on Ru-catalyzed tandem oxidative C–H olefination/aza-Michael and aza-Wacker reaction of *N*-arylbenzamides with  $\alpha,\beta$ -unsaturated esters to access isoindolinones (Scheme 1). Owing to the easy accessibility and preparation of  $\alpha,\beta$ -unsaturated esters, they have been widely used as coupling partners for oxidative C–H alkenylation<sup>3a–c</sup> and alkylation;<sup>4a,b</sup> however, only a limited number of studies have appeared in literature on their use for isoindoline synthesis.<sup>6a,b,8</sup> Employing  $\alpha,\beta$ -unsaturated esters as coupling partners provides the synthesized motifs, leading to their easy modification to access synthetically useful molecules.<sup>10</sup> Further, the choice of directing group is also crucial to induce reactivity and selectivity in reaction. The small difference in directing group can lead to change in reaction pathway. In most of the reported Ru-catalyzed reactions, use of directing groups is limited to *N*-alkylamides,<sup>8,9a</sup> while *N*-aromatic amides are shown to be less favorable due to steric interactions.<sup>8c</sup> In contrast, the present work makes effective use of *N*-pyridylamides as directing group to synthesize *N*-pyridylisoindolines which are documented to show potent antitumor activity and DNA binding features.<sup>11</sup> Inspired by the

Received: April 7, 2020

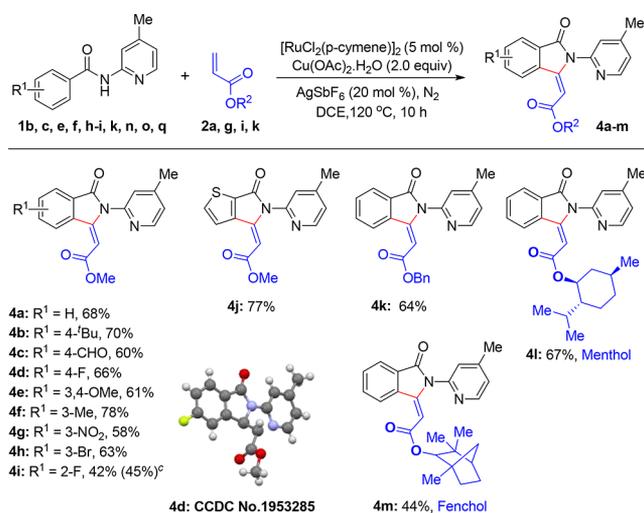
## Scheme 1. Synthesis of 3-Oxoisoindolinones

Scheme 2. Synthesis of Substituted 3-Oxoisoindolinyls<sup>a</sup>

<sup>a</sup>Reactions were performed using 0.5 mmol of **1**, 3.0 equiv of **2**, [ $\{\text{RuCl}_2(p\text{-cymene})\}_2$ ] (5.0 mol %), NaOAc (2.0 equiv), and  $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$  (2.0 equiv) in 2 mL of DCE under  $\text{O}_2$  at 120 °C for 6 h. <sup>b</sup>Isolated yield. <sup>c</sup>Recovered starting material.

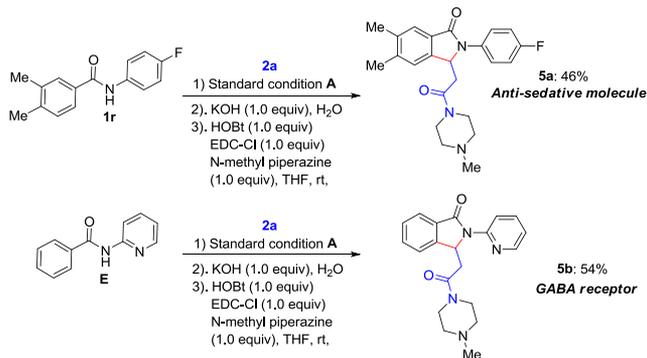
elegant contribution of momentous isoindolines in pharmaceuticals and our continuous efforts<sup>12</sup> in developing efficient methodologies for the synthesis of diversified heterocycles, we herein report the selective synthesis of valuable 3-oxoisoindolinyls and 3-oxoisoindolinylidenes by tandem oxidative C–H olefination/aza-Michael and aza-Wacker reaction of amides (Scheme 1). A plausible reaction cycle is provided, strongly supported by performed series of control experiments and characterization of reaction intermediates by mass spectrometry and X-ray diffraction.

During our investigation (for detailed information, see the Supporting Information) using *N*-(4-methylpyridin-2-yl)-benzamide **1a** with methyl acrylate **2a** as model substrate, we initially observed formation of two products, **3a** and **4a**. On

Scheme 3. Synthesis of Substituted 3-Oxoisoindolinylidenes<sup>a</sup>

<sup>a</sup>Reactions were performed using 0.5 mmol of **1**, 3.0 equiv of **2**, [ $\{\text{RuCl}_2(p\text{-cymene})\}_2$ ] (5.0 mol %),  $\text{AgSbF}_6$  (20 mol %), and  $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$  (2.0 equiv) in 2 mL of DCE under  $\text{N}_2$  at 120 °C for 10 h. <sup>b</sup>Isolated yield. <sup>c</sup>Recovered starting material.

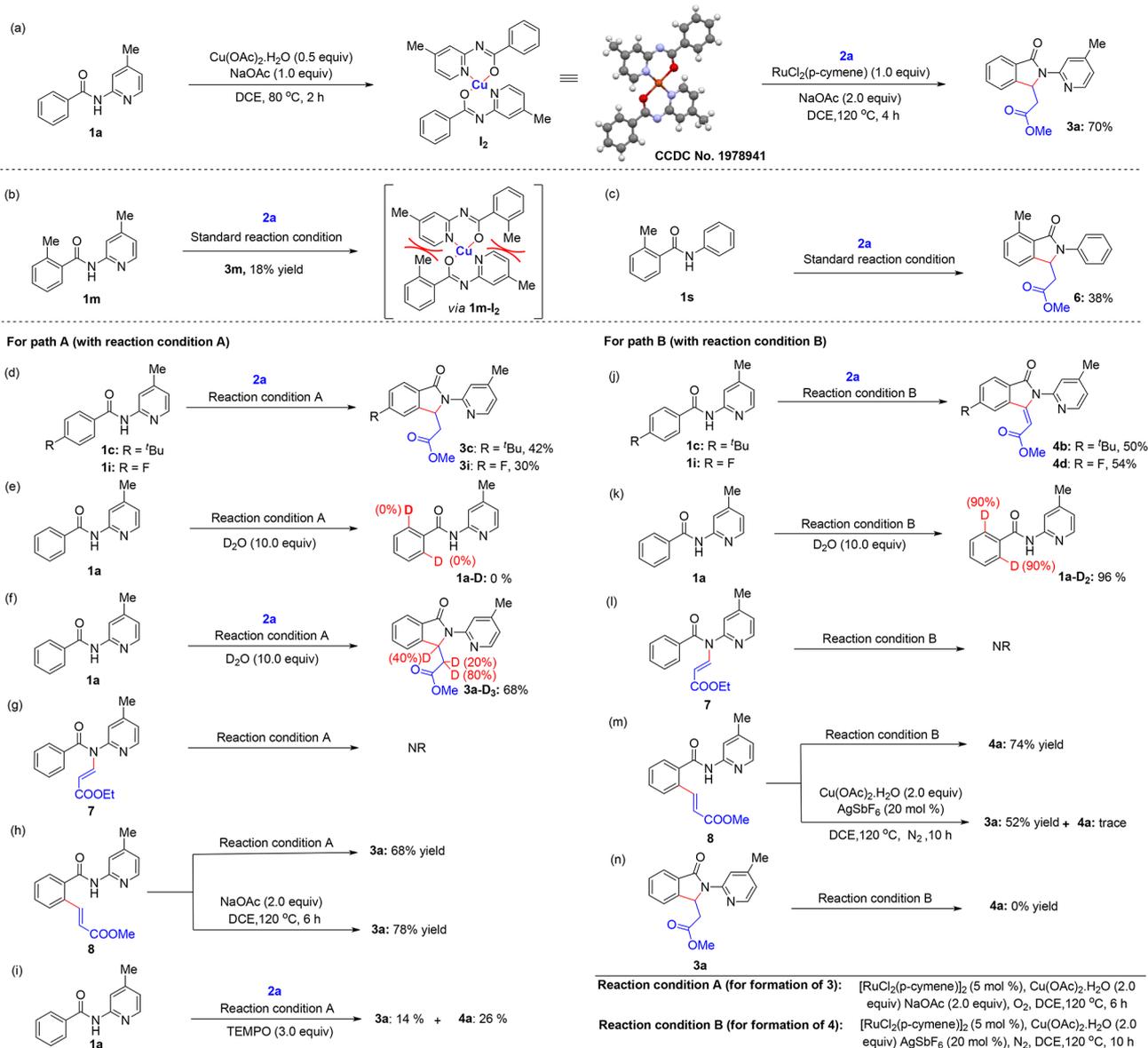
## Scheme 4. Synthesis of Biologically Active Molecules



further screening, we found that reaction **1a** with **2a** in the presence of [ $\{\text{RuCl}_2(p\text{-cymene})\}_2$ ] (5.0 mol %),  $\text{AgSbF}_6$  (20 mol %), and  $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$  (2.0 equiv) in DCE under  $\text{N}_2$  atmosphere at 120 °C for 10 h gave the product **4a** exclusively in 68% yield. As the product **3a** was supposed to be formed by tandem aza-Michael, we hypothesized that the addition of base could enhance the formation of **3a**. Pleasingly, addition of NaOAc (2.0 equiv) instead of  $\text{AgSbF}_6$  under  $\text{O}_2$  atmosphere delivers the product **3a** in 72% yield. With promising conditions, the scope of the reaction was evaluated for the synthesis of 3-oxoisoindolinyls (Scheme 2). Various substituted benzamides were subjected to react with methyl acrylate **2a** under optimal reaction conditions (Table S5, entry 3). The reaction of substrates **1b–d** bearing electron-releasing substituents such as –Me, –<sup>t</sup>Bu, and –OMe at the 4-position afforded the desired products **3b–d** in 58–76% yields.

It is worthwhile to note that substrate **1e** bearing –Me at the 3-position provided exclusively mono-olefinated products **3e** in 75% yield. Likewise, 3,4-OMe-containing substrate **1f** delivered the product **3f** in comparable yield. Further, substrates **1g–j** bearing strong electron-withdrawing –NO<sub>2</sub>, –CHO, and halogens, such as –F and –Br at the 4-position, were reacted smoothly to afford the corresponding products **3g–j** in 52–

## Scheme 5. Control Experiments



66% yields. The reaction is marginally affected by the presence of halogens such as  $-\text{Br}$  and  $-\text{Cl}$  at the 3-position of substrate **1k** and **1l**, providing the products **3k** and **3l** in 60% and 62% yields, respectively. Substrate **1m** having  $-\text{Me}$  at the 2-position also participated in the reaction and provided the desired product **3m** in 18% yield only. The reason might be the steric hindrance created by the methyl group, as the less sterically hindered substrate **1n** having  $-\text{F}$  at the 2-position affords the product **3n** in 40% yield. Notably, the tolerance of nitro, aldehyde, and halogen substituents facilitates the additional modification of the products. Additionally, heteroaromatic substrate **1o** participates well in the reaction to give the desired product **3o** in 87% yield. Probenecid, medicinally relevant molecule derived benzamide **1p**, successfully provided the desired product **3p**, albeit in moderate yield. Next, the scope of acrylates as a coupling partner was examined. The reaction of substrate **1b** with  $t$ butyl acrylate **2b** and acrylonitrile **2c** delivers the products **3q** and **3r** in 57% and 87% yields, respectively. Likewise, ethyl,  $n$ butyl, cyclohexyl, and benzyl acrylates **2d–g** with substrate **1a** gave the compounds **3s–v** in

68–74% yields. This method is not limited to the use of simple acrylates; natural product containing acrylates **2h–l** were also found to be viable, as demonstrated by the successful synthesis of corresponding compounds **3w–aa** in good yields. This also shows that steric bulk at the ester group can be tolerated in these reactions. According to the optimal reaction conditions (Table S4, entry 2), switching the catalytic conditions to the Ru/AgSbF<sub>6</sub> system led to the formation of 3-oxoisindolinylidene via aza-Wacker cyclization. Intrigued by the result, we proceeded to examine the scope of the reaction using differently substituted benzamides (Scheme 3). Electron-rich substrate **1c** containing  $-\text{tBu}$  at the 4-position gave the products **4b** in 70% yield. Electron-deficient substrates bearing a free  $-\text{CHO}$  group (**1h**) and  $-\text{F}$  (**1i**) at the 4-position were also compatible providing the desired product **4c** and **4d** in 60% and 66% yields, respectively. 3,4-OMe substituted substrate **1f** furnished the product **4e** selectively in 61% yield. Importantly, substrates **1e**, **1q**, and **1k** containing  $-\text{Me}$ ,  $-\text{NO}_2$ , and  $-\text{Br}$  at the 3-position were compatible, providing the corresponding products **4f–h** in 58–78% yields. The



formation of **4a** (Scheme 5m). After *ortho*-C–H alkenylation, Ru thus appears to involve in the catalytic cycle. Also, treatment of **3a** under the optimized reaction conditions failed to produce **4a** (Scheme 5n), therefore ruling out the possibility of intermediacy of **3a** and suggesting an aza-Wacker pathway.

On the basis of the results of control experiments and the literature precedents,<sup>16</sup> a plausible catalytic cycle is proposed in Scheme 6. At first, the coordination of **1a** with Cu(OAc)<sub>2</sub> resulted in the formation of square planar iminium complex **I**<sub>2</sub> via intermediate **I**<sub>1</sub>. Meanwhile, the 18e<sup>−</sup> complex of Ru was dissociated to species (I) (under reaction conditions A) or species (II) (under reaction conditions B)<sup>16a,c</sup> which gets ligated to the lone pair of iminium nitrogen of **I**<sub>2</sub> to give key intermediate **I**<sub>3</sub>. Intermediate **I**<sub>3</sub> gets dissociated to **I**<sub>4</sub> in which nitrogen atom of amide gets strongly coordinated to Ru with the release of AcOH. The alkene **2a** then coordinates to Ru in **I**<sub>4</sub> to afford **I**<sub>5</sub>, followed by its insertion between Ru–C bond to give intermediate **I**<sub>6</sub>. β-Hydride elimination of intermediate **I**<sub>6</sub> affords Ru ligated alkenylated benzamide **I**<sub>7</sub>. It is very interesting to note that intermediate **I**<sub>7</sub> reacts differently under oxygen and nitrogen atmosphere to yield two different products, **3a** and **4a**. In the case of oxygen atmosphere, O<sub>2</sub> gets inserted into Ru–H bond in **I**<sub>7</sub> via single-electron transfer to provide Ru-hydroperoxo [Ru–OOH] intermediate **I**<sub>8</sub>.<sup>15</sup> Following protonolysis, **I**<sub>8</sub> gives the *ortho*-alkenylated product **I**<sub>9</sub>, H<sub>2</sub>O<sub>2</sub>h and Ru species. The corresponding Ru species was further oxidized to Ru(II) active species I in the presence of Cu(OAc)<sub>2</sub> to again take part in the catalytic cycle and the alkenylated product **I**<sub>9</sub> undergoes aza-Michael in the presence of NaOAc to give the desired product **3a**. In contrast, under N<sub>2</sub> atmosphere, intermediate **I**<sub>7</sub> affords the alkenylated product **I**<sub>9</sub> in the presence of Cu(OAc)<sub>2</sub> and regenerates the Ru species (II).<sup>16c</sup> Subsequently, the amide group of **I**<sub>9</sub> then coordinates Ru species (II) to give intermediate **I**<sub>10</sub>.<sup>16a</sup> Intramolecular coordinative insertion in **I**<sub>10</sub> resulted into the formation of intermediate **I**<sub>11</sub>. The intermediate **I**<sub>11</sub> further undergoes β-Hydride elimination in the presence of Cu(OAc)<sub>2</sub> to provide the expected product **4a** and regenerate the Ru active species (II).<sup>16a</sup>

In conclusion, we have reported a Ru-catalyzed synthesis of isoindolinones with ample scope. Interestingly, fine-tuning of the reaction condition led to the formation of two different products by switching the reaction pathway. The findings of control experiments provide valuable support for the proposed mechanistic pathway. The isolation of an intermediate **I**<sub>2</sub> and experimental data led us to propose that the Ru species binds to the substrate in its imine form rather than in its amide form. A useful antisedative molecule and a GABA-inhibitor were synthesized successfully using the protocol.

## ■ ASSOCIATED CONTENT

### SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.0c01237>.

Data and spectral copies of <sup>1</sup>H, <sup>13</sup>C NMR and HRMS for target compounds (PDF)

## Accession Codes

CCDC 1953284–1953285 and 1978941 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif), or by emailing [data\\_request@ccdc.cam.ac.uk](mailto:data_request@ccdc.cam.ac.uk), or by

contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

## ■ AUTHOR INFORMATION

### Corresponding Author

Akhilesh K. Verma – Department of Chemistry, University of Delhi, Delhi 110007, India; [orcid.org/0000-0001-7626-5003](https://orcid.org/0000-0001-7626-5003); Email: [averma@acbr.du.ac.in](mailto:averma@acbr.du.ac.in)

### Authors

Manoj Kumar – Department of Chemistry, University of Delhi, Delhi 110007, India

Shalini Verma – Department of Chemistry, University of Delhi, Delhi 110007, India

Complete contact information is available at:

<https://pubs.acs.org/doi/10.1021/acs.orglett.0c01237>

### Author Contributions

†M.K. and S.V. contributed equally.

### Notes

The authors declare no competing financial interest.

## ■ ACKNOWLEDGMENTS

The research work was supported by SERB (EEQ/2018/000317) (M.K.) and (S.V.) are thankful to CSIR for fellowships.

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